



國立清華大學

化學工程學系

National Tsing Hua University  
Department of Chemical Engineering

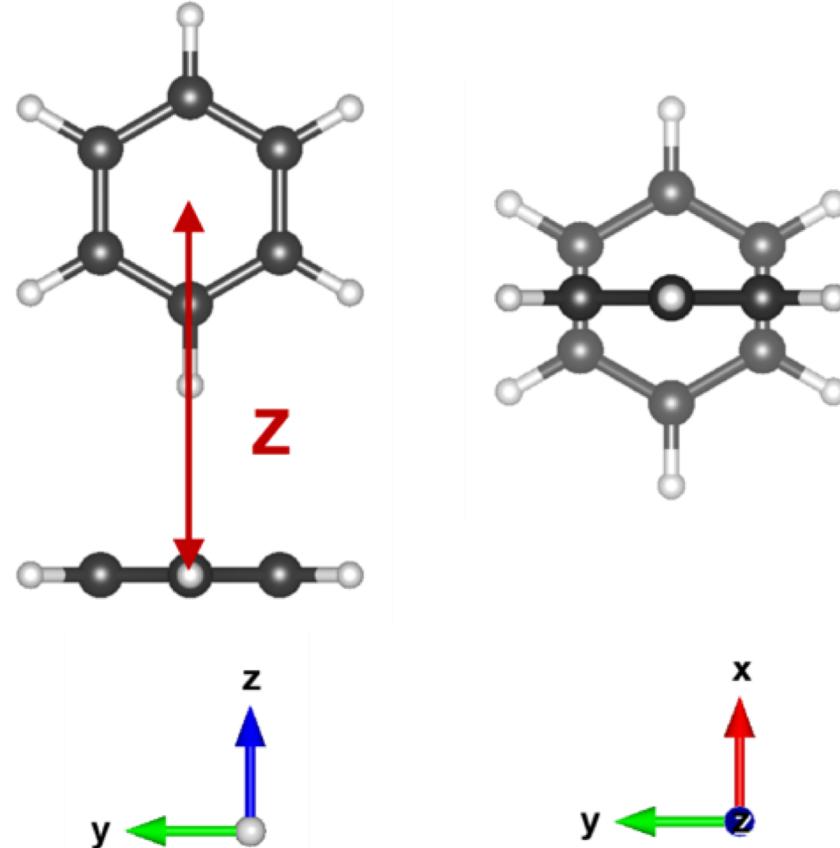
# 2022 Summer School

## Electronic Structure Calculations Using GFN2-xTB

Prof. Kun-Han Lin (林昆翰助理教授)  
Department of Chemical Engineering,  
National Tsing Hua University, Taiwan

# Homework 3

# Q1: T-shaped benzene dimer



## Tasks:

- Create a series of configurations of T-shaped dimers with different Z.
- Create the  $E_b$ -Z plot and determine the minimum.
- Evaluate the dimer interactions using multipole-expansion approximation.

# Rotate the benzene – rotate.py

```
1 import numpy as np  
2 from ase.io import read, write  
3 from ase.build import rotate  
4 from mpl_toolkits.mplot3d import Axes3D  
5 import matplotlib.pyplot as plt
```

```
7 # Read the xyz as Atom object in ASE  
8 xyz = read('xtbopt.xyz')
```

## Import necessary library

- Numpy (numerical analysis)
- Ase (atomic simulation environment)
- Matplotlib (MATLAB-like plotting)

Read xtbopt.xyz (optimized benzene geometry using xTB) and create an Atoms object of ASE

What is an Atoms object? <https://wiki.fysik.dtu.dk/ase/ase/atoms.html>

# The Atoms object

## The Atoms object

The `Atoms` object is a collection of atoms. Here is how to define a CO molecule:

```
from ase import Atoms
d = 1.1
co = Atoms('CO', positions=[(0, 0, 0), (0, 0, d)])
```

Here, the first argument specifies the type of the atoms and we used the `positions` keyword to specify their positions. Other possible keywords are: `numbers`, `tags`, `momenta`, `masses`, `magmoms` and `charges`.

# Extract information from the Atoms object

The get method

<code>get_atomic_numbers()</code>	<code>set_atomic_numbers()</code>
<code>get_initial_charges()</code>	<code>set_initial_charges()</code>
<code>get_charges()</code>	
<code>get_chemical_symbols()</code>	<code>set_chemical_symbols()</code>
<code>get_initial_magnetic_moments()</code>	<code>set_initial_magnetic_moments()</code>
<code>get_magnetic_moments()</code>	
<code>get_masses()</code>	<code>set_masses()</code>
<code>get_momenta()</code>	<code>set_momenta()</code>
<code>get_forces()</code>	
<code>get_positions()</code>	<code>set_positions()</code>
<code>get_potential_energies()</code>	
<code>get_scaled_positions()</code>	<code>set_scaled_positions()</code>
<code>get_stresses()</code>	
<code>get_tags()</code>	<code>set_tags()</code>
<code>get_velocities()</code>	<code>set_velocities()</code>

# Align the plane norm to z-axis

```
10 # Get the moment of inertia of the compound
11 I = xyz.get_moments_of_inertia(vectors=True)
12
13 # Get the center of mass before translation
14 COM = xyz.get_center_of_mass()
15 print('Original center of mass: ',COM)
16
17 # Get the center of mass after translation COM to (0,0,0)
18 xyz.translate(-COM)
19 COM = xyz.get_center_of_mass()
20 print('Center of mass after translation: ', COM)
21
22 # Align the shortest molecular axis to z-axis
23 xyz.rotate(I[1][2],'z','COM', rotate_cell=False)
```

]

Obtain the moments of inertia of the benzene.

]

Obtain the COM of the benzene.

]

Move the COM to the origin (0,0,0) using translate. And print the new COM to check the new COM.

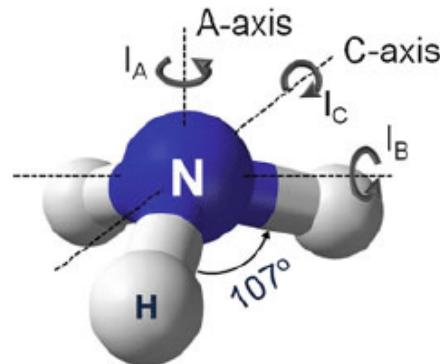
]

Rotate the benzene so that its plane norm I[1][2] align with the z-axis.

# Print the moments of inertia

Shortest molecular axis with the largest eigen value. Why?

```
(array([ 87.45954357, 87.4597337 , 174.91927726]),  
 array([[ 0.19390466, -0.06025610,  0.97916811],  
 [-0.98017909,  0.02942734,  0.19591576],  
 [-0.04061943, -0.99774908, -0.05335567]]))
```



It's a tuple object with the length of 2 (like a  $2 \times 1$  vector).

The first entry is an array containing  $I_A$ ,  $I_B$  and  $I_C$ .

The second entry is an array containing three arrays (eigen vectors) corresponding to the three eigen values ( $I_A$ ,  $I_B$  and  $I_C$ ).

# Align the plane norm to z-axis

```
25 C3 = xyz.get_positions()[2]
26 C9 = xyz.get_positions()[8]
27
28 v = C3 - C9
29
30 # Align the C3-C9 vector to x-axis
31 xyz.rotate(v, 'x', 'COM', rotate_cell=False)
32
33 # Write out the final structure as new.xyz
34 xyz.write('aligned_xz.xyz')
```

Write out the aligned benzene as aligned\_xz.xyz

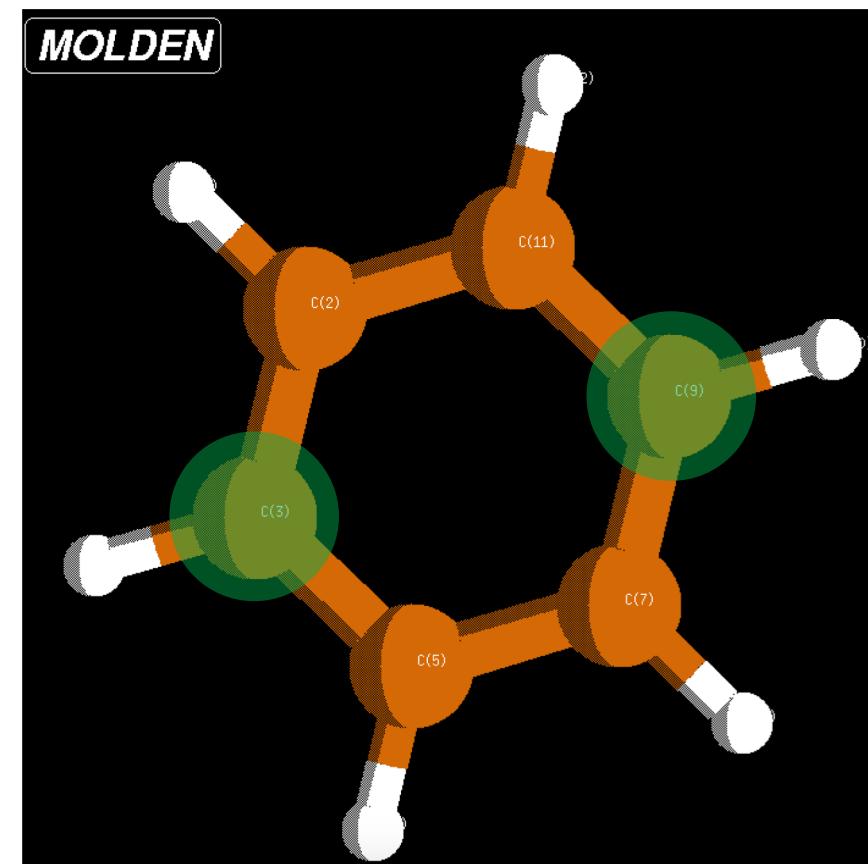
]

Obtain Cartesian coordinates of two carbons at the *para* position.

