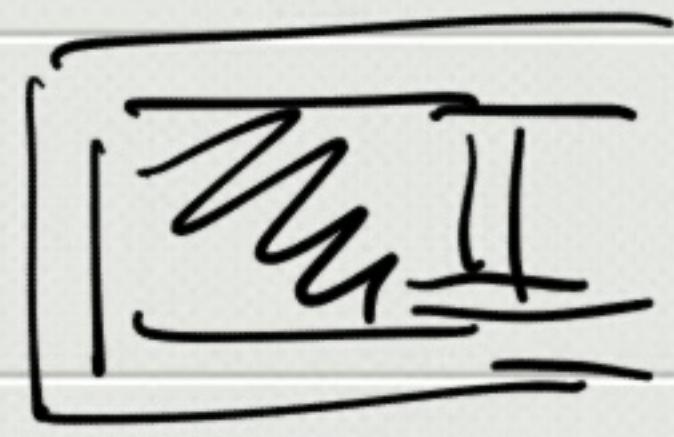


$m=2$ mono



A: $T_A = 600\text{ K}$ AB: isot. rev. $V_B = 2V_A$

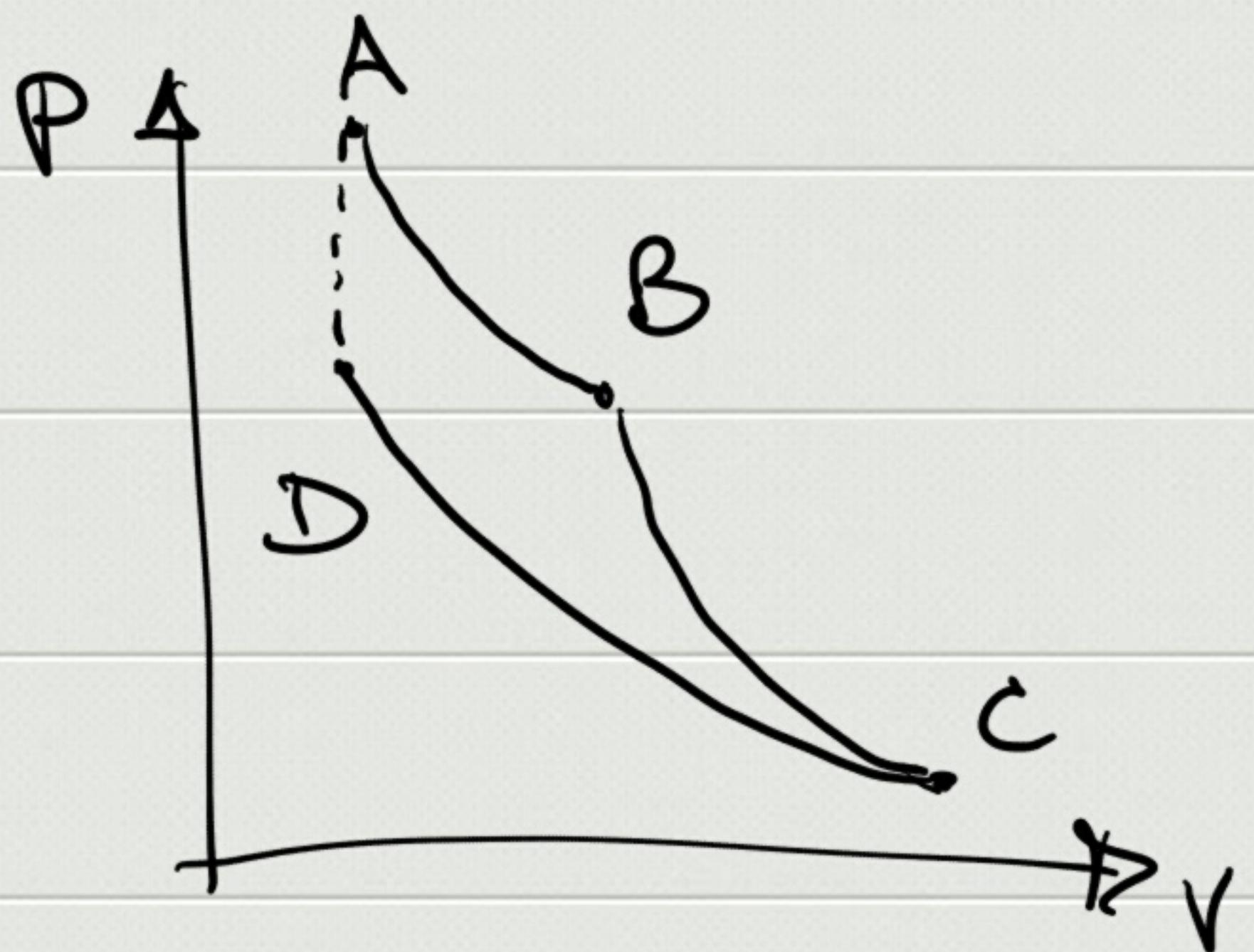
BC: adiab. rev. $V_C = 2V_B$

CD: isot. rev. $V_D = V_A$

(T_A)

