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CO₂ emission sources, greenhouse gases, and the global warming effect

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Abbreviations

CCS carbon capture and storage

CCSU carbon capture, storage and utilization

CFC chlorofluorocarbon
EDD ecological debt day
EOD Earth overshoot day
FCC fluid catalytic cracking

GHG greenhouse gas Gt gigatons

HFC hydrofluourocarbon

Mt metric tons ppm parts per million

ppmv parts per million volume

SG smart gridsUN United Nations

UNFCC United Nations Framework Convention on Climate Change

1.1 Introduction

One common challenge confronting many developing countries in recent times is environmental degradation as a result of the indiscriminate emission of CO₂ into the atmosphere [1]. Industrial activities have led to an increasing level of carbon emissions in the atmosphere

because of the growing level of industrialization and urbanization in many developing countries. This has also led to a significant increase in the global atmospheric concentration of anthropogenic greenhouse gases, such as CO_2 , leading to global warming and climate change. Climate change results in a decline in global agricultural output due to low rainfall, fluctuation in seasons, and temperature rise [2]. Many parts of the world are experiencing drought and becoming no longer suitable for commercial farming due to climate change. Incessant temperature and precipitation changes are also likely to lead to an increase in soil and water degradation. However, adaptive behavior has the potential to alleviate these impacts, as land use and management have been reported as having a greater impact on soil conditions than the indirect impact of climate change [3]. Even if all emissions from human activities suddenly stop, climatic conditions would continue to change [4]. The current level of anthropogenic pollution and the indiscriminate emission of greenhouse gases into the atmosphere could aggravate global warming, ocean acidification, desertification, and changing weather conditions. However, food security, rising sea levels and severe storms affecting coastal areas, health problems, migration, and the increasing economic damage are just some of the immediate implications of climate change.

As the knowledge base in the area of mitigating CO₂ emission is expanding, there is a need to update it with recent advances and future prospects in this field that could lead to a clean environment [5]. More efforts are also needed to develop strategies to address the root cause of anthropogenic climate change, as well as proffering solutions to some of the most serious effects of climate change and environmental degradation. According to Ahmed et al. [6, 7], environmental degradation is a disturbance in the balance of the natural ecosystem by which the existence of different life form is hindered, thereby affecting the survival of other lives in the ecosystem. On the other hand, the international disaster reduction strategy arm of the United Nations defines environmental degradation as reduction in the ability of the environment to meet environmental goals, as well as social and ecological needs [8].

In this study, environmental degradation is defined as the depletion of environmental resources such as air, water, and soil; the elimination of ecosystems; extinction of wildlife; and pollution, perceived to be harmful to the ecosystem. Environmental degradation is one of the 10 threats affecting the world today as documented by the United Nations high-level panel on threats, challenges, and climate change [9]. Environmental degradation exists in different forms [10]. For example, when natural territories in the ecosystem are destroyed and resources depleted, the environment deteriorates. Attempts have been made so far to combat this global problem through the protection of the environment and the management of resources. There are many examples of environmental degradation globally. A current example is the burning of Amazon which constitutes 60% of all tropical forests. Amazon is regarded as the lungs of the Earth, and its destruction poses a great threat to the environment globally [11]. Furthermore, deforestation is another example of environmental degradation and its effect leaves devastating effects on the world around us. The constant felling of trees is eliminating our oxygen supply, as well as CO₂ absorption by plants. With the continuation of deforestation, the world will have less oxygen, which could be a problem that is detrimental to human health. Another problem that results from deforestation is the excessive consumption and waste of paper products that come from the trees. The waste normally produced during

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deforestation is not usually recycled, therefore, an immense amount of waste is created. Excessive felling of trees could lead to soil degradation and constant deforestation makes the soil less nutritious, making it difficult to use again [12]. The main purpose of this chapter is to address environmental problems attributed to the increasing levels of CO₂ in the atmosphere. So far, the combustion of fossil fuel is well known as the major emitter of CO₂ to the atmosphere [13]. It is worthy to note that although other fossil fuel-linked sources of CO₂ emission (e.g., burning of oils and naturals gas) also contribute to the increasing concentration of greenhouse gases in the atmosphere, none has contributed more CO₂ to the atmosphere than coal combustion. As far as could be ascertained, the combustion of coal, especially for power generation, is the major contributor of CO₂ to the atmosphere [14–16]. The high concentration of CO_2 in the atmosphere is predicted to continuously increase if the problem of CO_2 emission is not addressed. For instance, in 2019 it was reported that the average concentration of CO₂ in the atmosphere (414.7 ppm) was 45% higher than that obtained between 1980 and 1990 [17]. It has also been reported in open literature that the concentration of CO₂ has increased by over 2ppm/year in the last decade [18]. In addition, global CO_2 emissions due to human activities have increased by over 400% since 1950 [19]. As a result, the concentration of atmospheric CO_2 has increased to 410 ppm, against 280 ppm reported since 1750 (around the start of the first industrial revolution) [20]. According to the 2015 Paris agreement, the rise in temperature of the Earth's interior should be kept below 2°C in comparison to the preindustrial levels, and the increase in Earth's temperature should be limited to below 1.5°C. According to a study by Van-Soest et al. [21], attaining the target set in the Paris agreement requires the capture and storage of at least 1 gigaton (Gt) of CO₂ annually until 2030.

In addition, it has also been documented in open literature that fossil fuel-fired power plants contribute about 33%–40% of the total CO_2 emission globally, with coal-fired power plants being the main contributor [22, 23]. For instance, a typical $500\,\mathrm{MW}$ power plant is capable of emitting about 2–3 million tons of CO_2 per year. Furthermore, as the call for the minimization of global CO_2 emissions intensifies, the need to reduce anthropogenic CO_2 emissions from power plants is becoming increasingly important. Carbon capture and storage (CCS) technology has been proposed as a potential practice that allows the continued use of fossil fuels in power plants and at the same time minimizes CO_2 emissions to the atmosphere [24, 25]. The primary purpose of capturing and storing CO_2 is to curtail CO_2 emissions.

Anthropogenic CO₂ emission occurs from direct human activities on forestry and other land use, such as deforestation, land clearing for agriculture, and soil degradation [26, 27]. Other sources of anthropogenic CO₂ identified in this study are human respiration, automobiles, power plant, and airplane emissions. However, it is noteworthy that the exhalation of CO₂ during human respiration does not make any significant contribution toward the depletion of the ozone layer (a layer in the Earth's stratosphere that prevents most of the sun's ultraviolet radiation from reaching the Earth), global warming, and climate change [28]. CO₂ emitted into the atmosphere does not lead to climate change directly. First, it depletes the ozone layer, exposes the Earth's surface to direct ultraviolet radiations from the sun which leads to an increase in the temperature of the Earth (global warming), then an incessant shift in global or regional climatic patterns of the Earth (climate change). A schematic illustration of the trend for climate change evolution is presented in Fig. 1.1.

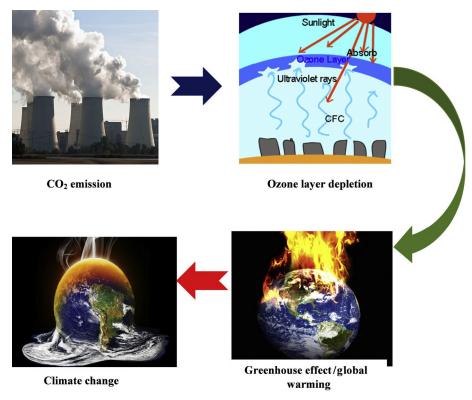


FIG. 1.1 Evolution of global warming and climate change from the combustion of fossil fuels. *Modified from How Carbon Capture Works, HowStuffWorks, 2008. 09 July. Available from: https://science.howstuffworks.com/environmental/green-science/carbon-capture.htm (Accessed 24 December 2019).*

Global attention has expressly shifted toward addressing the issue of climate change resulting from CO_2 emissions. Most studies reported in this field from the beginning of the 21st century to date focused mainly on CO_2 emission [29–32], its impacts [33–36], and remediation [37–41]. To this point, it is evident that scientists have made substantial contributions toward providing solutions to the problem of climate change by developing and testing new materials that can effectively capture CO_2 [42–47]. On a wider scope, researchers have recommended two major ways to minimize the emission of CO_2 as depicted in Fig. 1.2.

The first option shown in Fig. 1.2 will require the deployment of alternative renewable energies, or improving the energy efficiency of existing systems. However, the second solution may require the application of carbon capture and sequestration technologies to close the carbon cycle. This chapter seeks to explore the second option which suggests CO₂ capture as a more reliable alternative to minimizing CO₂ emissions to the atmosphere, thereby saving the Earth from extreme weather conditions. Carbon capture and storage (CCS) technology is widely embraced as one of the leading and easiest ways of mitigating carbon emissions to the atmosphere [48].

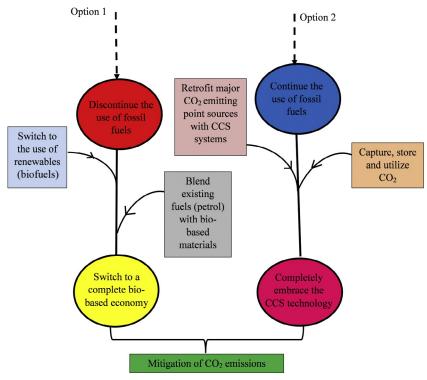


FIG. 1.2 A general roadmap toward minimizing global CO₂ emissions. *Modified from K.O. Yoro, Numerical Simulation of CO₂ Adsorption Behaviour of Polyaspartamide Adsorbent for Post-Combustion CO₂ Capture (MSc thesis), University of the Witwatersrand Johannesburg, South Africa, 2017.*

1.2 Major greenhouse gases and their forms of emission

1.2.1 CH₄ emission

The atmospheric concentration of methane has increased due to agriculture (rice and live-stock farming), coal mining, including human activities; oil and gas production and distribution; biomass combustion; and municipal landfills [49, 50]. This means that unless immediate action is taken in the aforementioned sectors, methane emission is expected to increase even beyond 2030. The energy sector, which includes coal mining, natural gas systems, oil systems as well as fixed and mobile combustion systems, is the largest source of methane emissions into the atmosphere. In 2009, approximately 303 tons of methane was emitted from these sources [51].

