

## ***Solutions Electrolytes And Concentration Lab***

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### **Solutions Electrolytes And Concentration Lab**

Solutions, Electrolytes, and Concentration Pre-Lab Study Questions 1. Why does an oil and vinegar salad dressing have two separate layers? This is because oil is non polar and water is polar 2. What is meant by mass/mass percent concentration of a solution? The amount of grams of solute in 100 grams of solution 3.

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### **Lab G-2 DM - Solutions Electrolytes and Concentration Pre ...**

Solutions, Electrolytes, and Concentration Report Sheet -Lab 15 Name Date Section Instructor Team Pre-Lab Study Questions 1. Why does an oil-and-vinegar salad dressing have two separate layers? 2. What is meant by the mass/mass percent concentration of a solution? 3. Why are some electrolytes strong, while others are weak? 4. What is molarity? A.

### **Solved: Solutions, Electrolytes, And Concentration Report ...**

Electrolytes are substances that dissolve by breaking into ions in solution and conduct electricity. Electrolyte solutions can conduct electricity. Electrolyte solutions can conduct electricity.

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Worksheet 5 — Solutions, Electrolytes and Concentration The name electrolyte is given to substances whose aqueous solutions contain ions, because ions are charged particles which, when they move through a solution, conduct electricity. Strong electrolytes dissociate completely into ions. This includes the strong

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In the on-campus lab, the conductivity of electrolytes is indicated by the brightness of a light bulb. 110 volts is directed through the solution and then through the light bulb. If the solution contains an adequate concentration of a strong electrolyte (like HCl or NaCl), electricity will flow through the solution and the bulb will be bright.

### **Lab 11 - Chemistry Land Intro**

A solution having less solute concentration than the solution in which it is being compared. What is a hypertonic solution? A solution having more solute concentration than the solution in which it is being compared Q.5 State whether each of the following are isotonic, hypotonic, or hypertonic: a.

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