Total No. of Questions: 6

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## Faculty of Engineering

End Sem (Odd) Examination Dec-2019 AU3ES09/FT3ES09/ME3ES09 Engineering Materials Programme: B.Tech. Branch/Specialisation: AU/FT/ME

**Duration: 3 Hrs. Maximum Marks: 60** 

	-	estions are compulsory. Internal should be written in full instead of	choices, if any, are indicated. Answers of only a, b, c or d.	of
Q.1	i.	Which of the following is the most common instrument for photographic recording of diffraction patterns?  (a) Debye-Scherrer powder camera		
		(b) Gamma camera		
		(c) Geiger tube		
		(d) Scintillation counter		
	ii.	Gold has crystal struct	ture.	1
		•	c) BCC (d) SCC	
	iii.	is usually defined as re	resistance of a material to penetration.	1
		-	b) Compressive strength	
			d) Fatigue strength	
	iv.	Lack of ductility is		1
		(a) Malleability (b)	b) Porosity	
		(c) Stiffness (d	d) Brittleness	
	v.	Process decompose	e the martensite into a ferrite-	1
		cementite mixture.		
		(a) Annealing (b	b) Normalising	
		(c) Tempering (c	d) None of these	
	vi.	icomprise of alternate layers of ferrite and cementite in stee		
		(a) Austenite (b)	b) Pearlite	
		(c) Ledeburite (c	d) None of these	
	vii.	Brass is an alloy of		1
			b) Copper and Tin	
		(c) Copper and Zinc (d	d) Nickel and Zinc	
			D (T	$\sim$

P.T.O.

	viii.	German silver is an alloy consisting of		1
		(a) Cu, Al, Ni	(b) Cu, Pb	
		(c) Cr, Cu	(d) Cu, Zn, Ni	
	ix.			1
		(a) Argillaceous	(b) Calcarious	
		(c) Both (a) and (b)	(d) None of these	
	х.	PVC stand for		1
		(a) Polythene vinyl chloride	(b) Phosphorus-vanadium-carbide	
		(c) Polyvinyl chloride	(d) Plastic very compact	
Q.2	i.	What are miller indices? Write indices for planes.	e the procedure for finding out miller	3
	ii.		and FCC crystal structure. Discuss the on number and atomic packing factor	7
OR	iii.	Classify crystal imperfection and explain point defect in detail.		
Q.3	i.	Draw labelled stress-strain diagram for ductile material, brittle <b>4</b> material, ceramic and polymer		4
	ii.	Explain the mechanism of cre	ep failure and fatigue failure.	6
OR	iii.	Explain and compare Izod and Charpy test. 6		
Q.4 i. Draw Bi-Cd Eutectic phase diagram and any one composition other than eutemperature		•	-	3
	ii.	Classify various heat treatmed doing full annealing and norm	ent processes. Explain the process of palising.	7
OR	iii.	Draw neat and clean Iron-iron carbide phase diagram indicating various temperature, carbon-content and various zones. Write three important reactions also.		
Q.5	i. ii.	Differentiate between white an (a) Classify steel with examp (b) Explain Maraging Steapplication)	•	<b>4 6</b>

OR	iii.	Write composition, properties and application of two important Aluminium based alloys.	6
Q.6	i.	Attempt any two:  Describe the structure, properties and application of the following engineering thermoplastic polymers  (a) Acrylonitrile Butadiene styrene (ABS)	5
	ii. iii.	(b) Nylons What are the General properties and application of ceramic materials? Write brief note on: (a) Laminates (b) Clad metals	5

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## **Marking Scheme**

## AU3ES09/FT3ES09/ME3ES09 Engineering Materials

Q.1	i.	Which of the following is the most common instrument for photographic recording of diffraction patterns?			
		(a) Debye-Scherrer powder camera			
	ii.	Gold has crystal structure.		1	
		(b) FCC			
	iii.	is usually defined as resistance of a mater	ial to penetration.	1	
		(c) Hardness			
	iv.	Lack of ductility is		1	
		(d) Brittleness			
	v.	Process decompose the martensite	into a ferrite-	1	
		cementite mixture.			
		(c) Tempering			
	vi.	comprise of alternate layers of ferrite and cer	mentite in steel.	1	
		(b) Pearlite			
	vii.	Brass is an alloy of		1	
		(c) Copper and Zinc			
	viii.	German silver is an alloy consisting of		1	
		(d) Cu, Zn, Ni			
	ix.	In rocks clay or alumina is the main constituent.			
		(a) Argillaceous			
	х.	PVC stand for		1	
		(c) Polyvinyl chloride			
Q.2	i.	Definition of miller indices	1 mark	3	
		Procedure for finding out miller indices for planes	2 marks		
	ii.	Unit cell diagram for BCC and FCC	1 mark	7	
		Number of atoms	2 marks		
		Coordination number	2 marks		
		Atomic packing factor	2 marks		
OR	iii.	Crystal imperfection		7	
		Classification	2 marks		
		Explanation of five types of point defect with diagr	am		
		-	5 marks		

Q.3	i.	Stress-strain diagram for ductile material, brittle and polymer	material, ceramic	4
		1 mark for each	(1 mark * 4)	
	ii.	Mechanism of creep failure	,	6
		Explanation	2 marks	
		Diagram	1 mark	
		Mechanism of fatigue failure		
		Explanation	2 marks	
		Diagram	1 mark	
OR	iii.	Izod Explanation	2 marks	6
		Diagram	1 mark	
		Charpy test Explanation	2 marks	
		Diagram	1 mark	
		2 Mg. um	1 IIIWIII	
Q.4	i.	Bi-Cd Eutectic phase diagram	2 marks	3
		Grain formation at four to five temperature	1 mark	
	ii.	Classification of heat treatment processes	3 marks	7
		Process of doing full annealing	2 marks	
		Process of normalising	2 marks	
OR	iii.	Iron-iron carbide phase diagram indicating		7
		Correct zones	2 marks	
		Correct temperatures	1 mark	
		Correct compositions	1 mark	
		Three important reactions	3 marks	
		1		
Q.5	i.	Differentiate between white and grey cast-iron.		4
		1 mark for each difference	(1 mark * 4)	
	ii.	(a) Classify steel with examples.	3 marks	6
		(b) Maraging Steel		
		Composition	1 mark	
		Properties	1 mark	
OR	:::	Application Two important Aluminium based allows	1 mark	6
OK	iii.	Two important Aluminium based alloys	2	U
		Composition Properties	2 marks 2 marks	
		Application	2 marks	
			- 111W111U	
Q.6		Attempt any two:		

i.	Structure, properties and application of thermoplastic polymers		5
	(a) Acrylonitrile Butadiene styrene (ABS)	2.5 marks	
	(b) Nylons	2.5 marks	
ii.	General properties of ceramic materials		5
	At least three properties 1 mark for each	3 marks	
	Application of ceramic materials		
	At least four applications 0.5 mark for each	2 marks	
iii.	Write brief note on:		5
	(a) Laminates	2.5 marks	
	(b) Clad metals	2.5 marks	

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