[4]

Q.5	i.	Enlist the air-standard assumptions made for analysis of air-	2
		standard cycles. (at least four assumption)	
	ii.	Air enters the compressor of an air-standard Brayton cycle at 100	8
		kPa, 300 K with a volumetric flow ratio of 5 m ³ /s. The compressor	
		pressure ratio is 10 and turbine inlet temperature is 1400 K.	
		Determine:	
		(a) Thermal efficiency of cycle	
		(b) Back work ratio	
		(c) Net power developed in kW.	

OR iii. Compare Otto, diesel and dual cycle with help of p-v & T-s charts: 8

(a) For same compression ratio

(b) For same maximum temperature & maximum pressure.

Q.6 Attempt any two:

i. Express Dalton's law of partial pressure. Does this law hold 5 exactly for ideal gas mixtures?

ii. What do you understand by Maxwell relations? Also derive the 5 Maxwell relations.

iii. A vessel of volume 0.4 m³ contains 0.45 kg f carbon monoxide and 1 kg of air, at 15 °C. Calculate the partial pressure of each constituents and the total pressure in the vessel. The air contains 23.3% oxygen and 76.6% nitrogen by mass. Take the molar masses of carbon monoxide, oxygen and nitrogen as 28, 32 and 28 kg/k mol respectively.

Total No. of Questions: 6

Total No. of Printed Pages:4

Enrollment No.....



Faculty of Engineering End Sem (Odd) Examination Dec-2018 AU3CO04 / ET3CO04 / ME3CO04

AU3CO04 / FT3CO04 / ME3CO04 **Engineering Thermodynamics** Branch/Specialisation: AU/FT/ME Programme: B.Tech. **Duration: 3 Hrs. Maximum Marks: 60** Note: (i) All questions are compulsory. Internal choices, if any, are indicated. Answers of Q.1 (MCQs) should be written in full instead of only a, b, c or d. (ii) Use of steam table should be allowed. i. In a reversible adiabatic process, the work transfer is equal to 1 Q.1 (a) Decrease in enthalpy (b) Decrease in internal energy (c) Heat transfer (d) The product of pressure & change in volume ii. During a heating process, the temperature of an object rises by 20 1 °C. This temperature rise is equivalent to (a) 20 °F (b) 52 °F (c) 36 °F (d) 68 °F iii. In a thermal power plant, turbine does 10,000 kJ of work, pump 1 consumes 10 kJ of work. The boiler receives 30,000 kJ of heat. Thermal efficiency of the plant is (a) 27 % (b) 33.334% (c) 35 % (d) 40 % iv. A refrigerator & heat pump operates between same temperature 1 limits. If the COP of the refrigerator is four, what is the COP of heat pump? (a) 3 (b) 3.4 (c) 5 (d) 4v. The total enthalpy of steam at 10 bar is 2000 kJ/kg. The condition 1 of steam is (a) Wet (b) Dry & saturated (c) Superheated (d) None of these

P.T.O.

	V1.	Internal latent heat of steam is given by	1	
	vii.	(a) $h_{fg} + pv_g$ (b) $h_g - pv_g$ (c) $h_{sup} - pv_g$ (d) $h_{fg} - pv_g$ The efficiency of air standard Diesel cycle depends on	1	
		(a) Cut-off ratio in cycle(b) Compression ratio in the cycle(c) Both (a) and (b)		
		(d) None of these		
	viii.	For given compression, among Otto, Diesel and Dual cycles	1	
		(a) Diesel cycle is most efficient (b) Otto cycle is most efficient		
		(b) Otto cycle is most efficient(c) Dual cycle is most efficient		
		(c) None of these		
	ix.	Which one of the following is true statement	1	
	IX.	(a) $m = n M$ (b) $m = \sum x_i n_i$ (c) $m = \frac{n}{M}$ (d) None of these	_	
	х.	Partial pressure of a gas in a gas mixture is defined as	1	
		(a) Pressure exerted by a gas at mixture volume		
		(b) Pressure exerted by a gas at mixture temperature		
		(c) Pressure exerted by a gas at mixture volume & temperature		
		(d) None of these		
Q.2	i.	What are the limitations of the First Law of Thermodynamics?	2	
	ii.	A centrifugal pump delivers 50 kg of water per second. The inlet	8	
		and outlet pressure are 1 bar & 4.2 bar respectively. The suction is		
		2.2 m below the centre of pump and delivery is 8.5 m above the		
		same level. The suction and delivery pipe diameters are 20 cm &		
		10 cm respectively. Determine the capacity of electric motor to		
OR	iii.	run the pump. 0.2 m ³ of a mixture of a fuel and air at 1.2 bar and 60 °C is	8	
OK	111.	compressed till its pressure becomes 12 bar and temperature	đ	
		270 °C. Then it is ignited suddenly at constant volume and its		
		pressure becomes twice the pressure at the end of compression.		
		Find the maximum temperature reached and heat transfer during		
		compression process. Assume the mixture as perfect gas.		