The vector C3 – C9.

]

Align v with the x-axis.



# Generate the dimer structures – gen\_dimer.py

```
1 import sys  
2 import numpy as np  
3 from ase.io import read, write  
4 from ase.build import rotate
```

## Import necessary library

- Numpy (numerical analysis)
- Ase (atomic simulation environment)
- sys (the sys.argv can be used to pass the variable to python script)

For example:

```
1 import sys  
2  
3 print (sys.argv[0])  
4 print (sys.argv[1])
```

The name of your python code

First string you enter (separated by a space)

python test.py hello

Outeput:

```
test.py  
hello
```

Therefore, sys.argv[1], sys.argv[2],...sys.argv[n] could be used to store the 1<sup>st</sup>, 2<sup>nd</sup>,...n<sup>th</sup> variables in your command line.

# Generate the dimer structures – gen\_dimer.py

```
6 # Read the xyz as Atom object in ASE  
7 xyz = read('aligned_xz.xyz')  
8 xyz_original = xyz.copy()
```

Read aligned.xyz and create an Atoms object of ASE.  
Copy a xyz Atoms object and name it xyz\_original.

```
10 # rotate the molecule along y axis by 90 degree  
11 xyz.rotate(90.0,'y','COM', rotate_cell=False)
```

Rotate the aligned benzene along y-axis by 90 degree.

```
13 # Shift the molecules in z direction with P% of 5.0 Angstrom  
14 v = np.array([0,0,5.0])*float(sys.argv[1])/100.0  
15 xyz.translate(v)
```

Define a translation vector  $\mathbf{v} = [0, 0, z]$ ,  
where the length could be controlled  
externally by `sys.argv[1]` when we run the  
code!

Translate the rotated and aligned benzene by  $\mathbf{v}$ .

# Generate the dimer structures – gen\_dimer.py

```
17 xyz.extend(xyz_original)
18 # Write out the final structure
19 xyz.write(sys.argv[2])
```

Combine the original and the rotated benzene Atoms objects together using `extend()`.

Write out the final dimer Atoms object into xyz file. The file name is controlled externally using `sys.argv[2]`.

# Create folders and inputs, and run! – create.sh

```
1 for a in 70 71 72 73 74 75 80 85 90 95 100 105 110 115 120 130 140
  150 160 170 180 190 200
2 do
3     ## for each variable a ($a), create a folder
4     mkdir $a
5
6     ## copy the xtb running script to folder $a
7     cp xtb.sh $a
8
9     ## generate a dimer input xyz file in the folder. the distance is
10    $a% of 5.0 Angstrom
11    python gen_dimer.py $a ${a}/input.xyz
12
13    ## enter the folder and run the script
14    cd $a
15    echo "Running $a now..."
16    nohup bash xtb.sh &
17    ## leave the $a folder and do the next loop
18    cd ..
```

- The `for` loop in bash. It will repeatedly run the commands sandwiched between `do` and `done`.
- The repetition will go over all numbers defined for `a`.
- When you need the value of `a`, please use `$a`.

# Create folders and inputs, and run! – create.sh

```
1 for a in 70 71 72 73 74 75 80 85 90 95 100 105 110 115 120 130 140
  150 160 170 180 190 200
2 do
3   ## for each variable a ($a), create a folder
4   mkdir $a           Create a folder with the name a.
5
6   ## copy the xtb running script to folder $a
7   cp xtb.sh $a       Copy the xtb.sh script into folder a.
8
9   ## generate a dimer input xyz file in the folder. the distance is
10    $a% of 5.0 Angstrom
11   python gen_dimer.py $a ${a}/input.xyz      Run the gen_dimer.py code, given the 1st variable the
12   ## enter the folder and run the script          distance; and the 2nd variable the file name.
13   cd $a             Enter the folder a.
14   echo "Running $a now..."
15   nohup bash xtb.sh &      Run the xtb.sh script in background (&) without being interrupted (nohup).
16   ## leave the $a folder and do the next loop
17   cd ..            Go back to the folder in the previous layer.
18 done
```

13 # Shift the molecules in z direction with P% of 5.0 Angstrom  
14 v = np.array([0,0,5.0])\*float(sys.argv[1])/100.0  
15 xyz.translate(v)

19 xyz.write(sys.argv[2])

# Extract and plot the results – plot\_Eb.py

```
1 import numpy as np  
2 import matplotlib.pyplot as plt
```

## Import necessary library

- Numpy (numerical analysis)
- Matplotlib (MATLAB-like plotting)

```
4 # Figure Setting  
5 # Edit the font, font size, and axes width  
6 plt.rcParams['font.size'] = 14  
7 plt.rcParams['axes.linewidth'] = 2  
8 fig,ax = plt.subplots()
```

Set up the figure setting, font size and axes linewidth.

Create a figure object and axes objects.

## matplotlib.pyplot.subplots

```
matplotlib.pyplot.subplots(nrows=1, ncols=1, *, sharex=False,  
sharey=False, squeeze=True, subplot_kw=None, gridspec_kw=None, **fig_kw)
```

# Extract the energy from monomer

```
10 # Read energy from monomer
11 name = 'monomer/xtb.out'
12 f = open(name,'r')
13 lines = f.readlines()
14 for l in lines:
15     if 'TOTAL ENERGY' in l:
16         tmp = l.split()
17         Em = float(tmp[3])
```

Specify the path to the xtb.out of the monomer computation.  
Open the file use open().  
Generate a list. In this list, each element is a line from the file f.  
A for loop in python:  
For each line in the list lines,  
if the string ‘TOTAL ENERGY’ appears in the line  
split the line by space and read the value.

For example, a file f like below:

```
I have a pen.
I have an apple.
Umm ugh!
Apple pen!
```

The list lines by f.readlines() will be:

```
["I have a pen.", "I have an apple.", "Umm ugh!", "Apple pen!"]
```

/ref\_func\_open.asp

# Split a string using split()

For example, a line I like below:

```
| TOTAL ENERGY      -15.879640643370 Eh |
```

l.split() will split the line by space, giving you a list like:

```
["|", "TOTAL", "ENERGY", "-15.879640643370", "Eh", "|"]
```

```
10 # Read energy from monomer
11 name = 'monomer/xtb.out'
12 f = open(name,'r')
13 lines = f.readlines()
14 for l in lines:
15     if 'TOTAL ENERGY' in l:
16         tmp = l.split()
17         Em = float(tmp[3])
```

That's why we take tmp[3], which is the 4<sup>th</sup> entry of the list tmp. And the float() function will make the string “-15.879640643370” into a float -15.879640643370.

# Create lists and open files

```
19 # Read energy from dimer
20 D = []          Create empty lists for distance (D) and binding energy (Eb).
21 Eb = []
22 folder = [ 70, 71, 72, 73, 74, 75, 80, 85, 90, 95, 100, 105, 110, 115
             , 120, 130, 140, 150, 160, 170, 180, 190, 200] Create a list (folder). containing the names of folders.
23
24 # Write the Eb and D value into E-D.txt file
25 fout = open('E-D.txt','w')      Open the file E-D.txt, which will be used to write the E-D
                                    information into it.
```

# Create lists and open files

```
27 for a in folder:  
28     # Read xtb.out in 'a' folder  
29     name = str(a)+'/xtb.out'  
30     f = open(name,'r')  
31     lines = f.readlines()  
32     for l in lines:  
33         if 'TOTAL ENERGY' in l:  
34             tmp = l.split()  
35             Ed = float(tmp[3])  
36             D.append(float(a)*5.0/100.0)  
37             Eb.append( (Ed - 2.0*Em)*27.2114 )  
38             fout.write(str(float(a)*5.0/100.0)+ ' '+str((Ed - 2.0*Em)  
                         )*27.2114) + '\n')
```

A for loop in python:  
For each name in folder list,  
1. Open the xtb.out file and read the lines  
2. A for loop in python:  
For each line (*l*) in lines list,  
*if* the string 'TOTAL ENERGY' appears in the line  
1. split the line by space and read the value.  
2. store the distance into *D* by append()  
3. calculate binding energy and store into *Eb* by append()  
4. write out the distance and binding energy into *fout* using  
write().

# Create lists and open files

```
42 # Plot the figure
43 ax.plot([0,np.max(folder)*5.0/100.0],[0,0],ls=':',c='black') Plot the y=0 horizontal dotted line in black.
44 ax.scatter(D,Eb) Plot the scatter plot using D as x-axis and Eb as y-axis.
45
46 # set label for X and Y axis
47 ax.set_xlabel('Z (\u00c5)',fontsize=18)
48 ax.set_ylabel(r'E$ b$ (eV)',fontsize=18)
49
50 # set x range
51 ax.set_xlim(2.0,np.max(folder)*5.0/100.0)
52
53 fig.tight_layout()
54 plt.show()
```

Plot the y=0 horizontal dotted line in black.

