

Biology

Biology is the scientific study of <u>life</u>. [1][2][3] It is a <u>natural science</u> with a broad scope but has several unifying themes that tie it together as a single, coherent field. [1][2][3] For instance, all <u>organisms</u> are composed of at least one <u>cell</u> that processes hereditary information encoded in <u>genes</u>, which can be transmitted to future generations. Another major theme is <u>evolution</u>, which explains the unity and diversity of life. [1][2][3] <u>Energy processing</u> is also important to life as it allows organisms to <u>move</u>, grow, and <u>reproduce</u>. [1][2][3] Finally, all organisms can regulate their own <u>internal</u> environments. [1][2][3][4][5]



Biology is the science of life. It spans multiple levels from biomolecules and cells to organisms and populations.

Biologists can study life at multiple levels of organization, [1] from the molecular biology of a cell to the anatomy and

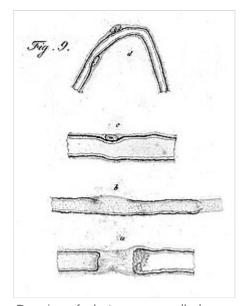
<u>physiology</u> of plants and animals, and the evolution of populations. Hence, there are multiple <u>subdisciplines</u> within biology, each defined by the nature of their <u>research questions</u> and the <u>tools</u> that they use. Like other scientists, biologists use the <u>scientific method</u> to make <u>observations</u>, pose questions, generate <u>hypotheses</u>, perform experiments, and form conclusions about the world around them.

Life on Earth, which emerged over 3.7 billion years ago, [10] is immensely diverse. Biologists have sought to study and classify the various life form, from <u>prokaryotic</u> organisms such as <u>archaea</u> and bacteria to <u>eukaryotic</u> organisms such as <u>protists</u>, fungi, plants, and animals. These organisms contribute to the <u>biodiversity</u> of an <u>ecosystem</u>, where they play specialized roles in the <u>cycling</u> of <u>nutrients</u> and energy through their <u>biophysical environment</u>.

History

The earliest of roots of science, which included medicine, can be traced to <u>ancient Egypt</u> and <u>Mesopotamia</u> in around 3000 to 1200 <u>BCE</u>. Their contributions shaped ancient Greek <u>natural philosophy</u>. Ancient Greek philosophers such as <u>Aristotle</u> (384–322 BCE) contributed extensively to the development of biological knowledge. He explored biological causation and the diversity of life. His successor, <u>Theophrastus</u>, began the scientific study of plants. Scholars of the medieval Islamic world who wrote on biology included <u>al-Jahiz</u> (781–869), <u>Al-Dīnawarī</u> (828–896), who wrote on botany, and <u>Rhazes</u> (865–925) who wrote on <u>anatomy</u> and <u>physiology</u>. Medicine was especially well studied by <u>Islamic scholars</u> working in Greek philosopher traditions, while natural history drew heavily on Aristotelian thought.

Biology began to quickly develop with Anton van Leeuwenhoek's dramatic improvement of the microscope. It was then that scholars discovered spermatozoa, bacteria, infusoria and the diversity of microscopic life. Investigations by Jan Swammerdam led to new interest in entomology and helped to develop techniques of microscopic dissection and staining. [19] Advances in microscopy had a profound impact on biological thinking. In the early 19th century, biologists pointed to the central importance of the cell. In 1838, Schleiden and Schwann began promoting the now universal ideas that (1) the basic unit of organisms is the cell and (2) that individual cells have all the characteristics of life, although they opposed the idea that (3) all cells come from the division of other cells, continuing to support spontaneous generation. However, Robert Remak and Rudolf Virchow were able to reify the third tenet, and by the 1860s most biologists accepted all three tenets which consolidated into cell theory. [20][21]



Drawing of what now are called Schwann cells by one of the founders of cell theory, Theodor Schwann.

Meanwhile, taxonomy and classification became the focus of natural historians. <u>Carl Linnaeus</u> published a basic <u>taxonomy</u> for the natural world in 1735, and in the 1750s introduced <u>scientific</u>

<u>names</u> for all his species. [22] <u>Georges-Louis Leclerc</u>, <u>Comte de Buffon</u>, treated species as artificial categories and living forms as malleable—even suggesting the possibility of common descent. [23]

Serious evolutionary thinking originated with the works of Jean-Baptiste Lamarck, who presented a coherent theory of evolution. The British naturalist Charles Darwin, combining the biogeographical approach of Humboldt, the uniformitarian geology of Lyell, Malthus's writings on population growth, and his own morphological expertise and extensive natural observations, forged a more successful evolutionary theory based on natural selection; similar reasoning and evidence led Alfred Russel Wallace to independently reach the same conclusions. [26][27]

The basis for modern genetics began with the work of <u>Gregor Mendel</u> in 1865. This outlined the principles of biological inheritance. However, the significance of his work was not realized until the early 20th century when evolution became a unified theory as the <u>modern synthesis</u> reconciled Darwinian evolution with <u>classical genetics</u>. In the 1940s and early 1950s, a <u>series of experiments</u> by <u>Alfred Hershey</u> and <u>Martha</u> Chase pointed to DNA as the component of chromosomes that held the



In 1842, <u>Charles Darwin</u> penned his first sketch of <u>On the Origin of</u> <u>Species</u>. [24]

trait-carrying units that had become known as <u>genes</u>. A focus on new kinds of model organisms such as <u>viruses</u> and bacteria, along with the discovery of the double-helical structure of DNA by <u>James Watson</u> and <u>Francis Crick</u> in 1953, marked the transition to the era of <u>molecular genetics</u>. From the 1950s onwards, biology has been vastly extended in the <u>molecular domain</u>. The <u>genetic code</u> was cracked by <u>Har Gobind Khorana</u>, <u>Robert W. Holley</u> and <u>Marshall Warren Nirenberg</u> after DNA was understood to contain <u>codons</u>. The <u>Human Genome Project</u> was launched in 1990 to map the human <u>genome</u>. [31]

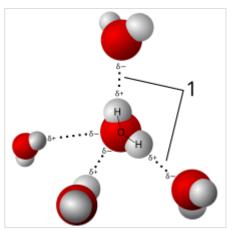
Chemical basis

Atoms and molecules

All organisms are made up of chemical elements; [32] oxygen, carbon, hydrogen, and nitrogen account for most (96%) of the mass of all organisms, with calcium, phosphorus, sulfur, sodium, chlorine, and magnesium constituting essentially all the remainder. Different elements can combine to form compounds such as water, which is fundamental to life. [32] Biochemistry is the study of chemical processes within and relating to living organisms. Molecular biology is the branch of biology that seeks to understand the molecular basis of biological activity in and between cells, including molecular synthesis, modification, mechanisms, and interactions.

Water

Life arose from the Earth's first ocean, which formed some 3.8 billion years ago. [33] Since then, water continues to be the most abundant molecule in every organism. Water is important to life because it is an effective solvent, capable of dissolving solutes such as sodium and chloride ions or other small molecules to form an aqueous solution. Once dissolved in water, these solutes are more likely to come in contact with one another and therefore take part in chemical reactions that sustain life. [33] In terms of its molecular structure, water is a small polar molecule with a bent shape formed by the polar covalent bonds of two hydrogen (H) atoms to one oxygen (O) atom (H_2O) . [33] Because the O–H bonds are polar, the oxygen atom has a slight negative charge and the two hydrogen atoms have a slight positive charge. [33] This polar property of water allows it to attract other water molecules via



Model of hydrogen bonds (1) between molecules of water

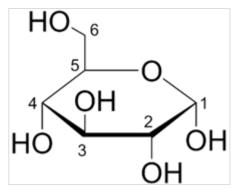
hydrogen bonds, which makes water <u>cohesive</u>. Surface tension results from the cohesive force due to the attraction between molecules at the surface of the liquid. Water is also <u>adhesive</u> as it is able to adhere to the surface of any polar or charged non-water molecules. Water is <u>denser</u> as a <u>liquid</u> than it is as a solid (or ice). This unique property of water allows ice to float above liquid water such as ponds, lakes, and oceans, thereby <u>insulating</u> the liquid below from the cold air above. Water has the capacity to absorb energy, giving it a higher <u>specific heat capacity</u> than other solvents such as <u>ethanol</u>. Thus, a large amount of energy is needed to break the hydrogen bonds between water molecules to convert liquid water into <u>water vapor</u>. As a molecule, water is not completely stable as each water molecule continuously dissociates into hydrogen and <u>hydroxyl</u> ions before reforming into a water molecule again. In <u>pure water</u>, the number of hydrogen ions balances (or equals) the number of hydroxyl ions, resulting in a pH that is neutral.

Organic compounds

<u>Organic compounds</u> are molecules that contain carbon bonded to another element such as hydrogen. With the exception of water, nearly all the molecules that make up each organism contain carbon. Carbon can form covalent bonds with up to four other atoms, enabling it to form diverse, large, and

complex molecules. [33][34] For example, a single carbon atom can form four single covalent bonds such as in methane, two double covalent bonds such as in carbon dioxide (CO₂), or a triple covalent bond such as in carbon monoxide (CO). Moreover, carbon can form very long chains of interconnecting carbon—carbon bonds such as octane or ring-like structures such as glucose.

The simplest form of an organic molecule is the <u>hydrocarbon</u>, which is a large family of organic compounds that are composed of hydrogen atoms bonded to a chain of carbon atoms. A hydrocarbon backbone can be substituted by other elements such as oxygen (O), hydrogen (H), phosphorus (P), and sulfur (S),



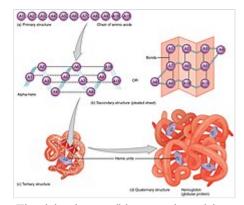
Organic compounds such as glucose are vital to organisms.

which can change the chemical behavior of that compound. [33] Groups of atoms that contain these elements (O-, H-, P-, and S-) and are bonded to a central carbon atom or skeleton are called <u>functional groups</u>. [33] There are six prominent functional groups that can be found in organisms: <u>amino group</u>, carboxyl group, carbonyl group, hydroxyl group, phosphate group, and sulfhydryl group. [33]

In 1953, the <u>Miller–Urey experiment</u> showed that organic compounds could be synthesized abiotically within a closed system mimicking the conditions of <u>early Earth</u>, thus suggesting that complex organic molecules could have arisen spontaneously in early Earth (see abiogenesis). [35][33]

Macromolecules

Macromolecules are large molecules made up of smaller subunits or monomers. [36] Monomers include sugars, amino acids, and nucleotides.[37] Carbohydrates include monomers and polymers of sugars. [38] Lipids are the only class of macromolecules that are not made up of polymers. They include steroids, phospholipids, and fats, [37] largely nonpolar and hydrophobic (water-repelling) substances.[39] Proteins are the most diverse macromolecules. They include enzymes, transport proteins, large signaling molecules, antibodies, and structural proteins. The basic unit (or monomer) of a protein is an amino acid. [36] Twenty amino acids are used in proteins. [36] Nucleic acids are polymers of nucleotides. [40] Their function is to store, transmit, and express hereditary information.[37]



The (a) primary, (b) secondary, (c) tertiary, and (d) quaternary structures of a hemoglobin protein

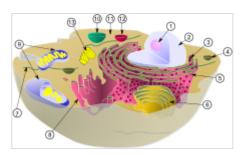
Cells

<u>Cell theory</u> states that <u>cells</u> are the fundamental units of life, that all living things are composed of one or more cells, and that all cells arise from preexisting cells through <u>cell division</u>. Most cells are very small, with diameters ranging from 1 to 100 <u>micrometers</u> and are therefore only visible under a <u>light</u> or electron microscope. There are generally two types of cells: eukaryotic cells, which contain a nucleus,

and <u>prokaryotic</u> cells, which do not. Prokaryotes are <u>single-celled organisms</u> such as <u>bacteria</u>, whereas eukaryotes can be single-celled or multicellular. In <u>multicellular organisms</u>, every cell in the organism's body is derived ultimately from a single cell in a fertilized egg.

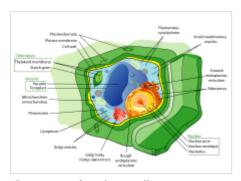
Cell structure

Every cell is enclosed within a <u>cell membrane</u> that separates its <u>cytoplasm</u> from the <u>extracellular space</u>. A cell membrane consists of a <u>lipid bilayer</u>, including <u>cholesterols</u> that sit between phospholipids to maintain their <u>fluidity</u> at various temperatures. Cell membranes are <u>semipermeable</u>, allowing small molecules such as oxygen, carbon dioxide, and water to pass through while restricting the movement of larger molecules and charged particles such as <u>ions</u>. Cell membranes also contain <u>membrane</u> proteins, including <u>integral membrane</u> proteins that go across the membrane serving as membrane transporters, and peripheral proteins that



Structure of an animal $\underline{\text{cell}}$ depicting various organelles

loosely attach to the outer side of the cell membrane, acting as enzymes shaping the cell. [45] Cell membranes are involved in various cellular processes such as cell adhesion, storing electrical energy, and cell signalling and serve as the attachment surface for several extracellular structures such as a cell wall, glycocalyx, and cytoskeleton.



Structure of a plant cell

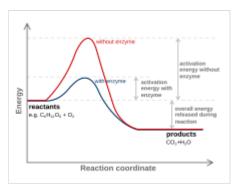
Within the cytoplasm of a cell, there are many biomolecules such as proteins and nucleic acids. [46] In addition to biomolecules, eukaryotic cells have specialized structures called organelles that have their own lipid bilayers or are spatially units. [47] These organelles include the cell nucleus, which contains most of the cell's DNA, or mitochondria, which generate adenosine triphosphate (ATP) to power cellular processes. Other organelles such as endoplasmic reticulum and Golgi apparatus play a role in the synthesis and packaging of proteins, respectively. Biomolecules such as proteins can be engulfed by lysosomes, another specialized organelle. Plant cells have additional

organelles that distinguish them from <u>animal cells</u> such as a cell wall that provides support for the plant cell, <u>chloroplasts</u> that harvest sunlight energy to produce sugar, and <u>vacuoles</u> that provide storage and structural support as well as being involved in reproduction and breakdown of plant seeds. <u>[47]</u> Eukaryotic cells also have cytoskeleton that is made up of <u>microtubules</u>, <u>intermediate filaments</u>, and <u>microfilaments</u>, all of which provide support for the cell and are involved in the movement of the cell and its organelles. <u>[47]</u> In terms of their structural composition, the microtubules are made up of <u>tubulin</u> (e.g., <u> α -tubulin</u>) whereas intermediate filaments are made up of fibrous proteins. <u>[47]</u> Microfilaments are made up of <u>actin</u> molecules that interact with other strands of proteins. <u>[47]</u>

Metabolism

All cells require energy to sustain cellular processes. <u>Metabolism</u> is the set of <u>chemical reactions</u> in an organism. The three main purposes of metabolism are: the conversion of food to energy to run cellular processes; the conversion of food/fuel to monomer building blocks; and the elimination of <u>metabolic</u> wastes. These enzyme-catalyzed reactions allow organisms to grow and reproduce, maintain their

structures, and respond to their environments. Metabolic reactions may be categorized as <u>catabolic</u>—the breaking down of compounds (for example, the breaking down of glucose to pyruvate by <u>cellular respiration</u>); or <u>anabolic</u>—the building up (synthesis) of compounds (such as proteins, carbohydrates, lipids, and nucleic acids). Usually, catabolism releases energy, and anabolism consumes energy. The chemical reactions of metabolism are organized into <u>metabolic pathways</u>, in which one chemical is transformed through a series of steps into another chemical, each step being facilitated by a specific enzyme. Enzymes are crucial to metabolism because they allow organisms to drive desirable reactions that require energy that will not occur

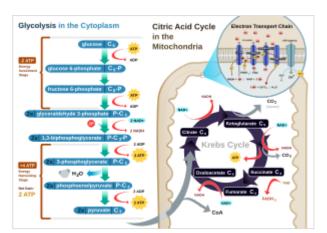


Example of an enzyme-catalysed exothermic reaction

by themselves, by <u>coupling</u> them to <u>spontaneous reactions</u> that release energy. Enzymes act as <u>catalysts</u>—they allow a reaction to proceed more rapidly without being consumed by it—by reducing the amount of <u>activation energy</u> needed to convert <u>reactants</u> into <u>products</u>. Enzymes also allow the <u>regulation</u> of the rate of a metabolic reaction, for example in response to changes in the cell's environment or to signals from other cells.

