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solution is available to use as the titrant. The student uses a 25.00 mL volumetric pipet to deliver the propanoic acid solution to a clean, dry flask. After adding an appropriate indicator to the flask, the student titrates the solution with the 0.173 M NaOH, reaching the end point after 20.52 mL of the base solution has been added.

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2014 HSC Chemistry - Board of Studies

2014 U.S. NATIONAL CHEMISTRY OLYMPIAD LOCAL SECTION EXAM Prepared by the American Chemical Society Chemistry Olympiad Examinations Task Force ... solution is prepared at 20 °C that contains 6.0 grams of $K_2Cr_2O_7$ in 50. mL of water. This solution is (A) dilute. (B) saturated.

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