Q.3	i.	Define refrigerator and heat pump with the help of line diagram.	2
	ii.	A heat pump maintains a space at 22 °C on a day, when outdoor	8
		air temperature is 0 °C. The heating requirement of space is	
		100,000 kJ/h and power consumed by the pump is 5 kW.	
		Calculate the rate at which heat is extracted from outside air and	
		COP of the heat pump. Also calculate the maximum COP. Does it	
		violate the second law of thermodynamics?	

- OR iii. A reversible refrigerator is used to maintain a space at the temperature of 0 °C, when it rejects heat to surroundings at 27 °C, If the heat removal rate from refrigerator is 90 MJ/h, determine the COP of the system.

 If the required input to run the refrigerator is supplied by a reversible engine which receives heat at 400 °C and rejects heat to surroundings, determine the overall COP of the system.
- Q.4 i. Define the terms: sub-cooled liquid, saturated liquid, saturated vapour and superheated vapours.
 - ii. A certain boiler generates 1000 kg of stem per hour at 10 bar and 0.97 dry. The steam is then taken to a super heater and is heated to a temperature of 280 °C keeping pressure constant. The feed water is available at 30 °C. Find
 - (a) Heat supplied per hour in boiler
 - (b) Heat supplied per hour in super heater
 - (c) Internal energy of super heated steam
 - (d) Change in internal energy of steam
- OR iii. A combined separating and throttling calorimeter was used to determine the dryness fraction of steam through steam main at a pressure of 9 bar. The pressure and temperature of steam after throttling were 1.25 bar and 115 °C respectively. The mass of steam condensed after throttling was 2.2 kg and mass of water collected in separator was 0.20 kg. Estimate the dryness fraction of steam in main.

Take specific heat for super heated steam as 2.1 kJ/kgK.

P.T.O.

Marking Scheme AU3CO04 / FT3CO04 / ME3CO04

Engineering Thermodynamics

Q.1	i.	In a reversible adiabatic process, the work transfer is equal to	1
	ii.	(b) Decrease in internal energy During a heating process, the temperature of an object rises by 20 °C. This temperature rise is equivalent to (d) 68 °F	1
	iii.	In a thermal power plant, turbine does 10,000 kJ of work, pump consumes 10 kJ of work. The boiler receives 30,000 kJ of heat. Thermal efficiency of the plant is (b) 33.334%	1
	iv.	A refrigerator & heat pump operates between same temperature limits. If the COP of the refrigerator is four, what is the COP of heat pump?	1
	v.	(c) 5The total enthalpy of steam at 10 bar is 2000 kJ/kg. The condition of steam is(a) Wet	1
	vi.	Internal latent heat of steam is given by (d) h _{fg} - pv _g	1
	vii.	The efficiency of air standard Diesel cycle depends on (c) Both (a) and (b)	1
	viii.	For given compression, among Otto, Diesel and Dual cycles (b) Otto cycle is most efficient	1
	ix.	Which one of the following is true statement (a) $m = n M$	1
	х.	Partial pressure of a gas in a gas mixture is defined as (c) Pressure exerted by a gas at mixture volume & temperature	1
Q.2	i.	At least four limitations of the First Law of Thermodynamics 0.5 marks for each (0.5 mark *4)	2
	ii.	0.5 marks for each (0.5 mark *4) Fully labelled diagram 2 marks Capacity of electric motor to run the pump = -22.2 kW 6 marks	8
		Omarks	