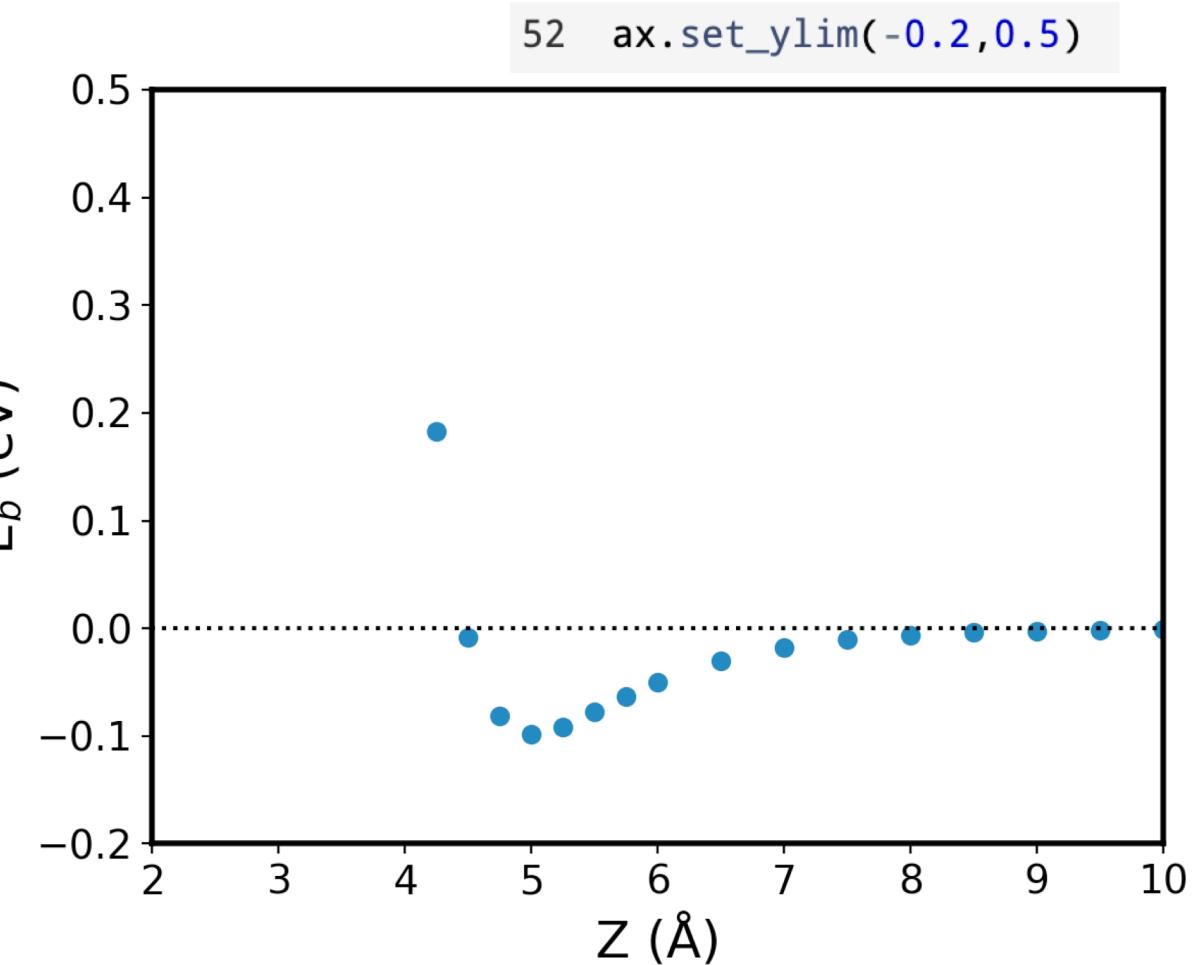
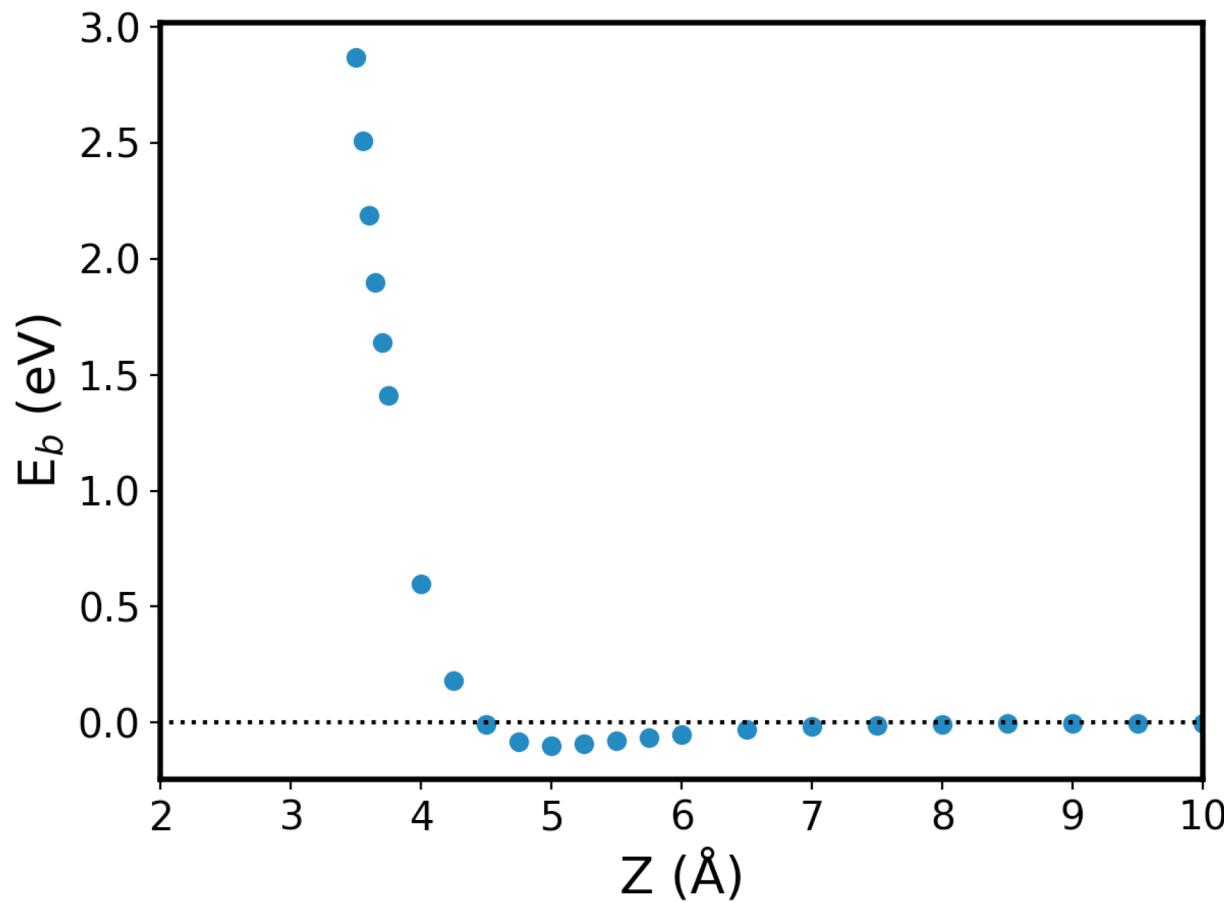
]

Set the label for x and y axis.

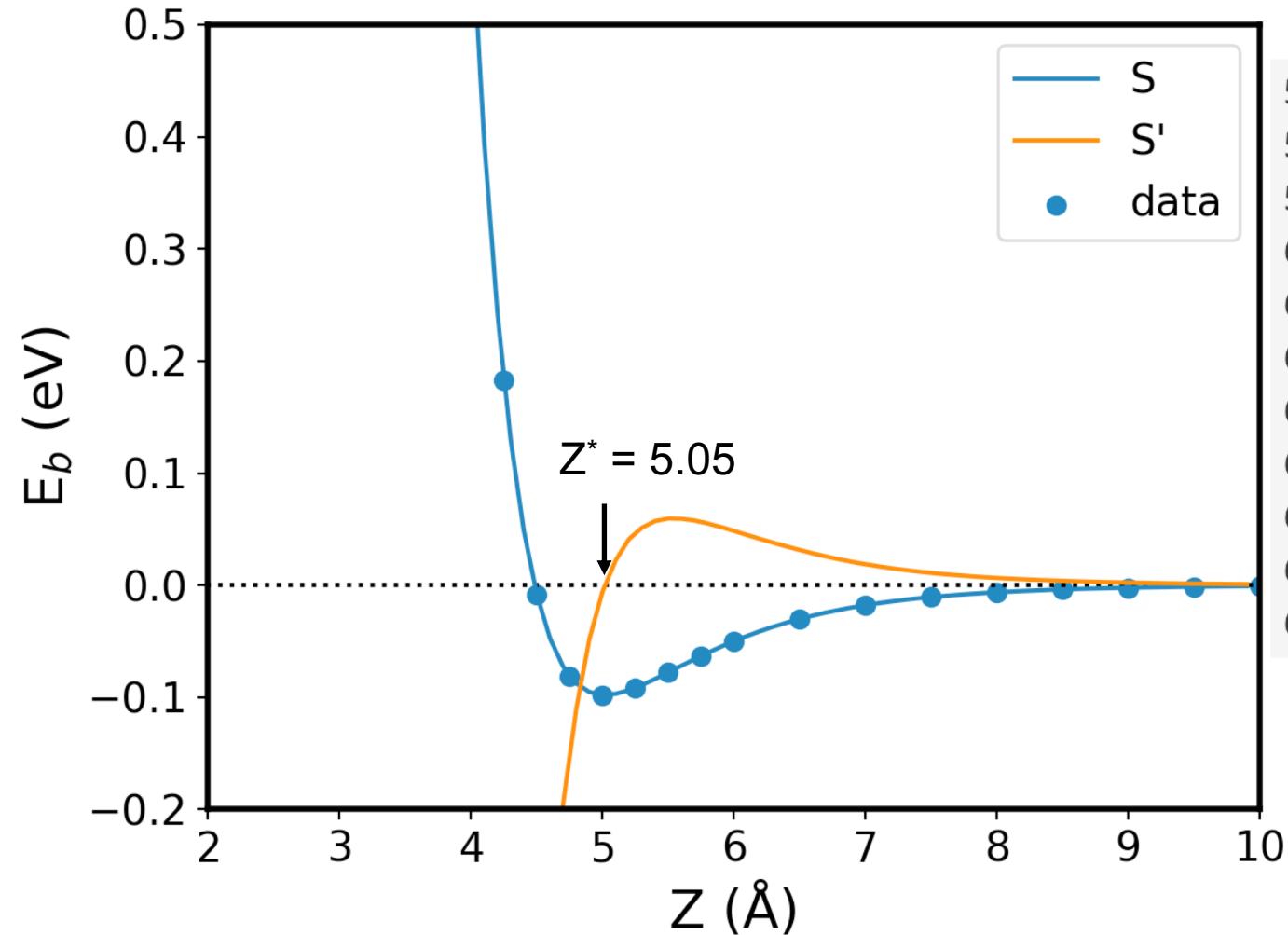
Set the range of display for x-axis.

