INDIAN INSTITUTE OF TECHNOLOGY KHARAGPUR

Mid-Autumn Semester Examination 2023-24

Subject Name: Computer aided Process Engineering

Duration: 2 hr

Full Marks: 30

[5]

Subject No.: CH31203

Department: Chemical Engineering

Specific charts, graph paper, log book etc., required: No

Instructions:

1. Attempt all questions

2. Assume, if necessary, clearly stating the reason

3. Answer all parts of a question together

A system of 3 chemical reactors is shown in Figure 1. In all 3 reactors, a first-order liquid phase chemical reaction, $A \rightarrow P$, takes place. The rate of disappearance of reactant A is given as: (-r) = kC. Here, the reaction rate constant, $k = 0.1 \text{ sec}^{-1}$.

(i) Derive the component A balance equations for all 3 reactors operated at steady state [5]

(ii) Find 3 concentrations of component $A(C_1, C_2 \text{ and } C_3)$ using the method of Gauss elimination

with backward sweep

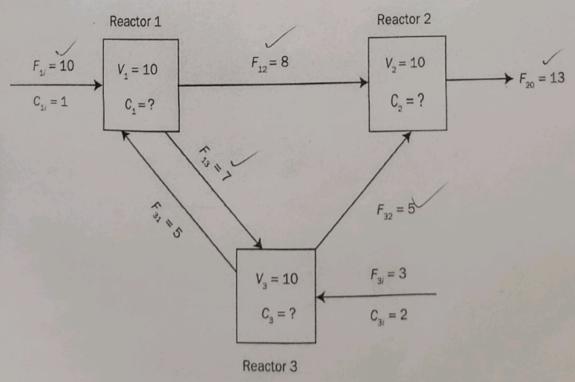


Figure 1. A system of 3 reactors [$F = \text{volumetric flow rate (m}^3/\text{sec)}$, $C = \text{concentration of reactant } A (\text{kg/m}^3)$, $V = \text{volume of liquid in the reactor (m}^3)$].

2. Find x at which the following function yields its minimum value:

$$f(x) = x + 2\sin x$$

[8]

Use N-R method with $x_0 = -2$ and desired tolerance = 10^{-5} .

- 3. (a) Discuss the relative merits and demerits of explicit and implicit Euler.
 - (b) Consider the following IVP:

$$\frac{dy}{dx} = x + y y(0) = 1$$

[2+4+6=12]

Compute y(0.5) with step size h = 0.1 using the following methods:

- (i) RK2
- (ii) RK4

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