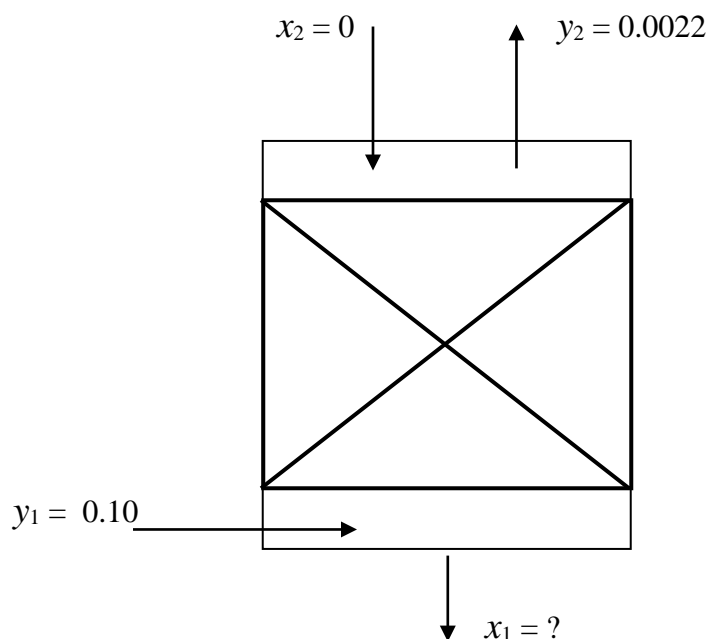


Solutions of 1st Problem of Tutorial Sheet #2 (2023)

1. Ammonia is to be removed from an ammonia-air mixture by water scrubbing in a 0.786 m diameter tower packed with 25 mm Raschig rings. The gas mixture is available at the rate of 600 m³/h (at 25°C and 1 atm) with 10% ammonia by volume. Pure water will be used as solvent at a rate twice the minimum. Film coefficients are $k_y a = 150 \text{ kmol/m}^3 \text{ h } \Delta y$ and $k_x a = 325 \text{ kmol/m}^3 \text{ h } \Delta x$. The equilibrium relation may be expressed $y^* = 1.02 x/(1 - x)$. Calculate the depth of the packing required for 98% removal of ammonia.

Solution:



$$M_{AV} (\text{inlet}) = (0.10 \times 17) + (0.90 \times 29) = 27.80$$

$$\rho_{G1} = \frac{P.M_{AV}(\text{inlet})}{R.T} = \frac{1 \times 27.80}{82.1 \times 10^{-3} \times 298} = 1.136 \text{ kg/m}^3$$

$$\text{Gas molar input rate} = G_1 = \frac{600 \times 1.136}{27.80} = 24.52 \text{ kmol/h}$$

$$\text{Molar flow rate of the carrier} = G_s = 24.52 \times 0.9 = 22.07 \text{ kmol/h}$$

$$G_2 = 22.07 + (24.52 - 22.07) \times 0.02 = 22.12 \text{ kmol/h}$$

The operating line equation may be written in terms of solute-free stream as follows:

$$G_s \left(\frac{y}{1-y} - \frac{y_2}{1-y_2} \right) = L_s \left(\frac{x}{1-x} - \frac{x_2}{1-x_2} \right)$$

For minimum liquid rate, the operating line will cut the equilibrium line at $y = y_1$ (= 0.10), and the corresponding maximum liquid phase concentration will be (from the equilibrium relationship) $x_{\max} = 0.089$. Putting $y = y_1$, and $x = x_{\max}$, in the operating line equation, we get,

$$22.07 \left(\frac{0.10}{1-0.10} - \frac{0.0022}{1-0.0022} \right) = L_{s,\min} \left(\frac{0.089}{1-0.089} \right)$$

$$\Rightarrow L_{s,\min} = 24.60 \text{ kmol/h } \text{HN}_3 - \text{free water}$$

Designed liquid rate, $L_s = 2 \times 24.6 = 49.20 \text{ kmol/h}$

The actual NH_3 concentration in the outlet water can be calculated again from the material balance equation as,

$$22.07 \left(\frac{0.10}{0.90} - \frac{0.0022}{0.9978} \right) = L_s \left(\frac{x_1}{1-x_1} \right) = 49.20 \left(\frac{x_1}{1-x_1} \right)$$

$$\Rightarrow x_1 = 0.046$$

The tower height can be calculated using the following equation:

$$Z = \frac{G'}{k_y a (1-y)_{iM}} \times \int_{y_2}^{y_1} \frac{(1-y)_{iM}}{(1-y)(y-y_i)} dy$$

$$= H_{tG} \times N_{tG}$$

The above equation can be simplified by replacing logarithmic average by arithmetic average. Thus,

$$(1-y)_{iM} = \frac{(1-y_i) - (1-y)}{\ln \left(\frac{1-y_i}{1-y} \right)} \approx \frac{(1-y_i) + (1-y)}{2}$$

$$\text{Therefore, } N_{tG} = \int_{y_2}^{y_1} \frac{dy}{y-y_i} + \frac{1}{2} \ln \frac{1-y_2}{1-y_1}$$

Evaluation of the first integral requires interfacial composition values. Now for any cross section of the tower, if we draw a line from any point on the operating line with a slope of

- $k_x a / k_y a$, then this line will intersect the equilibrium curve at a point, whose coordinates will give the interfacial composition. Also, if $k_x a$ and $k_y a$ are assumed to remain constant throughout the tower, then the ratio ($k_x a / k_y a$) will remain constant, that is, all such lines will be parallel.