Cellular respiration

Cellular respiration is a set of metabolic reactions and processes that take place in cells to convert chemical energy from nutrients into adenosine triphosphate (ATP), and then release waste products. [48] The reactions involved in respiration are catabolic reactions, which break large molecules into smaller ones, releasing energy. Respiration is one of the key ways a cell releases chemical energy to fuel cellular activity. The overall reaction occurs in a series of biochemical steps, some of which are redox reactions. Although cellular respiration is technically a combustion reaction, it clearly does not resemble one when it occurs in a cell because of the slow, controlled release of energy from the series of reactions.



Respiration in a eukaryotic cell

Sugar in the form of glucose is the main nutrient used by animal and plant cells in respiration. Cellular respiration involving oxygen is called aerobic respiration, which has four stages: glycolysis, citric acid cycle (or Krebs cycle), electron transport chain, and oxidative phosphorylation. Glycolysis is a metabolic process that occurs in the cytoplasm whereby glucose is converted into two pyruvates, with two net molecules of ATP being produced at the same time. Each pyruvate is then oxidized into acetyl-CoA by the pyruvate dehydrogenase complex, which also generates NADH and carbon dioxide. Acetyl-CoA enters the citric acid cycle, which takes places inside the mitochondrial matrix. At the end of the cycle, the total yield from 1 glucose (or 2 pyruvates) is 6 NADH, 2 FADH₂, and 2 ATP molecules. Finally, the next stage is oxidative phosphorylation, which in eukaryotes, occurs in the mitochondrial cristae. Oxidative phosphorylation comprises the electron transport chain, which is a series of four protein complexes that transfer electrons from one complex to another, thereby releasing energy from NADH and FADH₂ that is coupled to the pumping of protons (hydrogen ions) across the inner mitochondrial

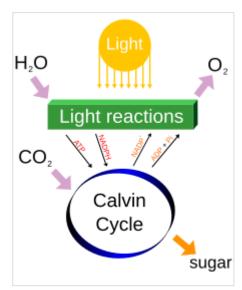
membrane (<u>chemiosmosis</u>), which generates a <u>proton motive force</u>. [49] Energy from the proton motive force drives the enzyme <u>ATP synthase</u> to synthesize more ATPs by <u>phosphorylating ADPs</u>. The transfer of electrons terminates with molecular oxygen being the final electron acceptor.

If oxygen were not present, pyruvate would not be metabolized by cellular respiration but undergoes a process of fermentation. The pyruvate is not transported into the mitochondrion but remains in the cytoplasm, where it is converted to waste products that may be removed from the cell. This serves the purpose of oxidizing the electron carriers so that they can perform glycolysis again and removing the excess pyruvate. Fermentation oxidizes NADH to NAD⁺ so it can be re-used in glycolysis. In the absence of oxygen, fermentation prevents the buildup of NADH in the cytoplasm and provides NAD⁺ for glycolysis. This waste product varies depending on the organism. In skeletal muscles, the waste product is lactic acid. This type of fermentation is called lactic acid fermentation. In strenuous exercise, when energy demands exceed energy supply, the respiratory chain cannot process all of the hydrogen atoms joined by NADH. During anaerobic glycolysis, NAD⁺ regenerates when pairs of hydrogen combine with pyruvate to form lactate. Lactate formation is catalyzed by lactate dehydrogenase in a reversible reaction. Lactate can also be used as an indirect precursor for liver glycogen. During recovery, when oxygen becomes available, NAD⁺ attaches to hydrogen from lactate to form ATP. In yeast, the waste products are ethanol and carbon dioxide. This type of fermentation is known as alcoholic or ethanol fermentation. The ATP generated in this process is made by substrate-level phosphorylation, which does not require oxygen.

Photosynthesis

Photosynthesis is a process used by plants and other organisms to convert light energy into chemical energy that can later be released to fuel the organism's metabolic activities via cellular respiration. This chemical energy is stored in carbohydrate molecules, such as sugars, which are synthesized from carbon dioxide and water. [50][51][52] In most cases, oxygen is released as a waste product. Most plants, algae, and cyanobacteria perform photosynthesis, which is largely responsible for producing and maintaining the oxygen content of the Earth's atmosphere, and supplies most of the energy necessary for life on Earth. [53]

Photosynthesis has four stages: <u>Light absorption</u>, electron transport, ATP synthesis, and <u>carbon fixation</u>. Light absorption is the initial step of photosynthesis whereby light energy is absorbed by <u>chlorophyll</u> pigments attached to proteins in the <u>thylakoid membranes</u>. The absorbed light energy is used to remove electrons from a donor (water) to a primary electron acceptor, a <u>quinone</u> designated as Q. In the second stage, electrons move from the quinone primary electron acceptor through a series



Photosynthesis changes sunlight into chemical energy, splits water to liberate O_2 , and fixes CO_2 into sugar.

of electron carriers until they reach a final electron acceptor, which is usually the oxidized form of $NADP^+$, which is reduced to NADPH, a process that takes place in a protein complex called <u>photosystem I</u> (PSI). The transport of electrons is coupled to the movement of protons (or hydrogen) from the stroma to the thylakoid membrane, which forms a pH gradient across the membrane as hydrogen becomes more concentrated in the lumen than in the stroma. This is analogous to the proton-motive force generated across the inner mitochondrial membrane in aerobic respiration. [49]

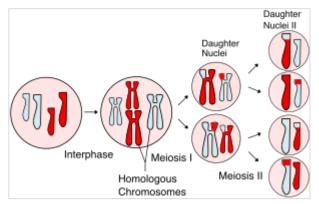
During the third stage of photosynthesis, the movement of protons down their <u>concentration gradients</u> from the thylakoid lumen to the stroma through the ATP synthase is coupled to the synthesis of ATP by that same ATP synthase. The NADPH and ATPs generated by the <u>light-dependent reactions</u> in the second and third stages, respectively, provide the energy and electrons to drive the synthesis of glucose by fixing atmospheric carbon dioxide into existing organic carbon compounds, such as <u>ribulose</u> <u>bisphosphate</u> (RuBP) in a sequence of light-independent (or dark) reactions called the <u>Calvin cycle</u>.

Cell signaling

Cell signaling (or communication) is the ability of cells to receive, process, and transmit signals with its environment and with itself. [55][56] Signals can be non-chemical such as light, electrical impulses, and heat, or chemical signals (or ligands) that interact with receptors, which can be found embedded in the cell membrane of another cell or located deep inside a cell. [57][56] There are generally four types of chemical signals: autocrine, paracrine, juxtacrine, and hormones. [57] In autocrine signaling, the ligand affects the same cell that releases it. Tumor cells, for example, can reproduce uncontrollably because they release signals that initiate their own self-division. In paracrine signaling, the ligand diffuses to nearby cells and affects them. For example, brain cells called neurons release ligands called neurotransmitters that diffuse across a synaptic cleft to bind with a receptor on an adjacent cell such as another neuron or muscle cell. In juxtacrine signaling, there is direct contact between the signaling and responding cells. Finally, hormones are ligands that travel through the circulatory systems of animals or vascular systems of plants to reach their target cells. Once a ligand binds with a receptor, it can influence the behavior of another cell, depending on the type of receptor. For instance, neurotransmitters that bind with an inotropic receptor can alter the excitability of a target cell. Other types of receptors include protein kinase receptors (e.g., receptor for the hormone insulin) and G protein-coupled receptors. Activation of G protein-coupled receptors can initiate second messenger cascades. The process by which a chemical or physical signal is transmitted through a cell as a series of molecular events is called signal transduction.

Cell cycle

The cell cycle is a series of events that take place in a <u>cell</u> that cause it to divide into two daughter cells. These events include the <u>duplication of its DNA</u> and some of its <u>organelles</u>, and the subsequent partitioning of its cytoplasm into two daughter cells in a process called <u>cell division</u>. In <u>eukaryotes</u> (i.e., animal, plant, <u>fungal</u>, and <u>protist</u> cells), there are two distinct types of cell division: <u>mitosis</u> and <u>meiosis</u>. Mitosis is part of the cell cycle, in which replicated <u>chromosomes</u> are separated into two new nuclei. Cell division gives rise to genetically identical cells in which the total number of chromosomes is maintained. In general, mitosis (division of the nucleus) is preceded by the S stage of <u>interphase</u> (during which the DNA is replicated) and is often



In meiosis, the chromosomes duplicate and the homologous.chromosomes exchange genetic information during meiosis I. The daughter cells divide again in meiosis II to form haploid gametes.

followed by <u>telophase</u> and <u>cytokinesis</u>; which divides the <u>cytoplasm</u>, <u>organelles</u> and <u>cell membrane</u> of one cell into two new <u>cells</u> containing roughly equal shares of these cellular components. The different stages of mitosis all together define the mitotic phase of an animal cell cycle—the division of the mother

cell into two genetically identical daughter cells. [60] The cell cycle is a vital process by which a single-celled fertilized egg develops into a mature organism, as well as the process by which hair, skin, blood cells, and some internal organs are renewed. After cell division, each of the daughter cells begin the interphase of a new cycle. In contrast to mitosis, meiosis results in four haploid daughter cells by undergoing one round of DNA replication followed by two divisions. [61] Homologous chromosomes are separated in the first division (meiosis I), and sister chromatids are separated in the second division (meiosis II). Both of these cell division cycles are used in the process of sexual reproduction at some point in their life cycle. Both are believed to be present in the last eukaryotic common ancestor.

<u>Prokaryotes</u> (i.e., <u>archaea</u> and bacteria) can also undergo cell division (or <u>binary fission</u>). Unlike the processes of <u>mitosis</u> and <u>meiosis</u> in eukaryotes, binary fission in prokaryotes takes place without the formation of a <u>spindle apparatus</u> on the cell. Before binary fission, DNA in the bacterium is tightly coiled. After it has uncoiled and duplicated, it is pulled to the separate poles of the bacterium as it increases the size to prepare for splitting. Growth of a new cell wall begins to separate the bacterium (triggered by <u>FtsZ</u> polymerization and "Z-ring" formation). The new cell wall (<u>septum</u>) fully develops, resulting in the complete split of the bacterium. The new daughter cells have tightly coiled DNA rods, <u>ribosomes</u>, and plasmids.

Sexual reproduction and meiosis

Meiosis is a central feature of sexual reproduction in eukaryotes, and the most fundamental function of meiosis appears to be conservation of the integrity of the genome that is passed on to progeny by parents. [63][64] Two aspects of sexual reproduction, meiotic recombination and outcrossing, are likely maintained respectively by the adaptive advantages of recombinational repair of genomic DNA damage and genetic complementation which masks the expression of deleterious recessive mutations. [65]

The beneficial effect of genetic complementation, derived from outcrossing (cross-fertilization) is also referred to as hybrid vigor or heterosis. Charles Darwin in his 1878 book *The Effects of Cross and Self-Fertilization in the Vegetable Kingdom* at the start of chapter XII noted "The first and most important of the conclusions which may be drawn from the observations given in this volume, is that generally cross-fertilisation is beneficial and self-fertilisation often injurious, at least with the plants on which I experimented." Genetic variation, often produced as a byproduct of sexual reproduction, may provide long-term advantages to those sexual lineages that engage in outcrossing. [65]

Genetics

Inheritance

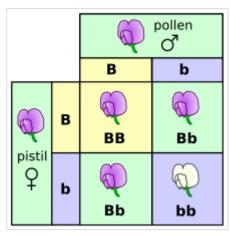
Genetics is the scientific study of inheritance. [67][68][69] Mendelian inheritance, specifically, is the process by which genes and traits are passed on from parents to offspring. [29] It has several principles. The first is that genetic characteristics, alleles, are discrete and have alternate forms (e.g., purple vs. white or tall vs. dwarf), each inherited from one of two parents. Based on the law of dominance and uniformity, which states that some alleles are dominant while others are recessive; an organism with at least one dominant allele will display the phenotype of that dominant allele. During gamete formation, the alleles for each gene segregate, so that each gamete carries only one allele for each gene. Heterozygotic individuals produce gametes with an equal frequency of two alleles. Finally, the law of independent assortment,

states that genes of different traits can segregate independently during the formation of gametes, i.e., genes are unlinked. An exception to this rule would include traits that are sex-linked. Test crosses can be performed to experimentally determine the underlying genotype of an organism with a dominant phenotype. [70] A Punnett square can be used to predict the results of a test cross. The chromosome theory of inheritance, which states that genes are found on chromosomes, was supported by Thomas Morgans's experiments with fruit flies, which established the sex linkage between eye color and sex in these insects. [71]

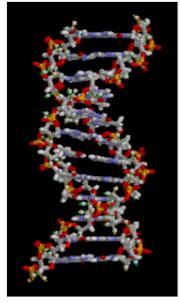
Genes and DNA

A gene is a unit of $\underline{\text{heredity}}$ that corresponds to a region of deoxyribonucleic acid (DNA) that carries genetic information that controls form or function of an organism. DNA is composed of

two polynucleotide chains that coil around each other to form a double helix. [72] It is found as linear chromosomes in eukaryotes, and circular chromosomes in prokaryotes. The set of chromosomes in a cell is collectively known as its genome. In eukaryotes, DNA is mainly in the cell nucleus. [73] In prokaryotes, the DNA is held within the nucleoid. [74] The genetic information is held within genes, and the complete assemblage in an organism is called its genotype. [75] DNA replication is a semiconservative process whereby each strand serves as a template for a new strand of DNA. $^{[72]}$ Mutations are heritable changes in DNA. $^{[72]}$ They can arise spontaneously as a result of replication errors that were not corrected by proofreading or can be induced by an environmental mutagen such as a chemical (e.g., nitrous acid, benzopyrene) or radiation (e.g., xray, gamma ray, ultraviolet radiation, particles emitted by unstable isotopes). [72] Mutations can lead to phenotypic effects such as loss-offunction, gain-of-function, and conditional mutations. [72] Some mutations are beneficial, as they are a source of genetic variation for evolution. [72] Others are harmful if they were to result in a loss of function of genes needed for survival.[72]



Punnett square depicting a cross between two pea plants heterozygous for purple (B) and white (b) blossoms



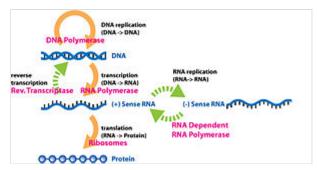
Bases lie between two spiraling DNA strands.