Emissions from agricultural activities (mainly cattle breeding) and waste management (mainly landfills) are other sources of emissions, which contribute about 216 tons of methane to the atmosphere [52]. Methane is a greenhouse gas with a global warming power estimated to be 28–36 times more than CO₂ over the last 100 years, thereby ranking it as the second most anthropogenic greenhouse gas emitted into the atmosphere [53]. Therefore, it is important to

study methane emissions to assess the impact of the biogas industry on climate change. Methane is also released in large quantities during biogas combustion [54]. However, a significant contribution to this emission comes from the waste generated during the conservation and management of biomass. On the other hand, other forms of biomass management strategies could also be considered in the future to reduce the biogenic emissions associated with methane.

In a study conducted by Poeschl et al. [55], methane emission was extensively discussed; in all cases examined by the authors, methane emission rates were below $5\,\mathrm{g\,kg^{-1}}$. In addition to cattle manure, a significant reduction in methane emission is associated with the proper processing and management of digestive products. Biomass is characterized by a high emission of methane gas after decomposition in the field without a prior treatment. In agriculture, the full implementation of a nondecomposing feeding strategy for animals can reduce the global methane emissions by 20% before 2030, while a full implementation of periodic aeration of permanently flooded rice shells can reduce methane emission by more than 30% [56]. Emissions from coal mining as well as the oil and gas sector could be reduced by more than 65% if gas leaks are prevented during transportation, distribution, recovery, and the use of gas at the production stage.

$1.2.2 N_2O$ emission

Besides carbon dioxide (CO_2) and methane (CH_4), nitrous oxide (N_2O) is another problematic greenhouse gas with a high potential of causing greenhouse effect. The emission of N_2O from biogas production processes contributes significantly to the global warming budget [57, 58]. The relative impact of N_2O on the environment largely depends on the choice of climate measures. Nonetheless, it is undeniable that the impact of N_2O can exceed the effects of CO_2 and CH_4 if the metric considered is a global temperature with a time change potential of more than 100 years [59].

Nitrous oxide can be emitted from wastewater containing organic-based nitrogen materials, such as materials from human or animal waste [60, 61]. Factors affecting the amount of nitric oxide produced by wastewater comprise temperature, acidity, biochemical oxygen demand, and nitrogen concentration. Natural sources of N_2O include soils, tundra, and oceans. Agriculture has important human resources (nitric oxide fertilizers, soil tillage), manure, biomass combustion, and industrial processes such as fertilizer production [62]. Fertilizers from nitrogen enrichment of soils are the primary human contributor to nitrous oxide emissions [63]. N_2O emission has grown by more than 35% from 2005 to date [64]. The wide uncertainty on estimates of global warming, as well as its mitigation potential largely depends on the evaluation of the emission rate of N_2O in storage and in use as a digestate fertilizer.

1.2.3 CO₂ emission

Carbon dioxide (CO_2) gets emitted when fossil fuels (e.g., coal, natural gas, and oils), solid waste, trees, and other biological materials are burnt [65]. Furthermore, CO_2 can be emitted as a result of certain chemical reactions (e.g., the production of cement). Human activities can also change the carbon cycle, thereby adding more CO_2 to the atmosphere and affecting

the capacity of natural sinks, such as forests. The burning of fossil fuels for energy and transport is the main human activity that leads to the emission of CO_2 , although certain changes in land use (e.g., deforestation) can also lead to the emission of CO_2 into the atmosphere [66]. Carbon dioxide is naturally removed from the atmosphere (or sequestered) when plants absorb it as part of the carbon life cycle. Alternatively, the emission of CO_2 from stationary point sources (e.g., power and cement production plants) can be minimized through CO_2 capture, storage, and utilization. Co-gasification of coal with biomass materials during combustion in coal-fired power plants is another proven option for mitigating CO_2 emission [67].

1.2.4 SO₂ emission

SO₂ is released during the combustion of carbonaceous fuels (coal, petroleum, and diesel) or other materials containing sulfur. The largest source of SO₂ emissions to the atmosphere is the combustion of fossil fuels from power plants and other industrial structures [68]. Smaller sources of SO₂ emissions include metal processing, foundry facilities, including natural sources such as volcanoes, locomotives (e.g., ships and other heavy vehicles), and fuel-burning equipment with high sulfur content. High concentrations of SO₂ are usually recorded near large industrial facilities and are associated with multiple effects on health and the environment [69]. According to a report by the US Environmental Protection Agency, short-term contact with SO₂ in the air can lead to a number of adverse health effects [70]. Several clinical, epidemiological, and toxicological studies confirmed the fundamental relationship between short-term exposure of SO₂ in the atmosphere and respiratory morbidity, whose observed health effects are respiratory tract infection and inflammation with many other respiratory problems [71]. Again, SO₂ combines with other air toxins to form sulfate particles, which are constituents of particulate matter (PM_{2.5}). Inhalation and exposure to PM_{2.5} [72] have different effects on cardiovascular and respiratory health problems.

In addition to the aforementioned health problems, many ecological problems are also associated with high concentrations of SO_2 in the atmosphere. For instance, SO_2 in combination with nitrogen oxides (NOx) in the air contributes to acid deposition and reduced visibility (regional turbidity indicator). Furthermore, high concentration of SO_2 can also destroy vegetation, (e.g., increased leaf damage, stunted plant growth, low crop yield, and decline in a variety of plant species in the ecosystem). A high concentration of SO_2 and other sulfur oxides in the atmosphere also leads to acid rain which can harm sensitive parts of the ecosystem. Finally, SO_2 can accelerate the corrosion of materials used in buildings, statues, and monuments (e.g., metals, concrete, limestone) [73]. Several techniques have been suggested for reducing the concentration of SO_2 in the atmosphere [74, 75]. A stepwise diagrammatic procedure to minimize SO_2 emission from power plants is presented in Fig. 1.3.

1.3 Greenhouse gases and the greenhouse effect

Greenhouse gases (GHG) are gaseous compounds that can emit ultraviolet radiation within a certain thermal infrared range [76]. Greenhouse gases retain high temperatures in the lower atmosphere, thus allowing less heat to escape back to space. This subsequently

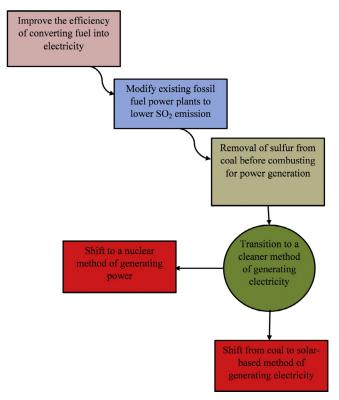


FIG. 1.3 Stepwise procedure to minimize SO₂ emission from power plants.

results in the greenhouse effect and global warming. The greenhouse effect is a natural process that warms the Earth's surface to a temperature above which it would be without the atmosphere. The intensity of the greenhouse effect depends largely on the temperature of the atmosphere, and on the presence of greenhouse gases in the atmosphere. Greenhouse gases are vital for supporting a habitable temperature for the Earth because if there were totally no greenhouse gases present in the atmosphere, the average surface temperature of the Earth would be about -18° C [77]. Common greenhouse gases present in the atmosphere include water vapor, chlorofluorocarbons (CFCs), hydrofluorocarbons (HFCs), carbon dioxide (CO_2) , methane (CH_4) , nitrous oxide (N_2O) , and ozone (O_3) . However, researchers have pointed out that the four main greenhouse gases attracting serious global attention today are CO_2 , SO_2 , CH_4 , and N_2O [78]. Although water vapor is arguably the most abundant greenhouse gas naturally present in the atmosphere, CO₂ is the most emitted greenhouse gas. Hence, different techniques have been reported to capture CO₂ [79–81]. A generalized classification of activities leading to the emission of greenhouse gases into the atmosphere is depicted in Fig. 1.4, while the main greenhouse gases and their major sources are presented in Table 1.1.

Information provided in Table 1.1 reveals that because of the high dependence on burning fossil fuels for power generation and other industrial activities, CO₂ is now the most emitted

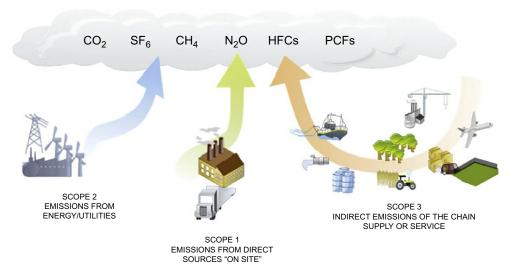


FIG. 1.4 Classified sources of greenhouse gas emission. *Adapted from B.R. Singh, Global Warming: Impacts and Future Perspective, BoD – Books on Demand, 2012.*

TABLE 1.1 Greenhouse gases and their major sources.

Greenhouse gases	Sources	% Emission in 2019
Carbon dioxide (CO ₂)	Fossil fuel combustion, deforestation	76
Methane (CH ₄)	Biomass combustion, agricultural wastes	13
Nitrous oxide (N ₂ O)	Fertilizer use	3
Sulfur dioxide (SO ₂)	Combustion of coal, oil, and diesel	7
Fluorinated gases (CFCs, HCFs)	Refrigeration	1

Modified from T.J. Blasing, Recent Greenhouse Gas Concentrations, Environmental System Science Data Infrastructure for a Virtual Ecosystem, Carbon Dioxide Information Analysis Center (CDIAC), Oak Ridge National Laboratory (ORNL), Oak Ridge, TN, United States, 2016.

greenhouse gas while fluorinated gases (CFCs and HCFs) are the least emitted greenhouse gases because the use of refrigerants has been phased out in many refrigeration systems in recent times [82].

1.4 CO₂ as a major contributor to global warming and climate change

From the beginning of the second industrial revolution that was characterized by mass production and electricity generation, energy production was largely contingent on the combustion of fossil fuels. Power generation, cement production, and the use of fossil fuels in many homes have resulted in the emission of significant amounts of CO₂ to the atmosphere over the past decades. Researchers have identified that between 1970 and 2002, global

emission of CO_2 increased by over 70% [83]. Since then, there have been great concerns over the ever-increasing atmospheric concentration of CO_2 as well as its hard effect on the environment [84]. Capturing CO_2 from large point sources is necessary to minimize the high concentration of CO_2 emitted into the atmosphere. According to a study reported by Schubert and Jahren [85], there are greater natural sources of CO_2 emissions (e.g., decomposition, ocean release, and respiration) than there are sources of CO_2 emission due to human activities such as the burning of oil, coal, and gas; deforestation; industrial manufacturing; and the use of aerosols. However, the natural sources of CO_2 are closely balanced by naturally occurring phenomena such as photosynthesis and weathering of rocks. This balance has made the atmospheric concentration of CO_2 to be as low as 260–280 ppmv for 10,000 years before the start of the industrial era [86]. To date, most anthropogenic (human) sources of CO_2 emission are still from everyday human activities that we do not even think about.

