

COGNITIVE NEUROSCIENCE FOR AI DEVELOPERS (SS 2023)

Week 03 - Neurons

Prof. Dr. Andreas M Kist

Single Choice Exercise // Only one answer is correct

Q1: At which magnitude is the resting membrane potential of neurons?

- A. V
- B. mV
- C. μ V
- D. nV
- E. pV

Q2: Which of the following ions is barely involved in membrane potential generation?

- A. K^+ (Potassium)
- B. Na^+ (Sodium)
- C. Cl^- (Chloride)
- D. A^- (Anions)
- E. Fe^{2+} (Iron)

Q3: Which statement is true in respect to neuron-neuron communication?

- A. Neurons are mainly connected electrically
- B. Neurotransmitters are released from the post to the presynapse
- C. Chemical connections allow signal adjustments
- D. Unmyelinated axons allow faster transmission
- E. Vesicles fuse with the presynapse upon sodium influx

Q4: Single perceptrons cannot compute

- A. Logical AND
- B. Logical OR
- C. Logical NOT AND
- D. Logical XOR
- E. Logical NOT

Q5: What statement is true w.r.t. neural anatomy?

- A. Neurons could have two or more axons
- B. Neurons always have only one dendrite
- C. Some neurons in the retina are bipolar
- D. Purkinje cells have no dendrites
- E. Oligodendrocytes are involved in the tripartite synapse

Neurobiology is the applied physics, chemics and electrical engineering. Discuss this statement with respect to the following three phenomena:

- Reversal potential
- Action potential
- Neural equivalent circuit

The Nernst equation is very important for calculating the equilibrium potential for ions. It is defined using the following formula:

$$E = \frac{RT}{zF} \ln \frac{[\text{ion outside cell}]}{[\text{ion inside cell}]}$$

Write down the individual constants and variables, their definition and their units.

R:

T:

z:

F:

[ion outside/inside cell]:

Calculate the E_{K^+} at 20°C for 5 mM K^+ outside and 140 mM K^+ inside: