



1. Properties of Nuclei Solutions

Practice Set-1

1. The radius of a ${}^{64}_{29}\text{Cu}$ nucleus is measured to be 4.8×10^{-13} cm. The radius of a ${}^{27}_{12}\text{Mg}$ nucleus can be estimated to be
- 2.86×10^{-13} cm
 - 5.2×10^{-13} cm
 - 3.6×10^{-13} cm
 - 8.6×10^{-13} cm

Solution:

$$\text{Since } R = R_0(A)^{1/3} \Rightarrow \frac{R_{\text{Mg}}}{R_{\text{Cu}}} = \left(\frac{A_{\text{Mg}}}{A_{\text{Cu}}}\right)^{1/3} = \left(\frac{27}{64}\right)^{1/3}$$

$$\Rightarrow \frac{R_{\text{Mg}}}{R_{\text{Cu}}} = \frac{3}{4} \Rightarrow R_{\text{Mg}} = \frac{3}{4} \times 4.8 \times 10^{-13} = 3.6 \times 10^{-13} \text{ cm.}$$

So the correct answer is **Option (c)**

2. The intrinsic electric dipole moment of a nucleus ${}^A_Z\text{X}$

- Increases with Z , but independent of A
- Decreases with Z , but independent of A
- Is always zero
- Increases with Z and A

Solution: So the correct answer is **Option (c)**

3. In deep inelastic scattering electrons are scattered off protons to determine if a proton has any internal structure. The energy of the electron for this must be at least

- 1.25×10^9 eV
- 1.25×10^{12} eV
- 1.25×10^6 eV
- 1.25×10^8 eV

Solution:

The internal structure of proton can only be determined if the wavelength of the incoming electron is nearly equal to the size of the proton

$$\text{i.e. } \lambda = R = 1.2A^{1/3}(\text{fm}) = 1.2\text{fm} = 1.2 \times 10^{-15} \text{ m}$$

$$\text{According to de-Broglie relation, } \lambda = \frac{h}{p} = \frac{h}{\sqrt{2mE}}$$

$$\text{This can be also written as } E^2 = h^2 \lambda^2 / c^2 + m_0^2 c^4$$

So the correct answer is **Option (b)**

4. The difference in the Coulomb energy between the mirror nuclei ${}^{49}_{24}\text{Cr}$ and ${}^{49}_{25}\text{Mn}$ is 6.0MeV. Assuming that the nuclei have a spherically symmetric charge distribution and that $\frac{e^2}{4\pi\epsilon_0}$ is approximately 1.0MeV – fm, the radius of the ${}^{49}_{25}\text{Mn}$ nucleus is (a) $4.9 \times 10^{-13} \text{ m}$ (b) $4.9 \times 10^{-15} \text{ m}$ (c) $5.1 \times 10^{-13} \text{ m}$ (d) $5.1 \times 10^{-15} \text{ m}$

Solution:

$$R = \frac{3e^2}{5 \cdot \Delta W} (Z_1^2 - Z_2^2) = \frac{3 \times 1 \times 10^{-15}}{5 \times 6} (25^2 - 24^2) \\ = 4.9 \times 10^{-15} \text{ m}$$

So the correct answer is **Option (b)**

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Practice Set-2

1. Inside a large nucleus, a nucleon with mass $939\text{MeV}c^{-2}$ has Fermi momentum 1.40fm^{-1} at absolute zero temperature. Its velocity is Xc , where the value of X is ——— (up to two decimal places). ($\hbar c = 197\text{MeV} - \text{fm}$)

Solution:

Here, Fermi-momentum or fermi radius, $k_F = 1.40\text{fm}^{-1}$ and $\hbar c = 197\text{MeV} - \text{fm}$

Now, Fermi velocity -

$$V_F = \frac{P}{m} = \frac{\hbar k_F}{m} = \frac{(\hbar c)k_F \cdot c}{mc^2} = \frac{(197) \times 1.40 \times c}{939} = \frac{275.8c}{939} = 0.29c$$

So the correct answer is **0.29**

2. The mean kinetic energy of a nucleon in a nucleus of atomic weight A varies as A^n , where n is ——— (upto two decimal places)

Solution:

$$\langle T \rangle = \frac{\int_0^R -\frac{\hbar^2}{2m} \left(\frac{d^2}{dr^2} + \frac{1}{r} \frac{d}{dr} \right) 4\pi r^2 dr}{\int_0^R 4\pi r^2 dr} = \frac{-\frac{\hbar^2}{2m} 4\pi \int_0^R (2+2) dr}{\int_0^R 4\pi r^2 dr} = \frac{-\frac{\hbar^2}{2m} 4\pi \times 4R}{4\pi R^3/3}$$

$$\Rightarrow \langle T \rangle \propto \frac{1}{R^2} = \frac{1}{(R_0 A^{1/3})^2} = \frac{1}{A^{2/3}} = A^{-2/3} \Rightarrow n = -\frac{2}{3} = -0.667 = -0.67$$

So the correct answer is **-0.67**

3. According to the Fermi gas model of nucleus, the nucleons move in a spherical volume of radius R ($= R_0 A^{1/3}$, where A is the mass number and R_0 is an empirical constant with the dimensions of length). The Fermi energy of the nucleus E_F is proportional to

a. R_0^2

b. $\frac{1}{R_0}$

c. $\frac{1}{R_0^2}$

d. $\frac{1}{R_0^3}$

Solution:

$$\text{Fermi energy } E_F = \frac{\hbar^2}{2m} \left(3\pi^2 \frac{N}{V} \right)^{2/3}$$

$$V = \frac{4\pi}{3} R^3 = \frac{4\pi}{3} (R_0 A^{1/3})^3 = \frac{4\pi}{3} R_0^3 A$$

$$\therefore E_F = \frac{\hbar^2}{2m} \left(\frac{3\pi^2 N}{\frac{4\pi}{3} R_0^3 A} \right)^{2/3} = \frac{\hbar^2}{2m} \left(\frac{9\pi N}{4A} \cdot \frac{1}{R_0^3} \right)^{2/3} \Rightarrow E_F \propto \frac{1}{R_0^2}$$

So the correct answer is **Option (c)**

4. The stable nucleus that has $\frac{1}{3}$ the radius of ^{189}Os nucleus is,

a. ^7Li

b. ^{16}O

c. ^4He

d. ^{14}N

Solution:

$$R = \frac{1}{3}R_{Os} \Rightarrow R_0(A)^{1/3} = \frac{1}{3}R_0(189)^{1/3} \Rightarrow A = 7$$

So the correct answer is **Option (a)**

5. The binding energy of the k -shell electron in a Uranium atom ($Z = 92, A = 238$) will be modified due to (i) screening caused by other electrons and (ii) the finite extent of the nucleus as follows:
- a. Increases due to (i), remains unchanged due to (ii)
 - b. Decreases due to (i), decreases due to (ii)
 - c. Increases due to (i), increases due to (ii)
 - d. Decreases due to (i), remains unchanged due to (ii)

Solution: So the correct answer is **Option (b)**

