

1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18

Tavola periodica degli elementi

	IA	Tavola periodica degli elementi																VIIIA	
1	1 H 1,008 (2,1)	IIA											IIIA	IVA	VA	VIA	VIIA	2 He 4,00	
2	3 Li 6,941 (1,0)	4 Be 9,012 (1,5)											5 B 10,81 (2,0)	6 C 12,01 (2,5)	7 N 14,00 (3,0)	8 O 16,00 (3,5)	9 F 19,00 (4,0)	10 Ne 20,18	
3	11 Na 22,99 (0,9)	12 Mg 24,31 (1,2)	IIIB	IVB	VB	VIB	VIIB	VIII B			IB	IIB	13 Al 26,98 (1,5)	14 Si 28,09 (1,8)	15 P 30,97 (2,1)	16 S 32,06 (2,5)	17 Cl 35,45 (3,0)	18 Ar 39,95	
4	19 K 39,10 (0,8)	20 Ca 40,08 (1,0)	21 Sc 44,96 (1,3)	22 Ti 47,90 (1,5)	23 V 50,94 (1,6)	24 Cr 52,00 (1,6)	25 Mn 54,94 (1,5)	26 Fe 55,85 (1,8)	27 Co 58,93 (1,8)	28 Ni 58,71 (1,9)	29 Cu 63,54 (1,9)	30 Zn 65,37 (1,6)	31 Ga 69,72 (1,6)	32 Ge 72,59 (1,8)	33 As 74,92 (2,0)	34 Se 78,96 (2,4)	35 Br 79,91 (2,8)	36 Kr 83,80	
5	37 Rb 85,47 (0,8)	38 Sr 87,62 (1,0)	39 Y 88,91 (1,2)	40 Zr 91,22 (1,4)	41 Nb 92,91 (1,5)	42 Mo 95,94 (1,8)	43 Tc 98,91 (1,9)	44 Ru 101,07 (2,2)	45 Rh 102,91 (2,2)	46 Pd 106,42 (2,2)	47 Ag 107,87 (1,9)	48 Cd 112,4 (1,7)	49 In 114,82 (1,7)	50 Sn 118,89 (1,8)	51 Sb 121,75 (1,9)	52 Te 127,68 (2,1)	53 I 126,90 (2,5)	54 Xe 131,30	
6	55 Cs 132,91 (0,7)	56 Ba 137,34 (0,9)	*57 La 138,91 (1,1)	72 Hf 178,49 (1,3)	73 Ta 180,95 (1,5)	74 W 183,85 (1,7)	75 Re 186,21 (1,9)	76 Os 190,23 (2,2)	77 Ir 192,22 (2,2)	78 Pt 195,09 (2,2)	79 Au 196,97 (2,4)	80 Hg 200,59 (1,9)	81 Tl 204,37 (1,8)	82 Pb 207,19 (1,8)	83 Bi 208,91 (1,9)	84 Po 210	85 At 210	86 Rn (222)	
7	87 Fr 223	88 Ra 226	**89 Ac 227	104 Rf 261	105 Db 262	106 Sg 266	107 Bh 107	108 Hs 265	109 Mt 266	110 Ds 281	111 Rg 282	112 Cn 285	113 Nh 286	114 Fl 289	115 Mc 290	116 Lv 293	117 Ts 294	118 Og 294	

Numero atomico

30
Zn
65,37
(1,6)

Simbolo

Massa atomica (peso atomico)

Elettronegatività
(secondo Pauling)

*Lantanoidi

58 Ce 140,12	59 Pr 140,907	60 Nd 144,24	61 Pm 146,92	62 Sm 150,35	63 Eu 151,36	64 Gd 157,25	65 Tb 158,924	66 Dy 162,50	67 Ho 164,930	68 Er 167,26	69 Tm 168,934	70 Yb 173,04	71 Lu 174,97
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**Attinoidi

90 Th 212,038	91 Pa 231,06	92 U 238,03	93 Np 237	94 Pu 242	95 Am 243	96 Cm 247	97 Bk 247	98 Cf 249	99 Es 254	100 Fm 253	101 Md 256	102 No 256	103 Lr 257
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Potenziali di Riduzioni Standard E° (volt), misurati a 1 atm e 25° C

