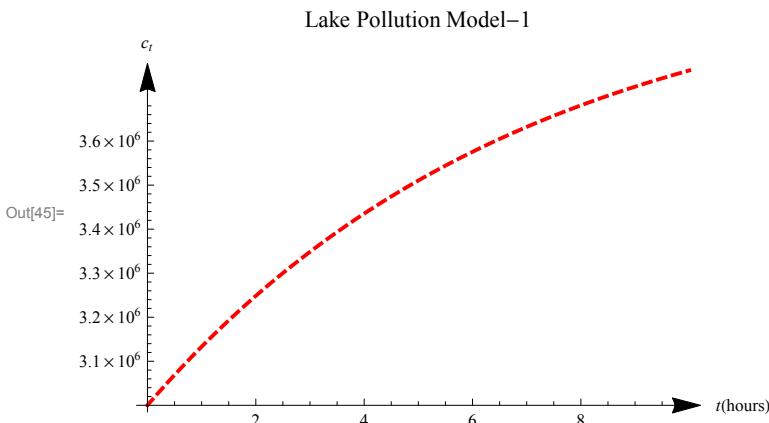


# PRACTICAL 1

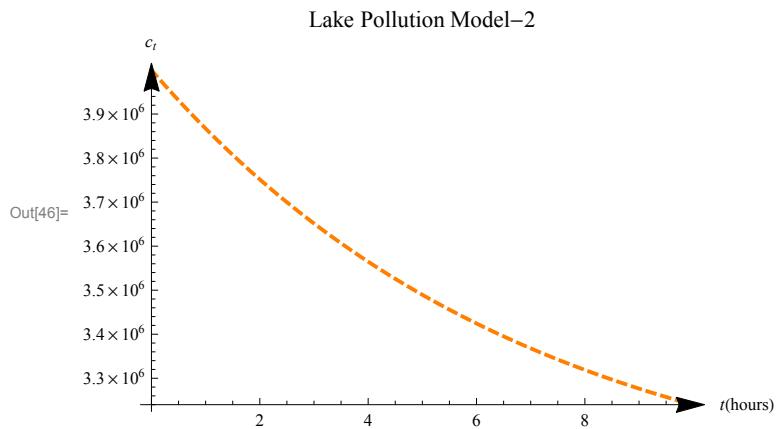
## (i) Lake Pollution Model

```
In[37]:= DE = C'[t] == F/V Cin - F/V C[t]
soln = DSolve[{DE, C[0] == c0}, C[t], t]
soln /. {cin -> 4*10^6, V -> 28*10^6, F -> 4*10^6, c0 -> 3*10^6}
Out[37]= C'[t] == -F C[t]/V + F C[in]/V
Out[38]= \{ \{ C[t] \rightarrow e^{-F t / V} (c0 - cin + e^{F t / V} cin) \} \}
Out[39]= \{ \{ C[t] \rightarrow e^{-t/7} (-1000000 + 4000000 e^{t/7}) \} \}

Plot[
Evaluate[C[t] /. soln /. {cin -> 4*10^6, V -> 28*10^6, F -> 4*10^6, c0 -> 3*10^6}],
{t, 0, 10}, AxesLabel \rightarrow {t[hours], C[t]}, AxesStyle \rightarrow Arrowheads[{0, 0.05}],
PlotStyle \rightarrow {Red, Thick, Dashed}, PlotLabel \rightarrow "Lake Pollution Model-1"]
```

