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Abstract. Recognizes that modern physical chemistry is based on quantum mechanics and that it makes pedagogical sense to begin with the atomic and molecular perspective and use it to build a firm understanding of macroscopic phenomena.

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vii This book provides solutions to nearly of the exercises and problems in Mathematics for Physical Chemistry, fourth edition, by Robert G. Mortimer. This edition is a revision of a third edition published by Elsevier/Academic Press in

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Chem 3322 -- Physical Chemistry II

Solution As with Example $\text{\textbackslash(PageIndex\{1\})}$, we recognize that unperturbed component of the problem (Equation $\text{\textbackslash(ref\{7.4.2\})}$) is the particle in an infinitely high well. For this system, the unperturbed Hamiltonian and solutions is the particle in an infinitely high box and the perturbation is a shift of the potential within half a box by $\text{\textbackslash(V_o)}$.

7.4: Perturbation Theory Expresses the Solutions in Terms ...

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(such as the Morse potential) Chemistry 312 Spring 2008 Donald A. McQuarrie and John D. Simon, Physical Chemistry: A Molecular Approach; University Science Books: Sausalito, CA; 1997

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McQuarrie and Simon have developed an excellent modern physical chemistry course that should inspire us to rethink our curriculum. Jeffrey Kovac Department of Chemistry University of Tennessee Knoxville, TN 37996-1600 Physical Chemistry: A Molecular Approach Donald A. McQuarrie and John D. Simon. University Sci-

Physical Chemistry Outline of Theoretical Chemistry

Physical Chemistry, A Molecular Approach, by Donald McQuarrie and John Simon, published by University Science Books, 0-935702-99-7, the definitive new textbook for physical chemistry.

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Mathematics for Physical Chemistry

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