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a) If you used 1 mole of N2, how many moles of NH3 could be produced? b) If 10 moles of NH3 were produced, how many moles of N2 would be required? c) If 3.00 moles of H2 were used, how many moles of NH3 would be made? d) If 0.600 moles of NH3 were produced, how many moles of H2 are required? Mole Ratio Worksheet. 1) Balance this equation:

Mole Ratio Worksheet - Community Unit School District 308

moles O2 x 2 mole H2O = 1 mole O2. 2) If 3.00 moles of H2O are produced, how many moles of oxygen must be consumed? 3) If 2.5 moles of H2O are produced, how many moles of hydrogen gas must be used,? 4) Suppose 4.00 grams of H2 were used? How many grams of water would be produced? Molar Ratio Practice Problems Solutions

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About This Quiz & Worksheet. Working with mole-to-mole ratios requires an understanding of stoichiometry, and this quiz and worksheet combination will test your understanding of this concept.

Quiz & Worksheet - Working with Mole-to-Mole Ratios ...

Tests & Worksheets; Online Lessons; ... Print Instructions NOTE: Only your test content will print. To preview this answer key, click on the File menu and select Print Preview. Click here to print this answer key! Click here to save or print this answer key as a PDF! ... What is the mole ratio of [math]CO_2 to C_6 H_12 O_6[/math] in the ...

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STOICHIOMETRY WORKSHEET ANSWER KEY 2 synthesis 0.5 10 2 1 ...

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