Scherrer's Quantum Mechanics Problems

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Chapter 1

The Origin Of Quantum Mechanics

- 1. Assume that a human body emits blackbody radiation at the standard body temperature.
 - (a) Estimate how much energy is radiated by the body in one hour.

The power emmitted by a blackbody is given by:

$$P = \sigma A T^4$$

where A is the surface area of the body, T is the temperature and σ is the Stefan-Boltzmann constant. Therefore the energy radiated by a body in a given time interval Δt :

$$E = \sigma A T^4 \Delta t$$

The surface area of the human body is approximately $2m^2$, the average body temperature is $36.1^{\circ}C = 309.25K$, and there are 3600s in an hour. Therefore:

$$E_{\text{hour}} = \left(5.67 * 10^{-8} \frac{\text{J}}{\text{s m}^2 \text{K}^4}\right) (2\text{m}^2) (309.25\text{K})^4 (3600\text{s})$$

= 3733kJ

(b) At what wavelength does this radiation reach a maximum

The formula for the maximum wavelength is:

$$\lambda_{\text{peak}} = \frac{w}{T}$$

where $w = 2.90 * 10^{-3}$ m K and T = 309.25K as before. Therefore the maximum wavelength is:

$$\lambda_{\text{peak}} = \frac{2.9 * 10^{-3} \text{m K}}{309.25 \text{K}} = 9.37 * 10^{-6} \text{m}$$

2. A distant red star is observed to have a blackbody spectrum with a maximum at a wavelength of $3500\text{\AA}[1\text{Å}=10^{-10}\text{ m}]$. What is the temperature of the star?

Inverting the formula from the pervious question:

$$T = \frac{w}{\lambda_{\text{peak}}}$$

giving:

$$T = \frac{2.9 * 10^{-3} \text{m K}}{3500 * 10^{-10} \text{m}} = 51428 \text{K}$$

- 3. The universe is filled with blackbody radiation at a temperature of 2.7K left over from the Big Bang. [This radiation was disvoered in 1965 by Bell Laboratory scientists who thought at one point that they were seeing interference from pigeon droppings on their microwave reciever.
 - (a) What is the total energy density of this radiation?

The energy density of the radiation is given by:

$$\rho = aT^4$$

where $a = 7.56 * 10^{-16} \frac{\text{J}}{\text{m}^{3}\text{K}^{4}}$. Therefore:

$$\rho = 7.56 * 10^{-16} \frac{J}{\text{m}^3 \text{K}^4} * (2.7 \text{K})^4 = 4.01 * 10^{-14} \frac{J}{\text{m}^3}$$

(b) What is the total energy density with wavelengths between 1mm and 1.01mm? Is the Rayleigh-Jeans formula a good approximation at these wavelengths?

Chapter 2

Math Interlude A: Complex Numbers and Linear Operators

1. Eavluate all of the following and express all of your final answers in the form a+bi:

(a)
$$i(2-3i)(3+5i)$$

$$i(2-3i)(3+5i) = i(6+10i-9i+15)$$

= $i(21+i)$
= $-1+21i$

(b)
$$i/i - 1$$

$$i/i - 1 = e^{i\pi/2}/(\sqrt{2}e^{3i\pi/4})$$

= $\frac{1}{\sqrt{2}}e^{-i\pi/4}$

(c)
$$(1+i)^{30}$$

$$(1+i)^{30} = (\sqrt{2}e^{i\pi/4})^{30}$$
$$= 2^{15}e^{15i\pi/2}$$
$$= 2^{15}e^{3i\pi/2}$$
$$= -2^{15}i$$

2.

3.

4. Suppose that a complex number z has the property that $z^*=z$. What does this indicate about z?

This indicates that z is a real number.

5. Reduce i^i to a real number

$$i^i = (e^{i\pi/2})^i = e^{-\pi/2}$$

6. What is wrong with the following argument?

$$\sqrt{\frac{1}{-1}} = \frac{\sqrt{1}}{\sqrt{-1}}$$

$$\sqrt{-1} = \frac{1}{i}$$

$$i = \frac{1}{i}$$

$$(i)(i) = 1$$

$$-1 = 1$$

The first line is a false equivalence.

$$\sqrt{\frac{1}{-1}} = \sqrt{-1} = i = e^{i\pi/2}$$

and

$$\frac{\sqrt{1}}{\sqrt{-1}} = \frac{1}{\sqrt{-1}} = e^{-i\pi/2}$$