

Carbon Nitride Supported High-Loading Fe Single-Atom Catalyst for Activating of Peroxymonosulfate to Generate $^1\text{O}_2$ with 100% Selectivity.

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Abstract

Singlet oxygen ($^1\text{O}_2$) is an excellent active species for the selective degradation of organic pollutions. However, it is difficult to achieve high efficiency and selectivity for the generation of $^1\text{O}_2$. In this work, we develop a graphitic carbon nitride supported Fe single-atoms catalyst (Fe_1/CN) containing highly uniform Fe-N₄ active sites with a high Fe loading of 11.2 wt%. The Fe_1/CN achieves generation of 100% $^1\text{O}_2$ by activating peroxymonosulfate (PMS), which shows an ultrahigh p-chlorophenol degradation efficiency. Density functional theory calculations results demonstrate that in contrast to Co and Ni single-atom sites, the Fe-N₄ sites in Fe_1/CN adsorb the terminal O of PMS, which can facilitate the oxidization of PMS to form $\text{SO}_5^{\bullet-}$, and thereafter efficiently generate $^1\text{O}_2$ with 100% selectivity. In addition, the Fe_1/CN exhibits strong resistance to inorganic ions, natural organic matter, and pH value during the degradation of organic pollutants in the presence of PMS. This work develops a novel catalyst for the 100% selective production of $^1\text{O}_2$ for highly selective and efficient degradation of pollutants.