Gene expression

Gene expression is the molecular process by which a <u>genotype</u> encoded in DNA gives rise to an observable <u>phenotype</u> in the proteins of an organism's body. This process is summarized by the <u>central dogma of molecular biology</u>, which was formulated by <u>Francis Crick</u> in 1958. [76][77][78] According to the Central Dogma, genetic information flows from DNA to RNA to protein. There are two gene expression processes: <u>transcription</u> (DNA to RNA) and <u>translation</u> (RNA to protein). [79]

Gene regulation

The regulation of gene expression by environmental factors and during different stages of <u>development</u> can occur at each step of the process such as <u>transcription</u>, <u>RNA splicing</u>, <u>translation</u>, and <u>post-translational modification</u> of a protein. [80] Gene expression can be influenced by positive or negative regulation, depending on which of the two types of regulatory proteins called <u>transcription factors</u> bind to the DNA sequence close to or at a promoter. [80] A cluster of genes that share the same promoter is called an <u>operon</u>, found mainly in prokaryotes and some lower



The extended <u>central dogma of molecular biology</u> includes all the processes involved in the flow of genetic information.

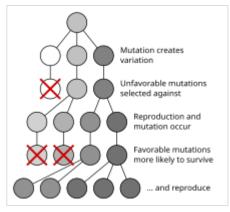
eukaryotes (e.g., *Caenorhabditis elegans*). [80][81] In positive regulation of gene expression, the activator is the transcription factor that stimulates transcription when it binds to the sequence near or at the promoter. Negative regulation occurs when another transcription factor called a repressor binds to a DNA sequence called an operator, which is part of an operon, to prevent transcription. Repressors can be inhibited by compounds called inducers (e.g., allolactose), thereby allowing transcription to occur. [80] Specific genes that can be activated by inducers are called inducible genes, in contrast to constitutive genes that are almost constantly active. [80] In contrast to both, structural genes encode proteins that are not involved in gene regulation. [80] In addition to regulatory events involving the promoter, gene expression can also be regulated by epigenetic changes to chromatin, which is a complex of DNA and protein found in eukaryotic cells. [80]

Genes, development, and evolution

Development is the process by which a multicellular organism (plant or animal) goes through a series of changes, starting from a single cell, and taking on various forms that are characteristic of its life cycle. [82] There are four key processes that underlie development: Determination, differentiation, morphogenesis, and growth. Determination sets the developmental fate of a cell, which becomes more restrictive during development. Differentiation is the process by which specialized cells arise from less specialized cells such as stem cells. $\frac{[83][84]}{}$ Stem cells are undifferentiated or partially differentiated cells that can differentiate into various types of cells and proliferate indefinitely to produce more of the same stem cell. [85] Cellular differentiation dramatically changes a cell's size, shape, membrane potential, metabolic activity, and responsiveness to signals, which are largely due to highly controlled modifications in gene expression and epigenetics. With a few exceptions, cellular differentiation almost never involves a change in the DNA sequence itself. [86] Thus, different cells can have very different physical characteristics despite having the same genome. Morphogenesis, or the development of body form, is the result of spatial differences in gene expression. [82] A small fraction of the genes in an organism's genome called the developmental-genetic toolkit control the development of that organism. These toolkit genes are highly conserved among phyla, meaning that they are ancient and very similar in widely separated groups of animals. Differences in deployment of toolkit genes affect the body plan and the number, identity, and pattern of body parts. Among the most important toolkit genes are the *Hox* genes. Hox genes determine where repeating parts, such as the many vertebrae of snakes, will grow in a developing embryo or larva.[87]

Evolutionary processes

Evolution is a central organizing concept in biology. It is the change in heritable characteristics of populations over successive generations. [88][89] In artificial selection, animals were selectively bred for specific traits. [90] Given that traits are inherited, populations contain a varied mix of traits, and reproduction is able to increase any population, Darwin argued that in the natural world, it was nature that played the role of humans in selecting for specific traits. [90] Darwin inferred that individuals who possessed heritable traits better adapted to their environments are more likely to survive and produce more offspring than other individuals. [90] He further inferred that this would lead to the accumulation of favorable traits over successive generations, thereby increasing the match between the organisms and their environment. [91][92][93][90][94]



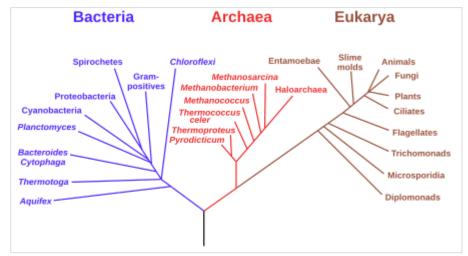
Natural selection for darker traits

Speciation

A species is a group of organisms that mate with one another and speciation is the process by which one lineage splits into two lineages as a result of having evolved independently from each other. For speciation to occur, there has to be reproductive isolation. Reproductive isolation can result from incompatibilities between genes as described by Bateson–Dobzhansky–Muller model. Reproductive isolation also tends to increase with genetic divergence. Speciation can occur when there are physical barriers that divide an ancestral species, a process known as allopatric speciation.

Phylogeny

phylogeny is an evolutionary history specific group of organisms or their genes. [96] It can be represented using phylogenetic tree, a diagram showing lines of descent among organisms or their genes. Each line drawn on the time axis of a tree represents a lineage of descendants of a particular species population. When a lineage divides into two, it represented as a fork or split



Phylogenetic tree showing the domains of $\underline{\text{bacteria}}$, $\underline{\text{archaea}}$, and $\underline{\text{eukaryotes}}$

on the phylogenetic tree. [96] Phylogenetic trees are the basis for comparing and grouping different

species. [96] Different species that share a feature inherited from a common ancestor are described as having homologous features (or synapomorphy). [97][98][96] Phylogeny provides the basis of biological classification. [96] This classification system is rank-based, with the highest rank being the domain followed by kingdom, phylum, class, order, family, genus, and species. [96] All organisms can be classified as belonging to one of three domains: Archaea (originally Archaebacteria), Bacteria (originally eubacteria), or Eukarya (includes the fungi, plant, and animal kingdoms). [99]

History of life

The history of life on Earth traces how organisms have evolved from the earliest emergence of life to present day. Earth formed about 4.5 billion years ago and all life on Earth, both living and extinct, descended from a <u>last universal common ancestor</u> that lived about 3.5 billion years ago. [100][101] Geologists have developed a <u>geologic time scale</u> that divides the history of the Earth into major divisions, starting with four eons (<u>Hadean, Archean, Proterozoic</u>, and <u>Phanerozoic</u>), the first three of which are collectively known as the <u>Precambrian</u>, which lasted approximately 4 billion years. [102] Each eon can be divided into eras, with the Phanerozoic eon that began 539 million years ago [103] being subdivided into <u>Paleozoic</u>, <u>Mesozoic</u>, and <u>Cenozoic</u> eras. [102] These three eras together comprise eleven <u>periods</u> (Cambrian, <u>Ordovician</u>, <u>Silurian</u>, <u>Devonian</u>, <u>Carboniferous</u>, <u>Permian</u>, <u>Triassic</u>, <u>Jurassic</u>, <u>Cretaceous</u>, Tertiary, and Quaternary).

The similarities among all known present-day <u>species</u> indicate that they have diverged through the process of <u>evolution</u> from their common ancestor. Biologists regard the ubiquity of the <u>genetic code</u> as evidence of universal <u>common descent</u> for all <u>bacteria</u>, <u>archaea</u>, and <u>eukaryotes</u>. Microbial mats of coexisting bacteria and archaea were the dominant form of life in the early <u>Archean</u> eon and many of the major steps in early evolution are thought to have taken place in this environment. The earliest evidence of <u>eukaryotes</u> dates from 1.85 billion years ago, and while they may have been present earlier, their diversification accelerated when they started using oxygen in their <u>metabolism</u>. Later, around 1.7 billion years ago, <u>multicellular organisms</u> began to appear, with differentiated cells performing specialised functions.

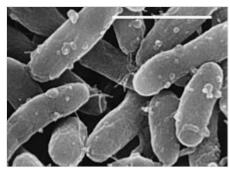
Algae-like multicellular land plants are dated back to about 1 billion years ago, [112] although evidence suggests that $\underline{\text{microorganisms}}$ formed the earliest $\underline{\text{terrestrial ecosystems}}$, at least 2.7 billion years ago. [113] Microorganisms are thought to have paved the way for the inception of land plants in the $\underline{\text{Ordovician}}$ period. Land plants were so successful that they are thought to have contributed to the $\underline{\text{Late Devonian}}$ extinction event. [114]

Ediacara biota appear during the Ediacaran period, [115] while vertebrates, along with most other modern phyla originated about 525 million years ago during the Cambrian explosion. [116] During the Permian period, synapsids, including the ancestors of mammals, dominated the land, [117] but most of this group became extinct in the Permian—Triassic extinction event 252 million years ago. [118] During the recovery from this catastrophe, archosaurs became the most abundant land vertebrates; [119] one archosaur group, the dinosaurs, dominated the Jurassic and Cretaceous periods. [120] After the Cretaceous—Paleogene extinction event 66 million years ago killed off the non-avian dinosaurs, [121] mammals increased rapidly in size and diversity. [122] Such mass extinctions may have accelerated evolution by providing opportunities for new groups of organisms to diversity. [123]

Bacteria and Archaea

Bacteria are a type of <u>cell</u> that constitute a large <u>domain</u> of <u>prokaryotic microorganisms</u>. Typically a few <u>micrometers</u> in length, bacteria have a <u>number of shapes</u>, ranging from <u>spheres</u> to <u>rods</u> and <u>spirals</u>. Bacteria were among the first life forms to appear on Earth, and are present in most of its <u>habitats</u>. Bacteria inhabit soil, water, <u>acidic hot springs</u>, <u>radioactive waste</u>, [124] and the deep <u>biosphere</u> of the <u>Earth's crust</u>. Bacteria also live in <u>symbiotic</u> and <u>parasitic</u> relationships with plants and animals. Most bacteria have not been characterised, and only about 27 percent of the <u>bacterial phyla</u> have species that can be grown in the laboratory. [125]

Archaea constitute the other domain of prokaryotic cells and were initially <u>classified</u> as bacteria, receiving the name archaebacteria (in the Archaebacteria <u>kingdom</u>), a term that has fallen out of use. [126] Archaeal cells have unique properties separating them from the other <u>two domains</u>, Bacteria and <u>Eukaryota</u>. Archaea are further divided into multiple recognized <u>phyla</u>. Archaea and bacteria are generally similar in size and shape, although a few archaea have very different shapes, such as the flat and square cells of <u>Haloquadratum walsbyi</u>. [127] Despite this morphological similarity to bacteria, archaea possess <u>genes</u> and several <u>metabolic pathways</u> that are more closely related to those of eukaryotes, notably for the <u>enzymes</u> involved in <u>transcription</u> and <u>translation</u>. Other aspects of archaeal biochemistry are unique, such as their



Bacteria – <u>Gemmatimonas</u> aurantiaca (-=1 Micrometer)



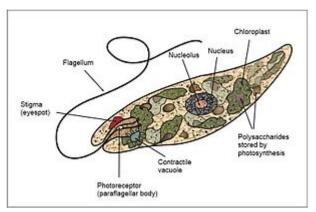
Archaea - Haloarchaea

reliance on ether lipids in their cell membranes, [128] including archaeols. Archaea use more energy sources than eukaryotes: these range from organic compounds, such as sugars, to ammonia, metal ions or even hydrogen gas. Salt-tolerant archaea (the Haloarchaea) use sunlight as an energy source, and other species of archaea fix carbon, but unlike plants and cyanobacteria, no known species of archaea does both. Archaea reproduce asexually by binary fission, fragmentation, or budding; unlike bacteria, no known species of Archaea form endospores.

The first observed archaea were <u>extremophiles</u>, living in extreme environments, such as <u>hot springs</u> and <u>salt lakes</u> with no other organisms. Improved molecular detection tools led to the discovery of archaea in almost every <u>habitat</u>, including soil, oceans, and <u>marshlands</u>. Archaea are particularly numerous in the oceans, and the archaea in <u>plankton</u> may be one of the most abundant groups of organisms on the planet.

Archaea are a major part of <u>Earth's life</u>. They are part of the <u>microbiota</u> of all organisms. In the <u>human microbiome</u>, they are important in the <u>gut</u>, mouth, and on the skin. Their morphological, metabolic, and geographical diversity permits them to play multiple ecological roles: carbon fixation; nitrogen cycling; organic compound turnover; and maintaining microbial symbiotic and <u>syntrophic</u> communities, for example. [130]

Eukaryotes



<u>Euglena</u>, a single-celled eukaryote that can both move and photosynthesize

Eukaryotes are hypothesized to have split from archaea, which was followed by their endosymbioses with bacteria (or symbiogenesis) that gave rise to mitochondria and chloroplasts, both of which are now part of modern-day eukaryotic cells.[131] The major lineages of eukaryotes diversified in the Precambrian about 1.5 billion years ago and can be classified into eight major clades: alveolates, excavates, stramenopiles, plants, rhizarians, amoebozoans, fungi. and animals.[131] Five of these clades are collectively known as protists, which are mostly microscopic eukaryotic organisms that are not plants, fungi, or animals. [131] While it is likely that protists share a

<u>common ancestor</u> (the <u>last eukaryotic common ancestor</u>), protists by themselves do not constitute a separate clade as some protists may be more closely related to plants, fungi, or animals than they are to other protists. Like groupings such as <u>algae</u>, <u>invertebrates</u>, or <u>protozoans</u>, the protist grouping is not a formal taxonomic group but is used for convenience. Most protists are unicellular; these are called microbial eukaryotes. [131]

Plantae, which would exclude fungi and some <u>algae</u>. Plant cells were derived by endosymbiosis of a <u>cyanobacterium</u> into an early eukaryote about one billion years ago, which gave rise to chloroplasts. [134] The first several clades that emerged following primary endosymbiosis were aquatic and most of the aquatic photosynthetic eukaryotic organisms are collectively described as algae, which is a term of convenience as not all algae are closely related. [134] Algae comprise several distinct clades such as <u>glaucophytes</u>, which are microscopic freshwater algae that may have resembled in form to the early unicellular ancestor of Plantae. [134] Unlike glaucophytes, the other algal clades such as <u>red</u> and <u>green</u> <u>algae</u> are multicellular. Green algae comprise three major clades: <u>chlorophytes</u>, <u>coleochaetophytes</u>, and stoneworts. [134]

<u>Fungi</u> are eukaryotes that digest foods outside their bodies, secreting digestive enzymes that break down large food molecules before absorbing them through their cell membranes. Many fungi are also <u>saprobes</u>, feeding on dead organic matter, making them important <u>decomposers</u> in ecological systems. [135]

