

QM206 - 2022
PARCIAL I

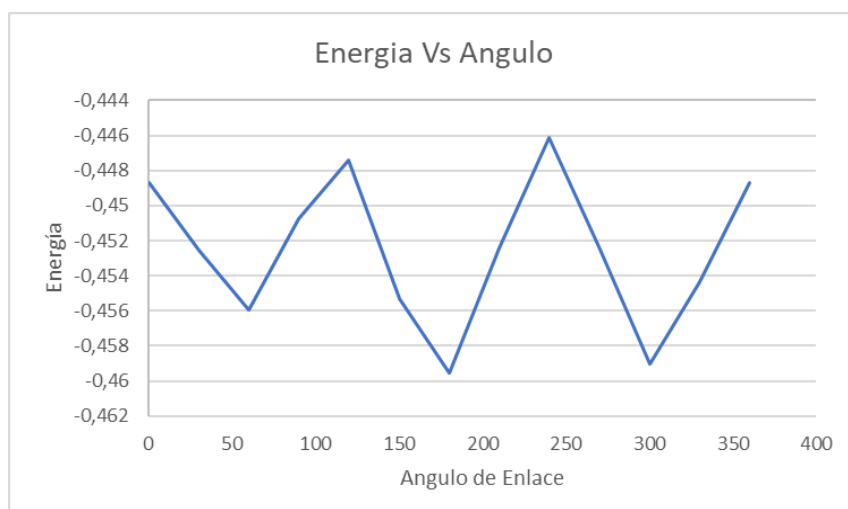
NOMBRE: Angelica Guerra 4-810-752

Realizar los siguientes cálculos utilizando los softwares utilizados en clases. Debe complementar sus cálculos con las imágenes correspondientes y anexar el enlace Github de los archivos generados.

PROBLEMA 1: DIAGRAMA DE ENERGÍA

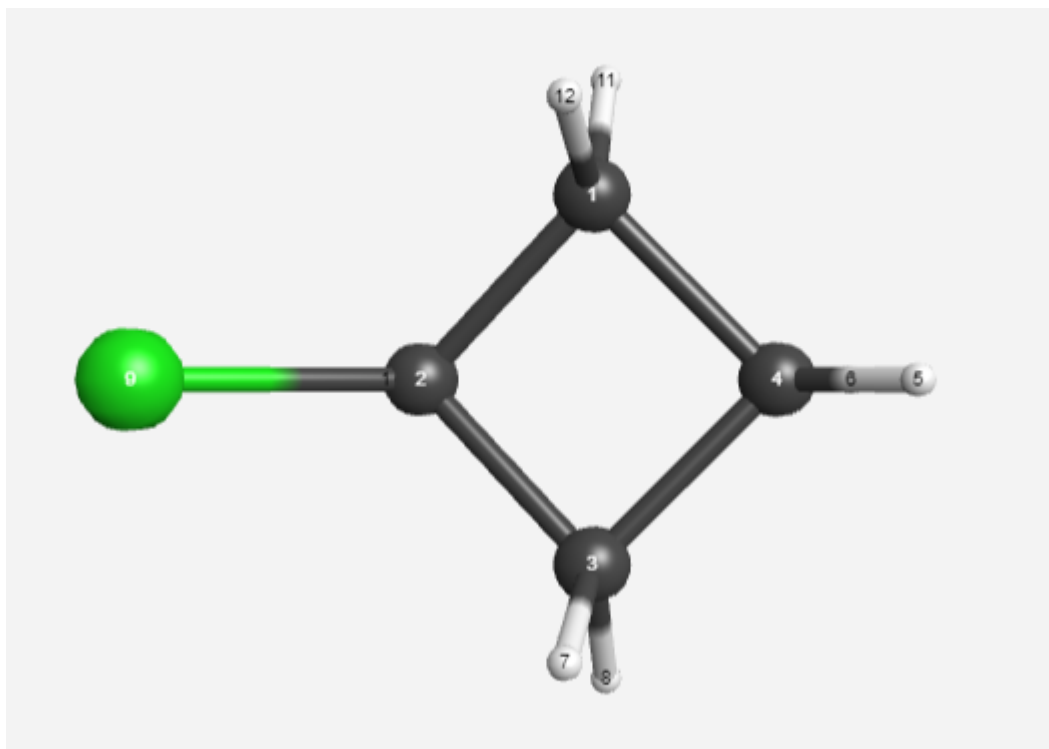
Construya un diagrama de energía vs ángulo de enlace para el 2,3-diclorobutano



