

SUPER STAR COACHING

CENTRE <u>Date:</u> 3/17/2019

House no 1552/A Ghosia colony Gulbahai

No.2, Karachi Guess paper XII

Prepared by: Mohammad Raza Ansari

SECTION 'B' (SHORT QUESTIONS' ANSWER)

INORGANIC

- 1) State periodic law, write its modification form. Why the law required modification.
- 2) Define PERIODS, GROUPS & BLOCK of long form of Periodic table
- 3) Give advantage and disadvantage of Mendeleev's Periodic table.
- 4) How is H2 prepared on large scale in pure state.(give any two preparations). \spadesuit
- 5) Why H is placed in group 1A suggest an alternate position, give reasons.
- 6) How is SODA ASH and other allied compound manufactured, write all the details of the process.
- 7) Write hydrolysis products of the following:
- (i) Boron nitride (ii) Calcium carbide (iii) Aluminum nitride (iv) Magnesium nitride
- 8) How is pure aluminum extracted from alumina OR Write the extraction of pure Cu from copper pyrite ore (CuFeS2).
- 9) Give reasons:
- (i) The ionization energy decreases down the groups of Periodic table.
- (ii) NH3 acts as ligand but NH4+ does not
- (iii) Why ionic radius of cation is smaller than its atomic radius.
- (iv) Why ΔHhyd decreases down group IA and IIA in periodic table.
- 10) Complete the reactions:

C + HNO3
$$\stackrel{\text{conc.}}{\longrightarrow}$$

S + H2SO4 $\stackrel{\text{conc.}}{\longrightarrow}$
K2MnO4 + Cl2 $\stackrel{\text{conc.}}{\longrightarrow}$
H2S + Fe3+ $\stackrel{\text{conc.}}{\longrightarrow}$

ORGANIC

- 1) How the following compounds are prepared from 2-Chloro propane.
 - (i) Propene (ii) 2-Propanol (iii) propane (iv) 2,3 di methyl butane
- 2) Write Markownikoff's Rule give its application in addition reactions of Alkene & Alkyne.
- 3) Define Electrophile write general mechanism of electrophilic substitution reaction
- 4) How an Alcohol can be converted into:
 - (i) Alkyl halide (ii) Carboxylic acid (iii) Ester (iv) Ether
- 5) Write one preparation of each of the following:
 - (i) Phenol (ii) Methyl Alcohol (iii) Acetone (iv) Acetic anhydride
- 6) Write distinction test between



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- (i) Ethene & Ethyne (ii) Benzene & Ethene (iii) Aldehyde & Ketone (iv) Aliphatic and Aromatic compound
- 7) Define and Classify any one of the following:
 - (i) Carbohydrate (ii)
 - (ii) Amino acid
- (iii) Lipids
- 8) Draw pyranose structure of α , D(+) & β -D(+) glucose.
- 9) Explain one of the following:
 - (i) Fertilizer (ii) Glass (iii) Detergent

SECTION 'C' (DETAILED QUESTIONS' ANSWER)

INORGANIC

- 1) How is H₂SO₄ or HNO₃ manufactured on large scale.
- 2) Compare the properties of Diamond & Graphite in detail.
- 3) Write short note on the following:
 - (i) Photography (ii) Bleaching Powder (iii) Corrosion and its prevention
- 4) Describe the properties of Transition metal
 - (i) Color of their compound
 - (ii) Interstitial compound
 - (iii) Variable oxidation state
 - (iv) Complex formation
- 5) Write one preparation of K₂Cr₂O₇, write one reaction each in which Potassium dichromate (K₂Cr₂O₇) & potassium Permanganate (KMnO₄) acts as an oxidizing agent.

ORGANIC

- 1) Explain any two of the following:
 - (i) Cracking Γ (ii) Polymerization & its types. (iii) Functional Group
- 2) How is ethylaicohol manufactured from starch write its conversion into following
- What are originard Reagent, with its reactivity and preparation. Convert G.R int Carboxylic Acid.
- 4) Copyert 1,2 dichloro ethane to ethene and ethyne, give the names of each reaction.
- 5) Explain Molecular Orbital theory of structure of Benzene.
- 6) Convert Benzene to
 - (i) O Nitro benzoic acid
 - (ii) Meta Nitro toluene
- 7) Describe S_N reaction, compare its two Mechanisms.

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