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in reaction $4\text{NH}_3 + 5\text{O}_2 \rightarrow 4\text{NO} + 6\text{H}_2\text{O}$, 18.0 g ammonia and 30.0 g of oxygen gas are combined. molar masses $\text{NH}_3=17.03$, $\text{NO}=30.01$. a. what is the theoretical yield of NO in grams, at 100% reaction.

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