

Dunite ANEOS in CTH with parameters suitable for molecular ANEOS in CTH 9.1. This EOS matches that used by Canup SPH simulations of the moon-forming impact. It is implemented in CTH 9.1 as DUNITE\_MOL.

Number	Value	Description
V1	3	Nel; number of elements in material (H,O)
V2	4	EOS; EoS Type; Solid-gas with electronic terms and detailed treatment of liquid/vapor region
V3	3.32 g/cm <sup>3</sup>	$\rho_0$ ; reference density
V4	0.	reference temperature defaults to 298 K.
V5	0	$p_0$ , reference pressure
V6	-6.6e5 cm/s	$S_0$ , from shock-particle velocity relation, $u_s = S_0 + S_1 u_p$ Neg sign required.
V7	0.82	$\Gamma_0$ , reference Gruneisen coefficient
V8	0.057 eV	Reference Debye temperature
V9	0.86	$S_1$ from linear shock-particle velocity relation, $u_s = S_0 + S_1 u_p$
V10	2	3 times the limiting value of $\Gamma$ at large compressions
V11	1.3e11 erg/g	$E_{sep}$ , Zero-T separation energy
V12	0.19 eV	$T_m$ , the melting temperature
V13	1.97e11	parameter $c_{53}$ , for low density modification to move the critical point
V14	0.8	parameter $c_{54}$ , for low density modification to move the critical point
V15	0	$H_0$ , thermal conductivity parameter; 0 if not included
V16	0	$c_{41}$ , thermal conductivity parameter; 0 if not included
V17	0.0	Lowest allowed solid density; defaults to $0.9\rho_0$
V18	4.65	Density at onset of high-pressure phase transition (hppt)
V19	4.9	Density at completion of hppt
V20	6.6e11 dyn/cm <sup>2</sup>	Pressure at center of hppt
V21	3.5e12 dyn/cm <sup>2</sup>	$dP/d\eta$ at end of hppt
V22	1.3e13 dyn/cm <sup>2</sup>	$d^2P/d\eta^2$ at end of hppt
V23	0	$H_{fus}$ , heat of fusion (not used)
V24	0	Density of liquid at the melting point (not used)
V25 ...	0	Parameters V25 through V32 not used in this type of EOS,
...V32	0	so type a whole row of zeroes
V33	1	Flag for ionization model. 0=Saha; 1=Thomas-Fermi
V34	0	$E_{shift}$ , shift energy for reactive chemistry modeling.
V35	0	$S_{shift}$ , shift entropy for reactive chemistry modeling.
V36 <sup>m</sup>	2	Number of atoms in molecular clusters (Melosh 2007)
V37 <sup>m</sup>	8.0 eV	Bond energy
V38 <sup>m</sup>	0	Rotational degrees of freedom
V39 <sup>m</sup>	1.5e-8 cm	Bond length
V40 <sup>m</sup>	0	Vibrational degrees of freedom
V41 <sup>m</sup>	0.1723 eV	=2000 K, Vibrational Debye temperature (probably not used)
V42 <sup>m</sup>	1	Flags use of Lennard-Jones (in this case, Mie) potential
V43 <sup>m</sup>	1.27	Exponent in Mie potential ( $P_{cold}$ ), should be 1.25 but CTH barfs
V44-V48	0	Not used

Z	f
8.	0.571
12.	0.286
14.	0.143