Coordinate Scan	
Coordinate	Energy (au)
0.0000	-0.44866
30.0000	-0.45254
60.0000	-0.45592
90.0000	-0.45074
120.0000	-0.44739
150.0000	-0.45530
180.0000	-0.45953
210.0000	-0.45241
240.0000	-0.44615
270.0000	-0.45233
300.0000	-0.45904
330.0000	-0.45434
360.0000	-0.44866
Animation speed 5	
Loop None	
















PROBLEMA 2: OPTIMIZACIÓN DE LA GEOMETRÍA
























a) Optimizar la geometría de la molécula de 1-clorociclobutano



Geometry Sequence Energies		
Step	Energy (au)	  
0	-608.008071102	
1	-608.121933559	
2	-608.229637548	
3	-608.271370538	
4	-608.278459234	
5	-608.278842557	
6	-608.279122288	
7	-608.279268196	
8	-608.279283947	
9	-608.279286234	
10	-608.279286291	
Animation speed		<input type="text" value="5"/>
Loop		<input type="text" value="None"/>

b) Encontrar las energías de los diferentes modos de vibración

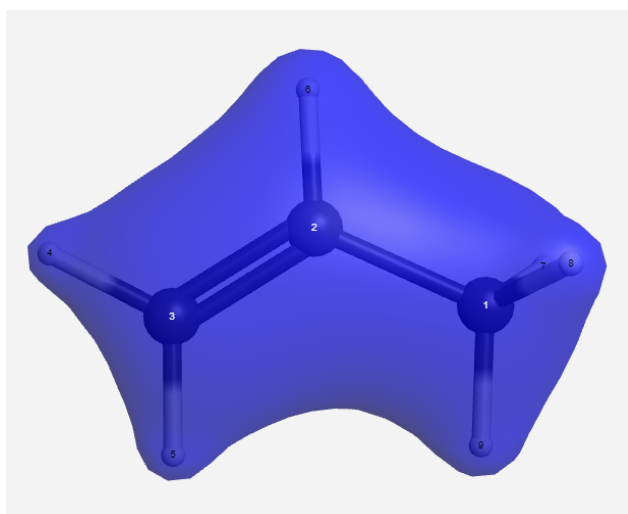
Vibrational Modes						
Show all						
Mode	Symmetry	Frequency (cm ⁻¹)	IR Intensity	Raman Intensity	Actions	
1	A'	95.79 (117.2508)	1.0483	0.2436		
2	A"	266.77 (326.5245)	3.2872	1.9047		
3	A'	323.87 (396.4117)	3.2959	2.8874		
4	A'	549.58 (672.6752)	5.6564	7.8453		
5	A'	712.08 (871.5799)	7.3258	3.6257		
6	A"	751.50 (919.8276)	0.2139	0.3271		
7	A'	820.95 (1004.8408)	18.7560	7.1956		
8	A'	928.64 (1136.6511)	9.2848	5.6033		
9	A"	931.90 (1140.6411)	0.2556	0.5356		
10	A"	951.02 (1164.0379)	0.0541	15.8326		
11	A'	1010.56 (1236.9187)	9.1000	11.6399		
12	A"	1042.28 (1275.7369)	0.7810	4.2080		
13	A'	1070.49 (1310.2717)	1.8116	3.6193		
14	A'	1181.67 (1446.3492)	3.0784	10.7544		
15	A"	1184.12 (1449.3539)	0.0052	9.6984		