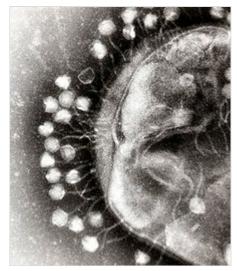
Animals are multicellular eukaryotes. With few exceptions, animals <u>consume organic material</u>, <u>breathe oxygen</u>, are <u>able to move</u>, can <u>reproduce sexually</u>, and grow from a hollow sphere of <u>cells</u>, the <u>blastula</u>, during <u>embryonic development</u>. Over 1.5 million <u>living</u> animal <u>species</u> have been <u>described</u>—of which around 1 million are <u>insects</u>—but it has been estimated there are over 7 million animal species in total. They have <u>complex interactions</u> with each other and their environments, forming intricate <u>food webs</u>. [136]

Viruses

Viruses are <u>submicroscopic</u> <u>infectious agents</u> that <u>replicate</u> inside the <u>cells</u> of <u>organisms</u>. [137] Viruses infect all types of <u>life forms</u>, from animals and plants to <u>microorganisms</u>, including bacteria and archaea. [138][139] More than 6,000 virus species have been described in detail. [140] Viruses are found in

almost every <u>ecosystem</u> on Earth and are the most numerous type of biological entity. [141][142]

The origins of viruses in the <u>evolutionary history of life</u> are unclear: some may have <u>evolved</u> from <u>plasmids</u>—pieces of DNA that can move between cells—while others may have evolved from bacteria. In evolution, viruses are an important means of <u>horizontal gene transfer</u>, which increases genetic diversity in a way analogous to <u>sexual reproduction</u>. [143] Because viruses possess some but not all characteristics of life, they have been described as "organisms at the edge of life", [144] and as <u>self-replicators</u>. [145]



Bacteriophages attached to a bacterial cell wall

Ecology

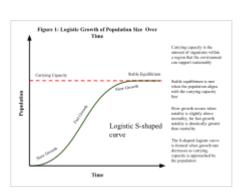
Ecology is the study of the distribution and abundance of life, the interaction between organisms and their environment. [146]

Ecosystems

The <u>community</u> of living (biotic) organisms in conjunction with the nonliving (abiotic) components (e.g., water, light, radiation, temperature, <u>humidity</u>, <u>atmosphere</u>, <u>acidity</u>, and soil) of their environment is called an <u>ecosystem</u>. [147][148][149] These biotic and abiotic components are linked together through <u>nutrient cycles</u> and energy flows. [150] Energy from the sun enters the system through <u>photosynthesis</u> and is incorporated into plant tissue. By feeding on plants and on one another, animals move <u>matter</u> and energy through the system. They also influence the quantity of plant and <u>microbial biomass</u> present. By breaking down dead <u>organic matter</u>, <u>decomposers</u> release <u>carbon</u> back to the atmosphere and facilitate <u>nutrient cycling</u> by converting nutrients stored in dead biomass back to a form that can be readily used by plants and other microbes. [151]

Populations

A population is the group of <u>organisms</u> of the same <u>species</u> that occupies an <u>area</u> and <u>reproduce</u> from generation to generation. [152][153][154][155][156] Population size can be estimated by multiplying population density by the area or volume. The <u>carrying capacity</u> of an <u>environment</u> is the maximum population size of a <u>species</u> that can be sustained by that specific environment, given the food, <u>habitat</u>, <u>water</u>, and other <u>resources</u> that are available. The carrying capacity of a population can be affected by changing environmental conditions such as changes in the availability of resources and the cost of maintaining them. In <u>human populations</u>, new <u>technologies</u> such as the <u>Green revolution</u> have helped increase the Earth's carrying capacity for

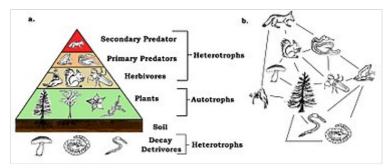


Reaching carrying capacity through a logistic growth curve

humans over time, which has stymied the attempted predictions of impending population decline, the most famous of which was by Thomas Malthus in the 18th century. [152]

Communities

A community is a group of populations of species occupying the same geographical area at the same time. [159] A biological interaction is the effect that a pair of organisms living together in a community have on each other. They can be either of the same species (intraspecific interactions), or of different species (interspecific interactions). These effects may be short-term, like pollination and or long-term; predation, both often



A (a) trophic pyramid and a (b) simplified food web. The trophic pyramid represents the biomass at each level. [158]

strongly influence the evolution of the species involved. A long-term interaction is called a symbiosis. Symbioses range from mutualism, beneficial to both partners, to competition, harmful to both partners. [160] Every species participates as a consumer, resource, or both in consumer-resource interactions, which form the core of food chains or food webs. [161] There are different trophic levels within any food web, with the lowest level being the primary producers (or autotrophs) such as plants and algae that convert energy and inorganic material into organic compounds, which can then be used by the rest of the community. [53][162][163] At the next level are the heterotrophs, which are the species that obtain energy by breaking apart organic compounds from other organisms. [161] Heterotrophs that consume plants are primary consumers (or herbivores) whereas heterotrophs that consume herbivores are secondary consumers (or carnivores). And those that eat secondary consumers are tertiary consumers and so on. Omnivorous heterotrophs are able to consume at multiple levels. Finally, there are decomposers that feed on the waste products or dead bodies of organisms. [161] On average, the total amount of energy incorporated into the biomass of a trophic level per unit of time is about one-tenth of the energy of the trophic level that it consumes. Waste and dead material used by decomposers as well as heat lost from metabolism make up the other ninety percent of energy that is not consumed by the next trophic level.[164]

Biosphere

In the global ecosystem or biosphere, matter exists as different interacting compartments, which can be biotic or abiotic as well as accessible or inaccessible, depending on their forms and locations. For example, matter from terrestrial autotrophs are both biotic and accessible to other organisms whereas the matter in rocks and minerals are abiotic and inaccessible. A <u>biogeochemical cycle</u> is a pathway by which specific <u>elements</u> of matter are turned over or moved through the biotic (<u>biosphere</u>) and the abiotic (<u>lithosphere</u>, <u>atmosphere</u>, and <u>hydrosphere</u>) compartments of Earth. There are biogeochemical cycles for nitrogen, carbon, and water.

Conservation

Conservation biology is the study of the conservation of Earth's <u>biodiversity</u> with the aim of protecting <u>species</u>, their <u>habitats</u>, and <u>ecosystems</u> from excessive rates of <u>extinction</u> and the erosion of biotic interactions. [167][168][169] It is concerned with factors that influence the maintenance, loss, and restoration of biodiversity and the science of sustaining evolutionary processes that engender <u>genetic</u>, population, species, and ecosystem diversity. [170][171][172][173] The concern stems from estimates suggesting that up

to 50% of all species on the planet will disappear within the next 50 years, [174] which has contributed to poverty, starvation, and will reset the course of evolution on this planet. [175][176] Biodiversity affects the functioning of ecosystems, which provide a variety of services upon which people depend. Conservation biologists research and educate on the trends of biodiversity loss, species extinctions, and the negative effect these are having on our capabilities to sustain the well-being of human society. Organizations and citizens are responding to the current biodiversity crisis through conservation action plans that direct research, monitoring, and education programs that engage concerns at local through global scales. [177][170][171][172]

See also

- Biology in fiction
- Glossary of biology
- Idiobiology
- List of biological websites
- List of biologists
- List of biology journals
- List of biology topics
- List of life sciences
- List of omics topics in biology
- National Association of Biology Teachers
- Outline of biology
- Periodic table of life sciences in <u>Tinbergen's four</u> questions
- Science tourism
- Terminology of biology

Atmospheric Carbon Net Land Increase Annual In

Fast carbon cycle showing the movement of carbon between land, atmosphere, and oceans in billions of tons per year. Yellow numbers are natural fluxes, red are human contributions, white are stored carbon. Effects of the slow carbon cycle, such as volcanic and tectonic activity, are not included. [165]

References

- 1. Urry, Lisa; Cain, Michael; Wasserman, Steven; Minorsky, Peter; Reece, Jane (2017). "Evolution, the themes of biology, and scientific inquiry". *Campbell Biology* (11th ed.). New York: Pearson. pp. 2–26. ISBN 978-0134093413.
- 2. Hillis, David M.; Heller, H. Craig; Hacker, Sally D.; Laskowski, Marta J.; Sadava, David E. (2020). "Studying life". *Life: The Science of Biology* (12th ed.). W. H. Freeman. ISBN 978-1319017644.
- 3. Freeman, Scott; Quillin, Kim; Allison, Lizabeth; Black, Michael; Podgorski, Greg; Taylor, Emily; Carmichael, Jeff (2017). "Biology and the three of life". *Biological Science* (6th ed.). Hoboken, N.J.: Pearson. pp. 1–18. ISBN 978-0321976499.

- Modell, Harold; Cliff, William; Michael, Joel; McFarland, Jenny; Wenderoth, Mary Pat; Wright, Ann (December 2015). "A physiologist's view of homeostasis" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC4669363). Advances in Physiology Education. 39 (4): 259–266. doi:10.1152/advan.00107.2015 (https://doi.org/10.1152%2Fadvan.00107.2015). ISSN 1043-4046 (https://search.worldcat.org/issn/1043-4046). PMC 4669363 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC4669363). PMID 26628646 (https://pubmed.ncbi.nlm.nih.gov/26628646).
- 5. Davies, PC; Rieper, E; Tuszynski, JA (January 2013). "Self-organization and entropy reduction in a living cell" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC3712629). Bio Systems. 111 (1): 1–10. Bibcode:2013BiSys.111....1D (https://ui.adsabs.harvard.edu/abs/20 13BiSys.111....1D). doi:10.1016/j.biosystems.2012.10.005 (https://doi.org/10.1016%2Fj.biosystems.2012.10.005). PMC 3712629 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC3712629). PMID 23159919 (https://pubmed.ncbi.nlm.nih.gov/23159919).
- 6. Based on definition from: "Aquarena Wetlands Project glossary of terms" (https://web.archive.org/web/20040608113114/http://www.bio.txstate.edu/~wetlands/Glossary/glossary.html). Texas State University at San Marcos. Archived from the original (http://www.bio.txstate.edu/~wetlands/Glossary/glossary.html) on 2004-06-08.
- 7. Craig, Nancy (2014). *Molecular Biology, Principles of Genome Function*. OUP Oxford. ISBN 978-0-19-965857-2.
- 8. Mosconi, Francesco; Julou, Thomas; Desprat, Nicolas; Sinha, Deepak Kumar; Allemand, Jean-François; Vincent Croquette; Bensimon, David (2008). "Some nonlinear challenges in biology". *Nonlinearity*. **21** (8): T131. Bibcode:2008Nonli..21..131M (https://ui.adsabs.harvar d.edu/abs/2008Nonli..21..131M). doi:10.1088/0951-7715/21/8/T03 (https://doi.org/10.1088/2F0951-7715%2F21%2F8%2FT03). S2CID 119808230 (https://api.semanticscholar.org/CorpusID:119808230).
- 9. Howell, Elizabeth (8 December 2014). "How Did Life Become Complex, And Could It Happen Beyond Earth?" (https://web.archive.org/web/20180817193332/https://www.astrobio.net/origin-and-evolution-of-life/life-become-complex-happen-beyond-earth/). *Astrobiology Magazine*. Archived from the original on 17 August 2018. Retrieved 14 February 2018.
- Pearce, Ben K.D.; Tupper, Andrew S.; Pudritz, Ralph E.; et al. (March 1, 2018).
 "Constraining the Time Interval for the Origin of Life on Earth". Astrobiology. 18 (3): 343–364. arXiv:1808.09460 (https://arxiv.org/abs/1808.09460). Bibcode:2018AsBio..18..343P (https://ui.adsabs.harvard.edu/abs/2018AsBio..18..343P). doi:10.1089/ast.2017.1674 (https://doi.org/10.1089%2Fast.2017.1674). PMID 29570409 (https://pubmed.ncbi.nlm.nih.gov/29570409). S2CID 4419671 (https://api.semanticscholar.org/CorpusID:4419671).
- 11. Lindberg, David C. (2007). "Science before the Greeks". *The beginnings of Western science: the European Scientific tradition in philosophical, religious, and institutional context* (2nd ed.). Chicago, Illinois: University of Chicago Press. pp. 1–20. ISBN 978-0-226-48205-7.
- 12. Grant, Edward (2007). "Ancient Egypt to Plato". *A History of Natural Philosophy: From the Ancient World to the Nineteenth Century* (https://archive.org/details/historynaturalph00gran). New York: Cambridge University Press. pp. 1 (https://archive.org/details/historynaturalph00gran/page/n16)–26. ISBN 978-052-1-68957-1.
- 13. *Handbook of the Historiography of Biology* (https://link.springer.com/book/10.1007/978-3-31 9-90119-0). Historiographies of Science. 2021. doi:10.1007/978-3-319-90119-0 (https://doi.org/10.1007%2F978-3-319-90119-0). ISBN 978-3-319-90118-3.
- 14. Magner, Lois N. (2002). <u>A History of the Life Sciences, Revised and Expanded</u> (https://books.google.com/books?id=YKJ6gVYbrGwC). CRC Press. ISBN 978-0-203-91100-6. Archived (https://web.archive.org/web/20150324125937/http://books.google.com/books?id=YKJ6gVYbrGwC) from the original on 2015-03-24.