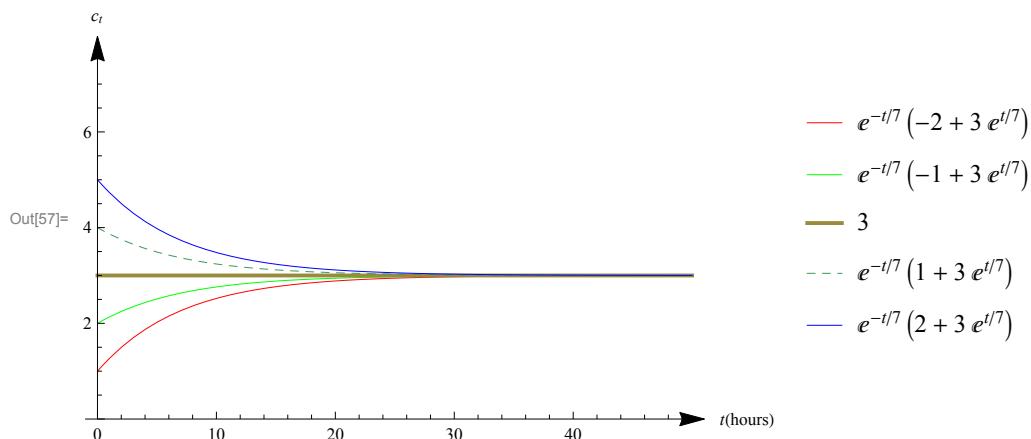


```
In[46]:= Plot[Evaluate[C[t] /. soln /. {cin → 3 * 106, v → 28 * 106, F → 4 * 106, c0 → 4 * 106}], {t, 0, 10}, AxesLabel → {t[hours], C[t]}, AxesStyle → Arrowheads[{0, 0.05}], PlotStyle → {Orange, Thick, Dashed}, PlotLabel → "Lake Pollution Model-2"]
```



```
In[54]:= DE = C'[t] == -F/V cin - F/V C[t]
soln = DSolve[{DE, C[0] == c0}, C[t], t]
soln /. {cin → 3, V → 28, F → 4, c0 → Range[1, 5]}
Plot[Evaluate[C[t] /. soln /. {cin → 3, V → 28, F → 4, c0 → Range[1, 5]}],
{t, 0, 50}, AxesLabel → {t[hours], C[t]}, PlotRange → {0, 8},
AxesStyle → Arrowheads[{0, 0.04}], PlotLegends → "Expressions",
PlotStyle → {Red, Green, Thick, Dashed, Blue}]
```

```
Out[54]= C'[t] == -F C[t]/V + F cin/V
Out[55]= {C[t] → e-F t/V (c0 - cin + eF t/V cin)}
Out[56]= {C[t] → {e-t/7 (-2 + 3 et/7), e-t/7 (-1 + 3 et/7), 3, e-t/7 (1 + 3 et/7), e-t/7 (2 + 3 et/7)}}
```



**Interpretation of Graph** :- As the time increase the concentration of pollutants in the lake will approach the cocentration of the polluted water entering the lake [  $C[t] = c_{in}$ ] Here  $c_0 = c_{in}$  then the level of pollution in the lake decreases monotonically to  $c_{in}$ . If  $c_0 < c_{in}$  then the level of pollution in the lake increase monotonically to  $c_{in}$ .

## (ii) Case of a single cold pill and a course of cold pills

Example :- Solve drug single cold pill model and plot its graph

```
In[108]:= eqn1 = x'[t] == -k1 x[t];
eqn2 = y'[t] == k1 x[t] - k2 y[t];
sol = DSolve[{eqn1, eqn2, x[0] == 1, y[0] == 0}, {x[t], y[t]}, t]
solx1 = x[t] /. sol /. {k1 → 1.3860, k2 → 0.1386}
solx2 = x[t] /. sol /. {k1 → 0.6931, k2 → 0.0231}
solx3 = x[t] /. sol /. {k1 → 0.6931, k2 → 0.08}

soly1 = y[t] /. sol /. {k1 → 1.3860, k2 → 0.1386}
soly2 = y[t] /. sol /. {k1 → 0.6931, k2 → 0.0231}
soly3 = y[t] /. sol /. {k1 → 0.6931, k2 → 0.08}

Plot[{solx1, solx2, solx3}, {t, 0, 15},
PlotStyle → {Thickness[0.02], {Red, Thickness[0.01]}, Thick},
Epilog → Inset[Framed[GI-Tract]], PlotLegends → "Expressions"]
Plot[{soly1, soly2, soly3}, {t, 0, 15},
PlotStyle → {Thickness[0.02], {Red, Thickness[0.01]}, Thick},
Epilog → Inset[Framed[Bloodstream]], PlotLegends → "Expressions"]
```

$$\text{Out[110]= } \left\{ x[t] \rightarrow e^{-k_1 t}, y[t] \rightarrow -\frac{e^{-k_1 t-k_2 t} \left(-e^{k_1 t} + e^{k_2 t}\right) k_1}{k_1 - k_2} \right\}$$