16	A''	1223.10 (1497.0657)	0.2214	4.4679		
17	A'	1239.55 (1517.1968)	38.8729	9.2284		
18	A''	1243.10 (1521.5472)	0.0181	0.3087		
19	A''	1294.26 (1584.1603)	1.1228	2.3906		
20	A'	1322.40 (1618.6018)	10.3787	4.2446		
21	A''	1460.47 (1787.5965)	0.1680	3.5089		
22	A'	1463.24 (1790.9969)	0.2330	20.3427		
23	A'	1486.37 (1819.2963)	0.0729	15.7621		
24	A'	2959.25 (3622.0956)	0.1661	23.8940		
25	A''	2962.95 (3626.6276)	0.0347	0.9415		
26	A'	2967.88 (3632.6609)	0.3754	85.1714		
27	A'	2996.52 (3667.7152)	0.0255	42.1254		
28	A'	3065.46 (3752.0880)	0.4817	27.9609		
29	A''	3068.64 (3755.9859)	0.4985	52.1627		
30	A'	3071.06 (3758.9428)	0.0769	14.5515		

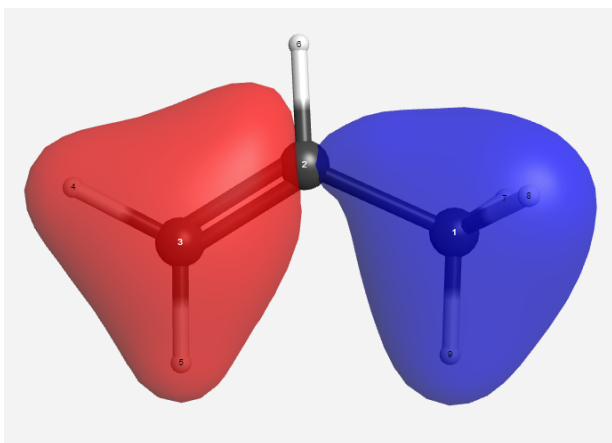
PROBLEMA 3: ORBITALES MOLECULARES

- a) Construya un diagrama de energía para los orbitales moleculares del propileno. Para cada nivel de energía presente la imagen con la estructura del OM correspondiente

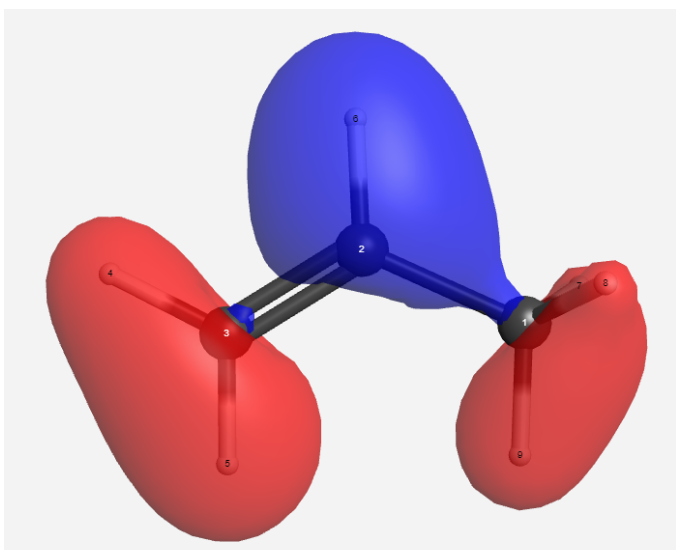
Molecular Orbitals						
Orbital	Symmetry	Occupancy	Spin	Energy (au)	Actions	
<input type="text"/>	<input type="text"/>	<input type="text"/>	<input type="text"/>	<input type="text" value="> -1.30286"/>	<input type="button" value="Show all"/>	
4	-	2	-	-1.00644		
5	-	2	-	-0.87781		
6	-	2	-	-0.70545		
7	-	2	-	-0.58906		
8	-	2	-	-0.53181		
9	-	2	-	-0.53101		
10	-	2	-	-0.46939		
11	-	2	-	-0.44136		
12	-	2	-	-0.30286		
13	-	0	-	0.32069		
14	-	0	-	0.5998		
15	-	0	-	0.65992		
16	-	0	-	0.67543		