- 15. Serafini, Anthony (2013). *The Epic History of Biology* (https://books.google.com/books?id=Z 3oECAAAQBAJ&q=biology%20egypt&pg=PA2). Springer. ISBN 978-1-4899-6327-7. Archived (https://web.archive.org/web/20210415122005/https://books.google.com/books?id=Z3oECAAAQBAJ&q=biology%20egypt&pg=PA2) from the original on 15 April 2021. Retrieved 14 July 2015.
- 16. Morange, Michel. 2021. *A History of Biology*. Princeton, NJ: Princeton University Press. Translated by Teresa Lavender Fagan and Joseph Muise.
- 17. One or more of the preceding sentences incorporates text from a publication now in the public domain: Chisholm, Hugh, ed. (1911). "Theophrastus". Encyclopædia Britannica (11th ed.). Cambridge University Press.
- 18. Fahd, Toufic (1996). "Botany and agriculture". In Morelon, Régis; Rashed, Roshdi (eds.). *Encyclopedia of the History of Arabic Science*. Vol. 3. Routledge. p. 815. ISBN 978-0-415-12410-2.
- 19. Magner, Lois N. (2002). *A History of the Life Sciences, Revised and Expanded* (https://books.google.com/books?id=YKJ6gVYbrGwC). CRC Press. pp. 133–44. ISBN 978-0-203-91100-6. Archived (https://web.archive.org/web/20150324125937/http://books.google.com/books?id=YKJ6gVYbrGwC) from the original on 2015-03-24.
- 20. Sapp, Jan (2003). "7". *Genesis: The Evolution of Biology*. New York: Oxford University Press. ISBN 978-0-19-515618-8.
- 21. Coleman, William (1977). *Biology in the Nineteenth Century: Problems of Form, Function, and Transformation*. New York: Cambridge University Press. ISBN 978-0-521-29293-1.
- 22. Mayr, Ernst. The Growth of Biological Thought, chapter 4
- 23. Mayr, Ernst. The Growth of Biological Thought, chapter 7
- 24. * Darwin, Francis, ed. (1909). *The foundations of The origin of species, a sketch written in 1842* (http://darwin-online.org.uk/converted/pdf/1909_Foundations_F1555.pdf) (PDF). Cambridge: Printed at the University Press. p. 53. LCCN 61057537 (https://lccn.loc.gov/61057537). OCLC 1184581 (https://search.worldcat.org/oclc/1184581). Archived (https://web.archive.org/web/20160304111606/http://darwin-online.org.uk/converted/pdf/1909_Foundations_F1555.pdf) (PDF) from the original on 4 March 2016. Retrieved 27 November 2014.
- 25. <u>Gould, Stephen Jay</u>. *The Structure of Evolutionary Theory*. The Belknap Press of Harvard University Press: Cambridge, 2002. ISBN 0-674-00613-5. p. 187.
- 26. Mayr, Ernst. *The Growth of Biological Thought*, chapter 10: "Darwin's evidence for evolution and common descent"; and chapter 11: "The causation of evolution: natural selection"
- 27. Larson, Edward J. (2006). "Ch. 3" (https://books.google.com/books?id=xzLRvxlJhzkC). Evolution: The Remarkable History of a Scientific Theory. Random House Publishing Group. ISBN 978-1-58836-538-5. Archived (https://web.archive.org/web/20150324124009/http://books.google.com/books?id=xzLRvxlJhzkC) from the original on 2015-03-24.
- 28. Henig (2000). *Op. cit* (http://archive.org/details/monkingardenlost00heni). pp. 134–138.
- 29. Miko, Ilona (2008). "Gregor Mendel's principles of inheritance form the cornerstone of modern genetics. So just what are they?" (https://www.nature.com/scitable/topicpage/gregor-mendel-and-the-principles-of-inheritance-593/). *Nature Education*. **1** (1): 134. Archived (https://web.archive.org/web/20190719224056/http://www.nature.com/scitable/topicpage/gregor-mendel-and-the-principles-of-inheritance-593) from the original on 2019-07-19. Retrieved 2021-05-13.
- 30. Futuyma, Douglas J.; Kirkpatrick, Mark (2017). "Evolutionary Biology". *Evolution* (4th ed.). Sunderland, Mass.: Sinauer Associates. pp. 3–26.
- 31. Noble, Ivan (2003-04-14). "Human genome finally complete" (https://news.bbc.co.uk/2/hi/science/nature/2940601.stm). *BBC News*. Archived (https://web.archive.org/web/20060614141 605/http://news.bbc.co.uk/2/hi/science/nature/2940601.stm) from the original on 2006-06-14. Retrieved 2006-07-22.

- 32. Urry, Lisa; Cain, Michael; Wasserman, Steven; Minorsky, Peter; Reece, Jane (2017). "The chemical context of life". *Campbell Biology* (11th ed.). New York: Pearson. pp. 28–43. ISBN 978-0134093413.
- 33. Freeman, Scott; Quillin, Kim; Allison, Lizabeth; Black, Michael; Podgorski, Greg; Taylor, Emily; Carmichael, Jeff (2017). "Water and carbon: The chemical basis of life". *Biological Science* (6th ed.). Hoboken, N.J.: Pearson. pp. 55–77. ISBN 978-0321976499.
- 34. Urry, Lisa; Cain, Michael; Wasserman, Steven; Minorsky, Peter; Reece, Jane (2017). "Carbon and the molecular diversity of life". *Campbell Biology* (11th ed.). New York: Pearson. pp. 56–65. ISBN 978-0134093413.
- 35. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "Carbon and molecular diversity of life". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 56–65. ISBN 978-1464175121.
- 36. Freeman, Scott; Quillin, Kim; Allison, Lizabeth; Black, Michael; Podgorski, Greg; Taylor, Emily; Carmichael, Jeff (2017). "Protein structure and function". *Biological Science* (6th ed.). Hoboken, N.J.: Pearson. pp. 78–92. ISBN 978-0321976499.
- 37. Urry, Lisa; Cain, Michael; Wasserman, Steven; Minorsky, Peter; Reece, Jane (2017). "The structure and function of large biological molecules". *Campbell Biology* (11th ed.). New York: Pearson. pp. 66–92. ISBN 978-0134093413.
- 38. Freeman, Scott; Quillin, Kim; Allison, Lizabeth; Black, Michael; Podgorski, Greg; Taylor, Emily; Carmichael, Jeff (2017). "An introduction to carbohydrate". *Biological Science* (6th ed.). Hoboken, N.J.: Pearson. pp. 107–118. ISBN 978-0321976499.
- 39. Freeman, Scott; Quillin, Kim; Allison, Lizabeth; Black, Michael; Podgorski, Greg; Taylor, Emily; Carmichael, Jeff (2017). "Lipids, membranes, and the first cells". *Biological Science* (6th ed.). Hoboken, N.J.: Pearson. pp. 119–141. ISBN 978-0321976499.
- 40. Freeman, Scott; Quillin, Kim; Allison, Lizabeth; Black, Michael; Podgorski, Greg; Taylor, Emily; Carmichael, Jeff (2017). "Nucleic acids and the RNA world". *Biological Science* (6th ed.). Hoboken, N.J.: Pearson. pp. 93–106. ISBN 978-0321976499.
- 41. Mazzarello, P. (May 1999). "A unifying concept: the history of cell theory". *Nature Cell Biology*. **1** (1): E13–15. doi:10.1038/8964 (https://doi.org/10.1038%2F8964). PMID 10559875 (https://pubmed.ncbi.nlm.nih.gov/10559875). S2CID 7338204 (https://api.semanticscholar.org/CorpusID:7338204).
- 42. Campbell, Neil A.; Williamson, Brad; Heyden, Robin J. (2006). *Biology: Exploring Life* (http://www.phschool.com/el_marketing.html). Boston: Pearson Prentice Hall. ISBN 978-0132508827. Archived (https://web.archive.org/web/20141102041816/http://www.phschool.com/el_marketing.html) from the original on 2014-11-02. Retrieved 2021-05-13.
- 43. Urry, Lisa; Cain, Michael; Wasserman, Steven; Minorsky, Peter; Reece, Jane (2017). "Membrane structure and function". *Campbell Biology* (11th ed.). New York: Pearson. pp. 126–142. ISBN 978-0134093413.
- 44. Alberts, B.; Johnson, A.; Lewis, J.; et al. (2002). *Molecular Biology of the Cell* (https://www.ncbi.nlm.nih.gov/books/NBK21054/?term=Molecular%20Biology%20of%20the%20Cell) (4th ed.). New York: Garland Science. <u>ISBN</u> 978-0-8153-3218-3. <u>Archived (https://web.archive.org/web/20171220092628/https://www.ncbi.nlm.nih.gov/books/NBK21054/?term=Molecular%20Biology%20of%20the%20Cell) from the original on 2017-12-20.</u>
- 45. Tom Herrmann; Sandeep Sharma (March 2, 2019). "Physiology, Membrane" (https://www.ncbi.nlm.nih.gov/books/NBK538211/). StatPearls. PMID 30855799 (https://pubmed.ncbi.nlm.nih.gov/30855799). Archived (https://web.archive.org/web/20220217034021/https://www.ncbi.nlm.nih.gov/books/NBK538211/) from the original on February 17, 2022. Retrieved May 14, 2021.

- 46. Alberts, Bruce; Johnson, Alexander; Lewis, Julian; Raff, Martin; Roberts, Keith; Walter, Peter (2002). "Cell Movements and the Shaping of the Vertebrate Body" (https://www.ncbi.nlm.nih.gov/books/NBK26863/). Molecular Biology of the Cell (4th ed.). Archived (https://web.archive.org/web/20200122055346/https://www.ncbi.nlm.nih.gov/books/NBK26863/) from the original on 2020-01-22. Retrieved 2021-05-13. The Alberts text discusses how the "cellular building blocks" move to shape developing embryos. It is also common to describe small molecules such as amino acids as "molecular building blocks (https://www.ncbi.nlm.nih.gov/entrez/query.fcgi?cmd=Search&db=books&doptcmdl=GenBookHL&term=%22all+cells%22+AND+mboc4%5Bbook%5D+AND+372023%5Buid%5D&rid=mboc4.section.4#23) Archived (https://web.archive.org/web/20200122055404/https://www.ncbi.nlm.nih.gov/books?cmd=Search&doptcmdl=GenBookHL&term=%22all%2Bcells%22%2BAND%2Bmboc4%5Bbook%5D%2BAND%2B372023%5Buid%5D&rid=mboc4.section.4#23) 2020-01-22 at the Wayback Machine".
- 47. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "Cells: The working units of life". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 60–81. ISBN 978-1464175121.
- 48. Bailey, Regina. "Cellular Respiration" (http://biology.about.com/od/cellularprocesses/a/cellre spiration.htm). Archived (https://web.archive.org/web/20120505043947/http://biology.about.c om/od/cellularprocesses/a/cellrespiration.htm) from the original on 2012-05-05.
- 49. Lodish, Harvey; Berk, Arnold.; Kaiser, Chris A.; Krieger, Monty; Scott, Matthew P.; Bretscher, Anthony; Ploegh, Hidde; Matsudaira, Paul (2008). "Cellular energetics". *Molecular Cell Biology* (6th ed.). New York: W.H. Freeman and Company. pp. 479–532. ISBN 978-0716776017.
- 50. "photosynthesis" (http://www.etymonline.com/index.php?term=photosynthesis&allowed_in_f rame=0). *Online Etymology Dictionary*. Archived (https://web.archive.org/web/20130307020 959/http://www.etymonline.com/index.php?term=photosynthesis&allowed_in_frame=0) from the original on 2013-03-07. Retrieved 2013-05-23.
- 51. φῶς (https://www.perseus.tufts.edu/hopper/text?doc=Perseus:text:1999.04.0057:entry=fw=s 2). Liddell, Henry George; Scott, Robert; *A Greek–English Lexicon* at the Perseus Project
- 52. σύνθεσις (https://www.perseus.tufts.edu/hopper/text?doc=Perseus:text:1999.04.0057:entry=su/nqesis). Liddell, Henry George; Scott, Robert; <u>A Greek–English Lexicon</u> at the <u>Perseus Project</u>
- 53. Bryant, D. A.; Frigaard, N. U. (Nov 2006). "Prokaryotic photosynthesis and phototrophy illuminated". *Trends in Microbiology*. **14** (11): 488–496. doi:10.1016/j.tim.2006.09.001 (http s://doi.org/10.1016%2Fj.tim.2006.09.001). PMID 16997562 (https://pubmed.ncbi.nlm.nih.go v/16997562).
- 54. Reece, J.; Urry, L.; Cain, M. (2011). *Biology* (https://archive.org/details/isbn_978125615876 9/page/235) (International ed.). Upper Saddle River, New Jersey: Pearson Education. pp. 235, 244 (https://archive.org/details/isbn_9781256158769/page/235). ISBN 978-0-321-73975-9. "This initial incorporation of carbon into organic compounds is known as carbon fixation."
- 55. Neitzel, James; Rasband, Matthew. "Cell communication" (https://www.nature.com/scitable/t opic/cell-communication-14122659/). *Nature Education*. Archived (https://web.archive.org/web/20100929110101/https://www.nature.com/scitable/topic/cell-communication-14122659/) from the original on 29 September 2010. Retrieved 29 May 2021.
- 56. "Cell signaling" (https://www.nature.com/scitable/topicpage/cell-signaling-14047077/).

 Nature Education. Archived (https://web.archive.org/web/20101031053612/https://www.nature.com/scitable/topicpage/cell-signaling-14047077/) from the original on 31 October 2010. Retrieved 29 May 2021.
- 57. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "Cell membranes and signaling". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 82–104. ISBN 978-1464175121.

- 58. Martin, E. A.; Hine, R. (2020). *A dictionary of biology* (6th ed.). Oxford: Oxford University Press. ISBN 978-0199204625. OCLC 176818780 (https://search.worldcat.org/oclc/176818780).
- 59. Griffiths, A. J. (2012). *Introduction to genetic analysis* (10th ed.). New York: W.H. Freeman. ISBN 978-1429229432. OCLC 698085201 (https://search.worldcat.org/oclc/698085201).
- 60. "10.2 The Cell Cycle Biology 2e | OpenStax" (https://openstax.org/books/biology-2e/page s/10-2-the-cell-cycle). openstax.org. 28 March 2018. Archived (https://web.archive.org/web/ 20201129223722/https://openstax.org/books/biology-2e/pages/10-2-the-cell-cycle) from the original on 2020-11-29. Retrieved 2020-11-24.
- 61. Freeman, Scott; Quillin, Kim; Allison, Lizabeth; Black, Michael; Podgorski, Greg; Taylor, Emily; Carmichael, Jeff (2017). "Meiosis". *Biological Science* (6th ed.). Hoboken, New Jersey: Pearson. pp. 271–289. ISBN 978-0321976499.
- 62. Casiraghi, A.; Suigo, L.; Valoti, E.; Straniero, V. (February 2020). "Targeting Bacterial Cell Division: A Binding Site-Centered Approach to the Most Promising Inhibitors of the Essential Protein FtsZ" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC7167804). Antibiotics. 9 (2): 69. doi:10.3390/antibiotics9020069 (https://doi.org/10.3390%2Fantibiotics9020069). PMC 7167804 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC7167804). PMID 32046082 (https://pubmed.ncbi.nlm.nih.gov/32046082).
- 63. Brandeis M. New-age ideas about age-old sex: separating meiosis from mating could solve a century-old conundrum. Biol Rev Camb Philos Soc. 2018 May;93(2):801-810. doi: 10.1111/brv.12367. Epub 2017 Sep 14. PMID 28913952
- 64. Hörandl E. Apomixis and the paradox of sex in plants. Ann Bot. 2024 Mar 18:mcae044. doi: 10.1093/aob/mcae044. Epub ahead of print. PMID 38497809
- 65. Bernstein H, Byerly HC, Hopf FA, Michod RE. Genetic damage, mutation, and the evolution of sex. Science. 1985 Sep 20;229(4719):1277-81. doi: 10.1126/science.3898363. PMID 3898363
- 66. Darwin, C. R. 1878. The effects of cross and self fertilisation in the vegetable kingdom. London: John Murray. darwin-online.org.uk
- 67. Griffiths, Anthony J.; Wessler, Susan R.; Carroll, Sean B.; Doebley, John (2015). "The genetics revolution". *An Introduction to Genetic Analysis* (11th ed.). Sunderland, Massachusetts: W.H. Freeman & Company. pp. 1–30. ISBN 978-1464109485.
- 68. Griffiths, Anthony J. F.; Miller, Jeffrey H.; Suzuki, David T.; Lewontin, Richard C.; Gelbart, William M., eds. (2000). "Genetics and the Organism: Introduction" (https://www.ncbi.nlm.nih.gov/books/bv.fcgi?rid=iga.section.60). *An Introduction to Genetic Analysis* (7th ed.). New York: W. H. Freeman. ISBN 978-0-7167-3520-5.
- 69. Hartl, D.; Jones, E (2005). *Genetics: Analysis of Genes and Genomes* (https://archive.org/details/genetics00dani) (6th ed.). Jones & Bartlett. ISBN 978-0-7637-1511-3.
- 70. Miko, Ilona (2008). "Test crosses" (https://www.nature.com/scitable/topicpage/test-crosses-5 85/). *Nature Education*. **1** (1): 136. Archived (https://web.archive.org/web/20210521003428/https://www.nature.com/scitable/topicpage/test-crosses-585/) from the original on 2021-05-21. Retrieved 2021-05-28.
- 71. Miko, Ilona (2008). "Thomas Hunt Morgan and sex linkage" (https://www.nature.com/scitable/topicpage/thomas-hunt-morgan-and-sex-linkage-452/). Nature Education. 1 (1): 143. Archived (https://web.archive.org/web/20210520234008/https://www.nature.com/scitable/topicpage/thomas-hunt-morgan-and-sex-linkage-452/) from the original on 2021-05-20. Retrieved 2021-05-28.
- 72. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "DNA and its role in heredity". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 172–193. ISBN 978-1464175121.
- 73. Russell, Peter (2001). *iGenetics* (https://archive.org/details/igenetics0000russ_v6o1). New York: Benjamin Cummings. ISBN 0-8053-4553-1.