Semireazione	E°	Semireazione	E°
$\text{Li}^+ + \text{e}^- \rightarrow \text{Li(s)}$	-3,040	$2\text{H}^+ + 2\text{e}^- \rightarrow \text{H}_2(\text{g})$	0,000
$\text{Rb}^+ + \text{e}^- \rightarrow \text{Rb(s)}$	-2,98	$\text{Sn}^{4+} + 2\text{e}^- \rightarrow \text{Sn}^{2+}$	+0,15
$\text{K}^+ + \text{e}^- \rightarrow \text{K(s)}$	-2,931	$\text{Cu}^{2+} + \text{e}^- \rightarrow \text{Cu}^+$	+0,153
$\text{Cs}^+ + \text{e}^- \rightarrow \text{Cs(s)}$	-2,92	$\text{H}_2\text{SO}_4 + 2\text{H}^+ + 2\text{e}^- \rightarrow \text{H}_2\text{SO}_3 + \text{H}_2\text{O}$	+0,17
$\text{Ba}^{2+} + 2\text{e}^- \rightarrow \text{Ba(s)}$	-2,912	$\text{Cu}^{2+} + 2\text{e}^- \rightarrow \text{Cu(s)}$	+0,337
$\text{Ca}^{2+} + 2\text{e}^- \rightarrow \text{Ca(s)}$	-2,868	$\text{O}_2(\text{g}) + 2\text{H}_2\text{O} + 4\text{e}^- \rightarrow 4\text{OH}^-$	+0,401
$\text{Na}^+ + \text{e}^- \rightarrow \text{Na(s)}$	-2,71	$\text{MnO}_4^- + \text{e}^- \rightarrow \text{MnO}_4^{2-}$	+0,564
$\text{Mg}^{2+} + 2\text{e}^- \rightarrow \text{Mg(s)}$	-2,372	$\text{O}_2(\text{g}) + 2\text{H}^+ + 2\text{e}^- \rightarrow \text{H}_2\text{O}_2$	+0,682
$\text{Be}^{2+} + 2\text{e}^- \rightarrow \text{Be(s)}$	-1,847	$\text{Fe}^{3+} + \text{e}^- \rightarrow \text{Fe}^{2+}$	+0,771
$\text{Al}^{3+} + 3\text{e}^- \rightarrow \text{Al(s)}$	-1,662	$\text{Ag}^+ + \text{e}^- \rightarrow \text{Ag(s)}$	+0,800
$\text{Ti}^{2+} + 2\text{e}^- \rightarrow \text{Ti(s)}$	-1,63	$\text{Hg}^{2+} + 2\text{e}^- \rightarrow \text{Hg(s)}$	+0,851
$\text{Mn}^{2+} + 2\text{e}^- \rightarrow \text{Mn(s)}$	-1,185	$2\text{Hg}^{2+} + 2\text{e}^- \rightarrow \text{Hg}_2^{2+}$	+0,92
$\text{Zn}^{2+} + 2\text{e}^- \rightarrow \text{Zn(s)}$	-0,763	$\text{NO}_3^- + 4\text{H}^+ + 3\text{e}^- \rightarrow \text{NO(g)} + 2\text{H}_2\text{O}$	+0,96
$\text{Cr}^{3+} + 3\text{e}^- \rightarrow \text{Cr(s)}$	-0,74	$\text{Br}_2(\text{l}) + 2\text{e}^- \rightarrow 2\text{Br}^-$	+1,09
$\text{Fe}^{2+} + 2\text{e}^- \rightarrow \text{Fe(s)}$	-0,447	$\text{Pt}^{2+} + 2\text{e}^- \rightarrow \text{Pt(s)}$	+1,118
$\text{Cr}^{3+} + \text{e}^- \rightarrow \text{Cr}^{2+}$	-0,41	$\text{Cr}_2\text{O}_7^{2-} + 14\text{H}^+ + 6\text{e}^- \rightarrow 2\text{Cr}^{3+} + 7\text{H}_2\text{O}$	+1,33
$\text{Cd}^{2+} + 2\text{e}^- \rightarrow \text{Cd(s)}$	-0,403	$\text{Cl}_2(\text{g}) + 2\text{e}^- \rightarrow 2\text{Cl}^-$	+1,36
$\text{Ti}^{3+} + \text{e}^- \rightarrow \text{Ti}^{2+}$	-0,37	$\text{Au}^{3+} + 3\text{e}^- \rightarrow \text{Au(s)}$	+1,498
$\text{Co}^{2+} + 2\text{e}^- \rightarrow \text{Co(s)}$	-0,28	$\text{MnO}_4^- + 8\text{H}^+ + 5\text{e}^- \rightarrow \text{Mn}^{2+} + 4\text{H}_2\text{O}$	+1,51
$\text{Ni}^{2+} + 2\text{e}^- \rightarrow \text{Ni(s)}$	-0,257	$\text{H}_2\text{O}_2 + 2\text{H}^+ + 2\text{e}^- \rightarrow 2\text{H}_2\text{O}$	+1,77
$\text{Sn}^{2+} + 2\text{e}^- \rightarrow \text{Sn(s)}$	-0,137	$\text{F}_2(\text{g}) + 2\text{e}^- \rightarrow 2\text{F}^-$	+2,87
$\text{Pb}^{2+} + 2\text{e}^- \rightarrow \text{Pb(s)}$	-0,126		

