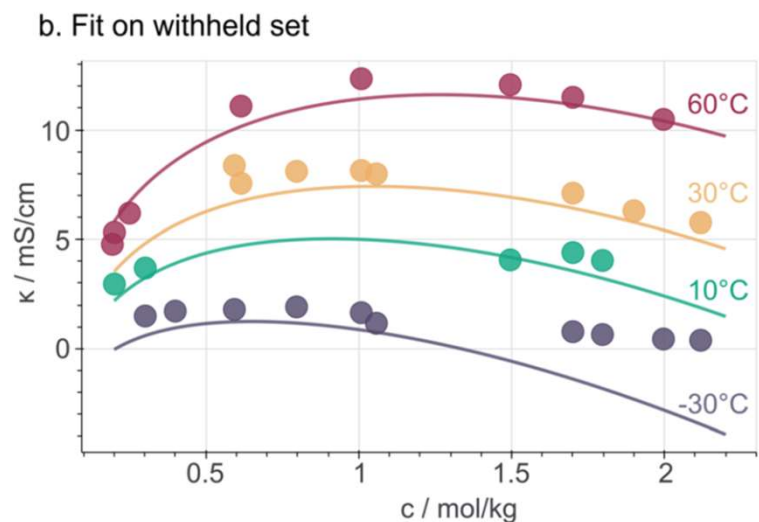


The background features abstract green geometric shapes, including triangles and polygons, in various shades of green, some overlapping and some semi-transparent, creating a modern, layered effect.

## Prediction of low temperature electrolyte conductivity

# Motivation



- Authors modeled electrolyte conductivity as a function of
  - Li salt loading
  - Fraction of propylene carbonate in electrolyte
  - Temperature
- Basically conductivity =  $f(\text{salt loading, PC fraction})$  at a given temperature
- Approach predicted good trends overall
  - Unable to capture the trends at low temperature

Digital  
Discovery

ROYAL SOCIETY  
OF CHEMISTRY

PAPER

[View Article Online](#)  
[View Journal](#) | [View Issue](#)

[Check for updates](#)

Cite this: *Digital Discovery*, 2022, 1, 440

## Learning the laws of lithium-ion transport in electrolytes using symbolic regression†

Eibar Flores,<sup>a\*</sup> Christian Wölke,<sup>b</sup> Peng Yan,<sup>c</sup> Martin Winter,<sup>b,c</sup> Tejs Vegge,<sup>a\*</sup> Isidora Cekic-Laskovic<sup>b</sup> and Arghya Bhowmik<sup>a\*</sup>

<sup>a</sup>Department of Chemistry, Technical University of Denmark, 2800 Lyngby, Denmark. E-mail: eibar.flores@dtu.dk

<sup>b</sup>Department of Chemistry, Technical University of Denmark, 2800 Lyngby, Denmark. E-mail: christian.wolke@dtu.dk

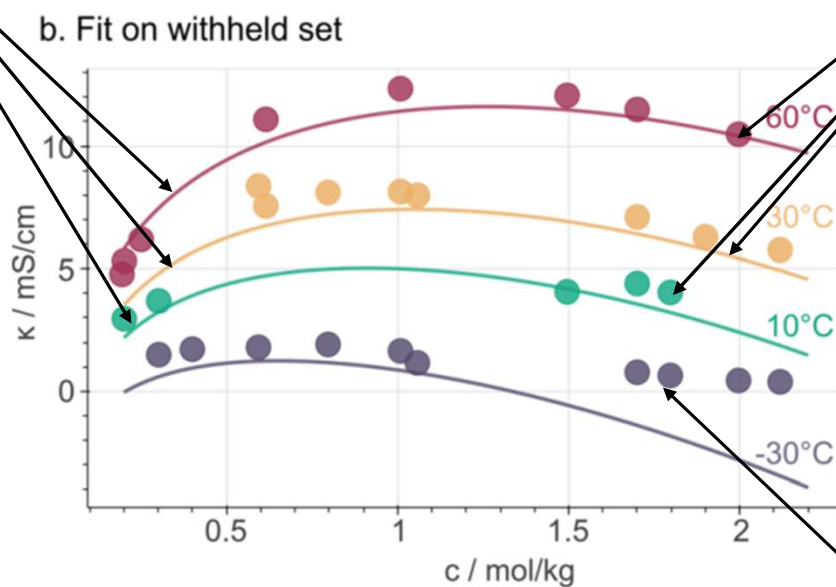
<sup>c</sup>Department of Chemistry, Technical University of Denmark, 2800 Lyngby, Denmark. E-mail: peng.yan@dtu.dk

†Full research paper is available free from [Digital Discovery](#) on 12th September 2022. See [DOI: 10.1039/D2DD00040A](#) for full text.

