



## Caffeine Extraction Lab Notes

Fundamentals of Chemistry I (University of Victoria)



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## Report for Laboratory Exercise #6

### Extraction of Caffeine from Tea <sup>1</sup>

Using Microsoft Word, students are to **insert responses in all yellow highlighted areas**. It is recommended that the report be completed without changing font size, column width, row width, margins and highlights. The completed report must be uploaded to the 101 CourseSpaces as a .pdf file within 2 calendar days of the end of the scheduled lab period.

Name: Lab Section: Quad: 1 Date: Nov/29/17

#### Abstract

(max 3 lines)

Through the means of a liquid liquid extraction process it was demonstrated that a single tea bag of Tetley Orange Pekoe tea contains 6.619% of caffeine. This percentage of caffeine was determined using the equation  $\% \text{ Caffeine} = \text{weight of caffeine(g)} * 100\% / \text{weight of tea leaves (g)}$

#### Data/Results

**Table 1.** Experimental data and calculated values

Extraction and isolation	
Weight of tea leaves	3.2634g
Weight of caffeine	0.2160g
Weight % caffeine in tea	6.619%
Lethal dose for a 700 g rat	0.622g, approx. 3 cups of tea.

#### Algebraic Equation

Weight % of caffeine in tea =

$[\text{Weight of Caffeine(g)} / \text{Weight of tea leaves(g)}] * 100\%$

#### Discussion Respond to the following:

How could this experiment be changed to give a more accurate amount of caffeine in a teabag? (max 3 lines). Using sublimation instead of a liquid liquid extraction process would yield a more accurate amount of caffeine in a tea bag as the impurities present in liquid liquid extraction would have a different sublimation temperature.

How does the amount of caffeine measured in tea bag compare with the amounts that might be lethal for a human? (max 3 lines). The lethal amount of caffeine for a human is approximately 150-200mg per kg of body mass, therefore the 216mg of caffeine rendered from one tea bag is less than 1% of the average 70kg person.

#### Conclusions

(max 1 line)

The quantity of caffeine found via liquid liquid extraction in one tea bag of Tetley Orange Pekoe tea is 0.216g.

### References

1. Properties of Materials, Laboratory Manual, Chemistry 101, pp.31-39. (University of Victoria: Victoria, B.C.). Fall 2017.
2. Tetley Orange Pekoe. Beverages Canada Inc, Etibicoke, Ontario, M9W 6L2, Lot #CTN304LR

Marks	max
<b>Pre-lab quiz</b>	3
<b>Laboratory Notebook:</b> Have all the relevant observations and data been recorded?	1
<b>Abstract:</b> Is the abstract consistent with content of the Exercise?	2
<b>Data/Results:</b> Is the table completed according to the lab notebook with correct numbers of significant figures?	1
<b>Algebraic equation:</b> Is the algebraic equation specific and correct?	1
<b>Discussion:</b> Have the questions been adequately addressed?	1
<b>Conclusions:</b> Are the conclusions consistent with the tabulated data?	1
<b>References:</b> Are the references complete and appropriate?	1
<b>Performance Evaluation:</b> Was the caffeine successfully extracted?	1
<b>Total mark</b>	12