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184 Guided Reading and Study Workbook CHAPTER 17,Thermochemistry (continued) 6. In
thermochemical calculations, is the direction of heat flow given from the point of view of the
system, or of the surroundings? 7. What universal law states that energy can neither be created nor
destroyed

SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505-510)

Thermochemistry Thermochemistry and Energy and Temperature Thermochemistry is study of
changes in energy (heat) associated ... notice final answer in problems above should be 3 sig fig
2.09x104 J or 20.9kJ . Thermochem 9 Calorimeter device to measure changes in heat Bomb (metal
chamber) Calorimeter shown below ...

Thermochemistry - University of Tennessee at Chattanooga

Heat = Transfer of Energy The 3 methods that heat can be transferred: Conduction - heat transfers
by direct contact Convection - is the process in which heat is carried from one place to another by
the bulk movement of a fluid. Radiation- heat transfers by electromagnetic waves.

Chapter 17 - Thermochemistry

Rules of Thermochemistry 1. The magnitude of ΔH is directly proportional to the amount of reactant
or product 2. ΔH for the reaction is equal in magnitude but opposite in sign for ΔH for the reverse of
the reaction 3. The value of ΔH is the same whether the reaction occurs in one step or as a series of

steps

Chapter 8 Thermochemistry - niu.edu.tw

How can you describe the specific heat of olive oil if it takes approximately 420 J of heat to raise the temperature of 7 g of olive oil by 30°C? a. greater than the specific heat of water c. equal to the specific heat of water b. less than the specific heat of water d. Not enough information is given. ____
32. The specific heat of silver is 0 ...

Ch 17 Thermochemistry Practice Test

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Science / Chapter 17 - thermochemistry (handouts)

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Thermochemistry Answers - Worksheet Number One

Thermochemistry. Showing top 8 worksheets in the category - Thermochemistry. Some of the worksheets displayed are Thermochemistry work key, Thermochemistry calculations work 1, The graph below shows a pure substance which is heated, Thermochemistry review, Chapter 8 thermochemistry, Ap chemistry unit 5, Chapter 05, Ap chemistry review work unit 4.

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