# Possible explanation for trends

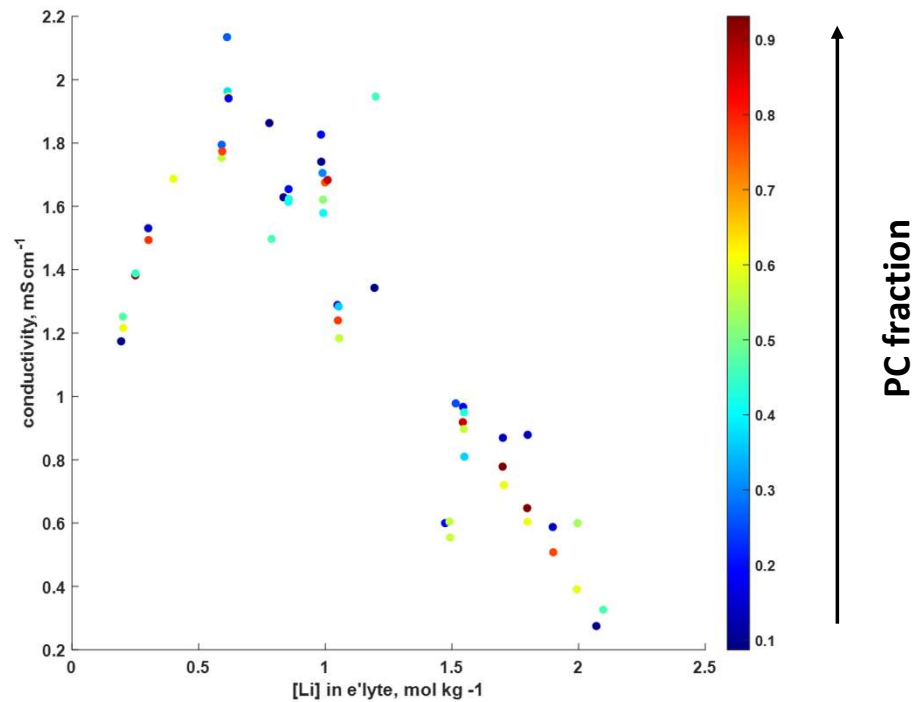
Increase with concentration i.e.  
“billiards” approximation

Falls due to ionic association



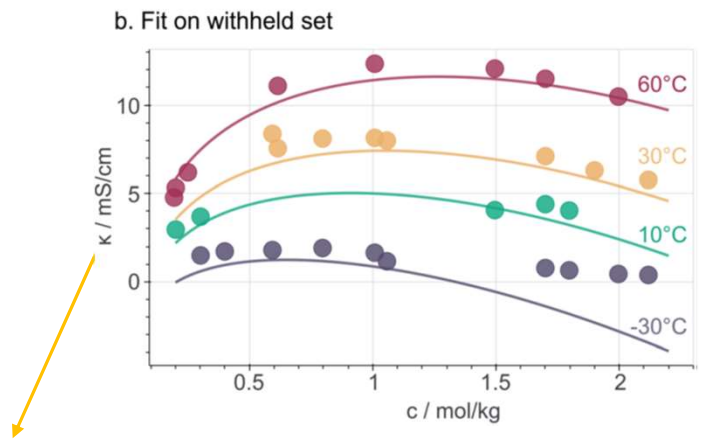
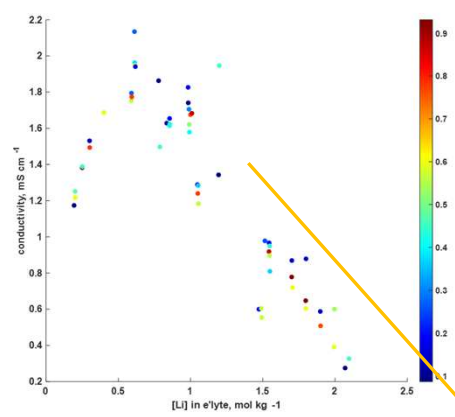
Ionic association ~ coagulation -> falls with lower temperature

How does conductivity vary with salt loading and PC fraction at 243K?



