

Generating Phase Diagrams via HLD-NAC

Andrew Nyland¹, Jack Heller², Nick Furth³, Dr. Adam Imel⁴, Dr. Thomas Zawodzinski⁴

¹Western Colorado University, Gunnison, CO; ²Lehigh University, Bethlehem, PA; ³New Jersey Institute of Technology, Newark, NJ; ⁴University of Tennessee, Knoxville, TN

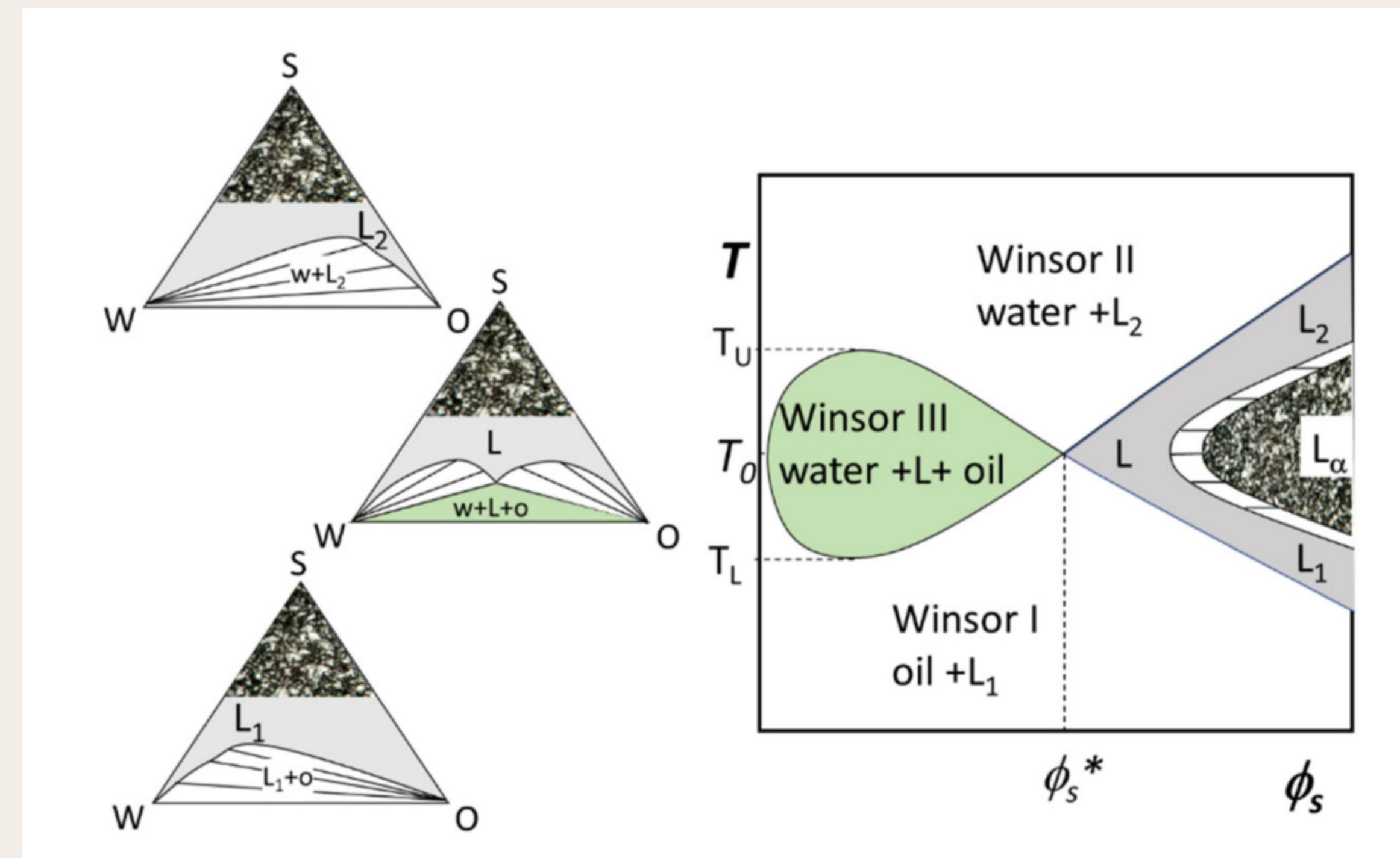


Figure 1: On the left is shown some example ternary phase diagrams, on the right is an example of a fish diagram. The ternary diagram was the first visualization to generate and can be seen in Figure 2. Creating a prism (or tent) diagram allowed for generation of the fish diagram from the HLD-NAC equation, seen in Figure 4. [1]