DA: isot. contatto termico

$W, Q, \gamma, \Delta S_0$



$$W_{AB} = Q_{AB} = nR T_A \ln \frac{V_B}{V_A} = 6915 \text{ J}$$

$$W_{BC} = -\Delta U_{BC} = -mc_V(T_C - T_B) = 5538 \text{ J}$$

$$T_B V_B^{\gamma-1} = T_C V_C^{\gamma-1} \Rightarrow T_C = T_B \left(\frac{V_B}{V_C} \right)^{\gamma-1} = 378 \text{ K}$$

$$W_{CD} = Q_{CD} = nR T_C \ln \frac{V_D}{V_C} = -8713 \text{ J}$$

$$W_{DA} = 0 \quad Q_{DA} = mc_V (T_A - T_D) = 5538 \text{ J}$$

$$\Rightarrow W = 3739 \text{ J} = Q$$

$$\eta = \frac{W}{Q_{\text{loss}}} = \frac{W}{Q_{\text{AB}} + Q_{\text{DA}}} = 0.3$$

$$\Delta S_U = ? = \Delta S_{U, \text{DA}} = \Delta S_{\text{gas, DA}} + \Delta S_{\text{comb, DA}} =$$

$$= m_C \ln \frac{T_A}{T_D} + \frac{\overset{\uparrow}{-Q_{\text{DA, gas}}}}{T_A} =$$

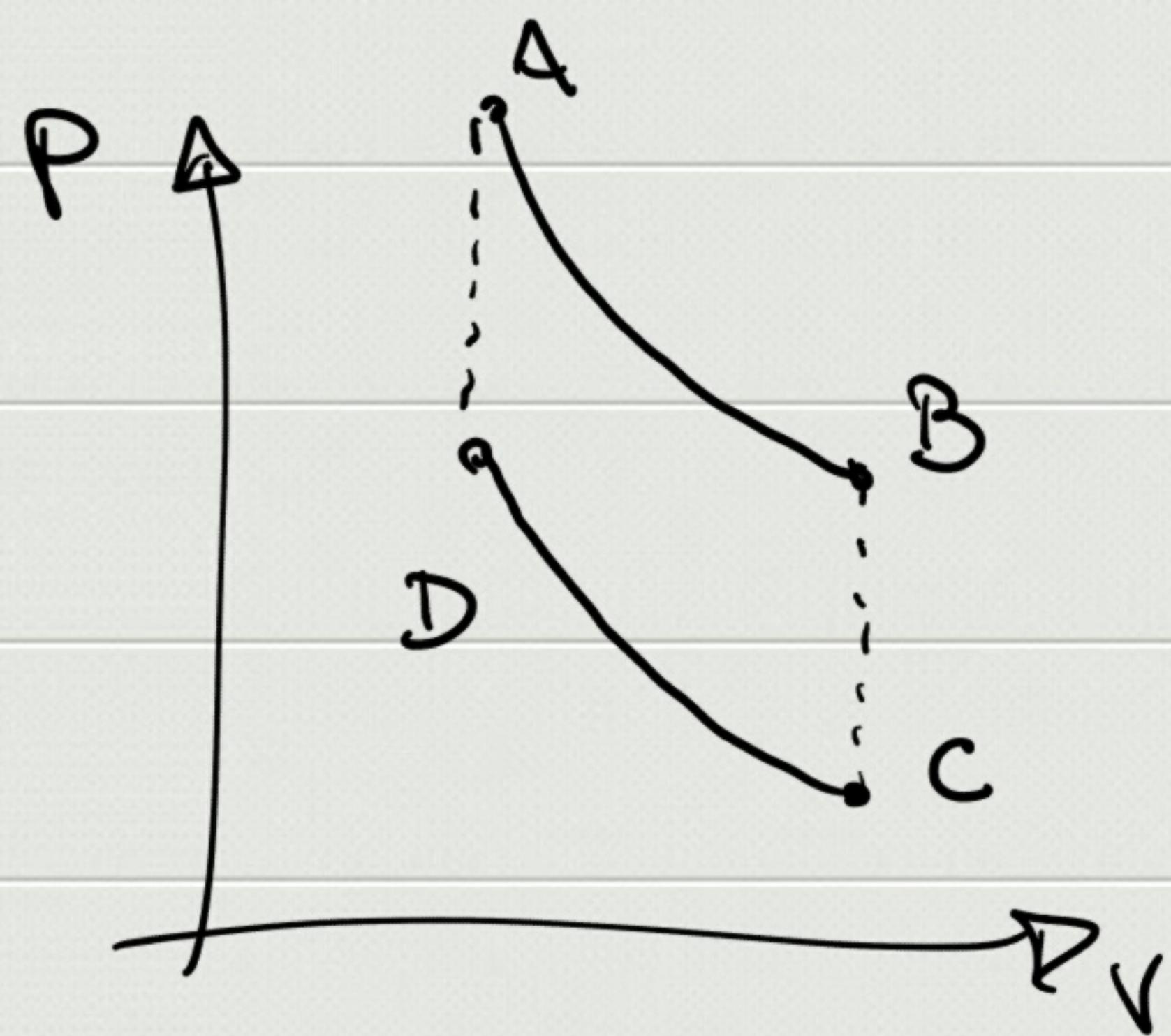
$$= m_C \ln \frac{T_A}{T_D} + \frac{-m_C (T_A - T_D)}{T_A} = 2.3 \text{ J/K}$$