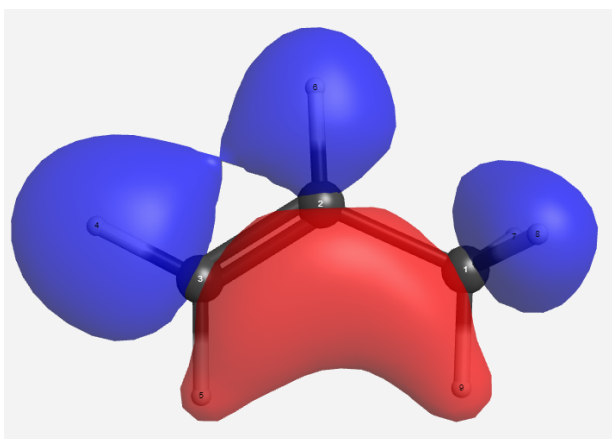
Orbital 4



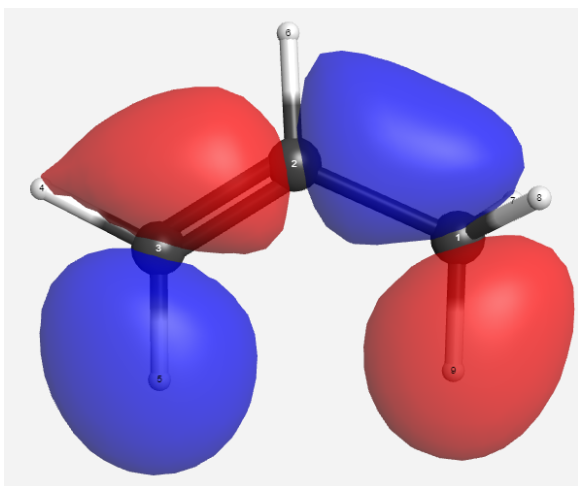
Orbital 5



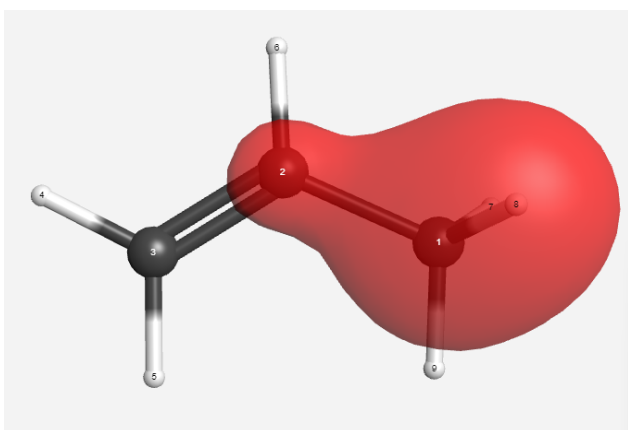
Orbital 6



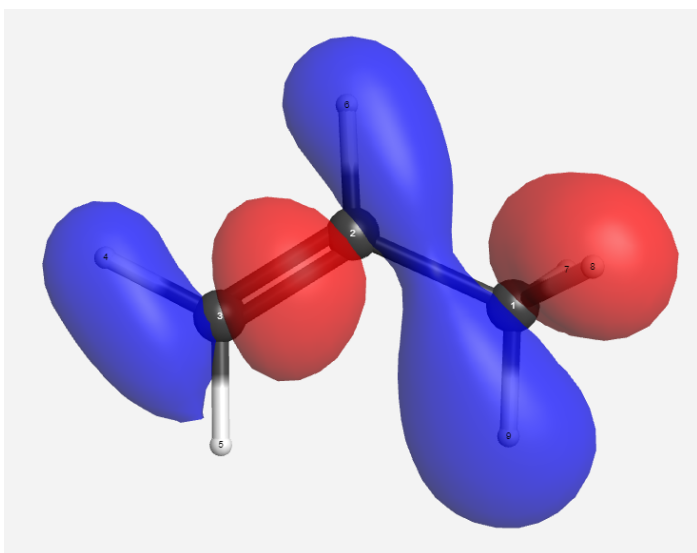
Orbital 7



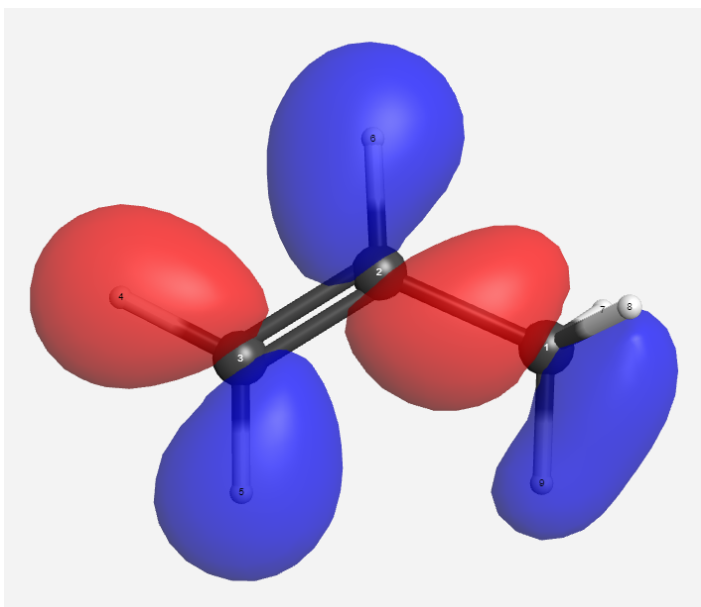
Orbital 8



