Solution Biology Definition

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Solution Biology Definition

Solution. A type of homogenous mixture in which the particles of one or more substances (the solute) are distributed uniformly throughout another substance (the solvent).

Solution - Biology-Online Dictionary | Biology-Online ...

Solution Definition A solution is a homogeneous mixture of solvent and solute molecules. A solvent is a substance that dissolves another substance by pulling the molecules apart through electrochemical interactions.

Solution - Definition, Types and Examples | Biology Dictionary

Isotonic Solution Definition An isotonic solution is one that has the same osmolarity , or solute concentration, as another solution. If these two solutions are separated by a semipermeable membrane , water will flow in equal parts out of each solution and into the other.

Isotonic Solution - Definition and Examples | Biology ...

A solution is a homogeneous mixture of solvent and solute molecules. A solvent is a substance that dissolves another substance by pulling the molecules apart through electrochemical interactions. Source: Noland, George B. 1983. General Biology, 11th Edition.

* Solution (Biology) - Definition, meaning - Online ...

noun, plural: solutes. (chemistry) 1. a component of a solution: in a solution, the dissolving substance is called a solvent whereas the dissolved substance is called a solute. 2. a substance (usually in lesser] amount) dissolved in another substance.

Solute - Biology-Online Dictionary | Biology-Online Dictionary

A hypertonic solution refers to a solution that has a greater concentration of solute than another solution. In the context of biology, when two aqueous solutions are separated by a cell membrane, if the concentration of solute is greater outside the cell than inside the membrane, the solution is called hypertonic.

Hypertonic Solution: Definition And Role In Cell Biology ...

The air we breathe, what we drink, as well as products in our household - these are a few examples of solutions we encounter every day. In this lesson, we will discuss what a solution is, the ...

What is a Solution in Science? - Definition & Examples ...

Biology I - Chapter 2-2. For BIO 1 class. This is Section 2 (Properties of Water) of Chapter 2 (The Chemistry of Life). Word Bank: cohesion, adhesion, mixture, solution, solute, solvent, suspension, pH scale, acid, base, buffer. cohesion. the attraction between molecules of the same substance. adhesion.

Biology I - Chapter 2-2 Flashcards | Quizlet

A solution is a homogeneous mixture of two substances—that is, it has the same distribution of particles throughout. Technically speaking, a solution consists of a mixture of one or more solutes dissolved in a solvent. The particles of solute and solvent are molecules or ions, with one or more solvent molecules bound to each solute particle.

Solution - definition of solution by The Free Dictionary

Overpopulation Definition Overpopulation refers to a population which exceeds its sustainable size within a particular environment or habitat. Overpopulation results from an increased birth rate, decreased death rate, the immigration to a new ecological niche with fewer predators, or the sudden decline in available resources.

Overpopulation: Definition, Effects, Causes, Solutions ...

Definition of Solution in Chemistry. The amount of solute that can be dissolved in solvent is called its solubility. For example, in a saline solution, salt is the solute dissolved in water as the solvent.

For solutions with components in the same phase, the substances present in lower concentration are solutes, while the substance present in highest abundance is the solvent.

Solution Definition in Chemistry - ThoughtCo

First, it may be considered an isotonic solution, meaning it has an equal amount of solute and water when compared to another solution. Second, it may be considered a hypertonic solution, meaning it has more solute and less water than another solution. Third, it may be considered a hypotonic solution,...

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