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Pideal = 1.0004 ˛P1 = n*R*T/(V - nb) P1 = 1.0029 ˛P2 = (a*n^2)/V^2 P2 = 0.0129 ˛Pwaals = P1 + P2 Pwaals = ˛1.0158 The ideal gas law predicts a pressure of 1.0004 atmospheres, while the van der Waals model predicts 1.0158 atmospheres.

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end disp(x) T4.5-1 The script le is angle = input(0Enter an angle in degrees.0)switch angle case 45 disp(0Angle is in first quadrant0)case -45 disp(0Angle is in second quadrant0)case 135 disp(0Angle is in third quadrant0)case -135 disp(0Angle is in fourth quadrant0)otherwise disp(0Quadrant is unknown.0)end T4.7-1 The 0.75 in the matrix should be replaced with 0.70.

Solutions to Problems in Chapter Four

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