During the industrial era, the demand for electricity was on the increase and power generation increased by almost 50% at the beginning of 2000 [87]. The global fossil fuel estimate of CO_2 emission also increased to about 25,000 metric tons in 2002. Fossil fuels will still remain the primary source of energy which will be on the increase by over 90% [88]. This shows that global energy demand has been on the increase, and it is expected to escalate by over 67% by 2030 [89]. In view of this, it is necessary to identify the major sources of CO_2 emission and how to prevent it. Table 1.2 presents various CO_2 emission sources and how to minimize emissions from such sources.

Information provided in Table 1.2 reveals that the anthropogenic emission of CO_2 has been the major source of CO_2 emission to the atmosphere with fossil fuel combustion engines and coal-fired power plants as the major contributor. Most of the major anthropogenic emissions are from stationary point sources, hence this study suggests carbon capture, storage, and

TABLE 1.2 Major sources of CO_2 emission and their preventive methods.

Sources	CO ₂ emission (billion Mt)	Proposed preventive option	References
Anthropogenic/human sources			
Fossil fuel combustion engines	392	CCSU	[90]
Cement production plants	113	CCSU	[91]
Power generation (coal-fired power plants)	279	CCSU, integration to methanol plant	[90]
Transportation	191	Blending fuels with biomass	[92]
Industrial manufacturing	178	CCSU	[93]
Land use changes	13	-	[94]
Nonanthropogenic/natural sources			
Plant, animal, and human respiration	7	-	[95]
Ocean-atmosphere exchange	7	_	[96]
Soil respiration and decomposition	1.54	-	[97]
Volcanic eruptions	0.15	-	[98]

utilization (CCSU) as the most suitable preventive option for CO₂ emission. CCSU simply entails the capture of waste carbon dioxide (CO₂) from large stationary point sources (e.g., coal or gas-fired power plants) and its deposition in a safe storage location where it will not gain access back to the atmosphere [99]. The release of CO₂ and other greenhouse gases from burning fossil fuels come into play in the following major areas:

- **Power industry**: Electricity generation in fossil fuel-fired power plants mostly comes from the burning of fossil fuels (e.g., coal and natural gas) for energy. If other cleaner alternatives for power generation such as nuclear and solar sources are not explored, the power industry will continue to emit large amounts of CO₂ to the atmosphere.
- Transport industry: The combustion of fossil fuels (e.g., coal, diesel, petrol) to run automobiles (e.g., cars, trucks, trains) and jets also contribute to the high levels of CO₂ in the atmosphere leading to global warming and climate change. Emissions from automobiles are nonstationary with high volumes of CO₂. So far, the blending of automobile fuels with bio-based materials [100] or switching to complete use of biofuels or electricity to power automobiles [101] has been the most promising option for mitigating CO₂ emissions from this sector.
- **Buildings**: The emissions of greenhouse gases (e.g., CO₂, CH₄, SO₂) from businesses, homes, and industries arise mainly from waste disposal, the combustion of fossil fuels for energy utilization, and the use of certain materials that contain greenhouse gases, which are the major sources of emissions from this sector. To date, no definite technology has been reported for mitigating CO₂ emission from homes and other businesses.

Power generation from the combustion of fossil fuels is the major contributor to CO_2 emission across major sectors of the economy in many developing countries. Furthermore, the agricultural sector has been proven to be the least contributor since 2018 [102]. A general representation of global CO_2 emission by sector is expressed in percentages and presented in Fig. 1.5.

If the random emission of important anthropogenic greenhouse gases like CO_2 is not reduced, the Earth will face the following devastating consequences [103–106]:

- (i) The atmospheric temperature in coastal areas will increase by 2°C by 2050 and 4°C by 2100.
- (ii) The interior temperature of the Earth will also increase by 4°C by 2050 and 7°C by 2100.

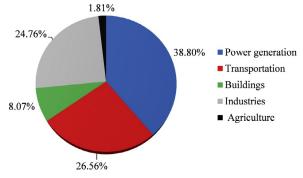


FIG. 1.5 Updated global CO₂ emission by sector in 2019.

- (iii) World food security will be adversely affected.
- (iv) Invasive plants will increase and negatively influence the water resources on Earth.
- (v) Health problems will be aggravated worldwide due to droughts and floods, and there will be an outbreak of diseases related to extreme weather conditions.
- (vi) Forests and various commercial plantations on Earth will be vulnerable to forest fires.

Although global warming has not been even across the Earth, the rising global average temperature shows that more regions are warming than cooling [107]. According to a recent global climate report, the combined temperature of land and ocean increased at an average rate of 0.07°C per decade since 1880 [106]. This means that the average rate of temperature increase from 1981 to date is more than double (0.17°C). Models predict that global surface temperatures will be 0.5°C warmer than the 1986–2005 average by 2020, regardless of the carbon emission pathway the world follows [108]. Furthermore, the high thermal capacity of water shows that the temperature of the ocean does not react instantly to the increase in heat trapped by greenhouse gases. Previous studies have shown that for the past 50 years, the global temperature rose at an average rate of about 0.13°C per decade, almost twice as fast as the increase of 0.07°C per decade observed over the past 50 years [109]. Researchers have predicted that the global average temperature will increase by about 0.2°C per decade over the next two decades [110]. It is envisaged in this study that by 2030, the warming imbalance caused by greenhouse gases will gradually overcome the thermal inertia of the oceans, and the projected temperature paths will start to diverge with unchecked carbon emissions, thereby leading to several additional degrees of warming by the end of the century.

1.5 Current CO₂ emission trends

Global CO₂ emissions have been growing on the average in recent times, with CO₂ emissions 1.9% higher in 2019 than in 2018 [111]. According to Janssens-Maenhout et al. [112], the growth in total global greenhouse gas (GHG) emissions (excluding land-use change) in 2018 was at a rate of 2%, equivalent to 51.8Gt of CO₂ with an annual growth rate of about 1.3%. Global greenhouse gas emissions in 2018 amounted to 55.6Gt CO₂ including those for land-use change. This increase occurred as the global economic growth in 2018 increased to an average annual rate of 3.5% since 2012 [113]. Current emissions of greenhouse gases that exclude those of land-use change today is approximately 57% higher than in 1990 and 43% in 2000.

In 2018, the 2% (1.0 Gt CO_2 equivalent) increase in global greenhouse gas emissions was primarily due to an increase in global fossil fuel combustion and those from noncombustion industrial processes, including cement production [114]. Global emissions of methane (CH₄) and nitrous oxide (N₂O) increased by 1.8% and 0.8%, respectively [115]. Global emissions of fluorinated gases continued (so-called F gases) increased by about 6% in 2019, contributing to the total growth of 2.0% greenhouse gas emissions. The 2.0% growth in greenhouse gas emissions is higher than the average annual increase of 1.2% since 2012 but less than the 2.5% increase in the first decade of this century (2010). Fossil fuel-related CO_2 emissions are the biggest source of greenhouse gas emissions, with a contribution of about 72%, followed by CH_4 (19%), N_2O (6%), and F gas (3%) [116]. The increase in greenhouse gas emissions in recent

years and the highest rate of 2.0% in 2018 is quite similar to the increase in CO_2 emissions in 2018, which contributes nearly 75% of the total greenhouse gas emissions (excluding those related to land-use change) [117].

Generally, 2019 was one of the five hottest years (2015–19) since recordings of global temperature rise started in 1880 [118]. Of the 10 warmest years since 1880, about nine have occurred after 2005. In 2019, temperatures in many parts of the world were warmer than average. Record warm temperatures have been measured across Europe, the Middle East, New Zealand, and parts of Asia. A heatwave of unprecedented intensity and duration hit Europe from April 18 to 22, 2019. France, Germany, and Switzerland had their hottest year in 2019 since 1880 while the Netherlands had its second-hottest year (2014 being the record) [119]. The deviation from the average global temperature level was 0.97°C above the average between 1880–1900, which is slightly lower than that of 2015–17 [120].

The incessant increase in CO_2 emission from 2018 has made the Earth to record the earliest Earth overshoot day in 2019 since 1970. Earth Overshoot Day (EOD), which is also referred to as Ecological Debt Day (EDD), is a calculated calendar date in which humanity's annual resource depletion surpasses the Earth's ability to produce resources [121]. EOD can be expressed as a quotient of the Earth's biocapacity (the amount of natural resources produced by the Earth within 365 days) and the Earth's ecological outline (annual human consumption of the Earth's natural resources), multiplied by 365 which is the number of days in a common Gregorian calendar year. From an economic standpoint, EOD can be defined as a day when humanity comes into environmental shortages, while in the field of ecology EOD is an example of the level at which human population transcends its environmental capabilities [122]. The EOD for 2019 was recorded on July 29, which is the earliest to date. Mathematically, Earth Overshoot Day (EOD) is expressed as

$$EOD = \frac{World\ biocapacity}{World\ ecological\ footprint} \times 365$$
 (1.1)

In addition, studies have shown that since the beginning of the second industrial revolution, global CO_2 emissions have continuously increased [123]. In view of this, an updated global CO_2 emission trend for 2019 is presented in this chapter as compiled from previous reports [124–128]:

- The current global CO₂ emission from the combustion of fossil fuel alone is about 58.7 billion tons in 2019.
- 2018 was the sixth consecutive year of large global increases in GHG emissions. In April and May 2018, atmospheric CO₂ concentrations averaged more than 410 ppm at the NOAA Mauna Loa Observatory, thereby surpassing another climate milestone.
- CO₂ emissions are responsible for an estimated 60% of humanity's demand on nature. As a result, Earth Overshoot Day is advancing each year, and humanity is now using nature 1.75 times as fast as the planet's ecosystems can regenerate.
- So far, CO₂ emissions have increased by 0.34% over the previous year, representing an increase of about 122 million tons over 2018, when CO₂ emission was about 35.6 billion tons.
- In 2015, CO₂ concentrations exceeded the symbolic reference by 400 parts per million (ppm) in the northern hemisphere.

- After a relatively decelerating trend since 2012 and 3 years without change, global energy-related CO₂ emissions increased by 1.4% in 2017 and 1.7% in 2018, reaching a record level of 33.1 Gt of CO₂.
- Global carbon emissions from the combustion of coal (the worst pollutant of all fossil fuel) unexpectedly declined by about 0.9% in 2019, However, this drop was offset by strong growth in the use of oil and natural gas worldwide. CO₂ emissions per capita worldwide is equivalent to 4.99 tons per person (based on a world population of 7.7 billion in 2019).
- The UNEP assessment data shows a large gap between emission levels and those corresponding to 2°C and 1.5°C paths, respectively, making it increasingly difficult to reach the 2030 emission targets.