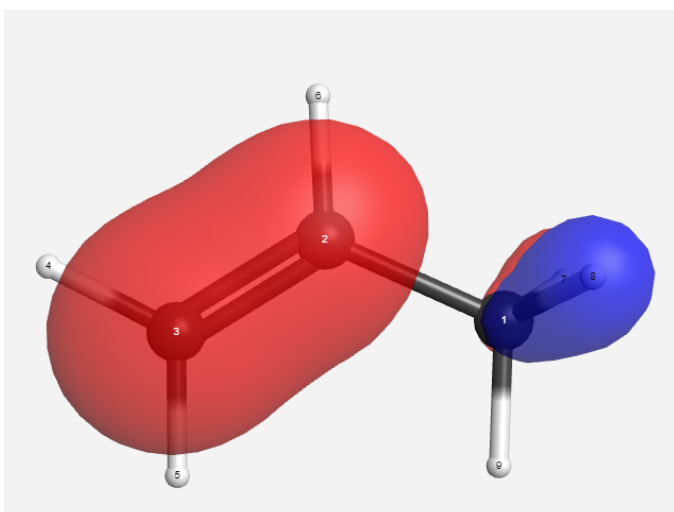
Orbital 9



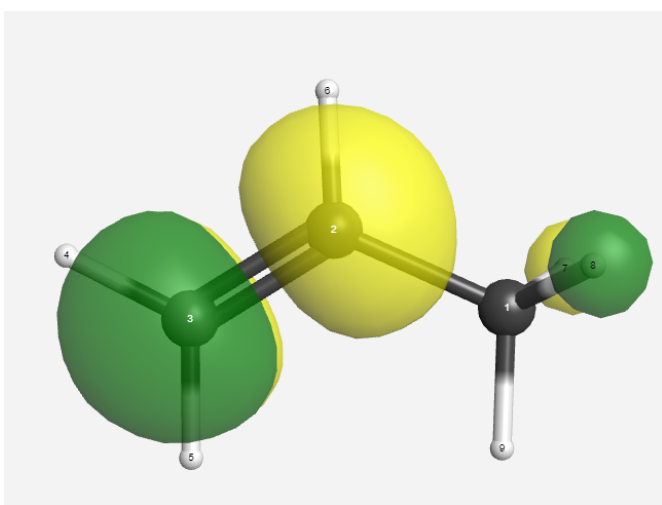
Orbital 10



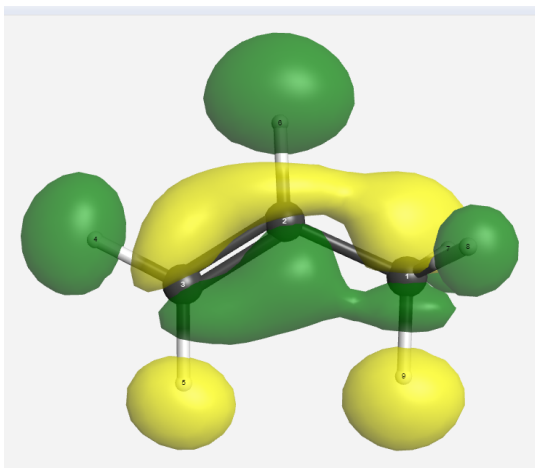
Orbital 11



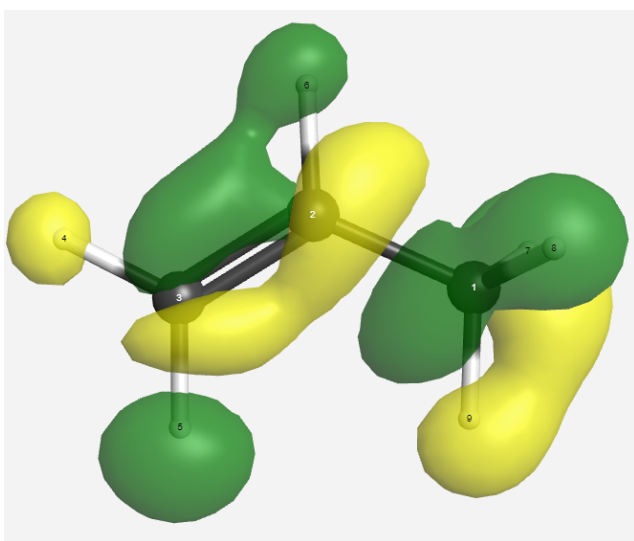
Orbital 12



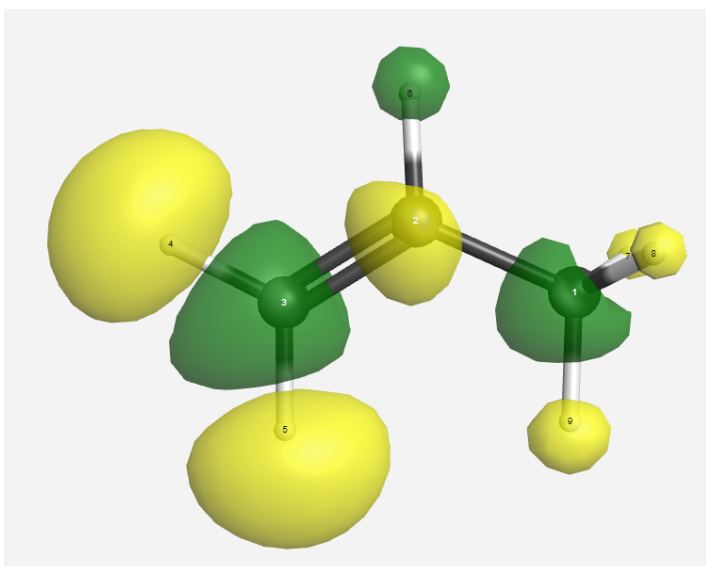
Orbital 13



Orbital 14