$$\text{Out[111]= } \{e^{-1.386 t}\}$$

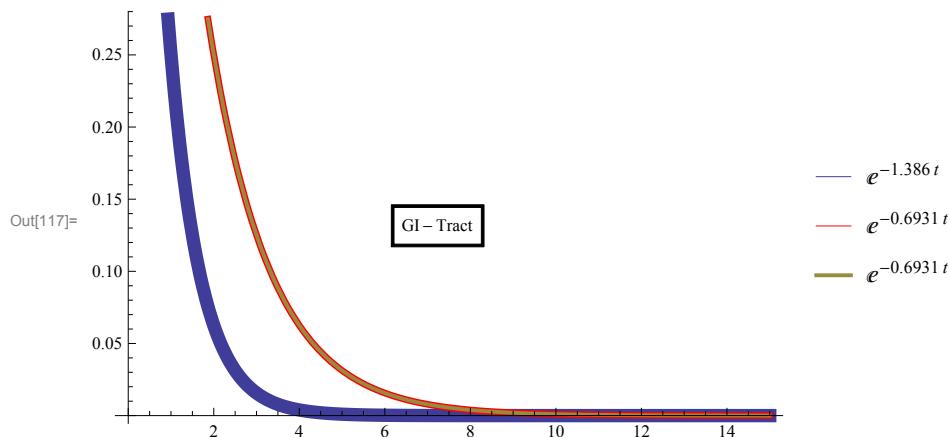
$$\text{Out[112]= } \{e^{-0.6931 t}\}$$

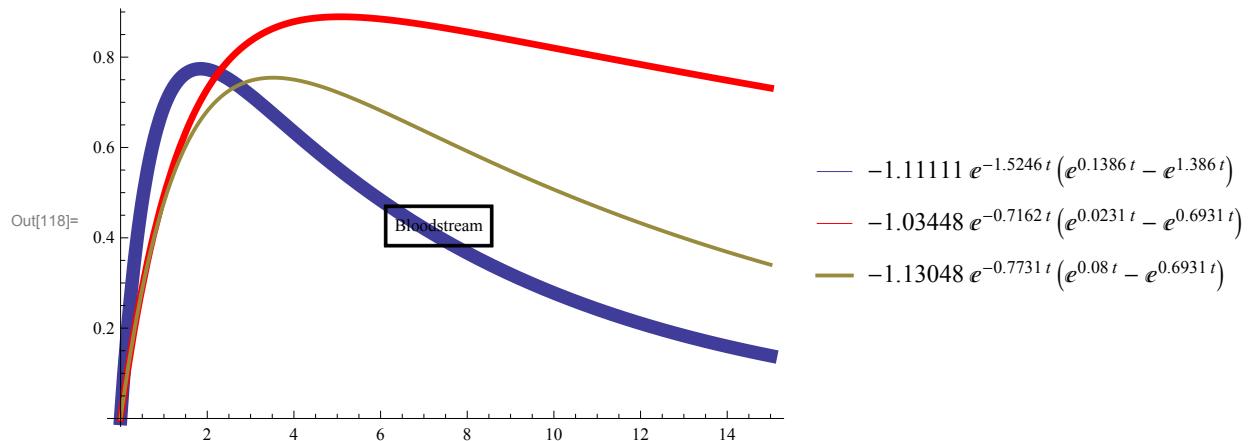
$$\text{Out[113]= } \{e^{-0.6931 t}\}$$

$$\text{Out[114]= } \{-1.11111 e^{-1.5246 t} (e^{0.1386 t} - e^{1.386 t})\}$$

$$\text{Out[115]= } \{-1.03448 e^{-0.7162 t} (e^{0.0231 t} - e^{0.6931 t})\}$$

$$\text{Out[116]= } \{-1.13048 e^{-0.7731 t} (e^{0.08 t} - e^{0.6931 t})\}$$





### Example :- Solve course of cold pills model and plot its graph.

```
In[119]:= Clear[eqn1, eqn2, sol]
eqn1 = x'[t] == I - k1 x[t]
eqn2 = y'[t] == k1 x[t] - k2 y[t]
sol = DSolve[{eqn1, eqn2, x[0] == 0, y[0] == 0}, {x[t], y[t]}, t]
sol /. {k1 → {1.3860, 0.6931}, k2 → {0.1386, 0.0231}, I → 1}

Out[120]= x'[t] == I - k1 x[t]

Out[121]= y'[t] == k1 x[t] - k2 y[t]

Out[122]= {x[t] → I e^-t k1 (-1 + e^t k1) / k1, y[t] → I e^-t k1-t k2 (e^t k1 k1 - e^t k1+t k2 k1 - e^t k2 k2 + e^t k1+t k2 k2) / (k2 (-k1 + k2))}

Out[123]= {x[t] → {0.721501 e^-1.386 t (-1 + e^1.386 t), 1.44279 e^-0.6931 t (-1 + e^0.6931 t)}, y[t] → {-5.78404 e^-1.5246 t (-0.1386 e^0.1386 t + 1.386 e^1.386 t - 1.2474 e^1.5246 t), -64.612 e^-0.7162 t (-0.0231 e^0.0231 t + 0.6931 e^0.6931 t - 0.67 e^0.7162 t)}]}
```