1.6 Mitigation of CO₂ emission in the industries

1.6.1 Power generation

In many countries today, electricity is generated by burning fossil fuels because it is cheap and the source of coal as a raw material is readily available [129]. Coal contains more carbon than petroleum or natural gas, which translates into higher volumes of CO_2 emissions per unit of electricity produced [130]. Since the power sector is widely regarded as the largest source of CO_2 emissions, it is imperative to explore ways to minimize CO_2 emissions from this important sector. The immediate option to mitigate carbon emission from coal-fired power plants could be achieved by

- Operating the most efficient power plants to reduce the fuel supply per unit of energy.
- Investing in technologies that can increase the efficiency of coal plants from 35% to 50%.
- The use of the combined cycle in coal-fired power plants by coal gasification.
- The conversion of gas turbines in open cycle mode to combined cycle mode to increase the efficiency of the installation. However, the real reduction in emissions can only be achieved by radically changing the combination of electricity production.

From the application of energy mainly powered by fossil fuels today, there is a need to move to a combination of nuclear and renewable energy. More wind and solar power plants, as well as nuclear power plants, are the only reasonable and cheapest solutions for power generation with reduced emissions [131].

As the search for knowledge and better alternatives to minimize CO₂ emission is increasing in this fourth industrial revolution era, researchers have proposed a new way to address this challenge. For example, Bari et al. [132] proposed the application of smart grid applications during power generation. A smart grid (SG) is a concept or framework for a modernized, scalable, and next-generation power supply system [133]. It is also portrayed as an integration of components, subsystems, and ancillary services controlled by highly intelligent control and management systems throughout the power system. In addition, a smart grid system is a combination of enabling technologies (hardware, software, or practices) that make the energy infrastructure more secure, reliable, efficient, and sustainable. Also called the smart or future network, the term SG is that of a "digital update." SG intelligence is applicable at the decision-making level of all relay-run computer programs, innovative electronic designs, mail automation systems, and control centers. SG technologies can contribute to reducing CO₂ emissions,

improving efficiency and storage by facilitating the integration of renewable energy sources, and enabling the use of hybrid electric vehicles [134].

1.6.2 Cement industry

The methodical approach used for mitigating CO_2 emission in the cement industry is similar to the approach used in the iron and steel industries [135]. The scenario for mitigating CO_2 emission in the cement industry is based on

- capital turnover with a continuous shift from wet production processes to dry production
- increased use of alternative additives to replace cement clinker
- fuel change from coal and petroleum coke to biomass or co-gasified coal + biomass

In addition, modern methods of dry production (i.e., rotary kilns) should replace the extruded production capacity with preheaters and precalciners. Similarly, the production of white cement via clinker production could be replaced with a new one in white cement factories. Clinker production is the largest part of the cement industry's direct release of CO₂. About 60% of carbon dioxide emitted from this sector is from calcination and another 40% from combustion [136, 137], thereby reducing the clinker content of the final cement produced in the process. Coal and petroleum coke continue to dominate gasoline in most cement industries. The supply of other fuels for cement production is expected to increase from 18% in 2010 to 40% by 2050, with 40% consisting of pure biomass. Carbon dioxide emission from the cement industry is calculated as the total carbon dioxide released from the process and from the fuel [138].

1.6.3 Petrochemical plants

Oil refineries produce different fuels and chemical feedstock by the distillation of crude oil, followed by reforming and cracking. CO₂ is produced in the streams of the petrochemical process mainly by reactions with oxygenates (mainly oxygen as in the oxidation of ethylene and water as in the reaction of water-gas change) [139]. The fluid catalytic cracking (FCC) unit is responsible for about 20%–50% of the CO₂ produced in a petrochemical plant [140]. Other petrochemical units such as the ethylene oxide and ammonia processes also produce high-purity CO₂ as a by-product. Most of the CO₂ in the petrochemical industry is emitted from the combustion of fuel oil and combustible gas [141]. CO₂ capture from the streams and flares of the petrochemical process has been recognized as one of the many strategies necessary to mitigate greenhouse gas emissions from petrochemical plants [142]. CO₂ can be captured in petrochemical plants without allowing it to emit into the atmosphere by retrofitting capture systems in the petrochemical plants. Absorption, adsorption, and membrane separation technologies are the most popular techniques that can be used to mitigate CO₂ emission in the petrochemical industry.

1.6.4 Iron and steel industry

The iron and steel industry is one of the largest industrial emitters of CO_2 accounting for about 7% global anthropogenic CO_2 emissions [143]. As a result, researchers are shifting attention toward reducing CO_2 emission from this important sector. Global steel production is

also expected to increase by 30% by 2050. Therefore, there is an urgent need to explore techniques that can effectively reduce CO_2 emissions from the iron and steel industry. The most relevant greenhouse gas emitted in the steel industry globally is CO_2 . On an average, about 1.83 tons of CO_2 were emitted for each ton of steel produced between 2017 and 2018 [144]. The iron and steel industry generates about 7%–9% direct CO_2 emission from the burning of fossil fuels [145]. Existing technologies, such as dry coke cooling, pressure recovery turbines, continuous casting, and furnace gas recovery facilities, can be used to improve plant efficiency and minimize CO_2 emissions.

The estimated potential to reduce CO_2 emissions in the iron and steel industry is around 27 Mt/year, and this can be achieved through:

- Energy optimization—By upgrading to newer technologies that use energy more efficiently.
- Fuel change—Using an alternative cleaner and less expensive fuel instead of coal.
- Recycling of scrap metal instead of forging new steel.
- Training and creating awareness among iron and steel workers on the different ways to optimize the use of machines, equipment, processes, and other steps to efficiently operate steel production with reduced CO₂ emission.

So far, different literature reports have revealed that some techniques for reducing CO_2 emission can be profitable, and generate other benefits (such as better environmental compliance, health benefits, etc.), although some could be expensive to implement on a large scale [146]. In addition, a transition to a more efficient industrial process or the application of carbon capture and storage technique to mitigate process emissions during operations in iron and steel industries could considerably reduce CO_2 emissions.

1.7 International treaties and limitations on the reduction of CO₂ emission

Climate change is a well-known global problem that is now affecting every country on every continent around the globe. It is detrimental to the national economy and also affects life and communities. Climate patterns are changing, sea levels are rising, weather events are becoming more intense, and greenhouse gas emissions have reached all-time highs because of human activities that are environmentally unfriendly [147].

The devastating consequence of climate change is likely to continue ravaging the Earth if activities leading to the emission of greenhouse gas to the environment are not addressed. Without any action toward mitigating global CO₂ emission, the average surface temperature of the world is likely to exceed 3°C in this century [148]. To proffer solution to the problem of global warming and climate change, different countries all over the world came together to sign treaties that pledge their allegiance toward minimizing the emission of CO₂ to the atmosphere. The signed treaties ranged from the 1979 Geneva Convention on long-distance cross-border air pollution [149], the Montreal Protocol of 1987 [150], 1997 Kyoto protocol [151], the 2012 Doha amendment [152], and the 2015 Paris agreement which became effective in November of 2016 [153, 154].

The Kyoto Protocol is an international accord spearheading the 1992 United Nations Framework Convention on Climate Change (UNFCCC), which obliges participating countries to reduce their greenhouse gas emissions through scientific agreements which are based on the broad facts that

- Global warming is real, and will gradually ravage the world if left unchecked.
- The anthropogenic emission of CO₂ into the environment is the leading cause of climate change.

The Protocol was adopted in Kyoto, Japan on December 11, 1997, and implemented on February 16, 2005. The Kyoto Protocol currently has about 192 signatory countries. The Treaty is based on the principle of joint but separate responsibility. The Kyoto protocol recognizes that different nations have diverse competencies to combat climate change, depending on their level of economic advancement. Hence, it places an obligation to limit carbon emissions by developed countries that have historically contributed to the present high level of greenhouse gas concentration in the atmosphere.

The paramount accounting era of the Kyoto Protocol began in 2008 and ended in 2012. The 36 nations that wholly took part in the first accounting era completed the conventions. Nevertheless, nine nations took advantage of its flexibility to fund emissions drops in additional countries since their carbon emissions were marginally exceeding their marks [155]. A significant reduction of greenhouse gases was recorded between 1990 and 1995 by countries in the former Eastern coalition after the collapse of the Soviet Union [156]. It is worth noting that although about 36 developing nations were able to drop down their emissions from fossil fuel combustion, worldwide carbon emissions still went up by 32% between 1990 and 2010 [157].

In 2012, another obligatory window, referred to as the Doha amendment to the Kyoto Protocol, was enacted, with 37 nation-states having mandatory objectives. Australia, Belarus, Iceland, Kazakhstan, Liechtenstein, Norway, Switzerland, Ukraine, and the European Union (comprising its 28 member states) declared intention to quit from the Kyoto Protocol or they may not implement the amendment in the second round commitment [158]. Japan, New Zealand, and Russia took part in the first cycle of the Kyoto Protocol but did not reach new targets during the second commitment period [159]. Canada (which withdrew from the Kyoto Protocol in 2012) and the United States (which have not ratified it) are among other developed countries that do not have goals in the second commitment period. By December 2019, about 135 countries had acknowledged being part of the Doha amendment, but its full enforcement requires acceptance from at least 144 countries [160]. So far, out of the 37 countries with binding commitments to the Doha amendment, 7 have ratified it. Furthermore, negotiations took place at the annual UNFCCC conference on decisions to be implemented at the culmination of the second obligatory period in 2020 [161]. This led to the enactment of the 2015 Paris Agreement, which is a discrete instrument under the UNFCCC instead of amending the Kyoto Protocol. As of April 2018, 175 countries had ratified the Paris Agreement, and 10 developing countries submitted preliminary editions of their climate change adaptation plans.

Prior to the 1997 Kyoto Protocol, two other international environmental agreements had been signed to indirectly address climate change [162]. Firstly, the 1987 Montreal Protocol on materials that affect the ozone layer which by law requires the parties to withdraw the use of chlorofluorocarbons in 1996 [163]. Secondly, the 1979 Geneva Convention on

long-range transboundary air pollution and its protocols regulate the emission of harmful gases, some of which are precursors of greenhouse gases. The main limitation, however, is that these agreements do not solve a complex series of interdependent climate problems.

