

# DATA SHEET FOR CHEM 101

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
<div>1 H 1.0079</div>																		<div>2 He 4.0026</div>
<div>3 Li 6.941</div>	<div>4 Be 9.0122</div>	<div>Atomic number Symbol Atomic weight</div>										<div>5 B 10.811</div>	<div>6 C 12.011</div>	<div>7 N 14.007</div>	<div>8 O 15.999</div>	<div>9 F 18.998</div>	<div>10 Ne 20.180</div>	
<div>11 Na 22.990</div>	<div>12 Mg 24.305</div>											<div>13 Al 26.982</div>	<div>14 Si 28.086</div>	<div>15 P 30.974</div>	<div>16 S 32.065</div>	<div>17 Cl 35.453</div>	<div>18 Ar 39.948</div>	
<div>19 K 39.098</div>	<div>20 Ca 40.078</div>	<div>21 Sc 44.956</div>	<div>22 Ti 47.867</div>	<div>23 V 50.942</div>	<div>24 Cr 51.996</div>	<div>25 Mn 54.938</div>	<div>26 Fe 55.845</div>	<div>27 Co 58.933</div>	<div>28 Ni 58.693</div>	<div>29 Cu 63.546</div>	<div>30 Zn 65.39</div>	<div>31 Ga 69.723</div>	<div>32 Ge 72.61</div>	<div>33 As 74.922</div>	<div>34 Se 78.96</div>	<div>35 Br 79.904</div>	<div>36 Kr 83.80</div>	
<div>37 Rb 85.468</div>	<div>38 Sr 87.62</div>	<div>39 Y 88.906</div>	<div>40 Zr 91.224</div>	<div>41 Nb 92.906</div>	<div>42 Mo 95.94</div>	<div>43 Tc [98]</div>	<div>44 Ru 101.07</div>	<div>45 Rh 102.91</div>	<div>46 Pd 106.42</div>	<div>47 Ag 107.87</div>	<div>48 Cd 112.41</div>	<div>49 In 114.82</div>	<div>50 Sn 118.71</div>	<div>51 Sb 121.76</div>	<div>52 Te 127.60</div>	<div>53 I 126.90</div>	<div>54 Xe 131.29</div>	
<div>55 Cs 132.91</div>	<div>56 Ba 137.33</div>	<div>57-70 †</div>	<div>71 Lu 174.97</div>	<div>72 Hf 178.49</div>	<div>73 Ta 180.95</div>	<div>74 W 183.84</div>	<div>75 Re 186.21</div>	<div>76 Os 190.23</div>	<div>77 Ir 192.22</div>	<div>78 Pt 195.08</div>	<div>79 Au 196.97</div>	<div>80 Hg 200.59</div>	<div>81 Tl 204.38</div>	<div>82 Pb 207.2</div>	<div>83 Bi 208.98</div>	<div>84 Po [209]</div>	<div>85 At [210]</div>	<div>86 Rn [222]</div>
<div>87 Fr [223]</div>	<div>88 Ra [226]</div>	<div>89-102 †</div>	<div>103 Lr [262]</div>	<div>104 Rf [261]</div>	<div>105 Db [262]</div>	<div>106 Sg [266]</div>	<div>107 Bh [264]</div>	<div>108 Hs [269]</div>	<div>109 Mt [268]</div>	<div>110 Ds [281]</div>	<div>111 Rg [272]</div>	<div>112 Cn [285]</div>	<div>113 Nh [286]</div>	<div>114 Fl [289]</div>	<div>115 Mc [289]</div>	<div>116 Lv [293]</div>	<div>117 Ts [293]</div>	<div>118 Og [294]</div>

