Mixed Stoichiometry Practice

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Mixed Stoichiometry Practice
Stoichiometry: Mixed Problems (KEY) 1) N2 + 3H2 → 2NH3 What volume of NH3 at STP is produced if 25.0 of N2 is reacted with an excess of H2? 3 3 3 2 3 2 2 2 40.0L NH 1mol NH 22.4L NH 1mol N 2mol NH 28.0g N 25.0g N 1mol N × × × = 2) 2KClO3 → 2KCl + 3O2 If 5.0g of KClO3 is decomposed, what volume of O2 is produced at STP? 2

Stoichiometry: Mixed Problems (KEY)
Mixed Stoichiometry Practice. Potassium Chlorate decomposes into potassium chloride and oxygen gas. Balanced Equation: ___KClO4 (___KCl + ____O2. How many grams of oxygen are produced when 3.0 moles of potassium chlorate decompose completely? Butane (C4H10) undergoes combustion.

Mixed Stoichiometry Practice - Socorro Independent School ...
Mixed Stoichiometry Problems . 1. 2H2 + O2 (2H2O. a). How many moles of H2 would be required to produce 5.0 moles of water? 5.0 moles water. b). What mass of H2O is formed when H2 reacts with 384 g of O2? 432g H2. 2. H2SO4 + 2NaOH (Na2SO4 + 2H2O. a). Balance this equation. Look above. b).

Mixed Stoichiometry Problems - Murrieta Valley Unified ...
Practice problems on a few mixed Stoichiometry problems. Mole to Mole, Grams to Moles, and Grams to Grams. You'll need a calculator and a periodic table befo...

Stoichiometry Mixed Practice Problems
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Mixed Stoichiometry Practice Name _____ Date _____ Period _____ Write and/or balance the following equations (remember the diatomic elements and to criss-cross charges for ionic compounds!!!) Use the mole ratios from the balanced equations to solve the following stoichiometry problems. Use units and labels in all conversions, and round your ...

Stoichiometry Worksheet - srvhs.org
Mixed Stoichiometry Problems. 1. Hydrogen and oxygen react under a specific set of conditions to produce water according to . the following: 2H2 (g) + O2 (g) (2H2O (g) A. How many moles of hydrogen would be required in order to produce 5.0 moles of water? B. How many moles of oxygen are required to produce 436 L of water vapor?

Mixed Stoichiometry Problems - Humble Independent School ...
Practice Work 53 – Stoichiometry-04 Mixed Stoichiometry Problems General Information You will need a periodic table, your stoichiometry notes, and Appendix 12 for this assignment. Sorry about the lack of format. I’m in a time crunch. 123.88 g/mol 70.90 g/mol 137.32 g/mol Reaction 1: P 4(s) + 6 Cl 2(g) 4 PCl 3(g)

Practice Work 53 Stoichiometry-04 Mixed Stoichiometry Problems
This quiz will give you some more practice in solving the various kinds of stoichiometric calculations. Remember that you cannot solve the questions without a balanced chemical equation (none will be provided this time) and the appropriate mole ratio. Review your notes and use them to help you ...

Stoichiometry : Stoichiometry VI: Mixed Problems Quiz
Practice Problems: Stoichiometry. Balance the following chemical reactions: Hint a. CO + O 2 CO 2 b. KNO 3 KNO 2 + O 2 c. O 3 O 2 d. NH 4 NO 3 N 2 O + H 2 O e. CH 3 NH 2 + O 2 CO 2 + H 2 O + N 2 Hint f. Cr(OH) 3 + HClO 4 Cr(ClO 4) 3 + H 2 O Write the balanced chemical equations of each

reaction:

Practice Problems: Stoichiometry - Department of Chemistry
Stoichiometry Practice. Potassium Chlorate decomposes into potassium chloride and oxygen gas. Balanced Equation: ___KClO4 (___KCl + ____O2. How many grams of oxygen are produced when 3.0 moles of potassium chlorate decompose completely?

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Mixed Stoichiometry Practice Name Date Period Write and/or balance the following equations (remember the diatomic elements and to criss- cross charges for ionic compounds!!!) Use the mole ratios from the balanced equations to solve the following stoichiometry problems. Use units and labels in all conversions, and round your answer to sig figs. 1.

www.srvhs.org
For the problems involving gases, assume that the reactions are being performed at STP. 1) Given the reaction: 4NH3(g) + 5O2(g) (4NO(g) + 6H2O(l)

Gas Stoichiometry Worksheet - Mrs. Klatt's Science Page
Chemistry--Unit 5: Stoichiometry Practice Problems I. Stoichiometry 1) Given the balanced equation 2KClO 3 (s) → 2KCl(s) + 3O2 (g), how many moles of O2 are produced from twelve moles of KClO

Chemistry--Unit 5: Stoichiometry Practice Problems I ...
Stoichiometry Practice Worksheet Solve the following stoichiometry grams-grams problems: 1) Using the following equation: 2 NaOH + H 2SO 4 2 H 2O + Na 2SO 4 How many grams of sodium sulfate will be formed if you start with 200.0

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Mixed Stoichiometry
Mixed Stoichiometry Practice Name _____ Date _____ Period _____ Write and/or balance the following equations (remember the diatomic elements and to criss-cross charges for ionic compounds!!!) Use the mole ratios from the balanced equations to solve the following stoichiometry problems.

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