1.8 Future prospects of heat and mass integration during CO₂ separation

Although the combustion of fossil fuels for power generation and other industrial activities have been publicized to be a major contributor of CO_2 to the atmosphere, it is currently the cheapest form of energy generation [164]. As a result, there is a need to develop new techniques to continue the combustion of fossil fuels in the power industry and yet keep a near-zero emission of CO_2 . In view of this, we suggest the retrofitting of CO_2 capture systems in existing fossil fuel-fired power plants. Many industries are refocusing on economically and environmentally benign designs to revamp their design and operational requirements to attain the objective of satisfying both environmental regulations and minimum energy consumptions in chemical processes [165]. Consequently, Several techniques (such as absorption, adsorption, membrane, and cryogenic separation) have been developed and tested for CO_2 capture in the past [136]. However, recent studies identified that these techniques are highly energy and material intensive in nature with high process cost, thereby rendering it an expensive technology to implement in most power plants [166]. Researchers have also suggested the use of phase change solvents, as well as additives and inhibitors (e.g., piperazine) to minimize energy consumption during CO_2 capture [167].

Although the aforementioned suggestions have recorded successes in minimizing energy usage during CO_2 capture to an extent, the major limitation identified with the approach is that it subsequently results in increasing process cost. Again, the aforementioned techniques have only shown effectiveness to an extent in selected CO₂ capture techniques such as gasliquid separation systems (e.g., absorption), and have not been fully extended to gas-solid separation systems such as adsorption, membrane, and cryogenic separation [168]. There is a need to develop new techniques that can tackle the high energy and material requirement in both absorptive and adsorptive CO₂ separation systems at an optimal process cost. As a result, we suggest the application of heat and mass exchanger network synthesis during CO_2 capture to achieve significant energy and material saving, and yet minimize CO_2 emission to the atmosphere. The proposed technique does not incur high process costs and it can be used to not only minimize energy consumption, but also curtail material requirements during CO₂ capture. So far, this approach is scantily reported in the literature. As far as could be ascertained, no study has adequately discussed the synthesis of a network (considering both heat and mass exchange) for CO₂ capture with the aim of optimizing the system toward attaining minimum consumption of energy and material during CO₂ capture, at an optimal operating cost.

Despite the very promising potentials of heat and mass integration in energy and material recovery studies, only a few reports have considered the application of process synthesis techniques through heat and mass exchange networks to abate excessive energy and material consumption in some industrial applications [169, 170]. This research field is an important area for future development in the use of process synthesis methodologies to handle energy

recovery tasks in environmental remediation studies, such as CCS. The application of the heat and mass integration techniques suggested in this chapter may be considered as a joint network of heat and mass exchangers connected to a regeneration network capable of taking energy savings and the material consumed into account. This research area has not been adequately explored in the past for CO_2 capture, thereby making it an interesting area to delve into. Furthermore, it is also envisaged in this chapter that a combination of synthesis techniques in a single procedure for heat and mass integration could be a more reliable approach for integrating heat and mass in CO_2 capture systems. This means that a fully optimized hybrid network approach can be attempted during the integration of heat and mass in future CO_2 capture research. A life-cycle assessment for heat and mass exchangers may also be explored in future R&D to gain an insight into their conservational influences on CCS research.

1.9 Conclusions and future outlook

 CO_2 emission, global warming and its resultant effect (climate change) are serious environmental problems that need urgent scientific attention, thereby calling for a reduction in CO₂ emissions in all sectors of human activity, including power generation. Furthermore, the literature reviewed in this study shows that researchers have put in tremendous efforts to reduce the emission of anthropogenic CO_2 into the atmosphere. Similarly, this chapter confirms that different techniques such as adsorption, absorption, and membrane separation have been proposed and tested in the literature for CO₂ capture. The available CO₂ capture techniques are highly energy and material intensive in nature, which makes CO_2 capture an expensive technology to implement in many countries. However, this chapter suggests that the application of process integration techniques through heat and mass exchanger network synthesis can be extended to reduce high energy and material requirement in the aforementioned CO₂ capture techniques. Also, this chapter revealed that CO₂ emission does not directly result in climate change. Instead, it first depletes the ozone layer, leading to global warming (the greenhouse effect), then climate change. In addition, the research efforts documented in this chapter reveal that human activities such as power generation, cement production, and transportation are the major contributors to greenhouse gas emission and global warming. This chapter establishes that the combustion of fossil fuels (e.g., coal) for power generation may still continue with the incorporation of CO₂ capture mechanism to the existing power plants. At the same time, introduction of CO₂ capture strategies to the existing power plants is possible at minimized cost through the application of CO₂ capture with integration of heat and mass exchanger networks in the capture process. Finally, this chapter reveals that some of the international treaties signed by participating countries pledging allegiance to reducing the emission of greenhouse gases are still faced with some limitations ranging from withdrawal by some developed countries, inadequate memberships for full implementation, and incomplete implementation of the signed agreement by some participating countries. Hence, the need to search for other alternatives to protect the environment and prevent future environmental disasters from increasing CO₂ emission resulting from human activities.

- S.I. Seneviratne, M.G. Donat, A.J. Pitman, R. Knutti, R.L. Wilby, Allowable CO₂ emissions based on regional and impact-related climate targets, Nature 529 (7587) (2016) 477–483.
- [2] P.T. Sekoai, K.O. Yoro, Biofuel development initiatives in sub-Saharan Africa: opportunities and challenges, Climate 4 (2) (2016) 33.
- [3] Y. Kang, S. Khan, X. Ma, Climate change impacts on crop yield, crop water productivity and food security—a review, Prog. Nat. Sci. 19 (12) (2009) 1665–1674.
- [4] S. Vijaya Venkata Raman, S. Iniyan, R. Goic, A review of climate change, mitigation and adaptation, Renew. Sust. Energ. Rev. 16 (1) (2012) 878–897.
- [5] R.G. Newell, Literature review of recent trends and future prospects for innovation in climate change mitigation. OECD Environment Working Papers, 9 (2009). https://doi.org/10.1787/218688342302.
- [6] K. Ahmed, M. Shahbaz, A. Qasim, W. Long, The linkages between deforestation, energy and growth for environmental degradation in Pakistan, Ecol. Indic. 49 (2015) 95–103.
- [7] N. Ahmed, T.I. Khan, A. Augustine, Climate change and environmental degradation: a serious threat to global security, Eur. J. Soc. Sci. Stud. 3 (1) (2018) 161–172.
- [8] Glossary of Basic Terminology on Disaster Risk Reduction UNESCO Digital Library, 2010. Available from: https://unesdoc.unesco.org/ark:/48223/pf0000225784. Accessed 24 December 2019.
- [9] H.G. Brauch, et al., Coping With Global Environmental Change Disasters and Security: Threats Challenges Vulnerabilities and Risks, Springer Science & Business Media, Berlin, Heidelberg, 2011.
- [10] A. Castiaux, Developing dynamic capabilities to meet sustainable development challenges, Int. J. Innov. Manag. 16 (6) (2012) 1240013.
- [11] H. Zhao, Climate change and sustainable development, in: H. Zhao (Ed.), The Economics and Politics of China's Energy Security Transition, Academic Press, Cambridge, MA, 2019, pp. 277–305.
- [12] L. Wenhua, Degradation and restoration of forest ecosystems in China, For. Ecol. Manag. 201 (1) (2004) 33-41.
- [13] R. Thiruvenkatachari, S. Su, H. An, X.X. Yu, Post combustion CO₂ capture by carbon fibre monolithic adsorbents, Prog. Energy Combust. Sci. 35 (5) (2009) 438–455.
- [14] A. Meisen, X. Shuai, Research and development issues in CO₂ capture, Energy Convers. Manag. 38 (1997) 37–42.
- [15] A. Williams, Combustion and Gasification of Coal. Routledge, New York, 2018. https://doi.org/ 10.1201/9781315139746.
- [16] H. Jin, K. Luo, O. Stein, H. Watanabe, X. Ku, Coal and biomass combustion, J. Combust. (2018) Available from: https://link.galegroup.com/apps/doc/A584177224/AONE?sid=lms. Accessed 24 December 2019.
- [17] F. Apadula, C. Cassardo, S. Ferrarese, D. Heltai, A. Lanza, Thirty years of atmospheric CO₂ observations at the plateau rosa station Italy, Atmosphere 10 (7) (2019) 418.
- [18] K. Kelektsoglou, Carbon capture and storage: a review of mineral storage of CO₂ in Greece, Sustainability 10 (12) (2018) 4400.
- [19] J. Maximillian, M.L. Brusseau, E.P. Glenn, A.D. Matthias, Pollution and environmental perturbations in the global system, in: M.L. Brusseau, I.L. Pepper, C.P. Gerba (Eds.), Environmental and Pollution Science, third ed., Academic Press, Cambridge, MA, 2019, pp. 457–476.
- [20] M. Santetti, A. Marqueti, H. Morrone, Technical progress in GDP production and CO₂ emissions in Brazil: 1970–2012, in: Digital Repository Economic Commission for Latin America and the Caribbean, 2018, pp. 119–132.
- [21] H.L. Van-Soest, et al., Early action on Paris Agreement allows for more time to change energy systems, Clim. Chang. 144 (2) (2017) 165–179.
- [22] Z.H. Tian, Z.L. Yang, Scenarios of carbon emissions from the power sector in Guangdong province, Sustainability 8 (9) (2016) 863.
- [23] Z. Zhang, Making the transition to a low-carbon economy: the key challenges for China, Asia Pac. Policy Stud. 3 (2) (2016) 187–202.
- [24] K.O. Yoro, Numerical Simulation of CO₂ Adsorption Behaviour of Polyaspartamide Adsorbent for Post-Combustion CO₂ Capture, (MSc thesis). University of the Witwatersrand, Johannesburg, South Africa, 2017.
- [25] B. Koerner, J. Klopatek, Anthropogenic and natural CO₂ emission sources in an arid urban environment, Environ. Pollut. 116 (2002) 45–51.
- [26] V.G. Azevedo, S. Sartori, L.M.S. Campos, CO₂ emissions: a quantitative analysis among the BRICS nations, Renew. Sust. Energ. Rev. 81 (2018) 107–115.

