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Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions Teacher Notes and Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

# 10 States of Matter - Ms. Agostine's Chemistry Page

Modern Chemistry • CHAPTER 9 HOMEWORK 9-1 (pp. 275–279) ... Modern Chemistry • CHAPTER 9 HOMEWORK 9-2 (pp. 280–282) ... Circle the letter of the best answer. 1. For the general chemical equation  $2A + 3B \not\equiv 4C + D$ , how many moles of A are required to produce 18.4 mol of C?

#### **HOMEWORK 9-1**

SHORT ANSWER Answer the following questions in the space provided. 1. Use the concept of potential energy to describe how a covalent bond forms between two atoms. As the atoms involved in the formation of a covalent bond approach each other, the electron-proton attraction is stronger than the electron-electron and proton-proton repulsions.

#### 6 Chemical Bonding - Effingham County Schools / Overview

Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

\_\_\_\_\_ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

#### Assessment Chapter Test B - clarkchargers.org

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# Modern Chemistry Chapter 2 Measurements and Calculations ...

Modern Chemistry 84 Chapter Test Chapter: States of Matter In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_\_ 1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

# Assessment Chapter Test A - Kettering City School District

SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation: C 3H 4(g) + xO 2(g)  $\rightarrow$  3CO 2(g) + 2H 2O(g) 4 a. What is the value of the coefficient x in this equation? 40.07 g/mol b. What is the molar mass of C 3H 4? 2 mol O 2:1 mol H 2O c. What is the mole ratio of O 2 to H 2O in the above equation? 0.20 ...

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Holt McDougal Modern Chemistry Chapter 11: Gases Chapter Exam Instructions. Choose your answers to the questions and click 'Next' to see the next set of questions. You can skip questions if you ...

# Holt McDougal Modern Chemistry Chapter 11: Gases ...

Modern Chemistry 125 Chapter Test Chapter: Acids and Bases In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

\_\_\_\_\_\_ 1. Which of the properties listed below is not characteristic of an acid? a. a sour taste b. a slippery feel c. the ability to conduct an electric current

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