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Pearson Chemistry Chapter 8. In covalent bonding, electrons sharing usually occurs so that atoms attain the electron configurations of noble gases. Combinations of atoms of the nonmetals and metalloids in groups 4A, 5A, 6A and 7A of the periodic table are likely to form covalent bonds. The combined atoms usually acquire a total of eight electrons or an octet.

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The Chemistry of Carbon Organic chemistry is the study of compounds with bonds between carbon atoms. Carbon atoms have four valence electrons, allowing them to form strong covalent bonds with many other elements, including hydrogen, oxygen, phosphorus, sulfur, and nitrogen. Living organisms are made up of molecules made of carbon and these

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A covalent bond in which the bonding electrons are most likely to be found in sausage-shaped regions above and below the bond axis of the bonded atoms Polar molecule A molecule in which one side of the molecule is slightly negative and the opposite side is slightly positive.

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8.1 Molecular Compounds > 38 Copyright © Pearson Education, Inc., or its affiliates. All Rights Reserved. • molecular compound: a compound that is composed of ...

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242 Chapter 8 • Covalent Bonding Single Covalent Bonds When only one pair of electrons is shared, such as in a hydrogen molecule, it is a single covalent bond. The shared electron pair is often referred to as the bonding pair. For a hydrogen molecule, shown in Figure 8.4, each covalently bonded atom equally attracts the pair of shared electrons.

Chapter 8: Covalent Bonding

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Chapter 8 Covalent Bonding and Molecular Structure 8-4 H₂ molecule. More sophisticated descriptions of chemical bonding will be discussed in Chapter 9. 8.3 Lewis Structures OWL Opening Exploration 8.X One of the most important tools chemists use to predict the properties of a chemical species is its Lewis structure.

Chapter 8: Covalent Bonding and Molecular Structure

PEARSON CHEMISTRY: Students are introduced to the patterns of electrons in atoms in Lesson 5.2 (pp. 134-137). In Lesson 6.1 (pp. 160-166) students are introduced to how the periodic table is organized and that elements with similar properties are in the same column in the periodic table.

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Chapter 8 Covalent Bonding 181 Section Review Objectives • Distinguish molecular compounds from ionic compounds • Identify the information a molecular formula provides Vocabulary Part A Completion Use this completion exercise to check your understanding of the concepts and terms

05 CTR ch08 7/12/04 8:12 AM Page 181 MOLECULAR COMPOUNDS 8

8.1 Molecular Compounds > 21 Copyright © Pearson Education, Inc., or its affiliates. All Rights Reserved. • The molecular structure of water shows how the oxygen ...

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