Purpose

Predicting and understanding solubilization and conductivity behaviors of surfactants has been an area in fuel cell research, and performing experiments can be excessively time consuming. The Hydrophilic-Lipophilic Deviation and Net Average Curvature (HLD-NAC) equations can deliver an acceptable prediction of these behaviors. Our approach to a solution of making this process possible in a reasonable time-frame was to create an interactive simulator which samples the HLD-NAC over a range of input values output to the user as various industry standard views. As the HLD-NAC is still evolving, our implementations have been written to proper software engineering standards so that the implementation can be updated by anyone when the HLD-NAC is improved.

An approximate HLD-NAC equation:

$$HLD = \ln(S) - K \times EACN - \alpha T (T - T^*) + Cc + f(A)$$

$$H_n = \left| \frac{1}{R_o} \right| - \left| \frac{1}{R_w} \right| = \frac{-HLD}{L} \quad H_a = \left(\left| \frac{1}{R_o} \right| + \left| \frac{1}{R_w} \right| \right) / 2 \geq \frac{1}{\xi}$$

Which considers aspects of the brine, oil, and surfactant used. The NAC expands and determines the curvature of the microemulsion – which allows for determination of the resulting Winsor type systems.

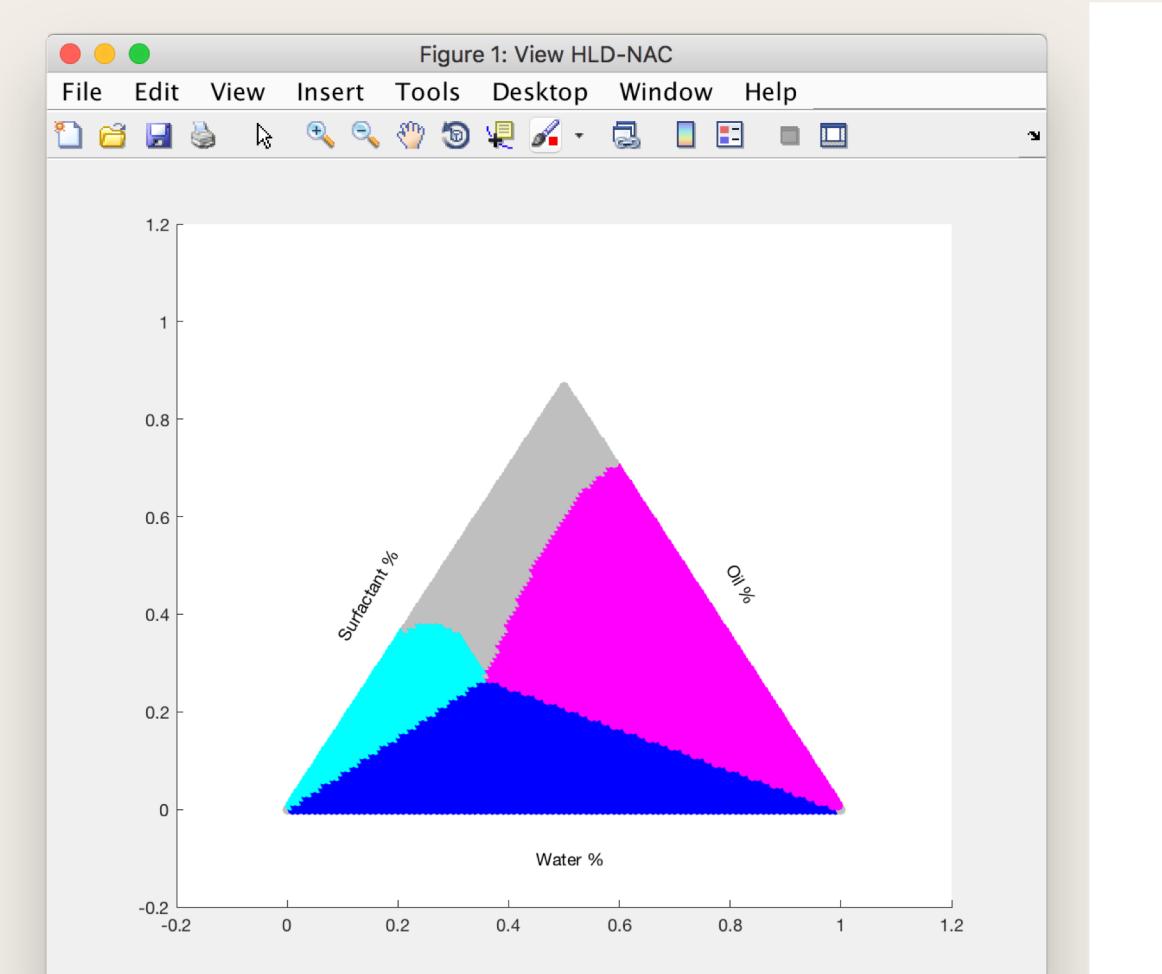


Figure 2: 2D Generated Phase Diagrams
Above: Filled-in scatterplot – allows for fine inspection of the data itself.
Right: A rendition of the same phase diagram for comparison to hand-drawn phase diagrams, or for ease of printing.