bisotto, AB, CD insieme inv.

BC: fascio contatto termico T_C

DA: " " " T_A

$$T_A = 2T_C \quad V_B = 2V_A \quad M = ?$$



$$Q_{AB}, Q_{BC}, Q_{CD}, Q_{DA} \geq 0?$$

| | | | |
|---|---|---|---|
| ✓ | ✗ | ✗ | ✓ |
| ○ | ○ | ○ | ○ |
| * | | | * |

$$M = \frac{W}{Q_{AB}}$$

$$W_{BC} = W_{DA} = 0$$

$$W_{AB} = Q_{AB} = \mu R T_A \ln \frac{V_B}{V_A}$$

$$W_{CD} = \mu R T_C \ln \frac{V_D}{V_C}$$

$$Q_{DA} = \mu c_v (T_A - T_D)$$

$$\mu R T_A \ln \frac{V_B}{V_A} + \mu R T_C \ln \frac{V_D}{V_C}$$

$$M = \frac{\mu R T_A \ln \frac{V_B}{V_A} + \mu c_v (T_A - T_D)}{Q_{AB}} =$$

$$\frac{\mu R T_A \ln \frac{V_B}{V_A} + \mu c_v (T_A - T_D)}{\frac{5}{2} R}$$

$$T_A = 2T_C$$

$$T_D = T_C$$

$$\begin{aligned}
 & \frac{2\gamma_C \ln \frac{V_B}{V_A} + \gamma_C \ln \frac{V_A}{V_B}}{2\gamma_C \ln \frac{V_B}{V_A} + \frac{5}{2}\gamma_C} = \frac{2 \ln \frac{V_B}{V_A} - \ln \frac{V_B}{V_A}}{2 \ln \frac{V_B}{V_A} + \frac{5}{2}} = \\
 & = \frac{\ln 2}{2 \ln 2 + \frac{5}{2}} = 0.18
 \end{aligned}$$

$$n=2 \quad c_p = 25,1 \text{ J/k mol}$$

$$\Delta B: \text{isotermo} \quad P_A = 10^5 \text{ Pa} \quad P_B = 2,7 \cdot 10^5 \text{ Pa}$$

