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Sample Solution Step 1 of 2 Draw the general representation of op-amp. From the figure, it is observed that the number terminals in a single op-amp is 5. Step 2 of 2 Calculate the minimum number of terminals required on an integrated circuit package containing four op-amps (Quad op-amp). In a guad op-amp circuit, the number of op-amps is 4.

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Solving for x: $x = 2.05 \times 10^{-6} M = [OH-]$ and the pOH is 5.69. The pH is 8.31 b) conjugate acid and anion What is the pH of a 0.5 M CH 3 NH 3 Cl solution. K b (CH 3 NH 3 +) = 4.38 x 10-4 K a (CH 3 NH 2) = 2.23 x 10-11 The first thing is to determine what is in solution. In this case, it is the acid, CH 3 NH 3 +. Set up a reaction table for

Review for Solving pH Problems: Base Ionization 1)

Calculate the masss of Ba(OH)2 required to prepare 2.50 L of a 0.0600M solution of baruim hydroxide. 18.1 M H2SO4 A sample of commercial sulfuric acid is 96.4% H2SO4 by mass, and its specific gravity is 1.84.

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