OR	iii.	Fully labelled pv diagram n = 1.27 R = 0.3 kJ/kgK m = 0.24 kg	2 marks	8
		Maximum temperature = 813 °C Cn = -0.36 kJ/kgK	3 marks	
		Heat transfer during comp. process = -14.2 kJ	3 marks	
Q.3	i.	Definition with line diagram of refrigerator	1 mark	2
		Definition with line diagram of heat pump	1 mark	
	ii.	Fully labelled block diagram	2 marks	8
		Rate of heat extraction = 22.78 kW	2 marks	
		COP hp = 5.56	1 mark	
		COP hp max. = 13.4	1 mark	
		As COP hp max. > COP hp, hence it does not viola		
OD			2 marks	•
OR	iii.	Fully labelled block diagram	2 marks	8
		$COP_R = 10.11$ $W_{in R} = W_{out HE} = 2.473 \text{ kW}$	2 marks	
		$\eta_{\text{rev}} = 55.42 \%$		
		$Q_h = 4.462 \text{ kW}$	2 marks	
		$COP_{overall} = \frac{\text{sole effect from system}}{\text{energy input to system}} = \frac{\text{Refrigerating effect}}{\text{heat input rate}}$		
		energy input to system heat input rate		
		$COP_{overall} = 5.60$	2 marks	
Q.4	i.	Define the terms: sub-cooled liquid, saturated l	iquid, saturated	2
		vapour and superheated vapours.		
		0.5 mark for each term	(0.5 mark *4)	
	ii.	For each correct answer give 2 marks	(2 marks *4)	8
		(a) heat supplied per hour in boiler = 2591.04 MJ/h	1	
		(b) heat supplied per hour in super heater = 406.26	MJ/h	
		(c) internal energy of super heated steam = 2759.89	9 kJ/kg	
		(d) change in internal energy of steam = 2635.4 kJ/m	=	
OR	iii.	$x_1 = 0.9167$	1 mark	8
		$x_2 = 0.9657$	4 marks	
		$x_{\text{final}} = 0.885$	3 marks	
		A IIIIai — 0.000	Jiiuixo	
Q.5	i.	Air-standard assumptions made for analysis of air-	standard cycles.	2
		(at least four assumption)		
		1 mark for each	(1 mark *4)	
			,	

			2 1	0
	ii.	Fully labelled pv diagram	2 marks	8
		(a) thermal efficiency of cycle = 48.2%	2 marks	
		$T_2 = 592.2 \text{ K},$		
		$T_4 = 721.1 \text{ K}$		
		(b) back work ratio = 0.417	2 marks	
		$density = 1.161 \text{ kg/m}^3$		
		mass flow rate = 5.8 kg/s		
		(c) net power developed = 2309.34 kW	2 marks	
OR	iii.	Compare Otto, diesel and dual cycle with help of p-	v & T-s charts:	8
		(a) For same compression ratio	4 marks	
		(b) For same maximum temperature & maximum p	ressure.	
			4 marks	
Q.6		Attempt any two:		
	i.	Express Dalton's law	3 marks	5
		Does this law hold exactly	2 marks	
	ii.	What do you understand	2 marks	5
		derive the Maxwell	3 marks	
	iii.	$P_{co} = 0.9623 \text{ bar}$	1 mark	5
		$P_{O2} = 0.4345 \text{ bar}$	1 mark	
		$P_{N2} = 1.64 \text{ bar}$	1 mark	
		$P_{\text{tot}} = 3.03 \text{ bar}$	2 marks	
		100 2.02 041	2 11141113	