Set the layout of figure and display the plot using show().

# E<sub>b</sub>-D curve



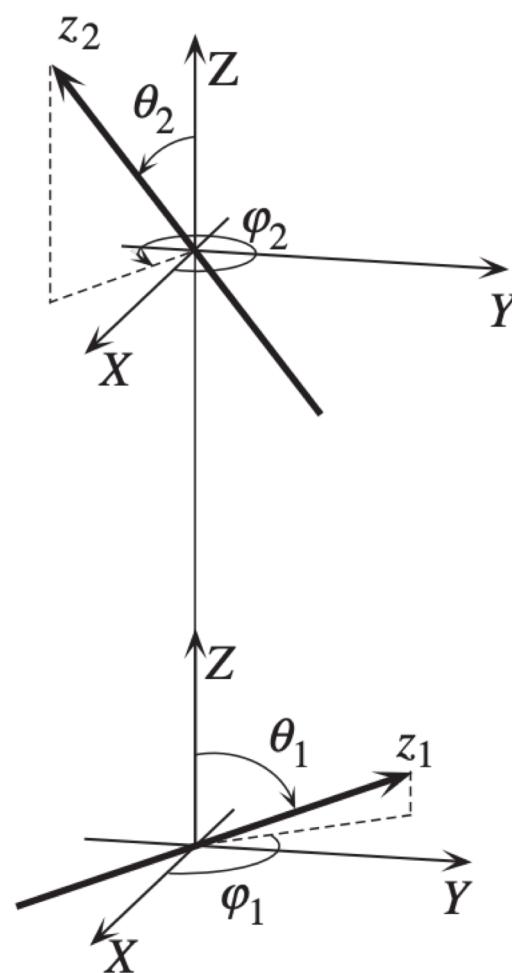
# Cubic spline



```
57  from scipy.interpolate import CubicSpline  
58  cs = CubicSpline(D, Eb)  
59  xs = np.arange(0, np.max(folder)*5.0/100.0, 0.1)  
60  
61  ax.plot(xs, cs(xs), label="S")  
62  ax.plot(xs, cs(xs,1), label="S'")  
63  pold = 0  
64  for n,p in enumerate(cs(xs,1)):  
65      if n > 0 and pold*p < 0:  
66          print (xs[n-1],xs[n], 0.5*(xs[n-1]+xs[n]))  
67  pold = p
```

Run a dimer with 5.05 Å separation again!

# Quadrupole-quadrupole interaction energy



$$U_{QQ} = \frac{Q^A Q^B}{4\pi\epsilon_0 R^5} \frac{3}{4} [1 - 5\cos^2\theta_A - 5\cos^2\theta_B - 15\cos^2\theta_A \cos^2\theta_B + 2(4\cos\theta_A \cos\theta_B - \sin\theta_A \sin\theta_B \cos\varphi)^2]$$

$Q^A$  and  $Q^B$  are the  $Q_{zz}$  of molecule A and B (note: this is only true for linear or highly-symmetrical molecules and when the molecular axis is aligned with z-axis ->

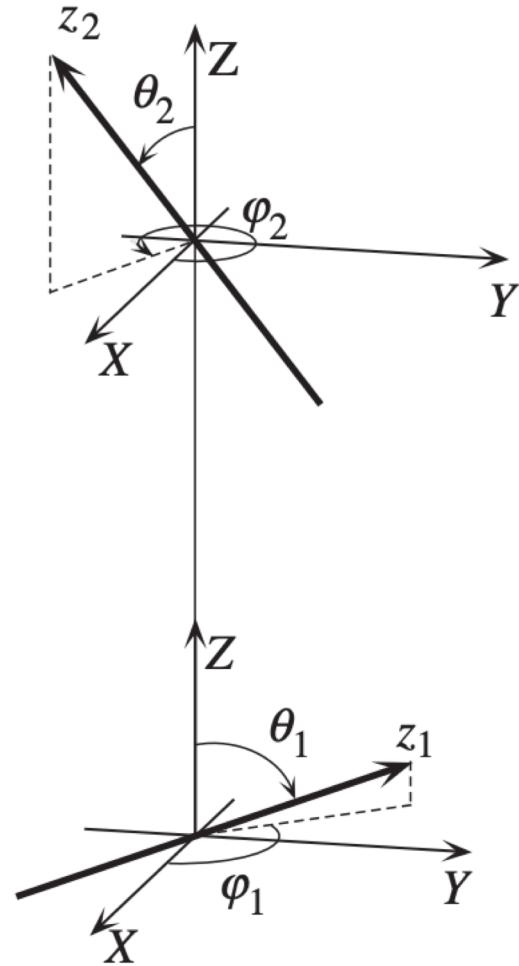
$$Q_{xx} = Q_{yy} = -2Q_{zz}; \text{ off-diagonal terms} = 0)$$

$$U_{QQ} = \frac{Q^A Q^B}{4\pi\epsilon_0 R^5} 6 \text{ for parallel-stacked}$$

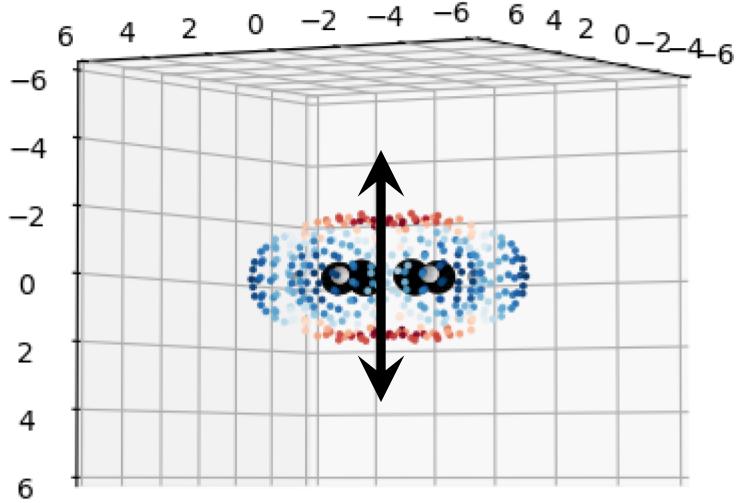
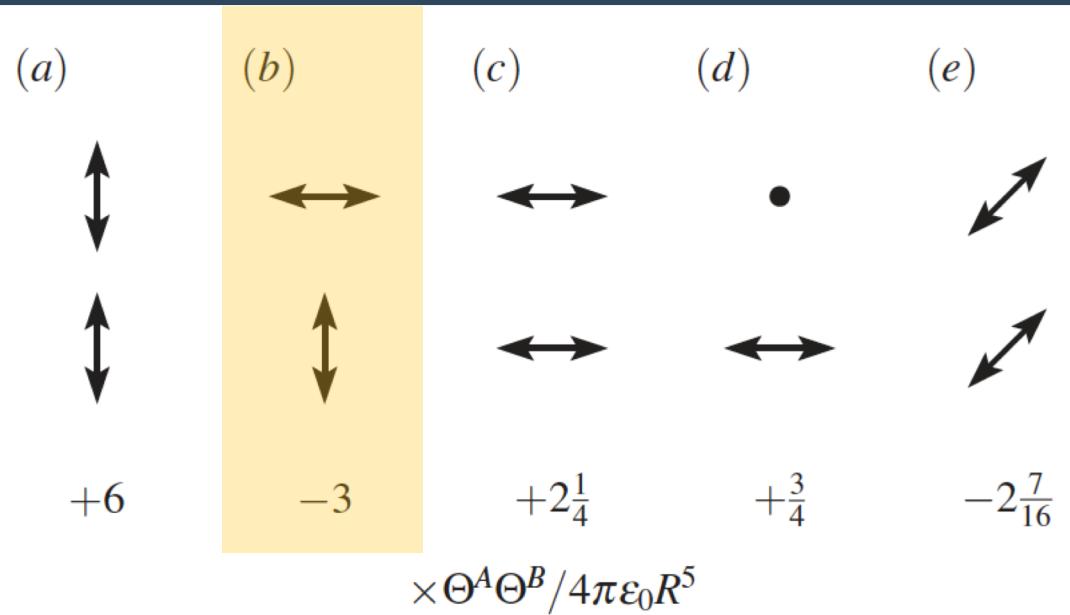
+6