†lanthanides

57 <b>La</b> 138.91	58 <b>Ce</b> 140.12	59 <b>Pr</b> 140.91	60 <b>Nd</b> 144.24	61 <b>Pm</b> [145]	62 <b>Sm</b> 150.36	63 <b>Eu</b> 151.96	64 <b>Gd</b> 157.25	65 <b>Tb</b> 158.93	66 <b>Dy</b> 162.50	67 <b>Ho</b> 164.93	68 <b>Er</b> 167.26	69 <b>Tm</b> 168.93	70 <b>Yb</b> 173.04
89 <b>Ac</b> [227]	90 <b>Th</b> 232.04	91 <b>Pa</b> 231.04	92 <b>U</b> 238.03	93 <b>Np</b> [237]	94 <b>Pu</b> [244]	95 <b>Am</b> [243]	96 <b>Cm</b> [247]	97 <b>Bk</b> [251]	98 <b>Cf</b> [251]	99 <b>Es</b> [252]	100 <b>Fm</b> [257]	101 <b>Md</b> [258]	102 <b>No</b> [259]

†actinides

Avogadro's Number ( $N_A$ )	$6.02 \times 10^{23} \text{ mol}^{-1}$
atomic mass unit (amu)	$1.661 \times 10^{-27} \text{ kg}$
electronic charge (e)	$1.602 \times 10^{-19} \text{ C}$
mass of an electron ( $m_e$ )	$9.109 \times 10^{-31} \text{ kg}$
speed of light in vacuum (c)	$3.00 \times 10^8 \text{ m s}^{-1}$
Planck's constant (h)	$6.63 \times 10^{-34} \text{ J s}$
Debye (D)	$3.34 \times 10^{-30} \text{ C m}$

$\Delta E = -2.18 \times 10^{-18} \text{ J} (1/n^2 - 1/n'^2)$
$v =  3.29 \times 10^{15} \text{ s}^{-1} (1/n^2 - 1/n'^2) $
$\lambda = h/mv$
$(\Delta x)(\Delta mv) \geq h/4\pi$
$c = \lambda v$
$\mu = Qr$
$1 \text{ J} = 1 \text{ kg m}^2 \text{ s}^{-2}$

G = giga = $10^9$
M = mega = $10^6$
k = kilo = $10^3$
m = milli = $10^{-3}$
$\mu$ = micro = $10^{-6}$
n = nano = $10^{-9}$
p = pico = $10^{-12}$

## Lattice Energies ( $\text{kJ mol}^{-1}$ )

LiF (s)	1037	LiCl (s)	852
NaF (s)	926	NaCl (s)	787
KF (s)	821	KCl (s)	717
CsF (s)	750	CsCl (s)	676
CaF <sub>2</sub> (s)	2347	CaCl <sub>2</sub> (s)	2255
NaH (s)	760	MgH <sub>2</sub> (s)	2490
KH (s)	701	CaH <sub>2</sub> (s)	2195

## Ionization Energies ( $\text{kJ mol}^{-1}$ )

	1 <sup>st</sup>	2 <sup>nd</sup>		1 <sup>st</sup>	2 <sup>nd</sup>
Cl	1251	2296	Mg	738	1451
F	1681	3375	Ca	590	
Li	513	7298	Ba	503	965
Na	496	4562	Rb	403	
P	1012	1907			
Cs	376	2420			
H	1312				

## Average Bond Energies ( $\text{kJ mol}^{-1}$ )

H-H	432	H-F	565	C-Cl	328
F-F	154	H-Cl	431	C-F	485
Cl-Cl	239	H-Br	366	C-H	413
Br-Br	193	H-I	299	O-H	467
I-I	151	N-H	388	C-O	358
O-O	146	O=O	495	C=O	800
N-N	163	N=N	409	N≡N	945
C-C	348	C=C	612	C≡C	837

## Average Bond Lengths (pm)

H-H	74	C-H	109	N-H	102	O-H	96
F-F	142	C-C	154	N-N	140	O-O	148
Cl-Cl	199	C=C	134	N=N	124	O=O	121
Br-Br	228	C≡C	120	N≡N	110		
I-I	267	C-N	143	N-O	136	C-O	143
H-F	92	C=N	138	N=O	122	C=O	123
H-Cl	127	C≡N	116	N≡O	106	C≡O	113
H-Br	141	C-Cl	176	S-O	168		
H-I	161	C=Cl	158	S=O	152		

## Electron Affinities ( $\text{kJ mol}^{-1}$ )

F -328	Cl -349	H -71	O -141	S -200	Li -60	Be >0
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