- Zoomed in on the conductivity scale
  - Clearly defined trend with Li salt loading
  - Haphazard behaviour with PC fraction

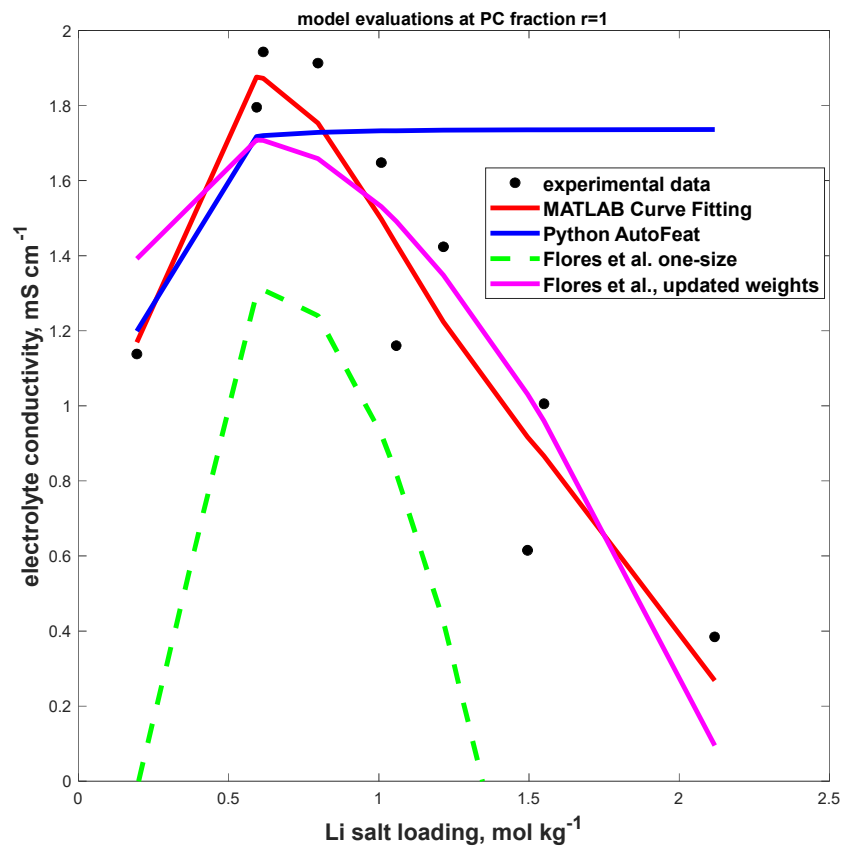
# Proposed corrections



Method	Rationale	Package	Name
Fit a high (≥2)degree polynomial in c, force through zero	Behaviour at constant temperature is essentially dependent on c and not r	MATLAB Curve Fitting Toolbox	polynomial fitting
Train a symbolic regression model in the lower temperature regime	Possible need to change weights due to shallow trends at higher concentration	Python Autofeat	symbolic regression, low T
Use features from Flores et al but update weights in a linear combination, force through zero		Python ski-kitlearn	updated weights

- $c$  = lithium salt loading, mol kg<sup>-1</sup>
- $r$  = fraction of propylene carbonate in electrolyte solvent (along with ethylene carbonate and ethyl methyl carbonate)
- $T$  = temperature (constant for this study)

# Comparing different approaches

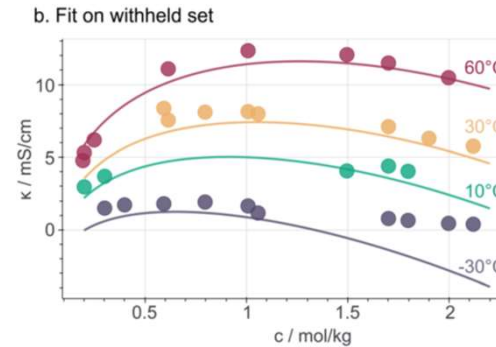


## How do the fitted coefficients change?

$$\kappa = \beta_1 c + \beta_2 T + \beta_3 r^{\frac{1}{2}} c^{\frac{5}{2}} + \beta_4 T^2 c^{\frac{1}{4}}$$

Not implemented since we're considering constant temperature

Coefficient	Flores et al	Current effort
$\beta_1$	-5.11	-2.804
$\beta_2$	-0.04	0
$\beta_3$	-0.35	-0.059
$\beta_4$	$2.73 \times 10^{-4}$	$7.437 \times 10^{-5}$



- The higher value of  $\beta_1$  moves the curve upwards
- Lower values for  $\beta_3$  and  $\beta_4$  help “shallow” the curve