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## TUTORIAL - HEAT ENGINE

### 1 cyclic Heat engine

$$T_1 (\text{source}) = 800^\circ\text{C} = 1073 \text{ K}$$

$$T_2 (\text{sink}) = 30^\circ\text{C} = 303 \text{ K}$$

$$\eta = \frac{W}{Q_1} ; \eta = 1 - \frac{T_2}{T_1} = \frac{1073 - 303}{1073} = 0.717$$

$$\therefore \frac{W}{Q_1} = \eta = 0.717$$

$$W = 1 \text{ kW}$$

$$Q_1 = 1.39 \text{ kW}$$

$Q_1$  is the amount of heat absorbed from the source.

$$W = Q_1 + Q_2$$

$$Q_2 (\text{heat rejected}) = 1.39 \text{ kW} - 1 \text{ kW} = 0.39 \text{ W}$$

least rate of heat rejection per kW is 0.39 W.

$$2 \quad T_1 = 90^\circ\text{C} = 363\text{K}$$

$$T_2 = 20^\circ\text{C} = 293\text{K}$$

1880 kJ/m<sup>2</sup>h of energy can be collected when plate is operating at 363K

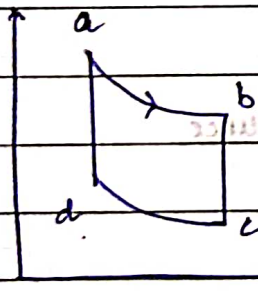
$$\eta_{\max} = \frac{W}{Q_{\min}} = 1 - \frac{T_2}{T_1} = 1 - \frac{293}{363} = \frac{363 - 293}{363} = \frac{70}{363} = 0.192$$

$$= Q_{\min} = \frac{1}{0.192} \text{ kJ/s}$$

$$Q_{\min} = 5.208 \text{ kJ/s} = \frac{5.208 \times 10^3 \text{ J/h}}{60 \times 60} = 18750 \text{ kJ/h}$$

$$\text{minimum area required} \Rightarrow \frac{18750}{1880} = 9.97 \text{ m}^2$$

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$$T_1 = 800\text{K}$$

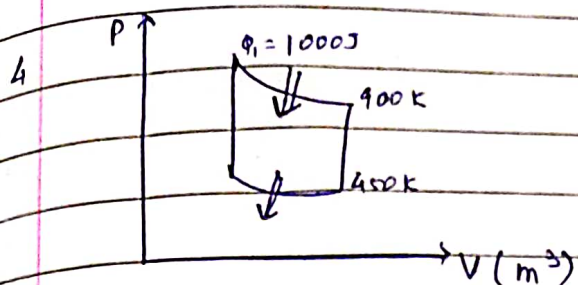
$$T_2 = 400\text{K}$$

$$Q_1 = 10,000 \text{ J}$$

$$\eta = \frac{W}{Q_1} = 1 - \frac{T_2}{T_1} = 1 - \frac{400}{800} = \frac{400}{800} = \frac{1}{2}$$

$$W = \frac{1}{2} \times 10000 = 5000 \text{ J} = \text{Work done}$$





$$Q_1 = 600 \text{ J}$$

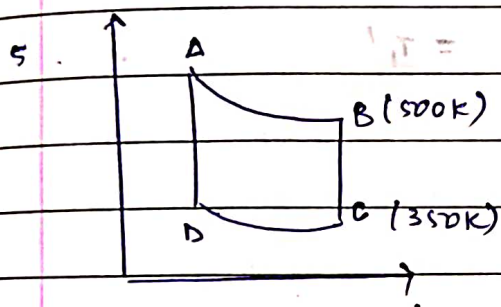
$$T_1 = 600 \text{ K}$$

$$T_2 = 400 \text{ K}$$

$$W = ?$$

$$\eta = \frac{W}{Q_1} = 1 - \frac{T_2}{T_1} = 1 - \frac{400}{600} = \frac{1}{3}$$

$$W = \frac{1}{3} \times 600 = 200 \text{ J}$$



$$T_1 = 500 \text{ K}$$

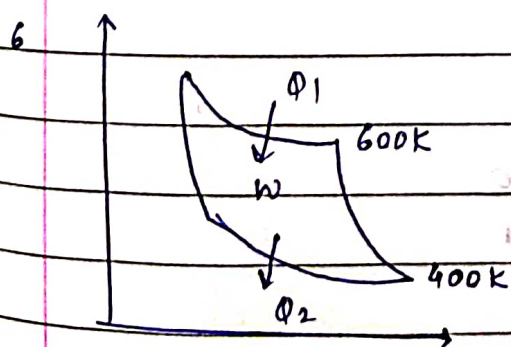
$$T_2 = 350 \text{ K}$$

$$\eta = ?$$

$$\eta = \frac{W}{Q_1} = 1 - \frac{T_2}{T_1} = 1 - \frac{350}{500}$$

$$= 1 - \frac{7}{10} = \frac{3}{10} = 0.3$$

$$\text{efficiency} = \frac{3}{10} \times 100 = 30\%$$



$$T_1 = 600 \text{ K}$$

$$T_2 = 400 \text{ K}$$

$$\eta = \frac{W}{Q_1} = 1 - \frac{T_2}{T_1} = 1 - \frac{400}{600} = \frac{200}{600} = \frac{1}{3}$$

$$Q_1 = 3 \text{ W J}$$

$$\text{Heat output} = Q_2 = Q_1 - W$$

$$= 3 \text{ W} - W$$

$$= 2 \text{ W J}$$

$$=$$

$$7 \quad T_1 = 800 \text{ K} ; \quad \eta = 50\%$$

$$T_1' = ? ; \quad \eta = 80\%$$

$$\eta = 1 - \frac{T_2}{T_1}$$

$$0.5 = 1 - \frac{T_2}{800} ; \quad T_2 = 0.5 \times 800$$

$$T_2 = 400 \text{ K}$$

$$\Rightarrow 0.80 = 1 - \left( \frac{T_1'}{400} \right)^{-1} = 1 - \frac{400}{T_1'}$$

$$T_1' = (0.20)^{-1} \times 400$$

$$= \frac{400}{0.20} \times 100 = 2000 \text{ K} = T_1'$$

$$8 \quad T_1 = 600 \text{ K}$$

$$\eta = 40\%$$

$$T_2 \rightarrow \text{constant}$$

$$0.4 = 1 - \frac{T_2}{600}$$

$$\frac{T_2}{600} = 0.6$$

$$T_2 = 360.0 \text{ K}$$

$$T_1' = ? , \quad \eta = 75\% \quad \Rightarrow \quad 0.75 = 1 - \frac{360}{T_1'}$$

$$\Rightarrow \frac{360}{T_1'} = \frac{25}{100}$$

$$T_1' = \frac{360}{0.25} = 1440 \text{ K}$$

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