# States Of Matter Class 11 Solutions

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#### States Of Matter Class 11

CBSE Class 11 Chemistry Notes: States of Matter. (ii) Boiling point This is the temperature at which the vapour" pressure of a liquid becomes equal to the atmospheric pressure. It increases in the presence of impurity and with rise in pressure. Boiling point of water is 100°C. (iii) Freezing point At this temperature,...

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States of Matter- CBSE Notes for Class 11 Chemistry CBSE NotesCBSE Notes ChemistryNCERT Solutions Chemistry • Intermolecular Forces Intermolecular forces are the forces of attraction and repulsion between interacting particles have permanent dipole moments. This interaction is stronger than the London forces but is weaker than ion-ion interaction because only partial charges are involved.

## States of Matter- CBSE Notes for Class 11 Chemistry ...

States of Matter Class 11: Overview. In our everyday life, we can observe four different states of matter, namely solid, liquid, gas and plasma. However, there are numerous other states that can be seen to exist but only under extreme conditions. The ones worth mentioning are glass and Bose-Einstein condensates.

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States of Matter Notes Class 11 Chemistry Chapter 5 Download in Pdf. 4. Boyle's Law : The pressure of fixed msss of gas varies inversely with the volume at constant T. P  $\alpha$  1/V(n,T const.) P 1 V 1 = 2 V 2 5. Charle's Law : At constant P, the volume of fixed amount of gas varies directly with its absolute temperature.

## States of Matter Notes Class 11 Chemistry Chapter 5 ...

This article is prepared about Chapter 5 States Of Matter Class 11 Notes of Chemistry. The notes on states of matter of class 11 chemistry have been prepared with great care keeping in mind the effectiveness of it for the students.

## States Of Matter Class 11 Notes Chemistry Chapter 5 ...

Chemistry States of Matter part 1 (Introduction) CBSE class 11 XI.

## Chemistry States of Matter part 1 (Introduction) CBSE class 11 XI

States of Matter, Chapter Notes, Class 11, Chemistry (IIT-JEE & AIPMT) Out of the three states of matter, the most simplest one is the gaseous state. The state is characterized by sensitivity of volume change with change of pressure and temperature. It is due to large distance between molecules as compared to their own dimensions.

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existence of the three states of matter. Now we will learn more about gaseous and liquid states and the laws which govern the behaviour of matter in these states. We shall deal with the solid state in class XII. 5.4 THE GASEOUS STATE This is the simplest state of matter. Throughout our life we remain immersed in the ocean of air which is a ...

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in three states is depicted as follows: We have already learnt the cause for the existence of the three states of matter. Now we will learn more about gaseous and liquid states and the laws which govern the behaviour of matter in these states. We shall deal with the solid state in class XII. 5.4 THE GASEOUS STATE This is the simplest state of ...

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