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The equilibrium state corresponds to the saturated vapor pressure curve $p = p_s(T)$, which is limited by a triple point (T_{tr}, P_{tr}) and a critical point (T_c, p_c) of the substance. (In the presence of surface tension forces at the phase interface, pressures values $p(l)$ and $p(g)$ in coexisting phases are not equal.) The Clausius-Clapeyron equation follows from Eq.

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