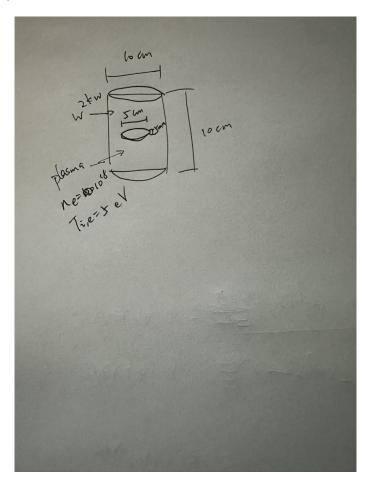
NPRE 321 HW 6

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Due: Nov20 Edit: November $20,\,2024$

1. a)



b)
$$S = \pi dh + 2(\frac{1}{2}d)^2\pi$$

$$= 0.01\pi + 0.005\pi$$

$$= 0.015\pi \text{ m}^{2}$$

$$\Gamma_{q} = \frac{P}{S}$$

$$= \frac{2000}{0.015\pi}$$

$$= 42441.3 \text{ W/m}^{2}$$

 $=0.01\pi + 0.005\pi$

c)

$$S_{Cu} = \pi dh + 2(\frac{1}{2}d)^{2}\pi$$

$$= 0.00025\pi + 0.00125\pi$$

$$= 0.0015\pi$$

$$V = 2(\frac{1}{2}d)^{2}\pi \times h$$

$$= 1.963495 \times 10^{-5} \text{ m}^{3}$$

$$Q = \Gamma_{q}tS$$

$$= 42441.3 \times (5 \times 60) \times 0.0015\pi$$

$$= 6 \times 10^{4} \text{ J}$$

$$Q = mC_{p}\Delta T$$

$$\Delta T = \frac{Q}{mC_{p}}$$

$$= \frac{Q}{\rho VC_{p}}$$

$$= \frac{6 \times 10^{4}}{8.9 \times 10^{3} \times 1.963495 \times 10^{-5} \times 390}$$

$$= 880.373 \text{ K}$$

d) Since for 5 min, the copper sheet is heated up 880 kelvine. Therefore, it will be mealting down if running on higher power or longer time. However, as we covered in Practicum 2, we chould run coolants like water to cool down the copper sheet.

$$\begin{split} \Gamma &= \bar{n}\bar{v} \\ &= \bar{n} \times \sqrt{\frac{8T_i}{\pi m_i}} \\ &= 10^{18} \times \sqrt{\frac{8 \times 5 \times 1.602 \times 10^{-19}}{\pi \times 34.948 \times 1.66 \times 10^{-27}}} \\ &= 5.929 \times 10^{21} \text{ m}^{-2} \text{s}^{-1} \end{split}$$

f)

$$\begin{split} \dot{n} &= \Gamma \times t \\ &= 5.929 \times 10^{21} \times 60^2 \\ &= 2.135 \times 10^{25} \text{ m}^{-2} \end{split}$$

g)

$$E_i = 6 \times T_i$$

= 30 eV > 3.5 eV

Thus, yes, the ion would be able to suppter copper away.

$$Y(100) = 0.5$$

$$Y(3.5) = 0$$
Assume linear locally
$$Y(30) = 0.1373$$

$$S_{Cu} = \pi dh + 2(\frac{1}{2}d)^{2}\pi$$

$$= 0.00025\pi + 0.00125\pi$$

$$= 0.0015\pi$$

$$\dot{n} = 2.135 \times 10^{25}$$

$$N = S_{Cu} \cdot \dot{n}$$

$$= 1.006 \times 10^{23}$$

$$N_{Cu} = N \times Y$$

$$= 1.381 \times 10^{22}$$

$$M_{\text{Remove},Cu} = N_{Cu} \times m_{Cu}$$

$$= 1.381 \times 10^{22} \times 63.546 \times 1.66 \times 10^{-27}$$

$$= 0.001457 \text{kg}$$

$$M_{\text{Target},Cu} = \rho V$$

$$= 8.9 \times 10^{3} \times 1.963495 \times 10^{-5} \text{ m}^{3}$$

2. a)

$$D+D \rightarrow T+p$$

$$D+D \rightarrow {}^3He+n$$

= 0.194751 kg

 $\frac{M_{\mathrm{Remove},Cu}}{}=0.8337\%$

 $M_{\text{Target},Cu}$

b)

$$E = mc^{2}$$

$$= 2(m_{D})c^{2}$$

$$= 2 \times (2.014102 \times 1.66 \times 10^{-27})(3 \times 10^{8})^{2}$$

$$= 6.01814 \times 10^{-10} \text{ J}$$

$$D+D \rightarrow T+p$$

$$\begin{split} E &= (3.016049 + 1.00727647) \times 1.66 \times 10^{-27} \times (3 \times 10^8)^2 \\ &= 6.01085 \times 10^{-10} \text{ J} \\ \Delta E &= 6.01814 \times 10^{-10} \text{ J} - 6.01085 \times 10^{-10} \text{ J} \\ &= 7.288 \times 10^{-13} \text{ J} \\ &= 4.5495 \times 10^6 \text{ eV} \end{split}$$

 $D+D \rightarrow {}^{3}He+n$

$$\begin{split} E &= (3.016029 + 1.008665) \times 1.66 \times 10^{-27} \times (3 \times 10^8)^2 \\ &= 6.01289 \times 10^{-10} \text{ J} \\ \Delta E &= 6.01814 \times 10^{-10} - 6.01289 \times 10^{-10} \\ &= 5.24394 \times 10^{-13} \text{ J} \\ &= 3.273 \times 10^6 \text{ eV} \end{split}$$

c) $D+D\to T+p$

$$\begin{split} \frac{T}{2D} &= \frac{3.016049}{2\times 2.014102} \\ &= 74.8733\% \\ \frac{p}{2D} &= \frac{1.00727647}{2\times 2.014102} \\ &= 25.0056\% \end{split}$$

 $D+D \rightarrow {}^{3}He+n$

$$\begin{split} \frac{^3He}{2D} &= \frac{3.016029}{2\times 2.014102} \\ &= 74.8728\% \\ \frac{n}{2D} &= \frac{1.00727647}{2\times 2.014102} \\ &= 25.0401\% \end{split}$$