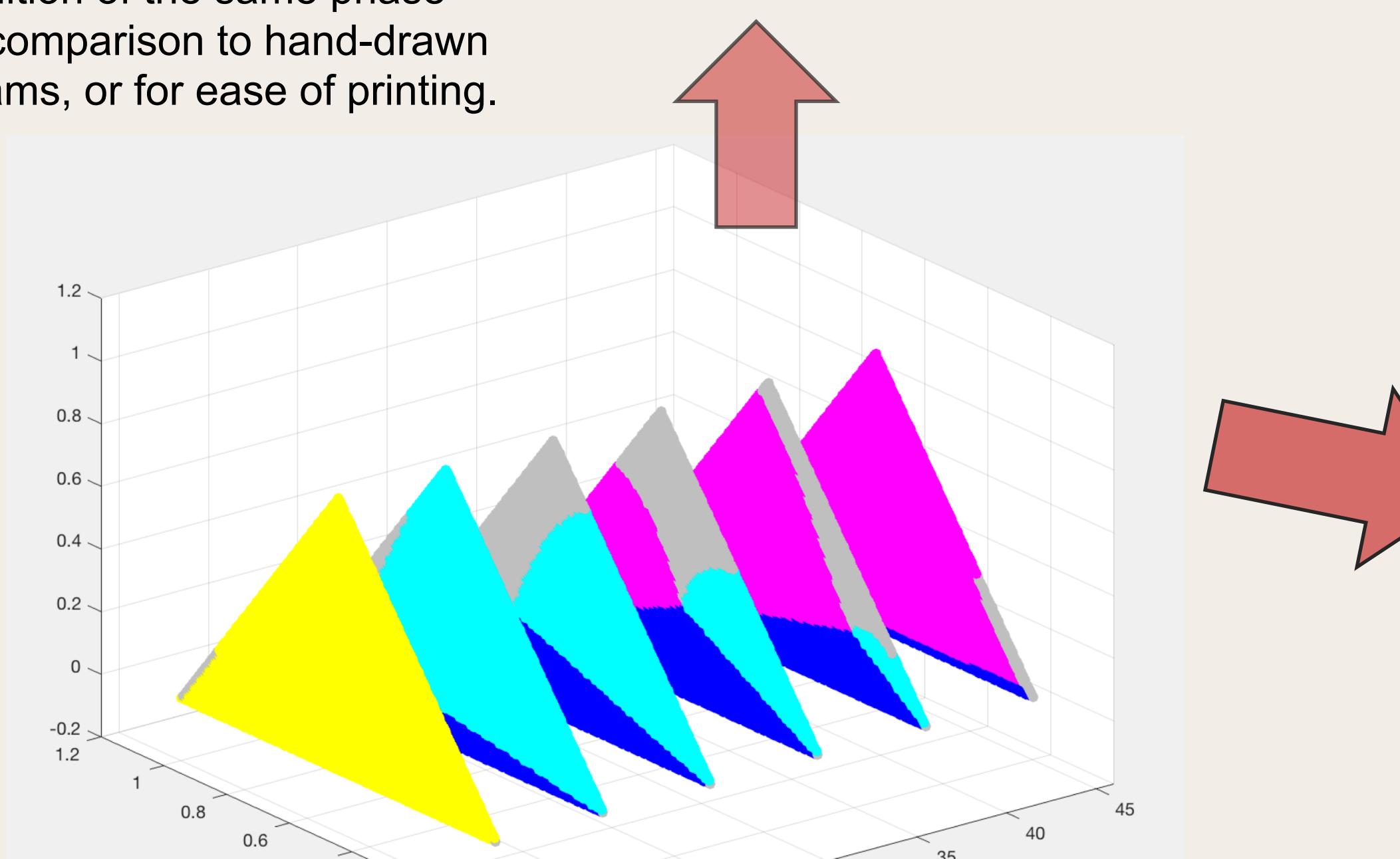