- 74. Thanbichler, M; Wang, SC; Shapiro, L (October 2005). "The bacterial nucleoid: a highly organized and dynamic structure" (https://doi.org/10.1002%2Fjcb.20519). Journal of Cellular Biochemistry. **96** (3): 506–21. doi:10.1002/jcb.20519 (https://doi.org/10.1002%2Fjcb.20519). PMID 15988757 (https://pubmed.ncbi.nlm.nih.gov/15988757). S2CID 25355087 (https://api.semanticscholar.org/CorpusID:25355087).
- 75. "Genotype definition Medical Dictionary definitions" (http://www.medterms.com/script/main/art.asp?articlekey=8472). Medterms.com. 2012-03-19. Archived (https://web.archive.org/web/20130921054803/http://www.medterms.com/script/main/art.asp?articlekey=8472) from the original on 2013-09-21. Retrieved 2013-10-02.
- 76. Crick, Francis H. (1958). "On protein synthesis". *Symposia of the Society for Experimental Biology.* **12**: 138–63. PMID 13580867 (https://pubmed.ncbi.nlm.nih.gov/13580867).
- 77. Crick, Francis H. (August 1970). "Central dogma of molecular biology". *Nature*. **227** (5258): 561–3. Bibcode:1970Natur.227..561C (https://ui.adsabs.harvard.edu/abs/1970Natur.227..561C). doi:10.1038/227561a0 (https://doi.org/10.1038%2F227561a0). PMID 4913914 (https://pubmed.ncbi.nlm.nih.gov/4913914). S2CID 4164029 (https://api.semanticscholar.org/CorpusID:4164029).
- 78. "Central dogma reversed". *Nature*. **226** (5252): 1198–9. June 1970.

 Bibcode:1970Natur.226.1198. (https://ui.adsabs.harvard.edu/abs/1970Natur.226.1198.).

 doi:10.1038/2261198a0 (https://doi.org/10.1038%2F2261198a0). PMID 5422595 (https://pubmed.ncbi.nlm.nih.gov/5422595). S2CID 4184060 (https://api.semanticscholar.org/Corpust D:4184060).
- 79. Lin, Yihan; Elowitz, Michael B. (2016). "Central Dogma Goes Digital" (https://doi.org/10.101 6%2Fj.molcel.2016.03.005). *Molecular Cell.* **61** (6): 791–792. doi:10.1016/j.molcel.2016.03.005 (https://doi.org/10.1016%2Fj.molcel.2016.03.005). PMID 26990983 (https://pubmed.ncbi.nlm.nih.gov/26990983).
- 80. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "Regulation of gene expression". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 215–233. ISBN 978-1464175121.
- 81. Keene, Jack D.; Tenenbaum, Scott A. (2002). "Eukaryotic mRNPs may represent posttranscriptional operons" (https://doi.org/10.1016%2Fs1097-2765%2802%2900559-2). *Molecular Cell.* **9** (6): 1161–1167. doi:10.1016/s1097-2765(02)00559-2 (https://doi.org/10.1016%2Fs1097-2765%2802%2900559-2). PMID 12086614 (https://pubmed.ncbi.nlm.nih.gov/12086614).
- 82. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "Genes, development, and evolution". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 273–298. ISBN 978-1464175121.
- 83. Slack, J.M.W. (2013) Essential Developmental Biology. Wiley-Blackwell, Oxford.
- 84. Slack, J.M.W. (2007). "Metaplasia and transdifferentiation: from pure biology to the clinic". *Nature Reviews Molecular Cell Biology*. **8** (5): 369–378. doi:10.1038/nrm2146 (https://doi.org/10.1038%2Fnrm2146). PMID 17377526 (https://pubmed.ncbi.nlm.nih.gov/17377526). S2CID 3353748 (https://api.semanticscholar.org/CorpusID:3353748).
- 85. Atala, Anthony; Lanza, Robert (2012). *Handbook of Stem Cells* (https://books.google.com/books?id=wm-K_dKpjBAC&pg=RA1-PA451). Academic Press. p. 452. ISBN 978-0-12-385943-3. Archived (https://web.archive.org/web/20210412065854/https://books.google.com/books?id=wm-K_dKpjBAC&pg=RA1-PA451) from the original on 2021-04-12. Retrieved 2021-05-28.

- 86. Yanes, Oscar; Clark, Julie; Wong, Diana M.; Patti, Gary J.; Sánchez-Ruiz, Antonio; Benton, H. Paul; Trauger, Sunia A.; Desponts, Caroline; Ding, Sheng; Siuzdak, Gary (June 2010). "Metabolic oxidation regulates embryonic stem cell differentiation" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC2873061). Nature Chemical Biology. 6 (6): 411–417. doi:10.1038/nchembio.364 (https://doi.org/10.1038%2Fnchembio.364). PMC 2873061 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC2873061). PMID 20436487 (https://pubmed.ncbi.nlm.nih.gov/20436487).
- 87. Carroll, Sean B. "The Origins of Form" (http://www.naturalhistorymag.com/features/061488/t he-origins-of-form). *Natural History*. Archived (https://web.archive.org/web/2018100915450 1/http://www.naturalhistorymag.com/features/061488/the-origins-of-form) from the original on 9 October 2018. Retrieved 9 October 2016. "Biologists could say, with confidence, that forms change, and that natural selection is an important force for change. Yet they could say nothing about how that change is accomplished. How bodies or body parts change, or how new structures arise, remained complete mysteries."
- 88. Hall, Brian K.; Hallgrímsson, Benedikt (2007). <u>Strickberger's Evolution</u> (https://books.google.com/books?id=jrDD3cyA09kC). Jones & Bartlett Publishers. pp. 4–6. <u>ISBN 978-1-4496-4722-3</u>. <u>Archived (https://web.archive.org/web/20230326093707/https://books.google.com/books?id=jrDD3cyA09kC)</u> from the original on 2023-03-26. Retrieved 2021-05-27.
- 89. "Evolution Resources" (http://www.nas.edu/evolution/index.html). Washington, D.C.: National Academies of Sciences, Engineering, and Medicine. 2016. Archived (https://web.archive.org/web/20160603230514/http://www.nas.edu/evolution/index.html) from the original on 2016-06-03.
- 90. Urry, Lisa; Cain, Michael; Wasserman, Steven; Minorsky, Peter; Reece, Jane (2017). "Descent with modifications: A Darwinian view of life". *Campbell Biology* (11th ed.). New York: Pearson. pp. 466–483. ISBN 978-0134093413.
- 91. Lewontin, Richard C. (November 1970). "The Units of Selection" (http://joelvelasco.net/teac hing/167/lewontin%2070%20-%20the%20units%20of%20selection.pdf) (PDF). *Annual Review of Ecology and Systematics*. 1: 1–18. doi:10.1146/annurev.es.01.110170.000245 (https://doi.org/10.1146%2Fannurev.es.01.110170.000245). JSTOR 2096764 (https://www.jstor.org/stable/2096764). S2CID 84684420 (https://api.semanticscholar.org/CorpusID:8468442 0). Archived (https://web.archive.org/web/20150206172942/http://joelvelasco.net/teaching/167/lewontin%2070%20-%20the%20units%20of%20selection.pdf) (PDF) from the original on 2015-02-06.
- 92. Darwin, Charles (1859). On the Origin of Species, John Murray.
- 93. Futuyma, Douglas J.; Kirkpatrick, Mark (2017). "Evolutionary biology". *Evolution* (4th ed.). Sunderland, Mass.: Sinauer Associates. pp. 3–26.
- 94. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "Processes of evolution". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 299–324. ISBN 978-1464175121.
- 95. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "Speciation". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 343–356. ISBN 978-1464175121.
- 96. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "Reconstructing and using phylogenies". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 325–342. ISBN 978-1464175121.
- 97. Kitching, Ian J.; Forey, Peter L.; Williams, David M. (2001). "Cladistics" (https://www.science direct.com/science/article/pii/B9780123847195000228). In Levin, Simon A. (ed.). Encyclopedia of Biodiversity (2nd ed.). Elsevier. pp. 33–45. doi:10.1016/B978-0-12-384719-5.00022-8 (https://doi.org/10.1016%2FB978-0-12-384719-5.00022-8). ISBN 9780123847201. Archived (https://web.archive.org/web/20210829234556/https://www.sciencedirect.com/science/article/pii/B9780123847195000228) from the original on 29 August 2021. Retrieved 29 August 2021.)

- 98. Futuyma, Douglas J.; Kirkpatrick, Mark (2017). "Phylogeny: The unity and diversity of life". *Evolution* (4th ed.). Sunderland, Mass.: Sinauer Associates. pp. 401–429.
- 99. Woese, CR; Kandler, O; Wheelis, ML (June 1990). "Towards a natural system of organisms: proposal for the domains Archaea, Bacteria, and Eucarya" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC54159). Proceedings of the National Academy of Sciences of the United States of America. 87 (12): 4576–79. Bibcode:1990PNAS...87.4576W (https://ui.adsabs.harvard.edu/abs/1990PNAS...87.4576W). doi:10.1073/pnas.87.12.4576 (https://doi.org/10.1073/pnas.87.12.4576). PMC 54159 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC54159). PMID 2112744 (https://pubmed.ncbi.nlm.nih.gov/2112744).
- 100. Montévil, M; Mossio, M; Pocheville, A; Longo, G (October 2016). "Theoretical principles for biology: Variation" (https://www.academia.edu/27942089). Progress in Biophysics and Molecular Biology. From the Century of the Genome to the Century of the Organism: New Theoretical Approaches. 122 (1): 36–50. doi:10.1016/j.pbiomolbio.2016.08.005 (https://doi.org/10.1016%2Fj.pbiomolbio.2016.08.005). PMID 27530930 (https://pubmed.ncbi.nlm.nih.gov/27530930). S2CID 3671068 (https://api.semanticscholar.org/CorpusID:3671068). Archived (https://web.archive.org/web/20180320150224/http://www.academia.edu/27942089/Theoretical_principles_for_biology_Variation) from the original on 2018-03-20.
- 101. De Duve, Christian (2002). *Life Evolving: Molecules, Mind, and Meaning* (https://archive.org/details/lifeevolvingmole00duve_331). New York: Oxford University Press. p. 44 (https://archive.org/details/lifeevolvingmole00duve_331/page/n60). ISBN 978-0-19-515605-8.
- 102. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "The history of life on Earth". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 357–376. ISBN 978-1464175121.
- 103. "Stratigraphic Chart 2022" (https://stratigraphy.org/ICSchart/ChronostratChart2022-02.pdf) (PDF). International Stratigraphic Commission. February 2022. Archived (https://web.archive.org/web/20220402100018/https://stratigraphy.org/ICSchart/ChronostratChart2022-02.pdf) (PDF) from the original on 2 April 2022. Retrieved 25 April 2022.
- 104. Futuyma 2005
- 105. Futuyma, DJ (2005). *Evolution* (https://archive.org/details/evolution0000futu). Sinauer Associates. ISBN 978-0-87893-187-3. OCLC 57311264 (https://search.worldcat.org/oclc/57311264).
- 106. Rosing, Minik T. (January 29, 1999). "¹³C-Depleted Carbon Microparticles in >3700-Ma Sea-Floor Sedimentary Rocks from West Greenland". <u>Science</u>. **283** (5402): 674–676. Bibcode:1999Sci...283..674R (https://ui.adsabs.harvard.edu/abs/1999Sci...283..674R). doi:10.1126/science.283.5402.674 (https://doi.org/10.1126%2Fscience.283.5402.674). PMID 9924024 (https://pubmed.ncbi.nlm.nih.gov/9924024).
- 107. Ohtomo, Yoko; Kakegawa, Takeshi; Ishida, Akizumi; et al. (January 2014). "Evidence for biogenic graphite in early Archaean Isua metasedimentary rocks". *Nature Geoscience*. **7** (1): 25–28. Bibcode:2014NatGe...7...250 (https://ui.adsabs.harvard.edu/abs/2014NatGe...7...250). doi:10.1038/ngeo2025 (https://doi.org/10.1038%2Fngeo2025).
- 108. Nisbet, Euan G.; Fowler, C.M.R. (December 7, 1999). "Archaean metabolic evolution of microbial mats" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC1690475). Proceedings of the Royal Society B. 266 (1436): 2375–2382. doi:10.1098/rspb.1999.0934 (https://doi.org/10.1098%2Frspb.1999.0934). PMC 1690475 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC 1690475).
- 109. Knoll, Andrew H.; Javaux, Emmanuelle J.; Hewitt, David; et al. (June 29, 2006). "Eukaryotic organisms in Proterozoic oceans" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC1578724). Philosophical Transactions of the Royal Society B. 361 (1470): 1023–1038. doi:10.1098/rstb.2006.1843 (https://doi.org/10.1098%2Frstb.2006.1843). PMC 1578724 (htt ps://www.ncbi.nlm.nih.gov/pmc/articles/PMC1578724). PMID 16754612 (https://pubmed.ncbi.nlm.nih.gov/16754612).