- [27] P. Freund, 1—Anthropogenic climate change and the role of CO₂ capture and storage (CCS), in: J. Gluyas, S. Mathias (Eds.), Geological Storage of Carbon Dioxide (CO₂), Woodhead Publishing, Sawston, Cambridge, 2013, pp. 3–25.
- [28] K.O. Yoro, P.T. Sekoai, The potential of CO₂ capture and storage technology in South Africa's coal-fired thermal power plants, Environment 3 (3) (2016) 24.
- [29] J.W. Sun, An analysis of the difference in CO₂ emission intensity between Finland and Sweden, Energy 25 (11) (2000) 1139–1146.
- [30] J.W. Sun, Is CO₂ emission intensity comparable? Energy Policy 28 (15) (2000) 1081–1084.
- [31] K. Liaskas, G. Mavrotas, M. Mandaraka, D. Diakoulaki, Decomposition of industrial CO₂ emissions: the case of European Union, Energy Econ. 22 (4) (2000) 383–394.
- [32] Z.X. Wang, Q. Li, Modelling the nonlinear relationship between CO₂ emissions and economic growth using a PSO algorithm-based grey Verhulst model, J. Clean. Prod. 207 (2019) 214–224.
- [33] X. Zheng, D. Streimikiene, T. Balezentis, A. Mardani, F. Cavallaro, H. Liao, A review of greenhouse gas emission profiles, dynamics, and climate change mitigation efforts across the key climate change players, J. Clean. Prod. 234 (2019) 1113–1133.
- [34] Y. Dong, X. Jiang, M. Ren, J. Yuan, Environmental implications of China's wind-coal combined power generation system, Resour. Conserv. Recycl. 142 (2019) 24–33.
- [35] M.A.H. Mondal, J. Mathur, M. Denich, Impacts of CO₂ emission constraints on technology selection and energy resources for power generation in Bangladesh, Energy Policy 39 (4) (2011) 2043–2050.
- [36] A. Mardani, D. Streimikiene, F. Cavallaro, N. Loganathan, M. Khoshnoudi, 'Carbon dioxide (CO₂) emissions and economic growth: a systematic review of two decades of research from 1995 to 2017, Sci. Total Environ. 649 (2019) 31–49.
- [37] B. Talbi, CO₂ emissions reduction in road transport sector in Tunisia, Renew. Sust. Energ. Rev. 69 (2017) 232–238.
- [38] Z.A. Manan, W.N.R. Mohd Nawi, S.R. Wan Alwi, J.J. Klemeš, Advances in process integration research for CO₂ emission reduction—a review, J. Clean. Prod. 167 (2017) 1–13.
- [39] P. Usubharatana, D. McMartin, A. Veawab, P. Tontiwachwuthikul, Photocatalytic process for CO₂ emission reduction from industrial flue gas streams, Ind. Eng. Chem. Res. 45 (8) (2006) 2558–2568.
- [40] E. Benhelal, G. Zahedi, E. Shamsaei, A. Bahadori, Global strategies and potentials to curb CO₂ emissions in cement industry, J. Clean. Prod. 51 (2013) 142–161.
- [41] A. Hasanbeigi, L. Price, E. Lin, Emerging energy-efficiency and CO₂ emission-reduction technologies for cement and concrete production: a technical review, Renew. Sust. Energ. Rev. 16 (8) (2012) 6220–6238.
- [42] K.O. Yoro, M.K. Amosa, P.T. Sekoai, M.O. Daramola, Modelling and experimental investigation of effects of moisture and operating parameters during the adsorption of CO₂ onto polyaspartamide, Int. J. Coal Sci. Technol. 6 (2) (2019) 225–234.
- [43] K.O. Yoro, M. Singo, J.L. Mulopo, M.O. Daramola, Modelling and experimental study of the CO₂ adsorption behaviour of polyaspartamide as an adsorbent during post-combustion CO₂ capture, Energy Procedia 114 (2017) 1643–1664.
- [44] K.O. Yoro, M.K. Amosa, P.T. Sekoai, J. Mulopo, M.O. Daramola, Diffusion mechanism and effect of mass transfer limitation during the adsorption of CO₂ by polyaspartamide in a packed-bed unit. Int. J. Sustain. Eng. (2019) https://doi.org/10.1080/19397038.2019.1592261.
- [45] J.M. Ngoy, M.O. Daramola, T.L. Chitsiga, R. Falcon, N. Wagner, CO₂ adsorption using water-soluble polyaspartamide, South Afr. J. Chem. Eng. 23 (2017) 139–144.
- [46] K. Osler, N. Twala, O.O. Oluwasina, M.O. Daramola, Synthesis and performance evaluation of chitosan/carbon nanotube (chitosan/MWCNT) composite adsorbent for post-combustion carbon dioxide capture, Energy Procedia 114 (2017) 2330–2335.
- [47] M.C. Singo, X.C. Molepo, O.O. Oluwasina, M.O. Daramola, Chitosan-impregnated sod-metal organic frameworks (sod-ZMOF) for CO₂ capture: synthesis and performance evaluation, Energy Procedia 114 (2017) 2429–2440.
- [48] M.L. Szulczewski, C.W. MacMinn, H.J. Herzog, R. Juanes, Lifetime of carbon capture and storage as a climate-change mitigation technology, Proc. Natl. Acad. Sci. USA 109 (14) (2012) 5185–5189.
- [49] D.E. Flores-Jiménez, et al., Atmospheric dispersion of methane emissions from sugarcane burning in Mexico, Environ. Pollut. 250 (2019) 922–933.
- [50] A.J. Turner, C. Frankenberg, E.A. Kort, Interpreting contemporary trends in atmospheric methane, Proc. Natl. Acad. Sci. USA 116 (8) (2019) 2805–2813.

- [51] J. Baillie, et al., Methane emissions from conventional and unconventional oil and gas production sites in south-eastern Saskatchewan, Canada, Environ. Res. Commun. 1 (1) (2019) 011003.
- [52] S.S. Yang, C.M. Liu, Y.L. Liu, Estimation of methane and nitrous oxide emission from animal production sector in Taiwan during 1990–2000, Chemosphere 52 (9) (2003) 1381–1388.
- [53] S.N. Riddick, et al., Methane emissions from oil and gas platforms in the North Sea, Atmos. Chem. Phys. 19 (15) (2019) 9787–9796.
- [54] A. Matuszewska, M. Owczuk, A. Zamojska-Jaroszewicz, J. Jakubiak-Lasocka, J. Lasocki, P. Orliński, Evaluation of the biological methane potential of various feedstock for the production of biogas to supply agricultural tractors, Energy Convers. Manag. 125 (2016) 309–319.
- [55] M. Poeschl, S. Ward, P. Owende, Environmental impacts of biogas deployment—part I: life cycle inventory for evaluation of production process emissions to air, J. Clean. Prod. 24 (2012) 168–183.
- [56] G.W. Mathison, E.K. Okine, T.A. McAllister, Y. Dong, J. Galbraith, O.I.N. Dmytruk, Reducing methane emissions from ruminant animals, J. Appl. Anim. Res. 14 (1) (1998) 1–28.
- [57] V. Paolini, F. Petracchini, M. Segreto, L. Tomassetti, N. Naja, A. Cecinato, Environmental impact of biogas: a short review of current knowledge, J. Environ. Sci. Health Part A 53 (10) (2018) 899–906.
- [58] K. Oshita, T. Okumura, M. Takaoka, T. Fujimori, L. Appels, R. Dewil, Methane and nitrous oxide emissions following anaerobic digestion of sludge in Japanese sewage treatment facilities, Bioresour. Technol. 171 (2014) 175–181.
- [59] P.T. Sekoai, et al., Microbial cell immobilization in biohydrogen production: a short overview, Crit. Rev. Biotechnol. 38 (2) (2018) 157–171.
- [60] M.J. Kampschreur, H. Temmink, R. Kleerebezem, M.S.M. Jetten, M.C.M. Van Loosdrecht, Nitrous oxide emission during wastewater treatment, Water Res. 43 (17) (2009) 4093–4103.
- [61] Y. Law, L. Ye, Y. Pan, Z. Yuan, Nitrous oxide emissions from wastewater treatment processes, Philos. Trans. R. Soc. B Biol. Sci. 367 (1593) (2012) 1265–1277.
- [62] P.T. Sekoai, K.O. Yoro, M.O. Daramola, Batch fermentative biohydrogen production process using immobilized anaerobic sludge from organic solid waste, Environment 3 (4) (2016) 38.
- [63] Y. Uchida, I. Von-Rein, Mitigation of nitrous oxide emissions during nitrification and denitrification processes in agricultural soils using enhanced efficiency fertilizers. in: Soil Contamination and Alternatives for Sustainable Development, 2018. https://doi.org/10.5772/intechopen.81548.
- [64] J. Zhang, X. Han, N₂O emission from the semi-arid ecosystem under mineral fertilizer (urea and superphosphate) and increased precipitation in northern China, Atmos. Environ. 42 (2) (2008) 291–302.
- [65] K.O. Yoro, A.J. Isafiade, M.O. Daramola, Sequential synthesis of mass exchanger networks for CO₂ capture, in: Lecture Notes in Engineering and Computer Science: Proceedings of The World Congress on Engineering and Computer Science, 23-25 October, 2018, San Francisco, USA, 2018, pp. 503–508.
- [66] S.S. Amaral, et al., CO₂, CO, hydrocarbon gases and PM_{2.5} emissions on dry season by deforestation fires in the Brazilian Amazonia, Environ. Pollut. 249 (2019) 311–320.
- [67] M. Ozonoh, T.C. Aniokete, B.O. Oboirien, B.C. Udeh, K.O. Yoro, M.O. Daramola, Prediction of emissions and profits from a biomass, tyre, and coal fired co-gasification CHP plant using artificial neural network: Nigerian and South African perspectives, J. Phys. Conf. Ser. 1378 (2019) 022021.
- [68] O.O. Sadare, M. Masitha, K.O. Yoro, M.O. Daramola, Removal of sulfur (e.g. DBT) from petroleum distillates using activated carbon in a continuous packed-bed adsorption column, in: Lecture Notes in Engineering and Computer Science: Proceedings of the World Congress on Engineering and Computer Science, 23-25 October, 2018, San Francisco, USA, 2018, pp. 509–513.
- [69] J.F. Rogers, S.J. Thompson, C.L. Addy, R.E. McKeown, D.J. Cowen, P. Decouflé, Association of very low birth weight with exposures to environmental sulfur dioxide and total suspended particulates, Am. J. Epidemiol. 151 (6) (2000) 602–613.
- [70] US EPA, Sulfur Dioxide Basics, US EPA, 2016. 02 June. Available from: https://www.epa.gov/SO2-pollution/sulfur-dioxide-basics. Accessed 25 December 2019.
- [71] M. Saygın, et al., To investigate the effects of air pollution (PM₁₀ and SO₂) on the respiratory diseases asthma and chronic obstructive pulmonary disease, Turk. Thorac. J. 18 (2) (2017) 33–39.
- [72] X.Q. Jiang, X.D. Mei, D. Feng, Air pollution and chronic airway diseases: what should people know and do? J. Thorac. Dis. 8 (1) (2016) 31–40.
- [73] Y. Zeng, K. Li, Influence of SO_2 on the corrosion and stress corrosion cracking susceptibility of supercritical CO_2 transportation pipelines. Corros. Sci. (2019). https://doi.org/10.1016/j.corsci.2019.108404. 108404.