Entalpie di formazione standard (kJ/mol)

$\text{Al}_2\text{O}_3(\text{s})$	-1676	CO(g)	-110,5
$\text{B}_2\text{H}_6(\text{g})$	35,61	$\text{CO}_2(\text{g})$	-393,5
$\text{B}_2\text{O}_3(\text{s})$	-1272	$\text{COCl}_2(\text{g})$	-220,1
$\text{Ba(OH)}_2(\text{s})$	-946,3	$\text{CS}_2(\text{g})$	280,3
$\text{BaCO}_3(\text{s})$	-1216	$\text{Fe}_2\text{O}_3(\text{s})$	-824,2
BaO(s)	-553,4	$\text{Fe}_3\text{O}_4(\text{s})$	-1118
$\text{CH}_4(\text{g})$	-74,87	FeO(s)	-266,7
$\text{C}_2\text{H}_2(\text{g})$	226,7	$\text{H}_2\text{O(g)}$	-241,8
$\text{C}_2\text{H}_4(\text{g})$	52,47	$\text{H}_2\text{O(l)}$	-285,8
$\text{C}_2\text{H}_6(\text{g})$	-84,68	$\text{H}_2\text{S(g)}$	-20,50
$\text{C}_2\text{N}_2(\text{g})$	309,1	HBr(g)	-35,38
$\text{C}_3\text{H}_6(\text{g})$	20,42	HCHO(g)	-115,9
$\text{C}_3\text{H}_8(\text{g})$	-103,8	HCl(g)	-93,31
$\text{C}_4\text{H}_{10}(\text{g})$	-126,1	HCN (g)	135,1
$\text{C}_6\text{H}_6(\text{l})$	49,04	HCOOH(l)	-424,8
$\text{C}_6\text{H}_{12}(\text{l})$	-156,2	HF(g)	-272,5
$\text{C}_6\text{H}_{14}(\text{l})$	-198,8	HI(g)	26,36
$\text{C}_6\text{H}_6(\text{l})$	49,04	$\text{Li}_2\text{O(s)}$	-598,7
$\text{C}_8\text{H}_{18}(\text{l})$	-250,2	LiOH(s)	-484,9
$\text{C}_{12}\text{H}_{26}(\text{l})$	-352,4	MgO(s)	-601,2
$\text{Ca(OH)}_2(\text{s})$	-986,1	$\text{N}_2\text{H}_4(\text{g})$	95,19
$\text{CaCO}_3(\text{s})$	-1207	$\text{N}_2\text{O(g)}$	82,05
CaO(s)	-635,1	$\text{Na}_2\text{CO}_3(\text{s})$	-1131
$\text{CCl}_4(\text{g})$	-100,4	$\text{Na}_2\text{O(s)}$	-418,0
$\text{CF}_4(\text{g})$	-933,2	NaCl(s)	-411,1
$\text{CH}_2\text{Cl}_2(\text{g})$	-95,52	NaOH(s)	-425,9
$\text{CH}_3\text{CH}_2\text{OH(l)}$	-277,0	$\text{NH}_3(\text{g})$	-45,94
$\text{CH}_3\text{CHO(g)}$	-166,4	NO(g)	90,29
$\text{CH}_3\text{Cl(g)}$	-86,32	$\text{NO}_2(\text{g})$	33,09
$\text{CH}_3\text{OH(l)}$	-238,6	$\text{SiO}_2(\text{s})$	-910,9
$\text{CHCl}_3(\text{g})$	-101,25	ZnO(s)	-348,3