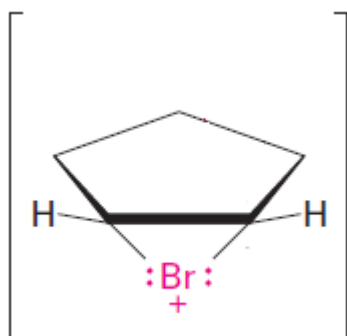


Orbital 15

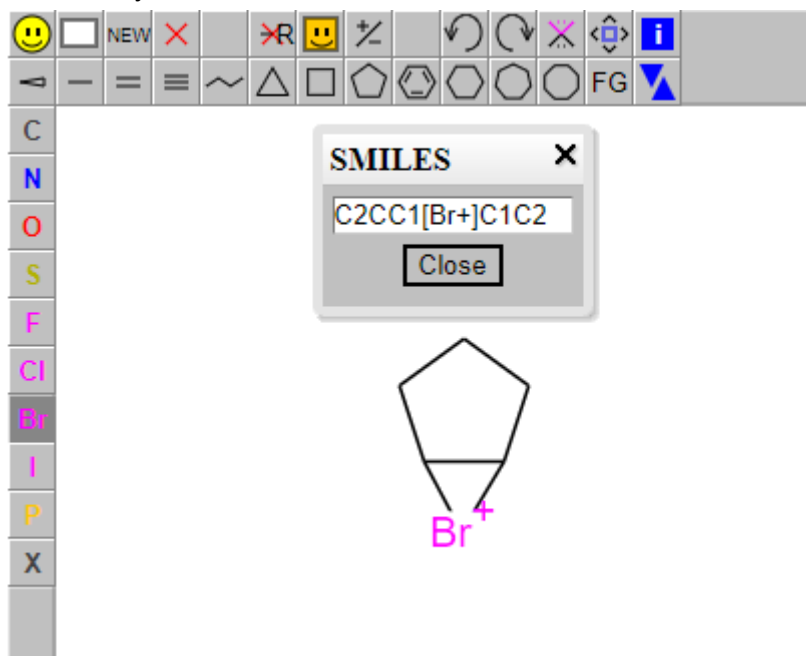


Orbital 16

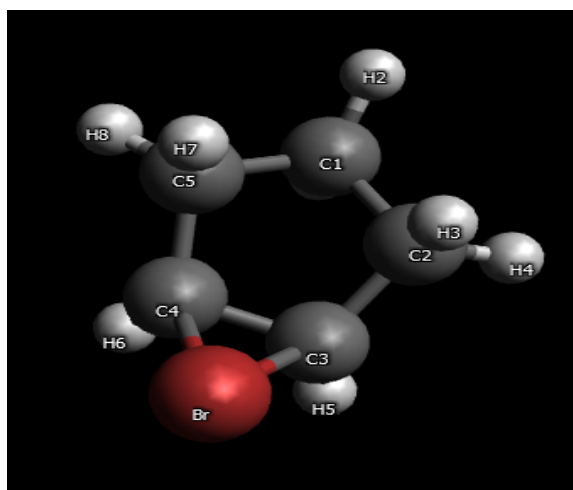
PROBLEMA 4: IÓN BROMONIO

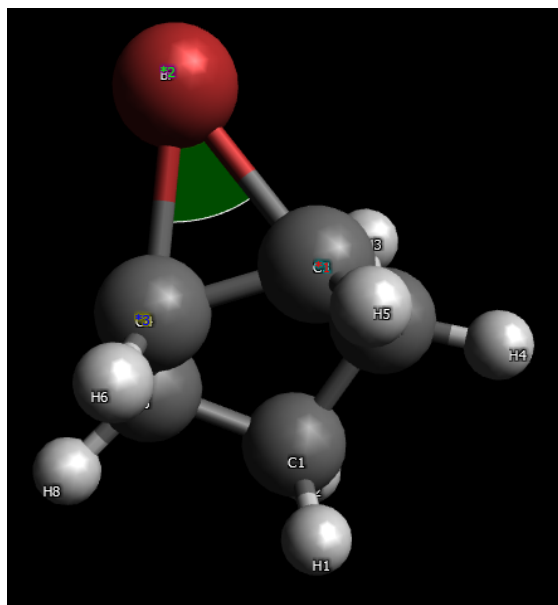


- a) Determine la estructura del ión bromonio indicado. Señale los ángulos correspondientes y distancia de enlaces C-Br



Estructura





Distancia: 1.982 Å
Ángulo C-Br: 43.7°

Angle:	43,7°	
Distance(s):	1,982 Å	1,982 Å

<https://github.com/angelica-017/QM206AngelicaGuerra/blob/main/ANGELICA%20GUERRA%20-%20QM206%20-%202022-parcial1.pdf>