

# Chem 101 Laboratory Exercise #2

## Laboratory Notebook

Moles, Concentration, Acid-Base Reactions and Quantitative Analysis by Titration

Name: Arfaz Hossain

Lab Section: B12??

Quad: ??

Date: October 5, 2023

In-lab Notes:

### EXPERIMENT 1:

Experimental Procedure for standardising a solution of NaOH using a titration:

Initial Without NaOH Solution Volume: 200mL

Final with added NaOH Solution: 212mL

1<sup>st</sup>

KHP Amount:	560g
Burette Reading Initial:	41.5mL
Burette Reading Final:	26.2mL

2<sup>nd</sup>

KHP Amount:	536g
Burette Reading Initial:	42.8mL
Burette Reading Final:	29.5mL

3<sup>rd</sup>

KHP Amount:	583g
Burette Reading Initial:	44.6mL
Burette Reading Final:	32.5mL

### EXPERIMENT 2 (Vineger):

1<sup>st</sup>

Vineger Amount:	23mL
Burette Reading Initial:	30.1mL
Burette Reading Final:	10.5mL

1<sup>st</sup>

Vineger Amount:	23mL
Burette Reading Initial:	41.4mL

Burette Reading Final: 24.7mL

1<sup>st</sup>

Vineger Amount: 23mL

Burette Reading Initial: 24.7mL

Burette Reading Final: 5.1mL

# Chem 101 Laboratory

## Exercise #2 Laboratory

### Notebook

**Moles, Concentration, Acid-Base Reactions and Quantitative Analysis by Titration**

**Name:** Arfaz Hossain

**Lab Section:** B12

**Quad:** 2

**Date:** October 5, 2023

**Experimental Procedures:**

**Abstract:**

**Data:**

Experiment 1	Trial #1	Trial #2	Trial #3
KHP Mass (g)	560 g	536 g	583 g
NaOH Initial (mL)	41.5 mL	42.8 mL	44.6 mL
NaOH Final (mL)	26.2 mL	29.5 mL	32.5 mL
NaOH Difference (mL)	15.3 mL	13.3 mL	12.1 mL
Experiment 2	Trial #1	Trial #2	Trial #3
Vinegar Volume (mL)	23.5 mL	27 mL	26 mL
NaOH Initial (mL)	30.1 mL	41.4 mL	24.7 mL
NaOH Final (mL)	10.5 mL	24.7 mL	5.1 mL
NaOH Difference (mL)	19.6 mL	16.7 mL	19.6 mL

**Results:**

Experiment 1	Trial #1	Trial #2	Trial #3
KHP Moles	2.74 moles	2.62 moles	2.85 moles
Volume of NaOH used	0.0153 L	0.0133 L	0.0121 L
Standardized NaOH Concentration (g/moles)	179.08 mol/L	196.99 mol/L	235.53 mol/L
Experiment 1	Trial #1	Trial #2	Trial #3
CH <sub>3</sub> COOH Moles	0.02003845 moles	0.0230229 moles	0.0221702 moles
Volume of NaOH used	0.0196 L	0.0167L	0.0196 L
Standardized NaOH Concentration (g/moles)	mol/L	mol/L	mol/L

**Algebraic Equations:**

**Discussion:**

**Conclusions:**

**References:**