MR. ALBERNAZ – PRE-AP (A DAY) AND REGULAR (B DAY) CHEMISTRY

AGENDAS FOR THE WEEK: *11/6 – 11/10*

	MONDAY (A DAY) 10:34AM-12:03PM	TUESDAY (B DAY) 10:34AM-12:03PM	WEDNESDAY (A DAY) 10:34AM-12:03PM	THURSDAY (B DAY) 10:34AM-12:03PM	FRIDAY (A DAY) 10:40AM-12:15PM
	Objective(s): SWBAT ***identify ions as cations and anions with proper ionic notation** ***use the octet rule to determine if an element will form a cation or anion**	**Career Fair Today**	**Test Today**	**Test Today**	**Objective(s): SWBAT** ***identify trends on the periodic table such as size, electronegative, etc.** ***use trends to compare elements based on their size and other properties**
P	**Students will begin with a warmup over lewis dot structures and valence electrons. Students will then consider metals/nonmetals through thte perspective of electrons.**				**Students will be prompted to consider the periodic table's layout. Why is it this way? Does it have to do with electrons? Properties?**
L **A**	**Students will complete a short set of notes over ions. Students will then work with cards to identify ions based on the number of electrons and protons in the element.**				**Students will complete a short page of guided notes on the periodicity. Students will then complete and activity on properties and magnitudes of different elements on the periodic table by cutting straws and placing them on an image of the periodic table.**
N	**Students will turn in their activity worksheets to be used as an evaluation. In addition, student questions throughout the lesson will gauge understanding as the activity progresses.**				**Students will turn in both their straws and the worksheets of periodicity.**