$$T_A = 300 \text{ K}$$

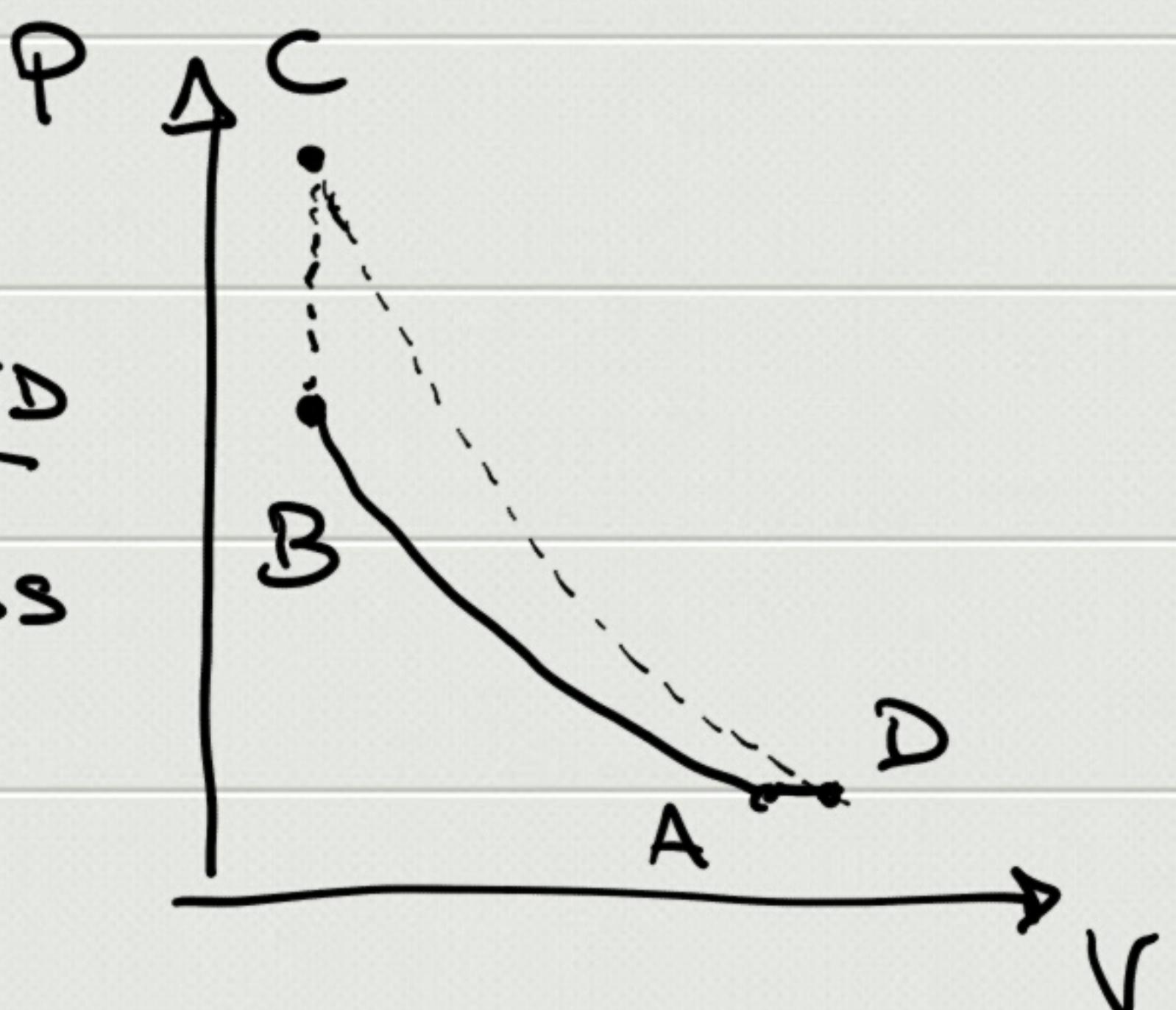
$$\Delta C: \text{isocore irr.} \quad Q_{BC} = 1,34 \cdot 10^4 \text{ J}$$

$$\Delta D: \text{adiabatica irr.} \quad P_D = P_A \quad V_D = 0,065 \text{ m}^3$$

$\Delta A: \text{isobare irr.}$

$$\gamma = ?$$

$$\gamma = \frac{W}{Q_{\text{ass}}} = 1 + \frac{Q_{CD}}{Q_{\text{ass}}}$$



$$Q_{AB} = W_{AB} = nR T_A \ln \frac{V_B}{V_A} = nR T_A \ln \frac{P_A}{P_B} = -5 \cdot 10^3 \text{ J} < 0$$

$$Q_{BC} = n c_v (T_C - T_B) > 0 \quad *$$

$$Q_{CD} = 0$$

$$Q_{DA} = n c_p (T_A - T_D) = \left\{ \begin{array}{l} \rightarrow P_A \\ T_D = \frac{P_D V_D}{n R} = 390 \text{ K} \\ = -4518 \text{ J} < 0 \end{array} \right.$$

$$\gamma = 1 + \frac{Q_{\Delta B} + Q_{\Delta A}}{Q_{BC}} = 0.29$$

$$\gamma_{max} = ?$$

$$Q_{BC} = n c_v (T_C - T_B) \Rightarrow T_C = T_B + \frac{Q_{BC}}{n c_v}$$

$$c_v = c_p - R \Rightarrow T_C = 700 \text{ K}$$

$$\gamma_{max} = 1 - \frac{T_A}{T_C} = 0.57$$