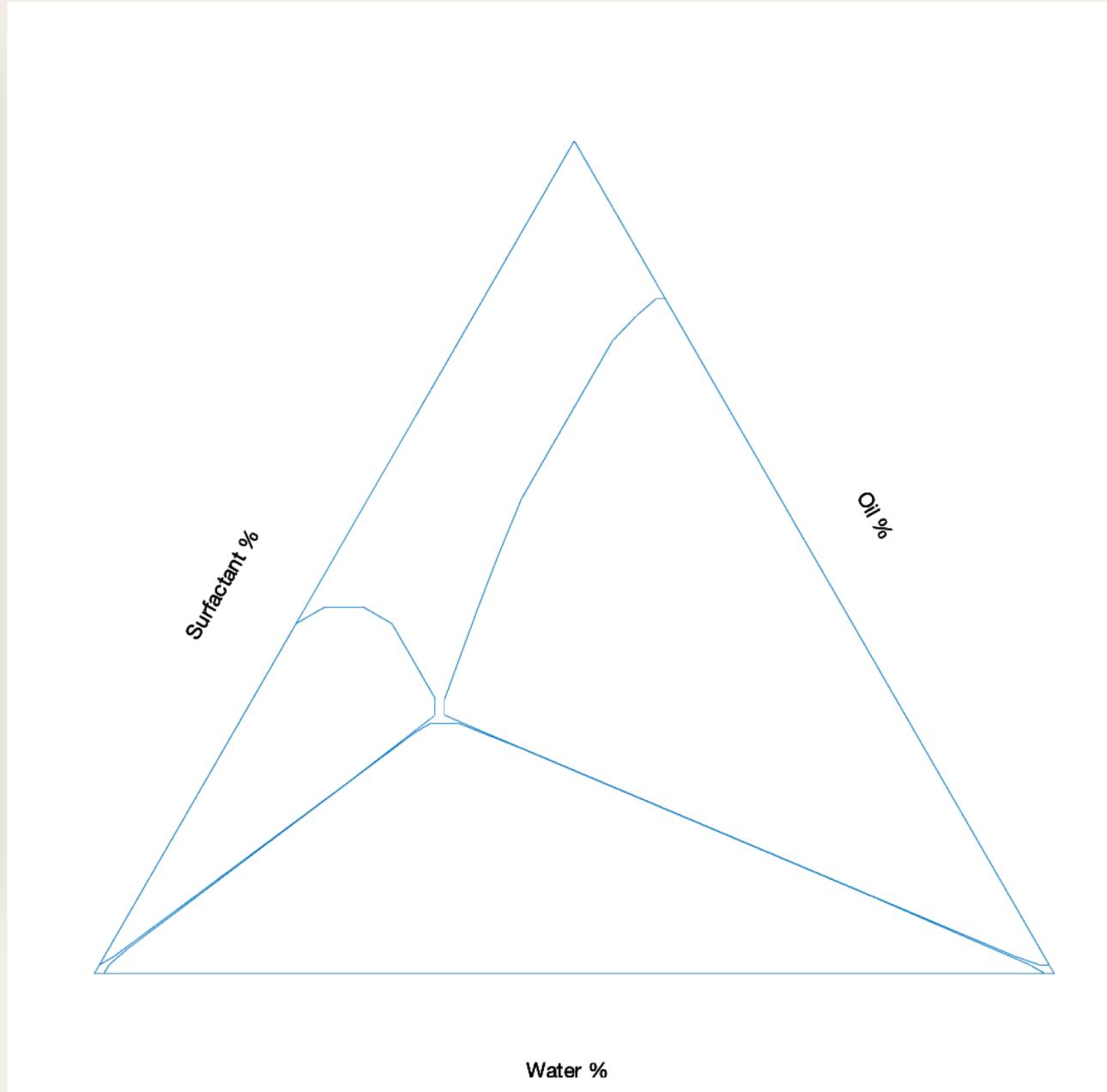