- [74] S.H. Joo, et al., Selective removal of SO₂ from coal-fired flue gas by alkaline solvents using a membrane contactor. Chem. Eng. Process. Process Intensif. (2019). https://doi.org/10.1016/j.cep.2019.107772. 107772.
- [75] X. Wang, Y. Li, W. Zhang, J. Zhao, Z. Wang, Simultaneous SO₂ and NO removal by pellets made of carbide slag and coal char in a bubbling fluidized-bed reactor. Process. Saf. Environ. Prot. 134 (2020) 83–94, https://doi.org/ 10.1016/j.psep.2019.11.022.
- [76] R. Johansson, S. Meyer, J. Whistance, W. Thompson, D. Debnath, Greenhouse gas emission reduction and cost from the United States biofuels mandate. Renew. Sust. Energ. Rev. 119 (2020) 109513, https://doi.org/10.1016/ j.rser.2019.109513.
- [77] O.A. Olaniyi, I.O. Olutimehin, O.A. Funmilayo, Review of climate change and its effect on Nigeria ecosystem, Int. J. Rural Dev. Environ. Health Res. 3 (3) (2019) 92–100.
- [78] A. Berrou, M. Raybaut, A. Godard, M. Lefebvre, High-resolution photoacoustic and direct absorption spectroscopy of main greenhouse gases by use of a pulsed entangled cavity doubly resonant OPO, Appl. Phys. B Lasers Opt. 98 (1) (2009) 217.
- [79] C.A. Grande, R.P.L. Ribeiro, E.L.G. Oliveira, A.E. Rodrigues, Electric swing adsorption as emerging CO₂ capture technique. Energy Procedia 1 (1) (2009) 1219–1225, https://doi.org/10.1016/j.egypro.2009.01.160.
- [80] E.I. Koytsoumpa, C. Bergins, E. Kakaras, The CO₂ economy: review of CO₂ capture and reuse technologies, J. Supercrit. Fluids 132 (2018) 3–16.
- [81] A.A. Olajire, CO₂ capture and separation technologies for end-of-pipe applications—a review, Energy 35 (6) (2010) 2610–2628.
- [82] S. Saleh, V. Pirouzfar, A. Alihosseini, Performance analysis and development of a refrigeration cycle through various environmentally friendly refrigerants, J. Therm. Anal. Calorim. 136 (4) (2019) 1817–1830.
- [83] Y. Alhorr, E. Eliskandarani, E. Elsarrag, Approaches to reducing carbon dioxide emissions in the built environment: low carbon cities, Int. J. Sustain. Built Environ. 3 (2) (2014) 167–178.
- [84] A.M. Singer, et al., The role of CO₂ emissions from large point sources in emissions totals, responsibility, and policy, Environ Sci Policy 44 (2014) 190–200.
- [85] B.A. Schubert, A.H. Jahren, The effect of atmospheric CO₂ concentration on carbon isotope fractionation in C₃ land plants, Geochim. Cosmochim. Acta 96 (2012) 29–43.
- [86] M. Liu, J. Wu, X. Zhu, H. He, W. Jia, W. Xiang, Evolution and variation of atmospheric carbon dioxide concentration over terrestrial ecosystems as derived from eddy covariance measurements, Atmos. Environ. 114 (2015) 75–87
- [87] X. Luo, J. Wang, M. Dooner, J. Clarke, Overview of current development in electrical energy storage technologies and the application potential in power system operation, Appl. Energy 137 (2015) 511–536.
- [88] J.K. He, China's INDC and non-fossil energy development, Adv. Clim. Chang. Res. 6 (3) (2015) 210–215.
- [89] L. Miller, R. Carriveau, Energy demand curve variables—an overview of individual and systemic effects, Sustainable Energy Technol. Assess. 35 (2019) 172–179.
- [90] K.R. Gurney, et al., High resolution fossil fuel combustion CO₂ emission fluxes for the United States, Environ. Sci. Technol. 43 (14) (2009) 5535–5541.
- [91] E. Worrell, L. Price, N. Martin, C. Hendriks, L.O. Meida, Carbon dioxide emissions from the global cement industry, Annu. Rev. Energy Environ. 26 (1) (2001) 303–329.
- [92] I.J. Lu, C. Lewis, S.J. Lin, The forecast of motor vehicle, energy demand and CO₂ emission from Taiwan's road transportation sector, Energy Policy 37 (8) (2009) 2952–2961.
- [93] D. González, M. Martínez, Changes in CO₂ emission intensities in the Mexican industry, Energy Policy 51 (2012) 149–163.
- [94] S.J. Davis, K. Caldeira, H.D. Matthews, Future CO₂ emissions and climate change from existing energy infrastructure, Science 329 (5997) (2010) 1330–1333.
- [95] Q. Cai, X. Yan, Y. Li, L. Wang, Global patterns of human and livestock respiration, Sci. Rep. 8 (1) (2018) 1–7.
- [96] V. Ramanathan, The role of ocean-atmosphere interactions in the CO₂ climate problem, J. Atmos. Sci. 38 (5) (1981) 918–930.
- [97] C. Oertel, J. Matschullat, K. Zurba, F. Zimmermann, S. Erasmi, Greenhouse gas emissions from soils—a review, Geochemistry 76 (3) (2016) 327–352.
- [98] H. Ritchie, M. Roser, CO₂ and Greenhouse Gas Emissions, Our World Data, (2017). Retrieved from, Accessed 25 December 2019https://ourworldindata.org/co2-and-other-greenhouse-gas-emissions.
- [99] F. Nocito, A. Dibenedetto, Atmospheric CO₂ mitigation technologies: carbon capture utilization and storage. Curr. Opin. Green Sustain. Chem. 21 (2020) 34–43, https://doi.org/10.1016/j.cogsc.2019.10.002.

- [100] A. Saravanan, M. Murugan, M. Sreenivasa Reddy, S. Parida, Performance and emission characteristics of variable compression ratio CI engine fueled with dual biodiesel blends of Rapeseed and Mahua. Fuel (2019) 116751, https://doi.org/10.1016/j.fuel.2019.116751.
- [101] S.K. Shukla, Renewable biofuels and their by-products for automotive applications. In: Encyclopedia of renewable and sustainable materials, 3 (2020) 180–186, https://doi.org/10.1016/B978-0-12-803581-8.11462-6.
- [102] B. Lin, B. Xu, Factors affecting CO₂ emissions in China's agriculture sector: a quantile regression, Renew. Sust. Energ. Rev. 94 (2018) 15–27.
- [103] M.S. Hossain, M. Arshad, L. Qian, M. Zhao, Y. Mehmood, H. Kächele, Economic impact of climate change on crop farming in Bangladesh: an application of Ricardian method, Ecol. Econ. 164 (2019) 106354.
- [104] I.R. Orimoloye, S.P. Mazinyo, A.M. Kalumba, O.Y. Ekundayo, W. Nel, Implications of climate variability and change on urban and human health: a review, Cities 91 (2019) 213–223.
- [105] M.G. Donat, J. Sillmann, E.M. Fischer, Chapter 3—changes in climate extremes in observations and climate model simulations. From the past to the future, in: J. Sillmann, S. Sippel, S. Russo (Eds.), Climate Extremes and Their Implications for Impact and Risk Assessment, Elsevier, Cambridge, MA, 2020, pp. 31–57.
- [106] A. Murdoch, C. Mantyka-Pringle, S. Sharma, The interactive effects of climate change and land use on boreal stream fish communities. Sci. Total Environ. 700 (2020) 134518, https://doi.org/10.1016/j.scitotenv.2019.134518.
- [107] M. Breckner, U. Sunde, Temperature extremes, global warming, and armed conflict: new insights from high resolution data, World Dev. 123 (2019) 104624.
- [108] S.C. Lewis, A.D. King, S.E. Perkins-Kirkpatrick, D.M. Mitchell, Regional hotspots of temperature extremes under 1.5°C and 2°C of global mean warming, Weather Clim. Extrem. 26 (2019) 100233.
- [109] C. Shi, Z.H. Jiang, W.L. Chen, L. Li, Changes in temperature extremes over China under 1.5°C and 2°C global warming targets, Adv. Clim. Change Res. 9 (2) (2018) 120–129.
- [110] N.S. Darmanto, A.C.G. Varquez, N. Kawano, M. Kanda, Future urban climate projection in a tropical megacity based on global climate change and local urbanization scenarios, Urban Clim. 29 (2019) 100482.
- [111] M. Fasihi, O. Efimova, C. Breyer, Techno-economic assessment of CO₂ direct air capture plants, J. Clean. Prod. 224 (2019) 957–980.
- [112] G. Janssens-Maenhout, et al., EDGAR v4.3.2 Global Atlas of the three major greenhouse gas emissions for the period 1970–2012, Earth Syst. Sci. Data 11 (3) (2019) 959–1002.
- [113] C. Wang, et al., Responses of greenhouse-gas emissions to land-use change from rice to jasmine production in subtropical China, Atmos. Environ. 201 (2019) 391–401.
- [114] Z. Cao, et al., Toward a better practice for estimating the CO₂ emission factors of cement production: an experience from China, J. Clean. Prod. 139 (2016) 527–539.
- [115] E. Hartung, G.J. Monteny, Emissions of methane (CH₄) and nitrous oxide (N₂O) from animal husbandry—a study of literature, Landtechnik 55 (4) (2000) 288–322.
- [116] M. Davis, M. Ahiduzzaman, A. Kumar, How will Canada's greenhouse gas emissions change by 2050? A disaggregated analysis of past and future greenhouse gas emissions using bottom-up energy modelling and Sankey diagrams, Appl. Energy 220 (2018) 754–786.
- [117] R.B. Jackson, et al., Warning signs for stabilizing global CO₂ emissions, Environ. Res. Lett. 12 (11) (2017) 110202.
- [118] M. Marzouk, E. Abdelakder, On the use of multi-criteria decision making methods for minimizing environmental emissions in construction projects', Decis. Sci. Lett. 8 (4) (2019) 373–392.
- [119] M.J. Menne, C.N. Williams, B.E. Gleason, J.J. Rennie, J.H. Lawrimore, The global historical climatology network monthly temperature dataset, version 4, J. Clim. 31 (24) (2018) 9835–9854.
- [120] B. Huang, et al., Extended reconstructed sea surface temperature, version 5 (ERSSTv5): upgrades, validations, and intercomparisons, J. Clim. 30 (20) (2017) 8179–8205.
- [121] A. Szilagyi, M. Mocan, A. Verniquet, A. Churican, D. Rochat, Eco-innovation, a business approach towards sustainable processes, products and services, Procedia Soc. Behav. Sci. 238 (2018) 475–484.
- [122] M. Wackernagel, F. Pearce, Day of reckoning, New Sci. 239 (3189) (2018) 20–21.
- [123] K. Hashimoto, Global temperature and atmospheric carbon dioxide concentration, in: K. Hashimoto (Ed.), Global Carbon Dioxide Recycling: For Global Sustainable Development by Renewable Energy, Singapore, Springer, 2019, pp. 5–17.
- [124] K. Greer, et al., Global trends in carbon dioxide (CO₂) emissions from fuel combustion in marine fisheries from 1950 to 2016, Mar. Policy 107 (2019) 103382.
- [125] F. Ziegler, O.R. Eigaard, R.W.R. Parker, P.H. Tyedmers, E.S. Hognes, S. Jafarzadeh, Adding perspectives to: "Global trends in carbon dioxide (CO₂) emissions from fuel combustion in marine fisheries from 1950-2016" Mar. Policy 107 (2019) 103488.