23

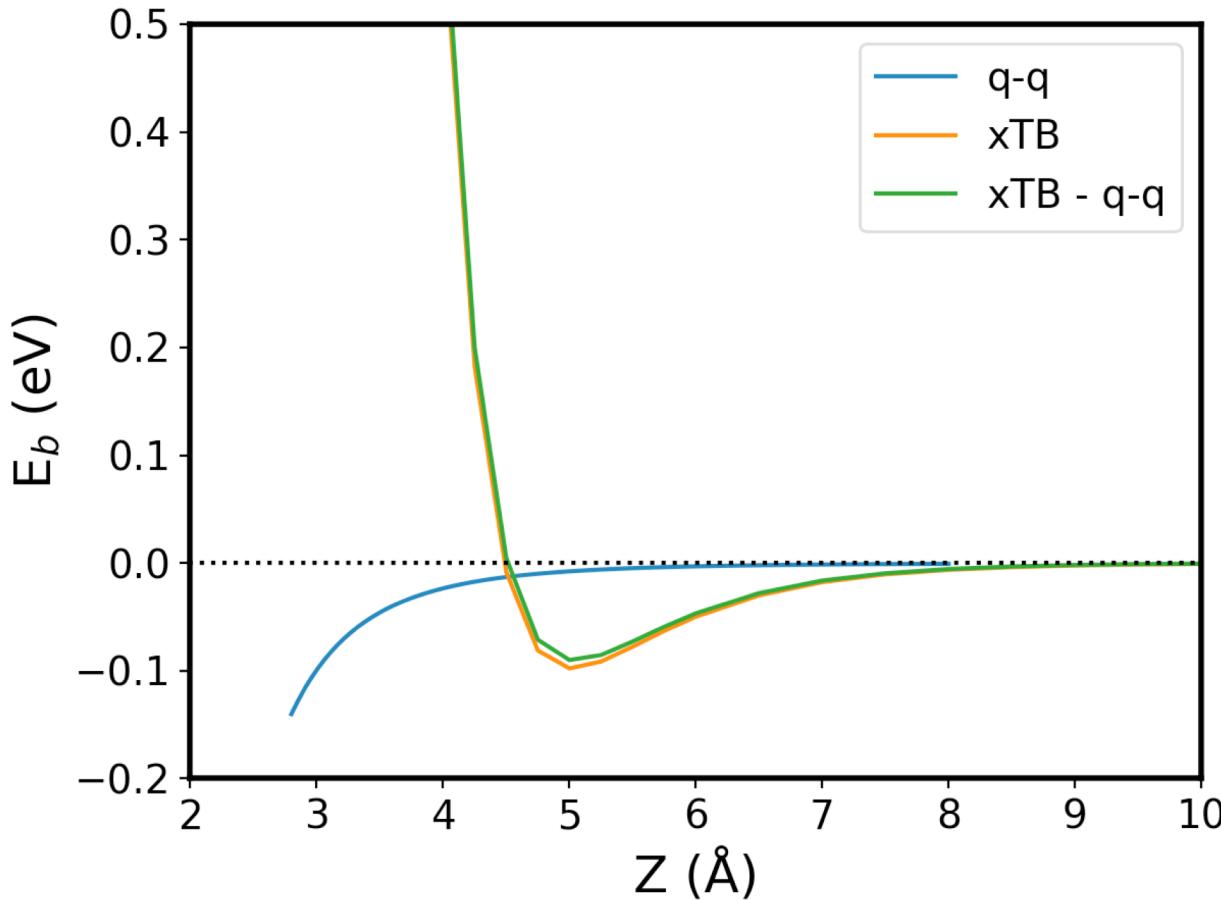
# T-shaped benzene dimer



$\theta_A = 0, \theta_B = 90^\circ$   
 $\varphi$  doesn't matter  
in this case



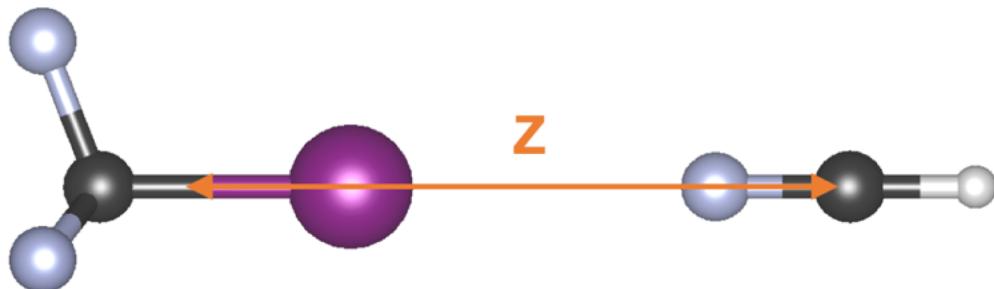
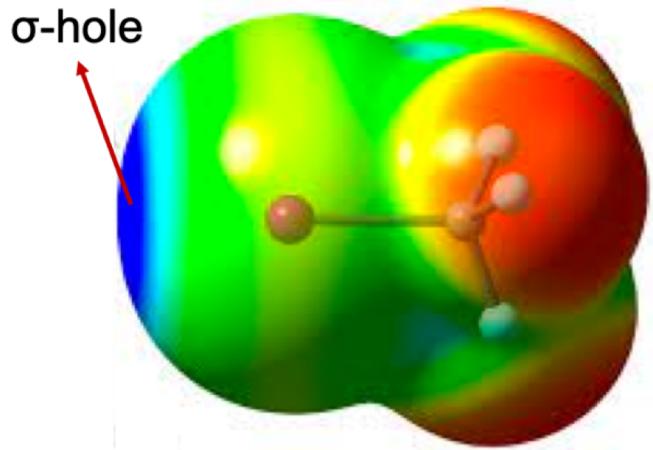
# Comparison between xTB and q-q



The script is similar to `plot_QQ.py`

- It is attraction instead of repulsion (parallel stacked) now!
- The magnitude is still low.

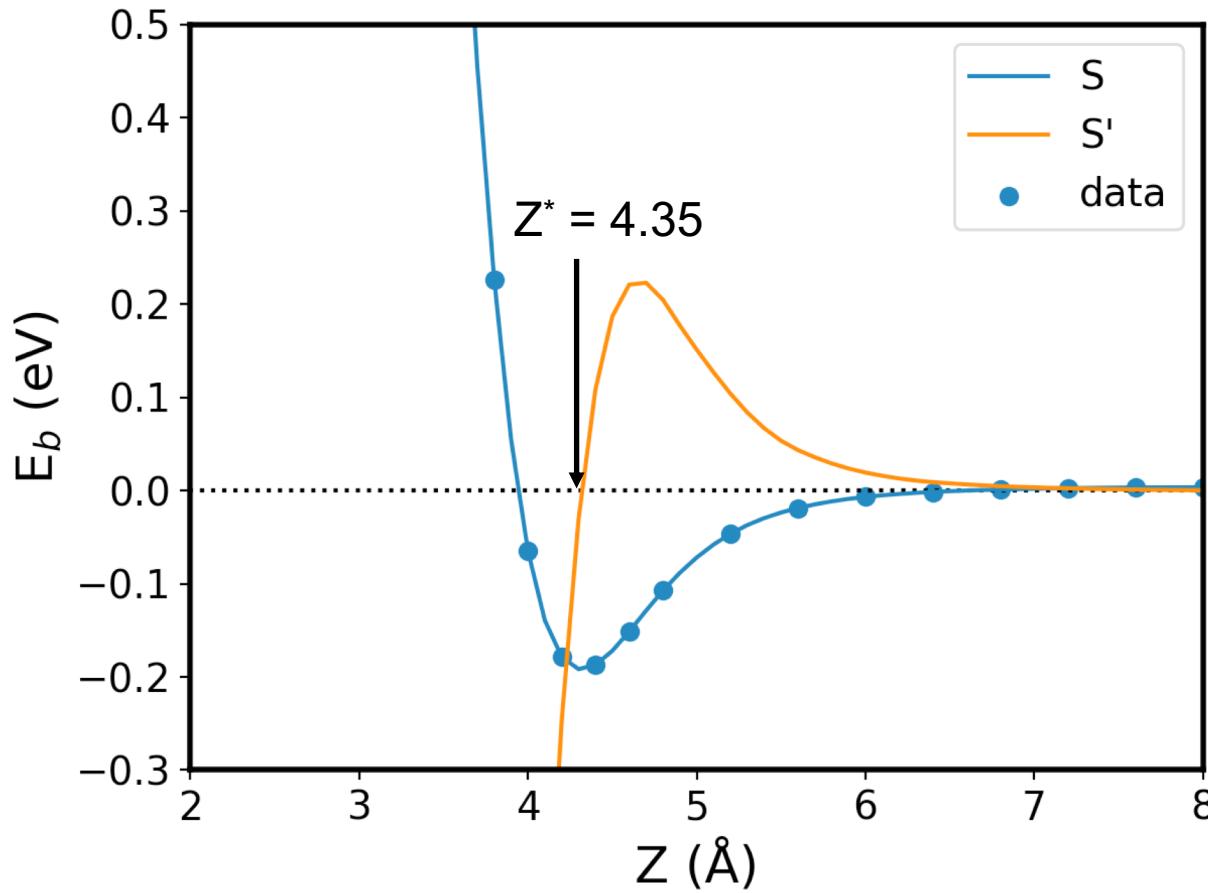
## Q2: Halogen bond pair



### Tasks:

- Create a series of configurations of ICF<sub>3</sub>-NCH dimer with different Z.
- Create the E<sub>b</sub>-Z plot and determine the minimum.
- Evaluate the dimer interactions using multipole-expansion approximation.
- Evaluate the dimer interactions using atomic charge model.