- 110. Fedonkin, Mikhail A. (March 31, 2003). "The origin of the Metazoa in the light of the Proterozoic fossil record" (https://web.archive.org/web/20090226122725/http://www.vend.paleo.ru/pub/Fedonkin_2003.pdf) (PDF). Paleontological Research. 7 (1): 9–41. doi:10.2517/prpsj.7.9 (https://doi.org/10.2517%2Fprpsj.7.9). S2CID 55178329 (https://api.semanticscholar.org/CorpusID:55178329). Archived from the original (http://www.vend.paleo.ru/pub/Fedonkin_2003.pdf) (PDF) on 2009-02-26. Retrieved 2008-09-02.
- 111. Bonner, John Tyler (January 7, 1998). "The origins of multicellularity". *Integrative Biology*. **1** (1): 27–36. doi:10.1002/(SICI)1520-6602(1998)1:1<27::AID-INBI4>3.0.CO;2-6 (https://doi.org/10.1002%2F%28SICI%291520-6602%281998%291%3A1%3C27%3A%3AAID-INBI4%3E 3.0.CO%3B2-6).
- 112. Strother, Paul K.; Battison, Leila; Brasier, Martin D.; et al. (May 26, 2011). "Earth's earliest non-marine eukaryotes". *Nature*. **473** (7348): 505–509. Bibcode:2011Natur.473..505S (https://ui.adsabs.harvard.edu/abs/2011Natur.473..505S). doi:10.1038/nature09943 (https://doi.org/10.1038%2Fnature09943). PMID 21490597 (https://pubmed.ncbi.nlm.nih.gov/21490597). S2CID 4418860 (https://api.semanticscholar.org/CorpusID:4418860).
- 113. Beraldi-Campesi, Hugo (February 23, 2013). "Early life on land and the first terrestrial ecosystems" (https://doi.org/10.1186%2F2192-1709-2-1). *Ecological Processes*. **2** (1): 1–17. Bibcode:2013EcoPr...2....1B (https://ui.adsabs.harvard.edu/abs/2013EcoPr...2....1B). doi:10.1186/2192-1709-2-1 (https://doi.org/10.1186%2F2192-1709-2-1).
- 114. Algeo, Thomas J.; Scheckler, Stephen E. (January 29, 1998). "Terrestrial-marine teleconnections in the Devonian: links between the evolution of land plants, weathering processes, and marine anoxic events" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC1692 181). Philosophical Transactions of the Royal Society B. 353 (1365): 113–130. doi:10.1098/rstb.1998.0195 (https://doi.org/10.1098%2Frstb.1998.0195). PMC 1692181 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC1692181).
- 115. Jun-Yuan, Chen; Oliveri, Paola; Chia-Wei, Li; et al. (April 25, 2000). "Precambrian animal diversity: Putative phosphatized embryos from the Doushantuo Formation of China" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC18256). Proc. Natl. Acad. Sci. U.S.A. 97 (9): 4457–4462. Bibcode:2000PNAS...97.4457C (https://ui.adsabs.harvard.edu/abs/2000PNAS...97.4457C). doi:10.1073/pnas.97.9.4457 (https://doi.org/10.1073%2Fpnas.97.9.4457). PMC 18256 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC18256). PMID 10781044 (https://pubmed.ncbi.nlm.nih.gov/10781044).
- 116. D-G., Shu; H-L., Luo; Conway Morris, Simon; et al. (November 4, 1999). "Lower Cambrian vertebrates from south China" (https://web.archive.org/web/20090226122732/http://www.bios.niu.edu/davis/bios458/Shu1.pdf) (PDF). Nature. 402 (6757): 42–46.

 Bibcode:1999Natur.402...42S (https://ui.adsabs.harvard.edu/abs/1999Natur.402...42S).
 doi:10.1038/46965 (https://doi.org/10.1038%2F46965). S2CID 4402854 (https://api.semanticscholar.org/CorpusID:4402854). Archived from the original (http://www.bios.niu.edu/davis/bios458/Shu1.pdf) (PDF) on 2009-02-26. Retrieved 2015-01-22.
- 117. Hoyt, Donald F. (February 17, 1997). "Synapsid Reptiles" (https://web.archive.org/web/2009 0520072737/http://www.csupomona.edu/~dfhoyt/classes/zoo138/SYNAPSID.HTML). ZOO 138 Vertebrate Zoology (Lecture). Pomona, Calif.: California State Polytechnic University, Pomona. Archived from the original (http://www.csupomona.edu/~dfhoyt/classes/zoo138/SY NAPSID.HTML) on 2009-05-20. Retrieved 2015-01-22.
- 118. Barry, Patrick L. (January 28, 2002). Phillips, Tony (ed.). "The Great Dying" (https://science.nasa.gov/science-news/science-at-nasa/2002/28jan_extinction/). Science@NASA. Marshall Space Flight Center. Archived (https://web.archive.org/web/20100410015208/https://science.nasa.gov/science-news/science-at-nasa/2002/28jan_extinction/) from the original on 2010-04-10. Retrieved 2015-01-22.

- 119. Tanner, Lawrence H.; Lucas, Spencer G.; Chapman, Mary G. (March 2004). "Assessing the record and causes of Late Triassic extinctions" (https://web.archive.org/web/200710252258 41/http://nmnaturalhistory.org/pdf_files/TJB.pdf) (PDF). Earth-Science Reviews. 65 (1–2): 103–139. Bibcode:2004ESRv...65..103T (https://ui.adsabs.harvard.edu/abs/2004ESRv...65.. 103T). doi:10.1016/S0012-8252(03)00082-5 (https://doi.org/10.1016%2FS0012-8252%280 3%2900082-5). Archived from the original (http://nmnaturalhistory.org/pdf_files/TJB.pdf) (PDF) on 2007-10-25. Retrieved 2007-10-22.
- 120. Benton, Michael J. (1997). *Vertebrate Palaeontology* (2nd ed.). London: Chapman & Hall. ISBN 978-0-412-73800-5. OCLC 37378512 (https://search.worldcat.org/oclc/37378512).
- 121. Fastovsky, David E.; Sheehan, Peter M. (March 2005). "The Extinction of the Dinosaurs in North America" (https://www.geosociety.org/gsatoday/archive/15/3/pdf/i1052-5173-15-3-4.pd f) (PDF). GSA Today. 15 (3): 4–10. doi:10.1130/1052-5173(2005)015<4:TEOTDI>2.0.CO;2 (https://doi.org/10.1130%2F1052-5173%282005%29015%3C4%3ATEOTDI%3E2.0.CO%3 B2). Archived (https://web.archive.org/web/20190322190338/https://www.geosociety.org/gsatoday/archive/15/3/pdf/i1052-5173-15-3-4.pdf) (PDF) from the original on 2019-03-22. Retrieved 2015-01-23.
- 122. Roach, John (June 20, 2007). "Dinosaur Extinction Spurred Rise of Modern Mammals" (https://web.archive.org/web/20080511161825/https://news.nationalgeographic.com/news/2007/06/070620-mammals-dinos.html). National Geographic News. Washington, D.C.: National Geographic Society. Archived from the original (https://news.nationalgeographic.com/news/2007/06/070620-mammals-dinos.html) on 2008-05-11. Retrieved 2020-02-21.
 - Wible, John R.; Rougier, Guillermo W.; Novacek, Michael J.; et al. (June 21, 2007). "Cretaceous eutherians and Laurasian origin for placental mammals near the K/T boundary". *Nature*. **447** (7147): 1003–1006. Bibcode:2007Natur.447.1003W (https://ui.a dsabs.harvard.edu/abs/2007Natur.447.1003W). doi:10.1038/nature05854 (https://doi.org/10.1038%2Fnature05854). PMID 17581585 (https://pubmed.ncbi.nlm.nih.gov/17581585). S2CID 4334424 (https://api.semanticscholar.org/CorpusID:4334424).
- 123. Van Valkenburgh, Blaire (May 1, 1999). "Major Patterns in the History of Carnivorous Mammals" (https://zenodo.org/record/890156). *Annual Review of Earth and Planetary Sciences*. 27: 463–493. Bibcode:1999AREPS..27..463V (https://ui.adsabs.harvard.edu/abs/1999AREPS..27..463V). doi:10.1146/annurev.earth.27.1.463 (https://doi.org/10.1146%2Fan nurev.earth.27.1.463). Archived (https://web.archive.org/web/20200229201201/https://zenodo.org/record/890156) from the original on February 29, 2020. Retrieved May 15, 2021.
- 124. Fredrickson, J. K.; Zachara, J. M.; Balkwill, D. L. (July 2004). "Geomicrobiology of high-level nuclear waste-contaminated vadose sediments at the Hanford site, Washington state" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC444790). *Applied and Environmental Microbiology*. **70** (7): 4230–41. Bibcode:2004ApEnM..70.4230F (https://ui.adsabs.harvard.edu/abs/2004ApEnM..70.4230F). doi:10.1128/AEM.70.7.4230-4241.2004 (https://doi.org/10.1128%2FAEM.70.7.4230-4241.2004). PMC 444790 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC444790). PMID 15240306 (https://pubmed.ncbi.nlm.nih.gov/15240306).
- 125. Dudek, N. K.; Sun, C. L.; Burstein, D. (2017). "Novel Microbial Diversity and Functional Potential in the Marine Mammal Oral Microbiome" (https://escholarship.org/content/qt1w91s 3vq/qt1w91s3vq.pdf?t=pghuwe) (PDF). Current Biology. 27 (24): 3752–3762.

 Bibcode:2017CBio...27E3752D (https://ui.adsabs.harvard.edu/abs/2017CBio...27E3752D).

 doi:10.1016/j.cub.2017.10.040 (https://doi.org/10.1016%2Fj.cub.2017.10.040).

 PMID 29153320 (https://pubmed.ncbi.nlm.nih.gov/29153320). S2CID 43864355 (https://api. semanticscholar.org/CorpusID:43864355). Archived (https://web.archive.org/web/20210308 145807/https://escholarship.org/content/qt1w91s3vq/qt1w91s3vq.pdf?t=pghuwe) (PDF) from the original on 2021-03-08. Retrieved 2021-05-14.

- 126. Pace, N. R. (May 2006). "Time for a change" (https://doi.org/10.1038%2F441289a). *Nature*. **441** (7091): 289. Bibcode:2006Natur.441..289P (https://ui.adsabs.harvard.edu/abs/2006Natur.441..289P). doi:10.1038/441289a (https://doi.org/10.1038%2F441289a). PMID 16710401 (https://pubmed.ncbi.nlm.nih.gov/16710401). S2CID 4431143 (https://api.semanticscholar.org/CorpusID:4431143).
- 127. Stoeckenius, W. (October 1981). "Walsby's square bacterium: fine structure of an orthogonal procaryote" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC216199). Journal of Bacteriology. 148 (1): 352–60. doi:10.1128/JB.148.1.352-360.1981 (https://doi.org/10.1128/JB.148.1.352-360.1981). PMC 216199 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC216199). PMID 7287626 (https://pubmed.ncbi.nlm.nih.gov/7287626).
- 128. "Archaea Basic Biology" (https://basicbiology.net/micro/microorganisms/archaea). March 2018. Archived (https://web.archive.org/web/20210428221114/https://basicbiology.net/microorganisms/archaea) from the original on 2021-04-28. Retrieved 2021-05-14.
- 129. Bang, C.; Schmitz, R. A. (September 2015). "Archaea associated with human surfaces: not to be underestimated" (https://doi.org/10.1093%2Ffemsre%2Ffuv010). FEMS Microbiology Reviews. **39** (5): 631–48. doi:10.1093/femsre/fuv010 (https://doi.org/10.1093%2Ffemsre%2 Ffuv010). PMID 25907112 (https://pubmed.ncbi.nlm.nih.gov/25907112).
- 130. Moissl-Eichinger. C.; Pausan, M.; Taffner, J.; Berg, G.; Bang, C.; Schmitz, R. A. (January 2018). "Archaea Are Interactive Components of Complex Microbiomes". *Trends in Microbiology*. **26** (1): 70–85. doi:10.1016/j.tim.2017.07.004 (https://doi.org/10.1016%2Fj.tim. 2017.07.004). PMID 28826642 (https://pubmed.ncbi.nlm.nih.gov/28826642).
- 131. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "The origin and diversification of eukaryotes". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 402–419. ISBN 978-1464175121.
- 132. O'Malley, Maureen A.; Leger, Michelle M.; Wideman, Jeremy G.; Ruiz-Trillo, Iñaki (2019-02-18). "Concepts of the last eukaryotic common ancestor". *Nature Ecology & Evolution*. **3** (3). Springer Science and Business Media LLC: 338–344. Bibcode:2019NatEE...3..338O (https://ui.adsabs.harvard.edu/abs/2019NatEE...3..338O). doi:10.1038/s41559-019-0796-3 (https://doi.org/10.1038%2Fs41559-019-0796-3). hdl:10261/201794 (https://hdl.handle.net/10261%2F201794). PMID 30778187 (https://pubmed.ncbi.nlm.nih.gov/30778187). S2CID 67790751 (https://api.semanticscholar.org/CorpusID:67790751).
- 133. Taylor, F. J. R. 'M. (2003-11-01). "The collapse of the two-kingdom system, the rise of protistology and the founding of the International Society for Evolutionary Protistology (ISEP)" (https://doi.org/10.1099%2Fijs.0.02587-0). International Journal of Systematic and Evolutionary Microbiology. 53 (6). Microbiology Society: 1707–1714.

 doi:10.1099/ijs.0.02587-0 (https://doi.org/10.1099%2Fijs.0.02587-0). PMID 14657097 (https://pubmed.ncbi.nlm.nih.gov/14657097).
- 134. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "The evolution of plants". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 420–449. ISBN 978-1464175121.
- 135. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "The evolution and diversity of fungi". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 451–468. ISBN 978-1464175121.
- 136. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "Animal origins and diversity". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 469–519. ISBN 978-1464175121.
- 137. Wu, K. J. (15 April 2020). "There are more viruses than stars in the universe. Why do only some infect us? More than a quadrillion quadrillion individual viruses exist on Earth, but most are not poised to hop into humans. Can we find the ones that are?" (https://web.archive.org/web/20200528154701/https://www.nationalgeographic.com/science/2020/04/factors-allow-viruses-infect-humans-coronavirus/). National Geographic Society. Archived from the original (https://www.nationalgeographic.com/science/2020/04/factors-allow-viruses-infect-humans-coronavirus/) on 28 May 2020. Retrieved 18 May 2020.