The x and y values of the operating line was obtained from the equation,

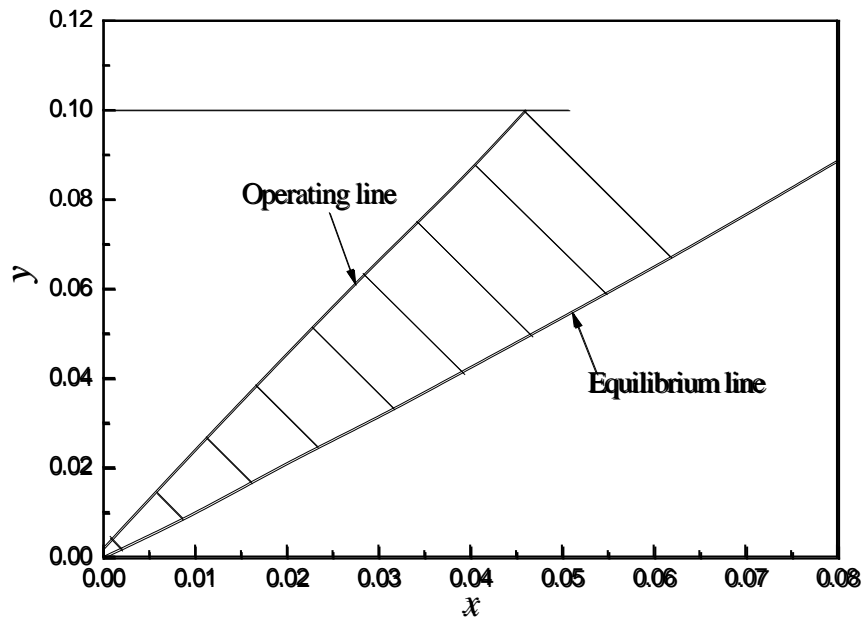
$$22.07 \left(\frac{y}{1-y} - \frac{0.0022}{1-0.0022} \right) = 49.20 \left(\frac{x}{1-x} \right)$$

$$\text{Or, } \frac{y}{1-y} = 2.23 \left(\frac{x}{1-x} \right) + 0.0022$$

x	0	0.01	0.02	0.03	0.04	0.046
y	0.002	0.024	0.0455	0.0664	0.0868	0.10

The interfacial concentrations were then found out as follows:

y	0.002	0.0142	0.0265	0.0387	0.051	0.063	0.075	0.0875	0.10
y_i	0.0005	0.008	0.017	0.024	0.033	0.041	0.0495	0.058	0.066
$\frac{1}{y - y_i}$	666.67	161.29	105.26	68.03	55.55	45.45	39.21	33.90	29.41



Determination of interfacial concentration

Therefore, the first integral is determined as

$$I = \frac{0.01225}{3} \{666.67 + 4 \times (161.29 + 68.03 + 45.45 + 33.90) + 2 \times (105.26 + 55.55 + 39.21) + 29.41\}$$

$$= 9.517$$

$$N_{tG} = 9.517 + \frac{1}{2} \ln \frac{1-0.002}{1-0.1} = 9.568$$

$$H_{tG} = \frac{G'}{k_y \cdot a \cdot (1-y)_{iM}}$$

where $(1-y)_{iM}$ can be taken as the arithmetic average of that at the bottom and at the top of the tower.

$$\text{At the tower bottom: } (1-y)_{iM1} = \frac{(1-0.066) - (1-0.1)}{\ln \frac{(1-0.066)}{(1-0.1)}} = 0.9168$$

$$\text{At the tower top: } (1-y)_{iM2} = \frac{(1-0.0005) - (1-0.002)}{\ln \frac{(1-0.0005)}{(1-0.002)}} = 0.9987$$

$$\text{Therefore, } (1-y)_{iM} = (0.9168 + 0.9987)/2 = 0.9577$$

$$H_{tG} = \frac{G'}{k_y \cdot a \cdot (1-y)_{iM}} = \frac{23.32 / 0.485}{150 \times 0.9577} = \frac{48.08}{143.655} = 0.335 \text{ m}$$

$$\text{Total packed height} = Z = H_{tG} \times N_{tG} = 0.335 \times 9.568 = \mathbf{3.2 \text{ m}}$$