# E<sub>b</sub>-D curve



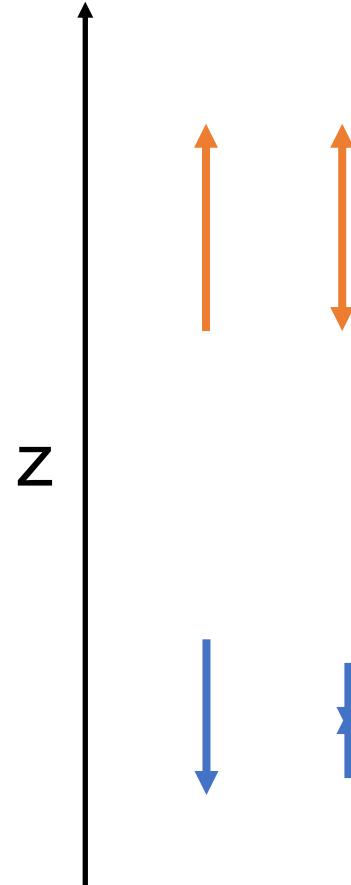
# Dipole and quadrupole of $\text{ICF}_3$ and $\text{NCH}$

$\text{NCH}$

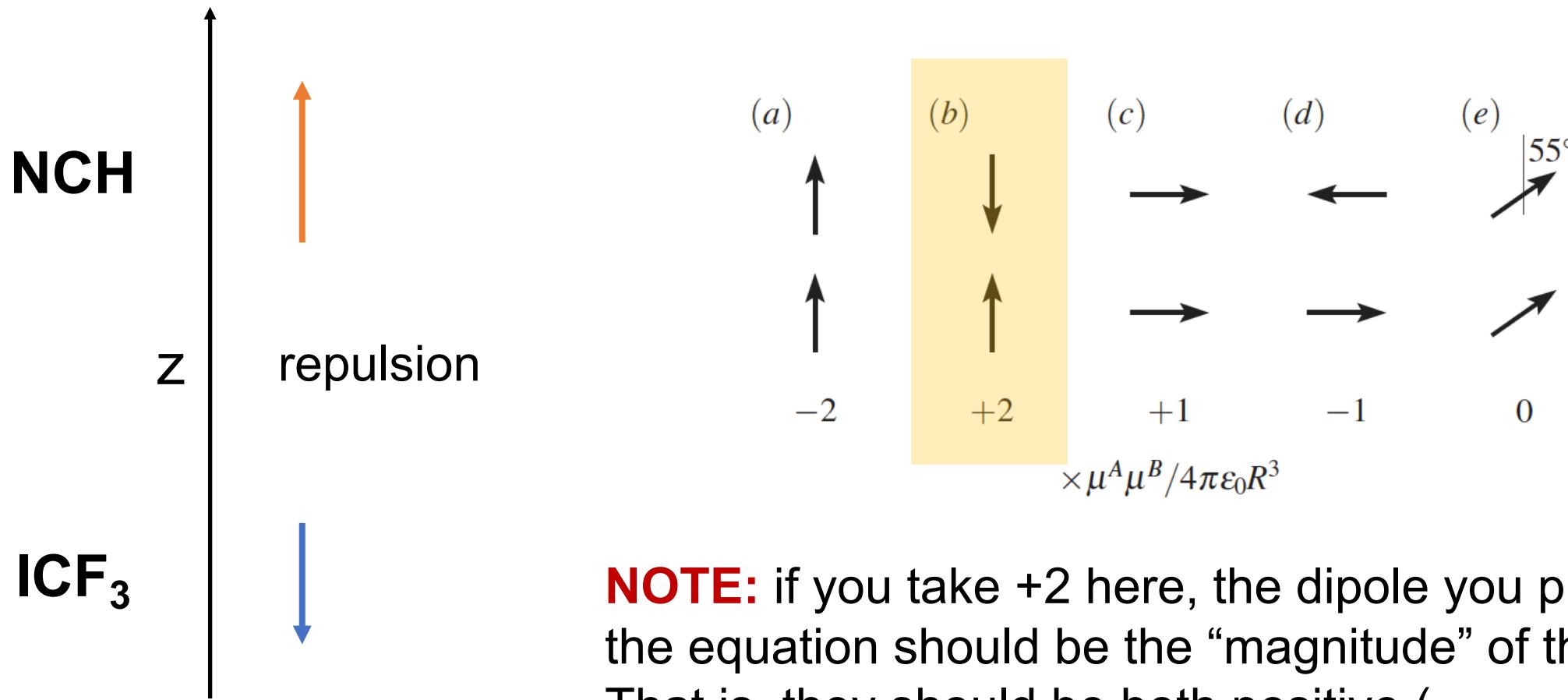
molecular dipole:					
	x	y	z	tot (Debye)	
q only:	-0.000	-0.000	0.505		
full:	-0.000	-0.000	1.031	2.622	
molecular quadrupole (traceless):					
	xx	xy	yy	xz	yz
q only:	-4.149	0.000	-4.149	-0.000	-0.000
q+dip:	-9.040	0.000	-9.040	-0.000	-0.000
full:	-8.090	-0.000	-8.090	-0.000	-0.000
					16.179

$\text{ICF}_3$

molecular dipole:					
	x	y	z	tot (Debye)	
q only:	-0.000	-0.000	0.387		
full:	-0.000	-0.000	-0.328	0.835	
molecular quadrupole (traceless):					
	xx	xy	yy	xz	yz
q only:	0.957	-0.000	0.957	-0.000	0.000
q+dip:	1.353	0.000	1.353	-0.000	0.000
full:	0.510	0.000	0.510	-0.000	0.000
					-1.020