- 138. Koonin, E. V.; Senkevich, T. G.; Dolja, V. V. (September 2006). <u>"The ancient Virus World and evolution of cells" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC1594570)</u>. *Biology Direct*. **1** (1): 29. doi:10.1186/1745-6150-1-29 (https://doi.org/10.1186%2F1745-6150-1-29). PMC 1594570 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC1594570). PMID 16984643 (https://pubmed.ncbi.nlm.nih.gov/16984643).
- 139. Zimmer, C. (26 February 2021). "The Secret Life of a Coronavirus An oily, 100-nanometerwide bubble of genes has killed more than two million people and reshaped the world. Scientists don't quite know what to make of it" (https://ghostarchive.org/archive/20211228/https://www.nytimes.com/2021/02/26/opinion/sunday/coronavirus-alive-dead.html). The New York Times. Archived from the original (https://www.nytimes.com/2021/02/26/opinion/sunday/coronavirus-alive-dead.html) on 2021-12-28. Retrieved 28 February 2021.
- 140. "Virus Taxonomy: 2019 Release" (https://ictv.global/taxonomy). talk.ictvonline.org. International Committee on Taxonomy of Viruses. Archived (https://web.archive.org/web/202 00320103754/https://talk.ictvonline.org/taxonomy) from the original on 20 March 2020. Retrieved 25 April 2020.
- 141. Lawrence C. M.; Menon S.; Eilers, B. J. (May 2009). "Structural and functional studies of archaeal viruses" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC2675988). The Journal of Biological Chemistry. 284 (19): 12599–603. doi:10.1074/jbc.R800078200 (https://doi.org/10.1074%2Fjbc.R800078200). PMC 2675988 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC2675988). PMID 19158076 (https://pubmed.ncbi.nlm.nih.gov/19158076).
- 142. Edwards, R.A.; Rohwer, F. (June 2005). "Viral metagenomics". *Nature Reviews. Microbiology*. **3** (6): 504–10. doi:10.1038/nrmicro1163 (https://doi.org/10.1038%2Fnrmicro1163). PMID 15886693 (https://pubmed.ncbi.nlm.nih.gov/15886693). S2CID 8059643 (https://api.semanticscholar.org/CorpusID:8059643).
- 143. Canchaya, C.; Fournous, G.; Chibani-Chennoufi, S. (August 2003). "Phage as agents of lateral gene transfer". *Current Opinion in Microbiology*. **6** (4): 417–24. doi:10.1016/S1369-5274(03)00086-9 (https://doi.org/10.1016%2FS1369-5274%2803%2900086-9). PMID 12941415 (https://pubmed.ncbi.nlm.nih.gov/12941415).
- 144. Rybicki, E. P. (1990). "The classification of organisms at the edge of life, or problems with virus systematics". *South African Journal of Science*. **86**: 182–86.
- 145. Koonin, E. V.; Starokadomskyy, P. (October 2016). "Are viruses alive? The replicator paradigm sheds decisive light on an old but misguided question" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC5406846). Studies in History and Philosophy of Biological and Biomedical Sciences. 59: 125–134. doi:10.1016/j.shpsc.2016.02.016 (https://doi.org/10.1016%2Fj.shpsc.2016.02.016). PMC 5406846 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC5406846). PMID 26965225 (https://pubmed.ncbi.nlm.nih.gov/26965225).
- 146. Begon, M; Townsend, CR; Harper, JL (2006). *Ecology: From individuals to ecosystems* (4th ed.). Blackwell. ISBN 978-1-4051-1117-1.
- 147. *Habitats of the world* (https://books.google.com/books?id=U-_mlcy8rGgC&pg=PA238). New York: Marshall Cavendish. 2004. p. 238. <u>ISBN</u> 978-0-7614-7523-1. <u>Archived (https://web.archive.org/web/20210415113154/https://books.google.com/books?id=U-_mlcy8rGgC&pg=PA238) from the original on 2021-04-15. Retrieved 2020-08-24.</u>
- 148. Tansley (1934); Molles (1999), p. 482; Chapin *et al.* (2002), p. 380; Schulze *et al.* (2005); p. 400; Gurevitch *et al.* (2006), p. 522; Smith & Smith 2012, p. G-5
- 149. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "The distribution of Earth's ecological systems". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 845–863. ISBN 978-1464175121.
- 150. Odum, Eugene P (1971). *Fundamentals of Ecology* (https://archive.org/details/fundamentals ofec0000odum) (3rd ed.). New York: Saunders. ISBN 978-0-534-42066-6.
- 151. Chapin III, F. Stuart; Matson, Pamela A.; Mooney, Harold A. (2002). "The ecosystem concept". *Principles of Terrestrial Ecosystem Ecology*. New York: Springer. p. 10. <u>ISBN</u> <u>978-</u>0-387-95443-1.

- 152. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "Populations". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 864–897. ISBN 978-1464175121.
- 153. Urry, Lisa; Cain, Michael; Wasserman, Steven; Minorsky, Peter; Reece, Jane (2017). "Population ecology". *Campbell Biology* (11th ed.). New York: Pearson. pp. 1188–1211. ISBN 978-0134093413.
- 154. "Population" (http://www.biology-online.org/dictionary/Population). Biology Online. Archived (https://web.archive.org/web/20190413145351/https://www.biology-online.org/dictionary/Population) from the original on 13 April 2019. Retrieved 5 December 2012.
- 155. "Definition of population (biology)" (https://web.archive.org/web/20160304104019/http://www.oxforddictionaries.com/definition/english/population?q=population). Oxford Dictionaries.

 Oxford University Press. Archived from the original (http://oxforddictionaries.com/definition/english/population?q=population) on 4 March 2016. Retrieved 5 December 2012. "a community of animals, plants, or humans among whose members interbreeding occurs"
- 156. Hartl, Daniel (2007). *Principles of Population Genetics*. Sinauer Associates. p. 45. ISBN 978-0-87893-308-2.
- 157. Chapman, Eric J.; Byron, Carrie J. (2018-01-01). "The flexible application of carrying capacity in ecology" (https://doi.org/10.1016%2Fj.gecco.2017.e00365). Global Ecology and Conservation. 13: e00365. Bibcode:2018GEcoC..1300365C (https://ui.adsabs.harvard.edu/abs/2018GEcoC..1300365C). doi:10.1016/j.gecco.2017.e00365 (https://doi.org/10.1016%2Fj.gecco.2017.e00365).
- 158. Odum, E. P.; Barrett, G. W. (2005). Fundamentals of Ecology (https://web.archive.org/web/2 0110820163059/http://www.cengage.com/aushed/instructor.do?disciplinenumber=1041&pro duct_isbn=9780534420666&courseid=BI03&codeid=2BF6&subTab=&mainTab=About_the_Book&mailFlag=true&topicName=) (5th ed.). Brooks/Cole, a part of Cengage Learning. ISBN 978-0-534-42066-6. Archived from the original (http://www.cengage.com/aushed/instructor.do?disciplinenumber=1041&product_isbn=9780534420666&courseid=BI03&codeid=2BF6&subTab=&mainTab=About_the_Book&mailFlag=true&topicName=) on 2011-08-20.
- 159. Sanmartín, Isabel (December 2012). "Historical Biogeography: Evolution in Time and Space" (https://doi.org/10.1007%2Fs12052-012-0421-2). *Evolution: Education and Outreach.* **5** (4): 555–568. doi:10.1007/s12052-012-0421-2 (https://doi.org/10.1007%2Fs12052-012-0421-2). hdl:10261/167031 (https://hdl.handle.net/10261%2F167031). ISSN 1936-6434 (https://search.worldcat.org/issn/1936-6434).
- 160. Wootton, JT; Emmerson, M (2005). "Measurement of Interaction Strength in Nature". <u>Annual Review of Ecology, Evolution, and Systematics</u>. **36**: 419–44. doi:10.1146/annurev.ecolsys.36.091704.175535 (https://doi.org/10.1146%2Fannurev.ecolsys.36.091704.175535). JSTOR 30033811 (https://www.jstor.org/stable/30033811).
- 161. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "Ecological and evolutionary consequences within and among species". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 882–897. ISBN 978-1464175121.
- 162. Smith, AL (1997). Oxford dictionary of biochemistry and molecular biology. Oxford [Oxfordshire]: Oxford University Press. p. 508. ISBN 978-0-19-854768-6. "Photosynthesis the synthesis by organisms of organic chemical compounds, esp. carbohydrates, from carbon dioxide using energy obtained from light rather than the oxidation of chemical compounds."
- 163. Edwards, Katrina. "Microbiology of a Sediment Pond and the Underlying Young, Cold, Hydrologically Active Ridge Flank". *Woods Hole Oceanographic Institution*.
- 164. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "Ecological communities". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 898–915. ISBN 978-1464175121.

- 165. Riebeek, Holli (16 June 2011). "The Carbon Cycle" (https://earthobservatory.nasa.gov/Features/CarbonCycle/?src=eoa-features). *Earth Observatory*. NASA. Archived (https://web.archive.org/web/20160305010126/http://earthobservatory.nasa.gov/Features/CarbonCycle/?src=eoa-features) from the original on 5 March 2016. Retrieved 5 April 2018.
- 166. Hillis, David M.; Sadava, David; Hill, Richard W.; Price, Mary V. (2014). "The distribution of Earth's ecological systems". *Principles of Life* (2nd ed.). Sunderland, Mass.: Sinauer Associates. pp. 916–934. ISBN 978-1464175121.
- 167. Sahney, S.; Benton, M. J (2008). "Recovery from the most profound mass extinction of all time" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC2596898). Proceedings of the Royal Society B: Biological Sciences. 275 (1636): 759–65. doi:10.1098/rspb.2007.1370 (https://doi.org/10.1098%2Frspb.2007.1370). PMC 2596898 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC2596898). PMID 18198148 (https://pubmed.ncbi.nlm.nih.gov/18198148).
- 168. Soulé, Michael E.; Wilcox, Bruce A. (1980). *Conservation biology: an evolutionary-ecological perspective*. Sunderland, Mass.: Sinauer Associates. ISBN 978-0-87893-800-1.
- 169. Soulé, Michael E. (1986). "What is Conservation Biology?" (https://web.archive.org/web/201 90412085412/http://www.michaelsoule.com/resource_files/85/85_resource_file1.pdf) (PDF). BioScience. 35 (11). American Institute of Biological Sciences: 727–34. doi:10.2307/1310054 (https://doi.org/10.2307%2F1310054). JSTOR 1310054 (https://www.jstor.org/stable/1310054). Archived from the original (http://www.michaelsoule.com/resource files/85/85 resource file1.pdf) (PDF) on 2019-04-12. Retrieved 2021-05-15.
- 170. Hunter, Malcolm L. (1996). *Fundamentals of conservation biology* (https://archive.org/details/fundamentalsofco00hunt). Oxford: Blackwell Science. ISBN 978-0-86542-371-8.
- 171. Meffe, Gary K.; Martha J. Groom (2006). *Principles of conservation biology* (3rd ed.). Sunderland, Mass.: Sinauer Associates. ISBN 978-0-87893-518-5.
- 172. Van Dyke, Fred (2008). Conservation biology: foundations, concepts, applications (https://books.google.com/books?id=Evh1UD3ZYWcC) (2nd ed.). New York: Springer-Verlag. doi:10.1007/978-1-4020-6891-1 (https://doi.org/10.1007%2F978-1-4020-6891-1). hdl:11059/14777 (https://hdl.handle.net/11059%2F14777). ISBN 978-1402068904. OCLC 232001738 (https://search.worldcat.org/oclc/232001738). Archived (https://web.archive.org/web/20200727115147/https://books.google.com/books?id=Evh1UD3ZYWcC) from the original on 2020-07-27. Retrieved 2021-05-15.
- 173. Sahney, S.; Benton, M. J.; Ferry, P. A. (2010). "Links between global taxonomic diversity, ecological diversity and the expansion of vertebrates on land" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC2936204). Biology Letters. 6 (4): 544–7. doi:10.1098/rsbl.2009.1024 (https://doi.org/10.1098%2Frsbl.2009.1024). PMC 2936204 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC2936204). PMID 20106856 (https://pubmed.ncbi.nlm.nih.gov/20106856).
- 174. Koh, Lian Pin; Dunn, Robert R.; Sodhi, Navjot S.; Colwell, Robert K.; Proctor, Heather C.; Smith, Vincent S. (2004). "Species coextinctions and the biodiversity crisis". *Science*. **305** (5690): 1632–4. Bibcode:2004Sci...305.1632K (https://ui.adsabs.harvard.edu/abs/2004Sci...305.1632K). doi:10.1126/science.1101101 (https://doi.org/10.1126%2Fscience.1101101). PMID 15361627 (https://pubmed.ncbi.nlm.nih.gov/15361627). S2CID 30713492 (https://api.semanticscholar.org/CorpusID:30713492).
- 175. Millennium Ecosystem Assessment (2005). *Ecosystems and Human Well-being: Biodiversity Synthesis.* World Resources Institute, Washington, D.C.[1] (http://www.millenniumassessment.org/documents/document.354.aspx.pdf) Archived (https://web.archive.org/web/20191014 033601/http://www.millenniumassessment.org/documents/document.354.aspx.pdf) 2019-10-14 at the Wayback Machine
- 176. Jackson, J. B. C. (2008). "Ecological extinction and evolution in the brave new ocean" (http s://www.ncbi.nlm.nih.gov/pmc/articles/PMC2556419). Proceedings of the National Academy of Sciences. 105 (Suppl 1): 11458–65. Bibcode:2008PNAS..10511458J (https://ui.adsabs.harvard.edu/abs/2008PNAS..10511458J). doi:10.1073/pnas.0802812105 (https://doi.org/10.1073%2Fpnas.0802812105). PMC 2556419 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC 2556419). PMID 18695220 (https://pubmed.ncbi.nlm.nih.gov/18695220).

177. Soule, Michael E. (1986). *Conservation Biology: The Science of Scarcity and Diversity*. Sinauer Associates. p. 584. ISBN 978-0-87893-795-0.

Further reading

- Alberts, B.; Johnson, A.; Lewis, J.; Raff, M.; Roberts, K.; Walter, P. (2002). <u>Molecular Biology of the Cell</u> (https://archive.org/details/molecularbiolog000wils) (4th ed.). Garland. <u>ISBN</u> 978-0-8153-3218-3. OCLC 145080076 (https://search.worldcat.org/oclc/145080076).
- Begon, M.; Townsend, C. R.; Harper, J. L. (2005). *Ecology: From Individuals to Ecosystems* (4th ed.). Blackwell Publishing Limited. <u>ISBN</u> <u>978-1-4051-1117-1</u>. <u>OCLC</u> <u>57639896</u> (https://search.worldcat.org/oclc/57639896).
- <u>Campbell, Neil</u> (2004). *Biology* (7th ed.). Benjamin-Cummings Publishing Company. ISBN 978-0-8053-7146-8. OCLC 71890442 (https://search.worldcat.org/oclc/71890442).
- Colinvaux, Paul (1979). Why Big Fierce Animals are Rare: An Ecologist's Perspective (https://archive.org/details/whybigfierceanim00paul) (reissue ed.). Princeton University Press. ISBN 978-0-691-02364-9. OCLC 10081738 (https://search.worldcat.org/oclc/10081738).
- Mayr, Ernst (1982). The Growth of Biological Thought: Diversity, Evolution, and Inheritance (https://books.google.com/books?id=pHThtE2R0UQC). Harvard University Press.
 ISBN 978-0-674-36446-2. Archived (https://web.archive.org/web/20151003080726/https://books.google.com/books?id=pHThtE2R0UQC) from the original on 2015-10-03. Retrieved 2015-06-27.
- Hoagland, Mahlon (2001). *The Way Life Works*. Jones and Bartlett Publishers inc. ISBN 978-0-7637-1688-2. OCLC 223090105 (https://search.worldcat.org/oclc/223090105).
- Janovy, John (2004). *On Becoming a Biologist* (2nd ed.). Bison Books. <u>ISBN</u> <u>978-0-8032-7620-8</u>. OCLC 55138571 (https://search.worldcat.org/oclc/55138571).
- Johnson, George B. (2005). *Biology, Visualizing Life* (https://archive.org/details/holtbiologyvisua00john). Holt, Rinehart, and Winston. <u>ISBN</u> 978-0-03-016723-2. <u>OCLC</u> 36306648 (https://search.worldcat.org/oclc/36306648).
- Tobin, Allan; Dusheck, Jennie (2005). *Asking About Life* (3rd ed.). Belmont, California: Wadsworth. ISBN 978-0-534-40653-0.

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- Biology Online Wiki Dictionary (https://www.biologyonline.com/dictionary)
- MIT video lecture series on biology (https://ocw.mit.edu/courses/biology/7-012-introduction-t o-biology-fall-2004/)
- OneZoom Tree of Life (https://www.onezoom.org/)
- Journal of the History of Biology (springer.com) (https://link.springer.com/journal/10739)

Journal links

- PLOS ONE
- PLOS Biology (https://journals.plos.org/plosbiology/) A peer-reviewed, open-access journal published by the Public Library of Science
- <u>Current Biology</u> (http://www.cell.com/current-biology/): General journal publishing <u>original</u> research from all areas of biology
- Biology Letters (https://royalsocietypublishing.org/journal/rsbl): A high-impact Royal Society journal publishing peer-reviewed biology papers of general interest

- Science (https://www.science.org/collections): Internationally renowned AAAS science journal see sections of the life sciences
- International Journal of Biological Sciences (https://www.ijbs.com/): A biological journal publishing significant peer-reviewed scientific papers
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