

Quantum Mechanics Solutions 3:

PARTICLE IN A BOX

[easy]

1. $\sin(\theta) = 0$ for $\theta = n\pi$, where n is any integer. Thus,

$$\psi(x=L) = \sin\left(\frac{2\pi L}{\lambda}\right) = 0 \text{ for } \frac{2\pi L}{\lambda} = n\pi, \text{ for any integer } n. \text{ Solving}$$

this equation for λ tells us that the wavelength can only take on certain discrete values, which we will denote as λ_n , where

$$\lambda_n = \frac{2L}{n}, \text{ for } n = 1, 2, 3, \dots. \text{ (Think about why we have excluded}$$

the value $n = 0$ and all negative integer values of n .)

[medium]

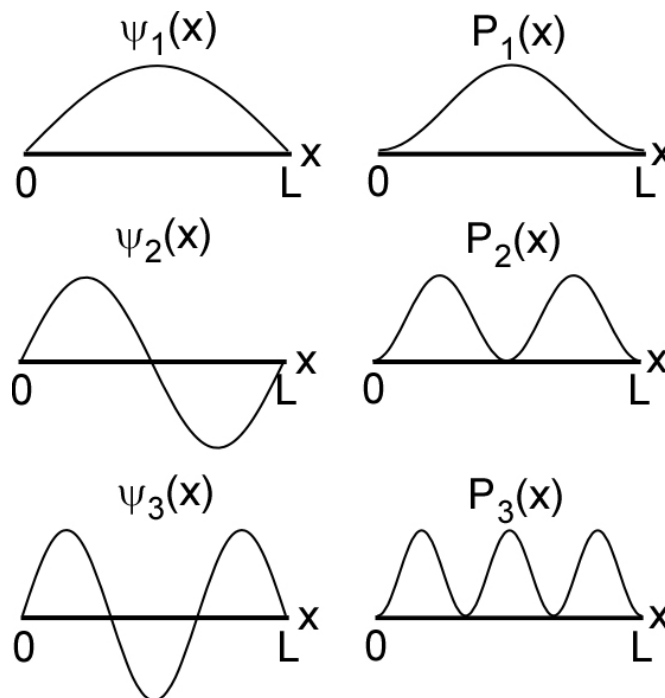
2. For these allowed values of wavelength, the corresponding

allowed wavefunctions are $\psi_n(x) = \sin\left(\frac{n\pi x}{L}\right)$. For $n = 1$ this is

$$\psi_1(x) = \sin\left(\frac{\pi x}{L}\right), \text{ for } n = 2 \text{ this is } \psi_2(x) = \sin\left(\frac{2\pi x}{L}\right), \text{ and so on.}$$

The corresponding probability functions are

$$P_n(x) = |\psi_n(x)|^2 = \sin^2\left(\frac{n\pi x}{L}\right). \text{ See sketch:}$$



[hard]

3. From the de Broglie relation $p = \frac{h}{\lambda}$ we have the allowed momentum values: $p_n = \frac{h}{\lambda_n} = \frac{hn}{2L}$, for $n = 1, 2, 3, \dots$. Thus, from the relation $E = \frac{p^2}{2m}$, we have the allowed energy values:
- $$E_n = \frac{p_n^2}{2m} = \frac{h^2 n^2}{8mL^2}.$$
- The lowest allowed energy (zero point energy) is therefore $E_{zeropoint} = E_1 = \frac{h^2}{8mL^2}$. (Note that in a Newtonian universe, Planck's constant would be zero ($h = 0$), and we would have $E_{zeropoint} = 0$). As the box is made smaller (L decreases), the zero point energy increases. Why would squeezing the box to make it smaller give it more energy?