### **Grade 8 Science**

Fluids Class 11

### **Flow Rate**

- Flow rate a measure of how quickly fluids move; measured in volume per unit time (L/s)
- Depends on:
  - The type of fluid that is flowing
  - The force pushing on the fluid
  - The size of the pipe or opening the fluid is flowing through
  - The type of surface over which the fluid is flowing



# Checkpoint



Find the flow rate if a 1L container takes 4 seconds to fill from your kitchen tap

# **Controlling Fluid Flow**

- Fluid mechanics study of how fluids behave when at rest and when moving
- Fluids dynamics the study of fluids in motion
  - Aerodynamics: the study of moving gases
  - Hydrodynamics: the study of moving liquids



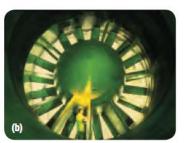




#### **Aeronautics**

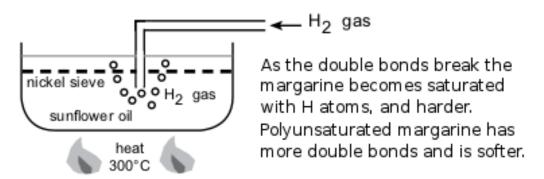
- The study of the science of flight
- Examples:
  - Wing designs to control airflow around wings
  - Parachute and paraglider designs
  - Wind tunnel used to study and control airflow around objects





# Fluid Control in the Food Industry

- Margarine and shortening made made by bubbling hydrogen gas through vegetable oil
- If hydrogen gas mixes too quickly, a substance called "trans fat" may be produced



- Ethylene gas can help ripen fruit
- Fruits are picked and transported before they are ripe because ripened fruit is firmer and less likely to damage
- Fruit is stored in a ripening room where it is exposed to ethylene gas

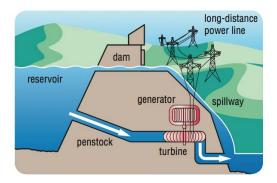






### **Water Dams**

- Dams are used to control the flow of water and to generate electricity
- Flow of the water spins the blades of the electric turbine thereby generating electricity



#### **Advantages**

- Does not emit air pollution
- Does not create radioactive wastes
- Renewable and clean energy source
- Reliable energy

#### **Disadvantages**

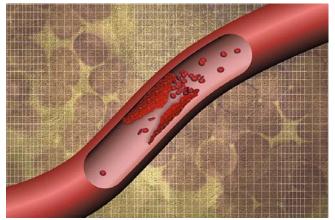
- · Loss of land due to flooding
- Construction of dam removes wetlands, agricultural land and lands for First Nations people
- Affects fish populations

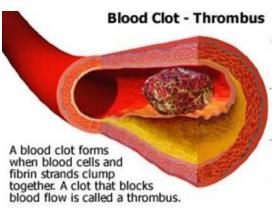




## **Controlling Blood Flow**

- Blood flow is one of the most important fluid movement within the body
  - Blood thinners medicines for patients whose blood clots too easily which can lead to heart attacks and strokes
  - Hemophilia genetic condition that causes the patient to bleed excessively; medicine needed to promote clotting
  - Artificial hearts transplanted into people who hearts are not strong enough to continuously pump blood





## **Volume**

Volume – how much space an object occupies

 $volume = length \times width \times height$ 

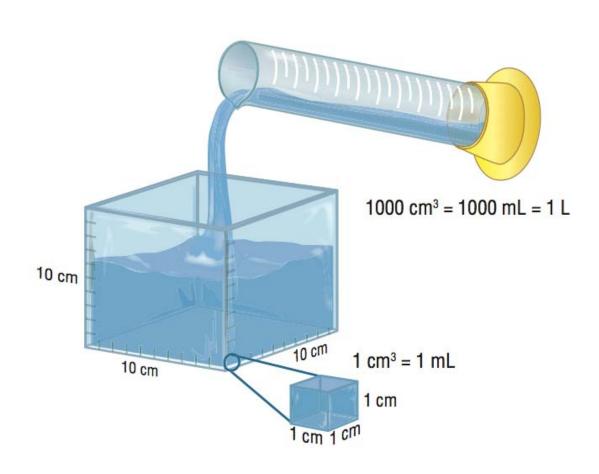
• For solids: cm<sup>3</sup>, m<sup>3</sup>

