

## Variable conductivity

Semiconductors in their natural state are poor conductors because a [current](#) requires the flow of electrons, and semiconductors have their [valence bands](#) filled, preventing the entry flow of new electrons. There are several developed techniques that allow semiconducting materials to behave like conducting materials, such as [doping](#) or [gating](#). These modifications have two outcomes: n-type and p-type. These refer to the excess or shortage of electrons, respectively. An unbalanced number of electrons would cause a current to flow through the material.<sup>[4]</sup>

## Heterojunctions

[Heterojunctions](#) occur when two differently doped semiconducting materials are joined together. For example, a configuration could consist of p-doped and n-doped [germanium](#). This results in an exchange of electrons and holes between the differently doped semiconducting materials. The n-doped germanium would have an excess of electrons, and the p-doped germanium would have an excess of holes. The transfer occurs until equilibrium is reached by a process called [recombination](#), which causes the migrating electrons from the n-type to come in contact with the migrating holes from the p-type. A product of this process is charged [ions](#), which result in an [electric field](#).<sup>[1][4]</sup>

### Excited electrons

A difference in electric potential on a semiconducting material would cause it to leave thermal equilibrium and create a non-equilibrium situation. This introduces electrons and holes to the system, which interact via a process called [ambipolar diffusion](#). Whenever thermal equilibrium is disturbed in a semiconducting material, the amount of holes and electrons changes. Such disruptions can occur as a result of a temperature difference or [photons](#), which can enter the system and create electrons and holes. The process that creates and annihilates electrons and holes are called [generation](#) and [recombination](#).<sup>[4]</sup>

### Light emission

In certain semiconductors, excited electrons can relax by emitting light instead of producing heat.<sup>[5]</sup> These semiconductors are used in the construction of [light-emitting diodes](#) and fluorescent [quantum dots](#).

### Thermal energy conversion

Semiconductors have large [thermoelectric power factors](#) making them useful in [thermoelectric generators](#), as well as high [thermoelectric figures of merit](#) making them useful in [thermoelectric coolers](#).<sup>[6]</sup>

# Materials<sup>[edit]</sup>

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