

ADITYA DEGREE COLLEGES: AU REGION

IV SEMESTER - MID - I - EXAMINATIONS

Course: B.Sc. (Chemistry Minor)

Max. Marks: 60

Time: 3 Hours

Subject: Physical Chemistry - II

SECTION - A

Answer any FIVE from the following questions:

5 X4 = 20 M

- 1. Explain the postulates of kinetic molecular theory of gasses.
- 2. Explain the Andrew's isothermals of carbon dioxide.
- 3. Explain the Joule Thomson effect.
- 4. Explain the Application of liquid crystals as LCD devices.
- 5. Explain the physical properties of liquids.
- 6. Explain the qualitative discussion of structure of water.
- 7. Explain the law of corresponding states.
- 8. Explain the continuity of states.

SECTION - A

Answer all the following questions:

4 X10 = 40 M

- 9. a) Derive the vanderwall's equation of state. (or)
 - b) Define co-efficient of viscosity? Explain the determination of viscocity?
- 10. a) Define critical phenomenon? Explain the relationship between critical constants and vanderwaal's constants. (or)
 - b) Explain the Temperature of viscosity of liquids and comparision with that of gases effect of addition of various solutes on viscocity.
- 11. a) Define Liquid crystals? Classification of liquid crystals and difference between liquid crystals and solid/liquid. (or)
 - b) Define vapor pressure? Explain the determination of vapour pressure.
- 12. a) Explain the deduction of gas laws from kinetic gas equation. (or)
 - b) Define surface tension? Factors affect the surface tension and determination of surface tension.