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PREFACE This book grew out of lecture notes for a graduate-level molecular spectroscopy course that I developed at Iowa State University between 1974 and 1987. It is intended to fill a pressing need for a concise introduction to the spectroscopy of atoms and molecules. I have tried to stress logical continuity throughout, with a view to developing readers' confidence in their physical intuition ...

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Spectroscopy Problems. In each of these problems you are given the IR, NMR, and molecular formula. Using this information, your task is to determine the structure of the compound. The best approach for spectroscopy problems is the following steps: Calculate the degree of unsaturation to limit the number of possible structures.

### **Spectroscopy Problems - Organic Chemistry at CU Boulder**

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This revision retains the features which have made it so popular with students and lecturers over the years. It remains an elementary and non-mathematical introduction to molecular spectroscopy that emphasizes the overall unity of the subject and offers a pictorial perception rather than a mathematical description of the principles of spectroscopy.

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Chapter 13: Nuclear Magnetic Resonance (NMR) Spectroscopy direct observation of the H's and C's of a molecules Nuclei are positively charged and spin on an axis; they create a tiny magnetic field + + Not all nuclei are suitable for NMR. <sup>1</sup>H and <sup>13</sup>C are the most important NMR active nuclei in organic chemistry Natural Abundance <sup>1</sup>H 99.9% <sup>13</sup>C 1.1%

### **Chapter 13: Nuclear Magnetic Resonance (NMR) Spectroscopy**

Fundamentals of Molecular Spectroscopy C. N. Banwell Snippet view - 1994. Fundamentals of Molecular Spectroscopy C. N. Banwell Snippet view - 1994. References to this book. Techniques and Mechanisms in Electrochemistry P.A. Christensen, A. Hamnet Limited preview - 1993.

### **Fundamentals of Molecular & Spectroscopy - Banwell ...**

Problems: 1. Suppose you are given two molecules (one is CH<sub>2</sub> and the other is CH<sub>2</sub>-but you don't know which is which). Both molecules have C<sub>2v</sub> symmetry. The CH bond length of molecule I is 1.121 Å and for molecule II it is 1.076 Å.

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