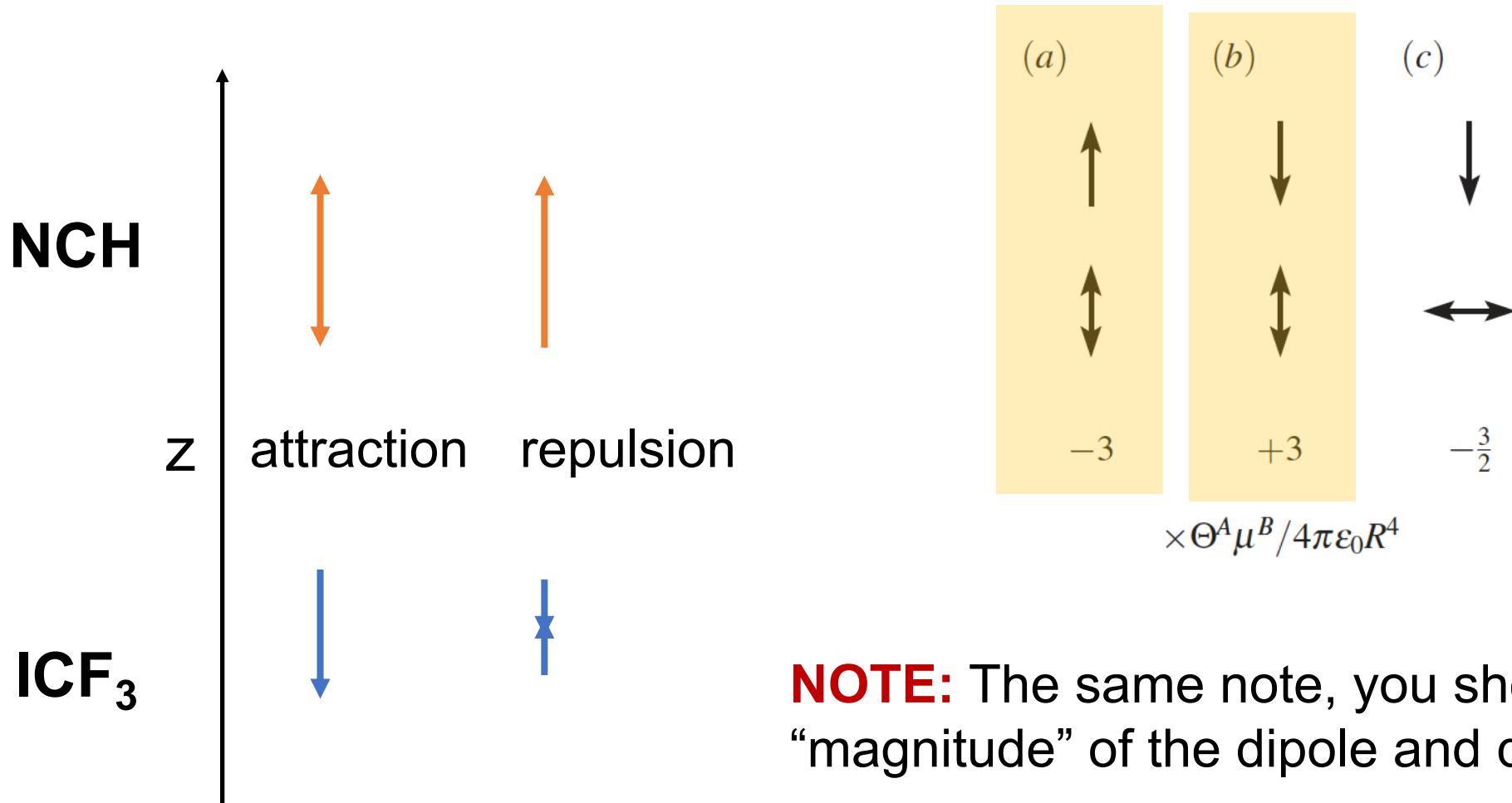


# Dipole-dipole interaction



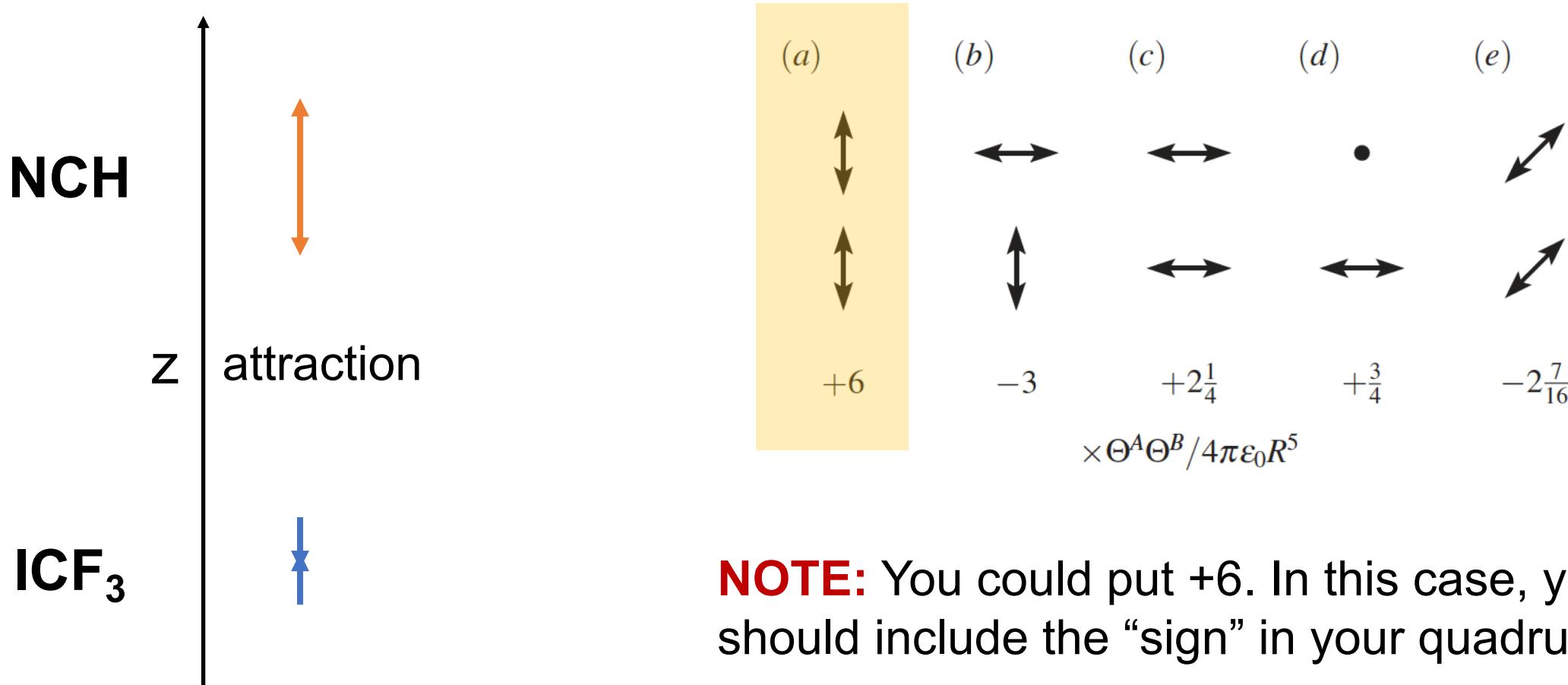
**NOTE:** if you take  $+2$  here, the dipole you put into the equation should be the “magnitude” of the dipole. That is, they should be both positive ( $\mu_A, \mu_B > 0$ ) .

