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thermodynamic properties of alloys as a function of alloy composition. In particular, the excess vibrational entropy of solid solution is calculated using the elastic constants of the system. Using this model, ... (Swalin, 1972). And at  $T \sim T_m$ , the Turnbull approximation ( $C_p \sim C_L$  at  $T \sim T_m$ )

### **Chapter 2 Thermodynamics of pure metals and alloys**

The solution procedure is thus: Select the derivative or variable to be computed; Select the potential representation that makes it easiest, or corresponds to variables you already have in hand. Manipulate the thermodynamic derivative you know to get the one you want. Easy as 1-2-3! We now turn to two methods to manipulate the thermodynamic ...

### **6. The solution of thermodynamic problems - Chemistry ...**

- The pure substances from which a solution may be prepared are called components, or constituents, of the solution.
- Solutions are not limited to liquids: for example air, a mixture of predominantly  $N_2$  and  $O_2$ , forms a vapor solution. Solid solutions such as the solid phase in the Si-Ge system are also common Jan-2012 M Subramanian [www ...](http://www...)

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Thermodynamics of Solution of  $SO_2(g)$  in Water and of Aqueous Sulfur Dioxide Solutions R. N. Goldberg and V. B. Parker National Bureau of Standards, Gaithersburg, MD 20899 Accepted: June 19, 1985 A consistent set of thermochemical property values,  $\Delta_f H$ ,  $\Delta_f G$ ,  $S^\circ$ , and  $C_p$ , at 298.15 K is given for the

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