Figure 3: A prism generated as slices, each of which is computed as a 2D ternary diagram. In the current view, the prism is sampled at 1% solubility increments and at 5 °C increments.

Methods

MATLAB was chosen for creating this project due to:

- 1) Interactive 2D/3D graphing environment provided
- 2) Compatibility with future researchers
- 3) Sufficient elapsed runtime, fast development time

The ultimate task of this project is to render a 4D situation onto a 3D plot, viewed via a 2D screen. First, a 3D cartesian space is generated, which is used to calculate slices of a prism. These slices are used to calculate ternary diagrams via the HLD-NAC/Flash algorithm. If needed, this data is converted to a volume.

Results

This project provides a user interactive (chosen view can be rotated and viewed alternatively, if wanted) simulator of the HLD-NAC equation. Many circumstances are considered, as well as various inputs dictated by the user.

Output views provided include:

- 1) 2D ternary phase diagrams (filled-in or "sketched")
- 2) 3D shell, tent, prism, and prism slice renditions
- 3) Fish diagram and Schinoda diagram
- 4) Volumes – these can sliced or further analyzed.

All of these views can be improved as the HLD-NAC logic evolves by altering only the relevant code file.

Figure 4: The fish diagram can be extracted from the prism to the left as a lengthwise vertical slice. The axes can vary and have different purposes – in this plot the x-axis is surfactant concentration (%) and the y-axis is temperature (°C).