• For gases: m<sup>3</sup>

Conversion:

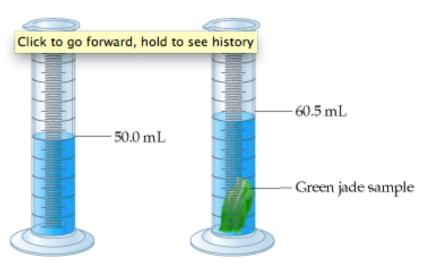
1cm<sup>3</sup> = 1mL

• For liquids: mL, L



## **Finding Volume by Displacement**

- Used to measure irregularly-shaped objects
- Displacement the volume that the object displaces



 The volume of objects too large to fit in a graduated cylinder can be measured using an overflow can



## **Density**

- Density the measure of the amount of matter in a given volume of a substance
- Characteristic property a property that makes a particular substance distinct from others

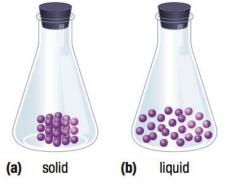
$$density = \frac{mass}{volume}$$

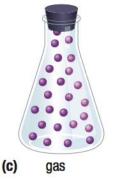
- For solids and liquids: g/cm<sup>3</sup> or g/mL
- For gases: kg/m<sup>3</sup>





- 1kg of styrofoam and 1kg of gold may have the same mass but styrofoam would require more volume
- Gold is more dense than styrofoam
- For the same volume, there would be more gold particles than styrofoam particles





- Particles of solids are usually closer together than liquids and gases
- Space between gas particles are larger than solids and liquids
- Gases are less dense and more compressible



# Checkpoint



An empty container has a mass of 50g. When 75mL of oil are placed in it, the total mass is 120g. Calculate the density of the oil.

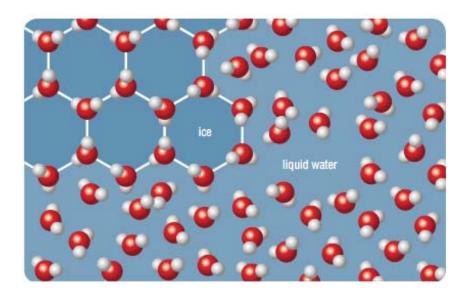
## **Density of Water**

Table 1 The Density of Some Common Materials

| Fluids                          | g/cm³ or<br>g/mL | kg/m³ |
|---------------------------------|------------------|-------|
| air                             | 0.001 3          | 1.3   |
| carbon<br>monoxide              | 0.001 45         | 1.45  |
| gasoline                        | 0.737            |       |
| distilled<br>water<br>(at 4 °C) | 1.0              |       |
| sea water                       | 1.03             |       |
| mercury<br>(a liquid<br>metal)  | 13.55            |       |
| Solids                          |                  |       |
| wood<br>(balsa)                 | 0.12             |       |
| ice                             | 0.92             |       |
| lead                            | 11.34            |       |

- Usually fluids become more dense as they cool because the particles move slower and closer together
- This is true for water only until 4°C; water is most dense at 4°C
- As cooling continues, water particles move farther apart, volume increases and forms ice

- Since the mass remains the same, but the volume increases, the density decreases
- Pure water is least dense at 0°C

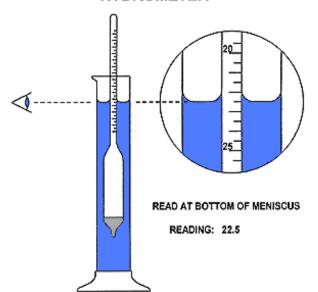


- Ice forms at the top of lakes and floats on liquid water
- If water behaved like other liquids, our lakes would freeze solid in the winter, killing the plants and animals



# **Hydrometer**

#### **HYDROMETER**



- A glass tube with a weight on one end floats in the liquid sample
- Take a reading at the bottom of the meniscus when the hydrometer is not touching the sides of the column