# Dipole-quadrupole interaction

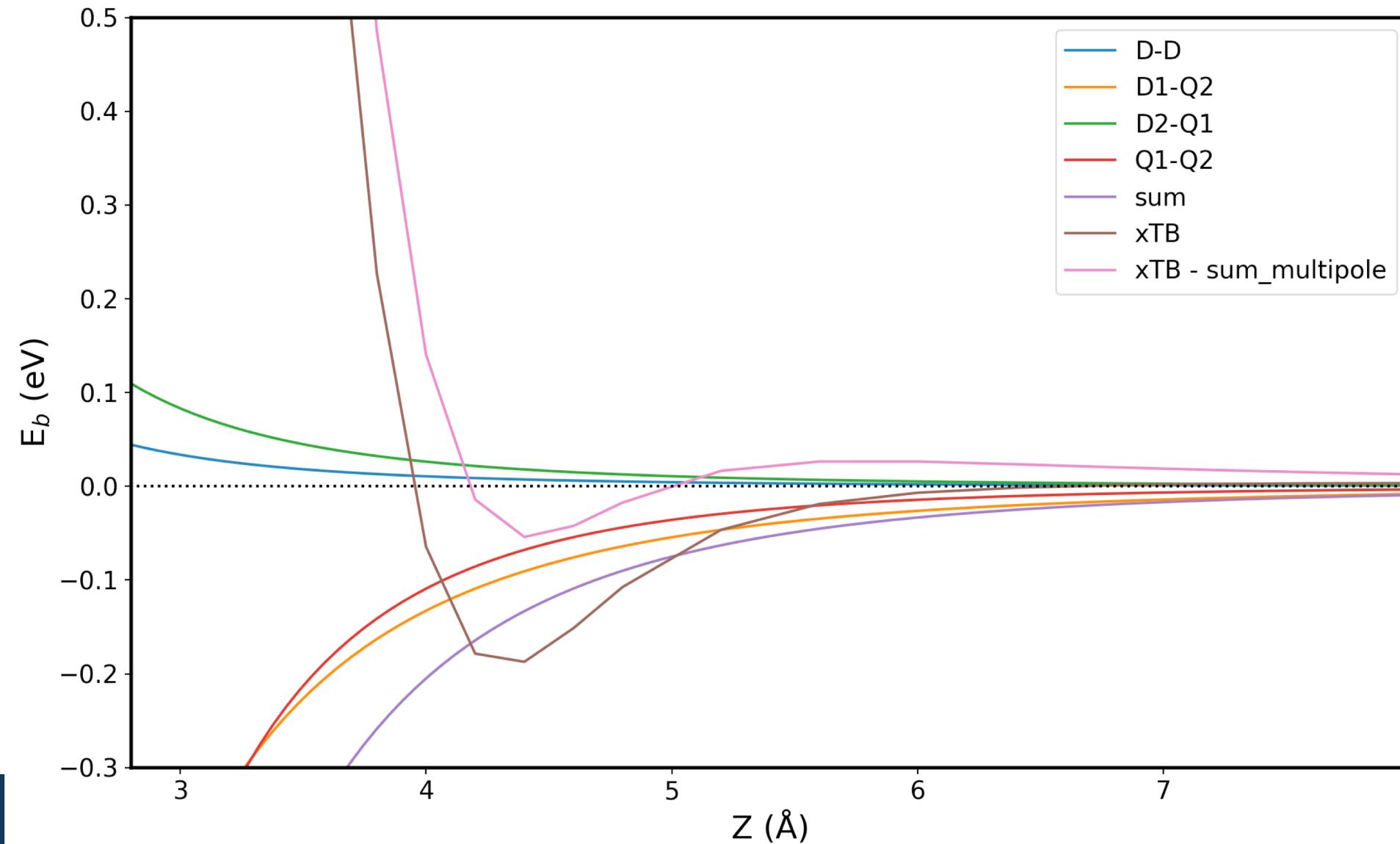


**NOTE:** The same note, you should put the “magnitude” of the dipole and quadrupole.

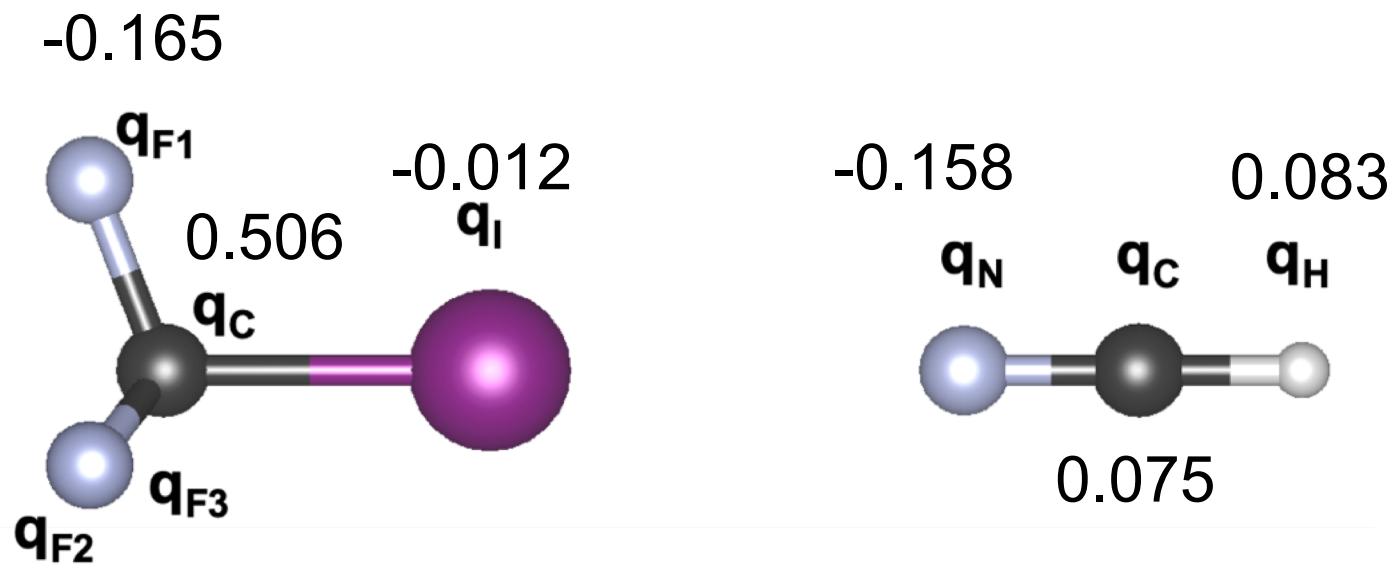
# Quadrupole-quadrupole interaction



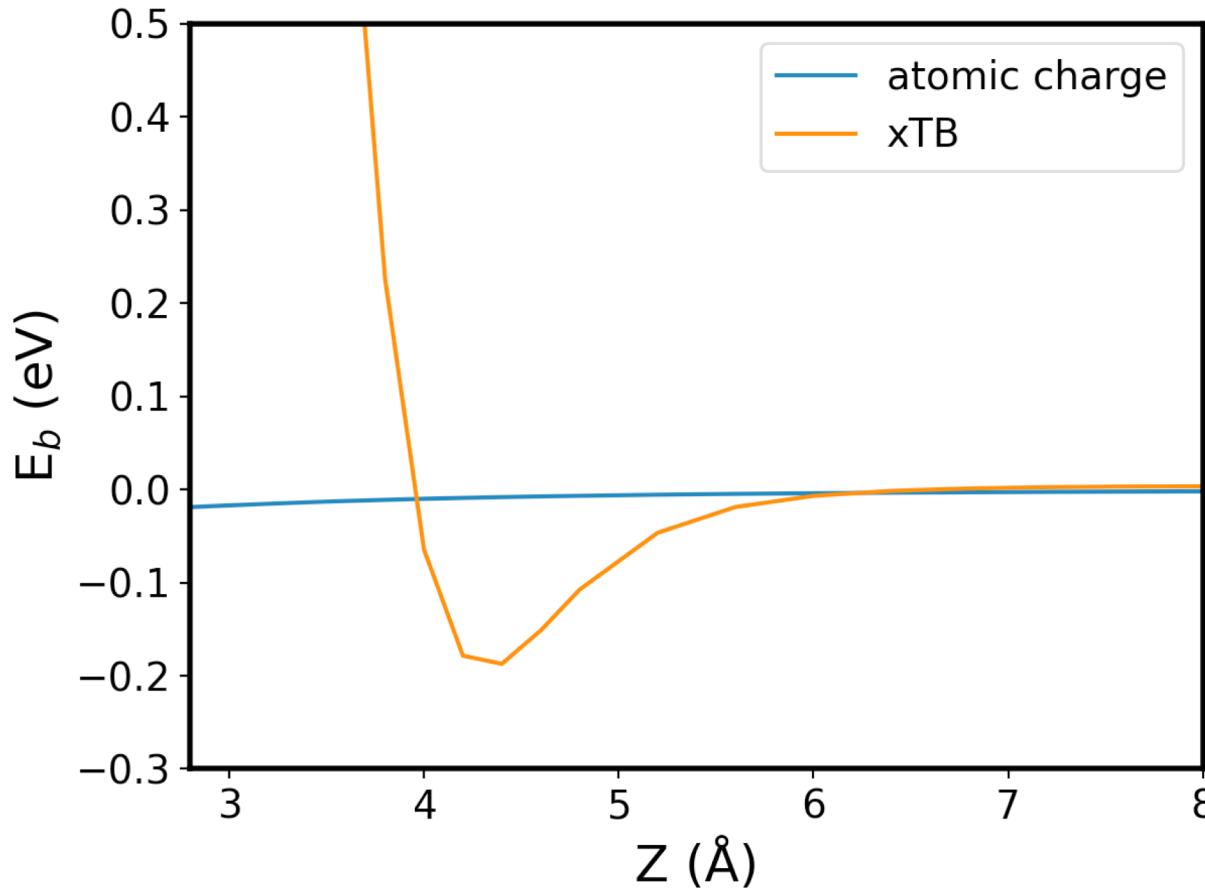
# Comparison between xTB and multipole-multipole



# Mulliken charge

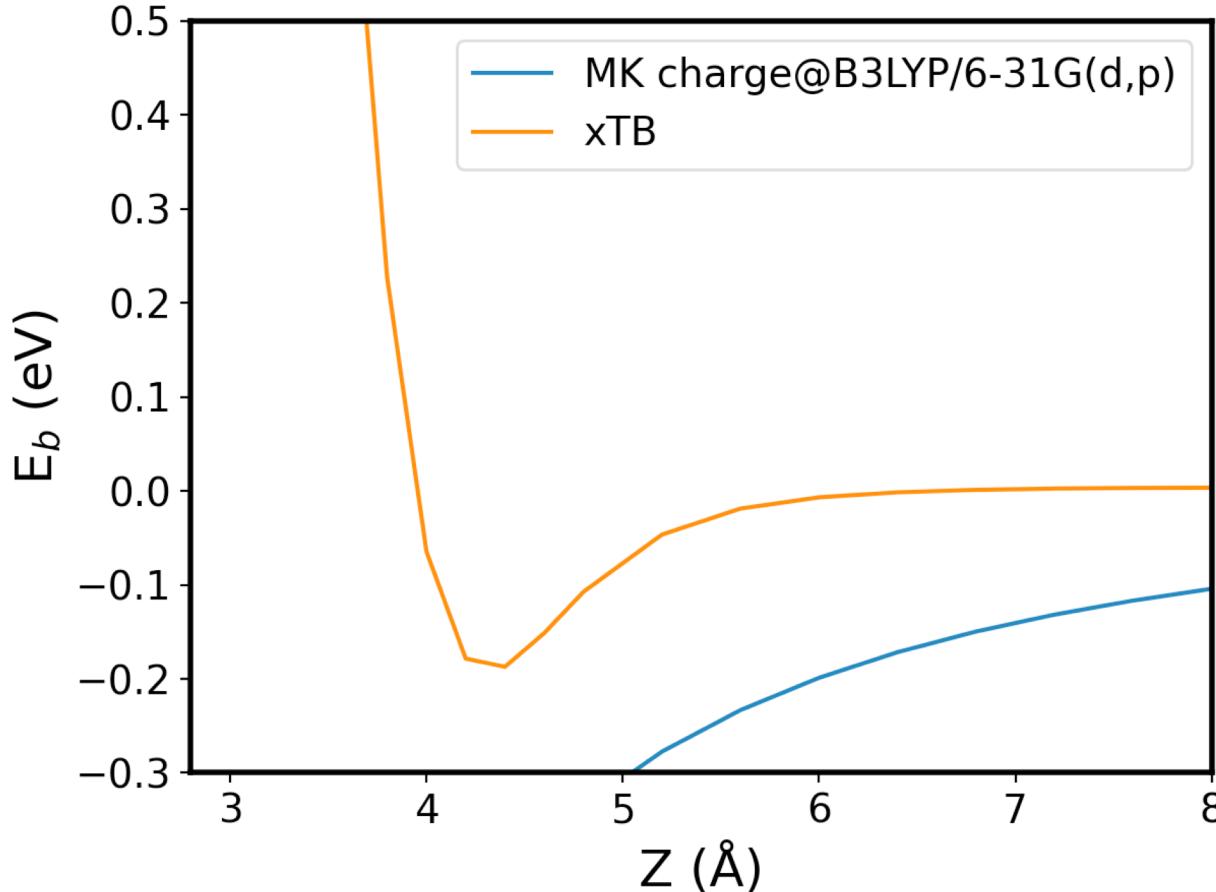


# Comparison between xTB and Mulliken charge



- Doesn't really work...
- Nearly not interacting based on atomic charge model.
- Mulliken charge is not good for this purpose.

# Comparison between xTB and MK charge



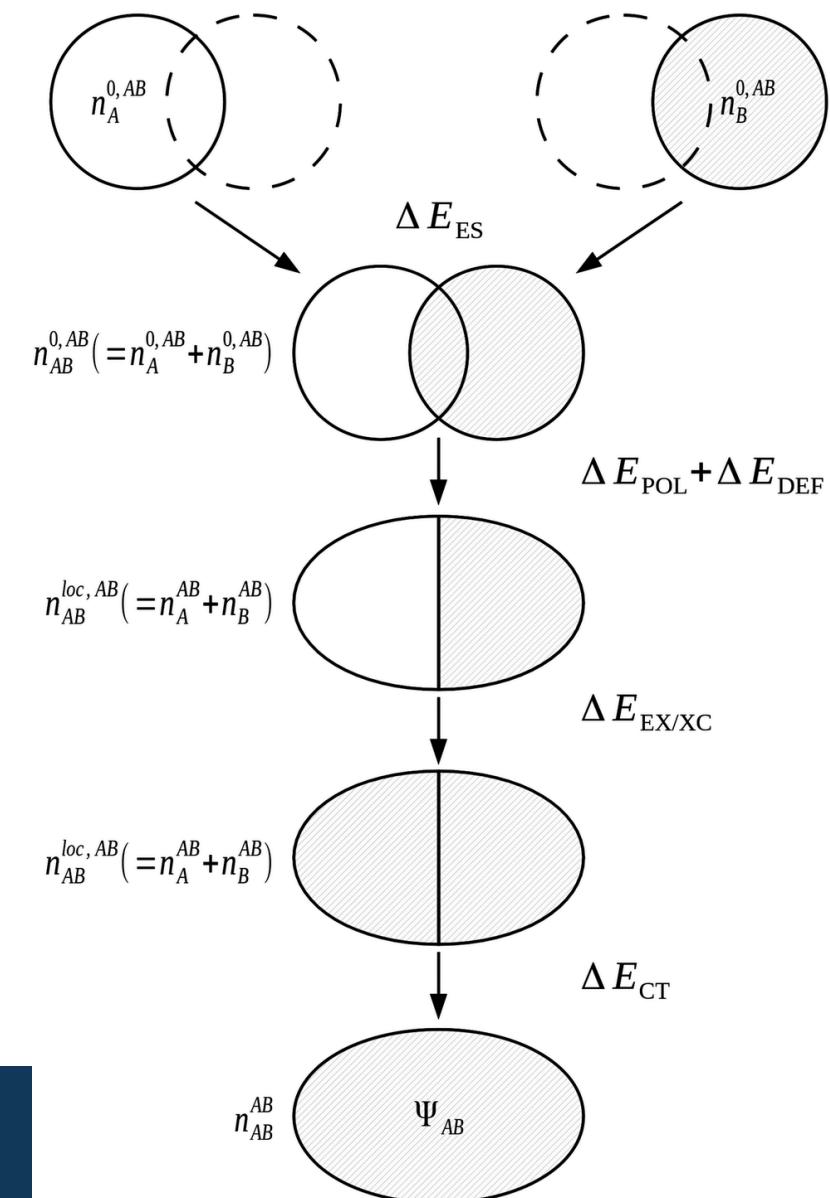
- Supposed to be more accurate than Mulliken charge.
- Common model used in forcefield MD simulations.
- Much stronger attraction.
- Goes to negative faster than multipole-multipole interaction.

# Evaluate electrostatic contribution using QM method?

In general, these methods are called energy decomposition analysis (EDA).

$$\Delta E = \Delta E_{\text{ES}} + \Delta E_{\text{POL}} + \Delta E_{\text{CT}} + \Delta E_{\text{EX/XC}} + \Delta E_{\text{DEF}}$$

Scheme of natural EDA.



# Readings

- Energy decomposition analysis
  - Chem. Soc. Rev., 2015, 44, 3177-3211
- Molecular interactions
  - The Theory of Intermolecular Forces, 2<sup>nd</sup> Ed., Anthony Stone