- [126] L. Zhang, B. Liu, J. Du, C. Liu, S. Wang, CO₂ emission linkage analysis in global construction sectors: alarming trends from 1995 to 2009 and possible repercussions, J. Clean. Prod. 221 (2019) 863–877.
- [127] L. Fraccascia, I. Giannoccaro, Analyzing CO₂ emissions flows in the world economy using global emission chains and global emission trees, J. Clean. Prod. 234 (2019) 1399–1420.
- [128] C. Holz, S. Kartha, T. Athanasiou, Fairly sharing 1.5: national fair shares of a 1.5 °C-compliant global mitigation effort, Int. Environ. Agreem. Polit. Law Econ. 18 (1) (2018) 117–134.
- [129] B. Liddle, P. Sadorsky, How much does increasing non-fossil fuels in electricity generation reduce carbon dioxide emissions? Appl. Energy 197 (2017) 212–221.
- [130] Q. Wang, S. Ge, Uncovering the effects of external demand on China's coal consumption: a global input–output analysis. J. Clean. Prod. 245 (2020) 118877, https://doi.org/10.1016/j.jclepro.2019.118877.
- [131] J.D. Jenkins, M. Luke, S. Thernstrom, Getting to zero carbon emissions in the electric power sector, Joule 2 (12) (2018) 2498–2510.
- [132] A. Bari, J. Jiang, W. Saad, A. Jaekel, Challenges in the smart grid applications: an overview, Int. J. Distrib. Sens. Netw. 10 (2) (2014) 1–11.
- [133] D.B. Avancini, J.J.P.C. Rodrigues, S.G.B. Martins, R.A.L. Rabêlo, J. Al-Muhtadi, P. Solic, Energy meters evolution in smart grids: a review, J. Clean. Prod. 217 (2019) 702–715.
- [134] E. Santacana, G. Rackliffe, L. Tang, X. Feng, Getting smart, IEEE Power Energ. Mag. 8 (2) (2010) 41–48.
- [135] W.R. Morrow, A. Hasanbeigi, J. Sathaye, T. Xu, Assessment of energy efficiency improvement and CO₂ emission reduction potentials in India's cement and iron & steel industries, J. Clean. Prod. 65 (2014) 131–141.
- [136] D.Y.C. Leung, G. Caramanna, M.M. Maroto-Valer, An overview of current status of carbon dioxide capture and storage technologies, Renew. Sust. Energ. Rev. 39 (2014) 426–443.
- [137] D.J. Barker, S.A. Turner, P.A. Napier-Moore, M. Clark, J.E. Davison, CO₂ capture in the cement industry, Energy Procedia 1 (1) (2009) 87–94.
- [138] W. Shen, L. Cao, Q. Li, W. Zhang, G. Wang, C. Li, Quantifying CO₂ emissions from China's cement industry, Renew. Sust. Energ. Rev. 50 (2015) 1004–1012.
- [139] M. Takht Ravanchi, S. Sahebdelfar, Carbon dioxide capture and utilization in petrochemical industry: potentials and challenges, Appl. Petrochem. Res. 4 (1) (2014) 63–77.
- [140] M. Al-Sabawi, J. Chen, S. Ng, Fluid catalytic cracking of biomass-derived oils and their blends with petroleum feedstocks: a review, Energy Fuel 26 (9) (2012) 5355–5372.
- [141] R. Vooradi, S.B. Anne, A.K. Tula, M.R. Eden, R. Gani, Energy and CO₂ management for chemical and related industries: issues, opportunities and challenges, BMC Chem. Eng. 1 (1) (2019) 7.
- [142] A. Waxman, A. Khomaini, B. Leibowicz, S. Olmstead, Emissions in the stream: estimating the greenhouse gas impacts of an oil and gas boom. Environ. Res. Lett. (2019). https://doi.org/10.1088/1748-9326/ab5e6f.
- [143] R. An, B. Yu, R. Li, Y.M. Wei, Potential of energy savings and CO₂ emission reduction in China's iron and steel industry, Appl. Energy 226 (2018) 862–880.
- [144] W. Sun, Y. Zhou, J. Lv, J. Wu, Assessment of multi-air emissions: case of particulate matter (dust), SO₂, NOx and CO₂ from iron and steel industry of China, J. Clean. Prod. 232 (2019) 350–358.
- [145] W. Xu, B. Wan, T. Zhu, M. Shao, CO₂ emissions from China's iron and steel industry, J. Clean. Prod. 139 (2016) 1504–1511.
- [146] L.M. Dion, M. Lefsrud, V. Orsat, Review of CO₂ recovery methods from the exhaust gas of biomass heating systems for safe enrichment in greenhouses, Biomass Bioenergy 35 (8) (2011) 3422–3432.
- [147] A. Dupont, The strategic implications of climate change, Survival 50 (3) (2018) 29–54.
- [148] G.C. Hegerl, S. Brönnimann, A. Schurer, T. Cowan, The early 20th century warming: anomalies, causes, and consequences, Wiley Interdiscip. Rev. Clim. Chang. 9 (4) (2018) 522.
- [149] A.A. Ojo, T.E. Ayo, A critical legal appraisal of the international environmental legal response mechanisms to the challenges of climate change in the world, J. Law Policy Glob. 73 (2018) 22.
- [150] R. Goyal, M.H. England, A.S. Gupta, M. Jucker, Reduction in surface climate change achieved by the 1987 Montreal Protocol, Environ. Res. Lett. 14 (12) (2019) 124041.
- [151] M. Miyamoto, K. Takeuchi, Climate agreement and technology diffusion: impact of the Kyoto Protocol on international patent applications for renewable energy technologies, Energy Policy 129 (2019) 1331–1338.
- [152] S.K. Nanda, From Rio to Doha: in search of cooperative action for climate change, in: B.C. Nirmal, R.K. Singh (Eds.), Contemporary Issues in International Law: Environment, International Trade, Information Technology and Legal Education, Singapore, Springer, 2018, pp. 161–171.
- [153] M. Mengel, A. Nauels, J. Rogelj, C.F. Schleussner, Committed sea-level rise under the Paris Agreement and the legacy of delayed mitigation action, Nat. Commun. 9 (1) (2018) 1–10.

- [154] T. Ourbak, A.K. Magnan, The Paris Agreement and climate change negotiations: small islands, big players, Reg. Environ. Chang. 18 (8) (2018) 2201–2207.
- [155] K. Begg, F. Van der Woerd, D. Levy, The Business of Climate Change: Corporate Responses to Kyoto, Greenleaf Publishing, Routledge, Sheffield, UK, 2018.
- [156] D. Lazăr, A. Minea, A.A. Purcel, Pollution and economic growth: evidence from Central and Eastern European countries, Energy Econ. 81 (2019) 1121–1131.
- [157] A. Hargrove, M. Qandeel, J.M. Sommer, Global governance for climate justice: a cross-national analysis of CO₂ emissions, Glob. Transit. 1 (2019) 190–199.
- [158] J. Caytas, The COP21 negotiations: one step forward, two steps back, J. Sustain. Dev. 19 (1) (2018) 1–16.
- [159] L.A. Henry, L.M. Sundstrom, Russia and the Kyoto Protocol: seeking an alignment of interests and image, Glob. Environ. Polit. 7 (4) (2007) 47–69.
- [160] D. Bodansky, The Paris climate change agreement: a new hope? Am. J. Int. Law 110 (2) (2016) 288–319.
- [161] A. Abdel-Latif, K.E. Maskus, R. Okediji, J.H. Reichman, P. Roffe, Overcoming the impasse on intellectual property and climate change at the UNFCCC: a way forward, Social Science Research Network, Policy Brief, 11(2011).
- [162] J. Gupta, A history of international climate change policy, Wiley Interdiscip. Rev. Clim. Chang. 1 (5) (2010) 636–653.
- [163] F.P.J. Landers, The black market trade in chlorofluorocarbons: the montreal protocol makes banned refrigerants a hot commodity, Ga. J. Int. Comp. Law 26 (2) (1997) 457–485.
- [164] C.F. You, X.C. Xu, Coal combustion and its pollution control in China, Energy 35 (11) (2010) 4467–4472.
- [165] X. Li, A. Kraslawski, Conceptual process synthesis: past and current trends, Chem. Eng. Process. Process Intensif. 43 (5) (2004) 583–594.
- [166] K.O. Yoro, A.J. Isafiade, M.O. Daramola, Multi-period heat exchanger network synthesis with temperature intervals and uncertain disturbances, Chem. Eng. Trans. 76 (2019) 1039–1044.
- [167] G. Rochelle, E. Chen, S. Freeman, D. Van Wagener, Q. Xu, A. Voice, Aqueous piperazine as the new standard for CO₂ capture technology, Chem. Eng. J. 171 (3) (2011) 725–733.
- [168] K.O. Yoro, P.T. Sekoai, A.J. Isafiade, M.O. Daramola, A review on heat and mass integration techniques for energy and material minimization during CO₂ capture, Int. J. Energy Environ. Eng. 10 (3) (2019) 367–387.
- [169] K.O. Yoro, N. Chiwaye, A.J. Isafiade, M.O. Daramola, Energy and material minimization during CO₂ capture using a combined heat and mass integration technique, in: The 10th Trondheim Conference on CO₂ Capture, Transport and Storage, vol. 4, 2019, pp. 111–119.
- [170] K.O. Yoro, A.J. Isafiade, M.O. Daramola, Synthesis of optimal heat exchanger networks with quantified uncertainties and non-isothermal mixing, J. Phys. Conf. Ser. 1378 (2019) 022018.