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SNC 2D1: Chemistry Workbook

1.13 understand that the Periodic Table is an arrangement of elements in order of atomic number
1.14 deduce the electronic configurations of the first 20 elements from their positions in the Periodic Table
1.15 deduce the number of outer electrons in a main group element from its position in the Periodic Table.

Specification and Sample Assessment Material

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Use the periodic table to write the electron configuration for silicon. Explain your thinking. 2. Use the periodic table to write the electron configuration for iodine. Explain your thinking. 3. Which group of elements is characterized by an s^2p^3 configuration? 4. Name the element that matches the following description.

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Reading the Periodic Table This figure shows one square from a detailed periodic table of the elements. • There is also a vertical column with the numbers 2, 8, and 3, which indicate the number of electrons in each occupied energy level of an aluminum atom.

What can you learn about each element from the periodic table?

table in Figure 6.5. Chapter 6 The Periodic Table 5'5 name, element symbol, atomic number, and average atomic mass. 12 4+4 The subatomic particles that play the key 52 Guided Reading and Study Workbook your answer. '. 2003 Standard Course of Study and common core State Standards for 364 Guided Reading and Study workbook

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The Periodic Table chapter of this Prentice Hall Chemistry Companion Course helps students learn the essential lessons associated with the periodic table.

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Chapter 6 The Periodic Table 51 SECTION 6.1 ORGANIZING THE ELEMENTS (pages 155–160) This section describes the development of the periodic table and explains the periodic law. It also describes the classification of elements into metals, nonmetals, and metalloids. Searching For An Organizing Principle (page 155) 1.

SECTION 6.1 ORGANIZING THE ELEMENTS (pages 155–160) (page 155)

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an element in group 8A of the periodic table; the s and p sublevels of the highest occupied energy level are filled. ... Chemistry Chapter 6 Vocabulary 17 terms. belkinsjr. ... Prentice Hall Chemistry Chapter 3 Vocab. 11 terms. jlopes. Prentice Hall Chemistry Chapter 7 10 terms. aja818. Features.

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