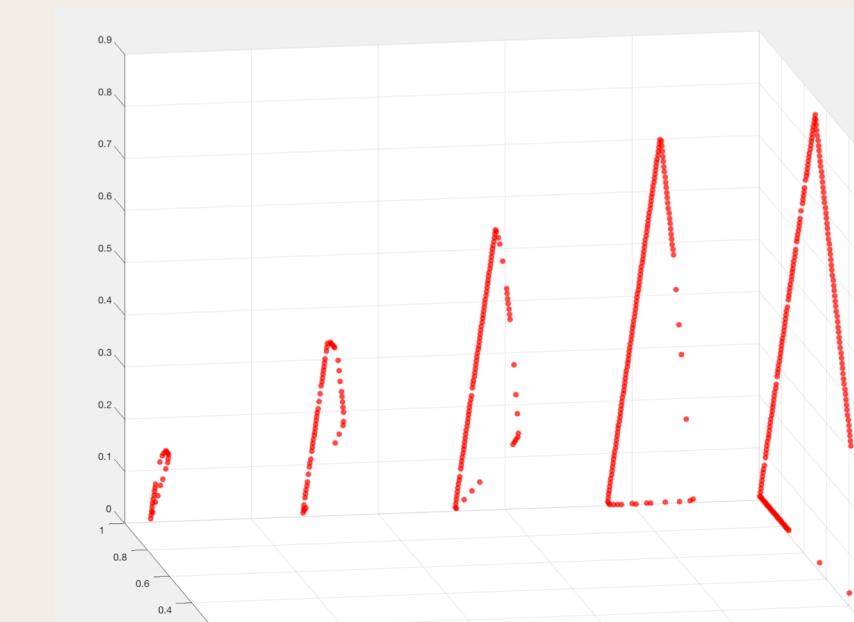
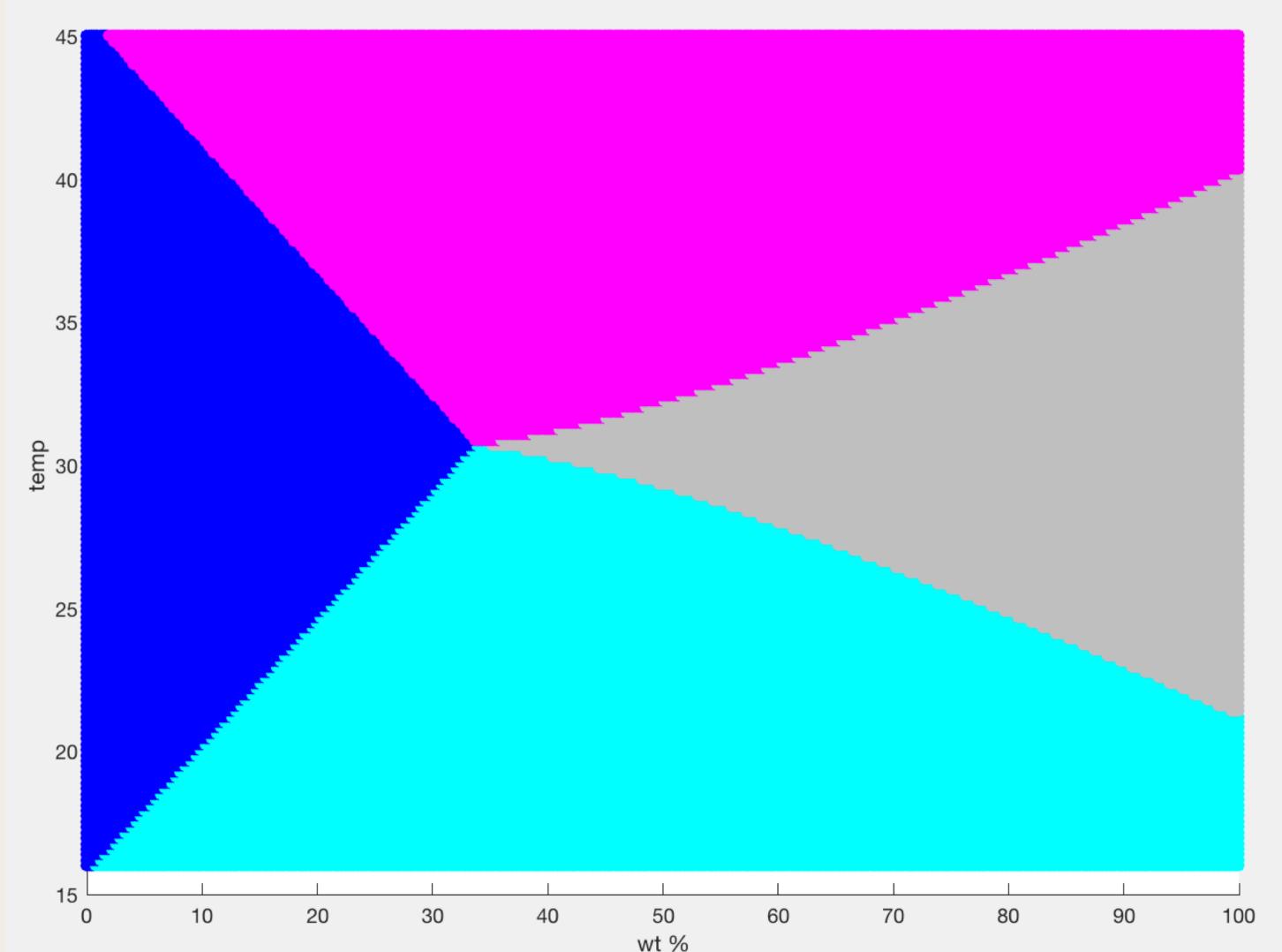
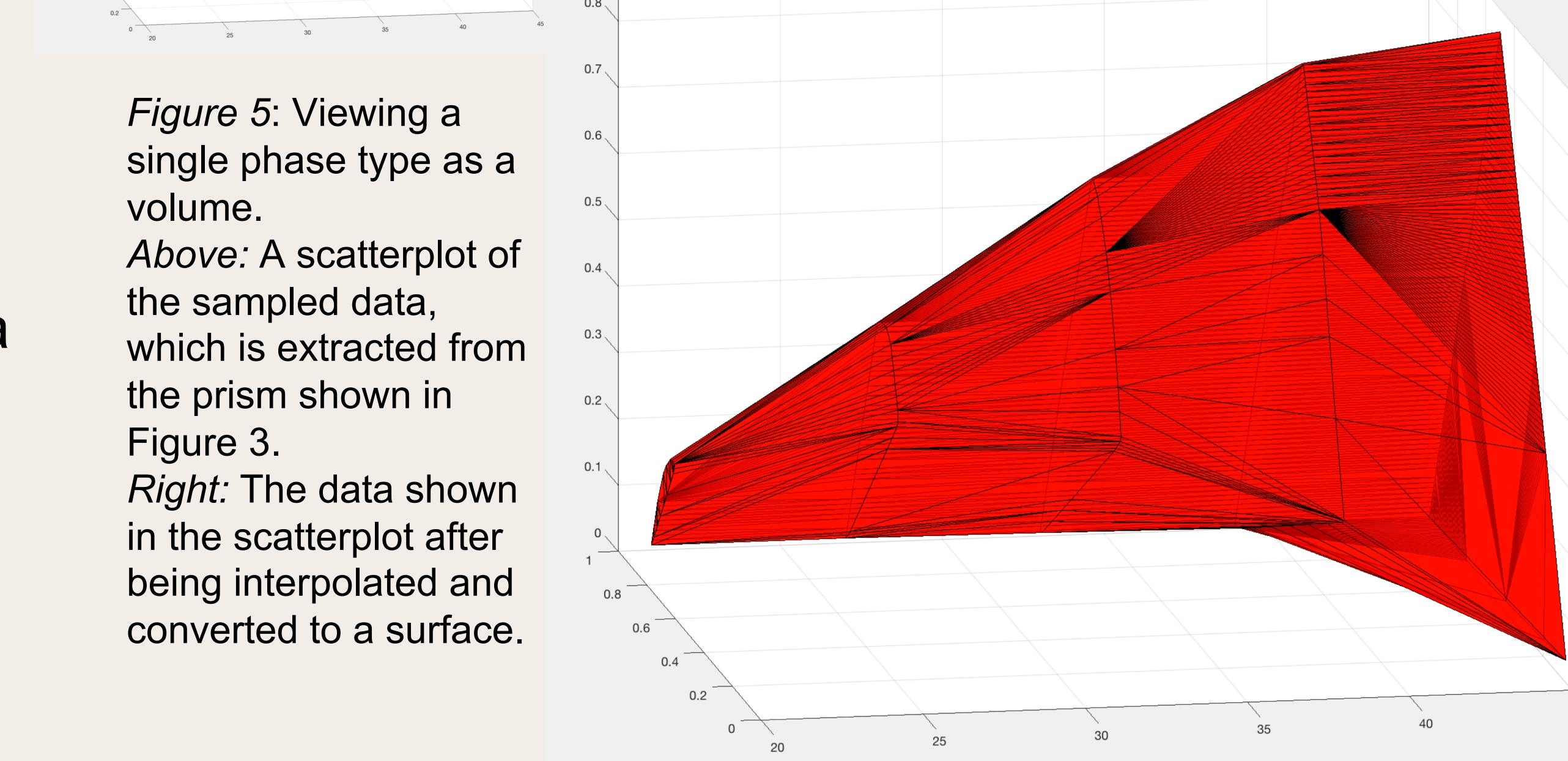


Figure 5: Viewing a single phase type as a volume.
Above: A scatterplot of the sampled data, which is extracted from the prism shown in Figure 3.
Right: The data shown in the scatterplot after being interpolated and converted to a surface.



1. Tartar, Giuseppe et al. *Microemulsion Microstructure(s): A Tutorial Review*. August 2020. *Nanomaterials* 1657(10). doi:10.3390/nano10091657

This work was performed during the 2021 Student Mentoring and Research Training (SMaRT) program that was jointly supported by the U.S. Department of Energy's Office of Energy Efficiency and Renewable Energy (EERE) through award number DE-EE0009177 provided to the University of Tennessee Oak Ridge Innovation Institute (UT-ORII) and the UT Science Alliance program.