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### CHAPTER 8 Chemical Equations and Reactions

Modern Chemistry 1 Chemical Equations and Reactions CHAPTER 8 REVIEW Chemical Equations and Reactions Teacher Notes and Answers Chapter 8 SECTION 1 SHORT ANSWER 1. a. d b. a c. b d. f e. e f. c 2. 8,4,9 3. a. 12 atoms b. 16 atoms c. 51 atoms d. 3 1024 atoms

### CHAPTER 8 REVIEW Chemical Equations and Reactions

Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions Teacher Notes and Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a

colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. Write empirical formulas to match the following molecular formulas: CH<sub>3</sub>CO<sub>2</sub> a. C<sub>2</sub>H<sub>6</sub>O<sub>4</sub> N<sub>2</sub>O<sub>5</sub> b. N<sub>2</sub>O<sub>5</sub> HgCl c. Hg<sub>2</sub>Cl<sub>2</sub> CH<sub>2</sub> d. C<sub>6</sub>H<sub>12</sub> 2. C<sub>4</sub>H<sub>8</sub> A certain hydrocarbon has an empirical formula of CH<sub>2</sub> and ...

### 7 Chemical Formulas and Chemical Compounds

CHAPTER 8 REVIEW Class S E-ci7b IV hemical Equations and Reactions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. b d A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products.

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

### 10 States of Matter - Ms. Agostine's Chemistry Page

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