

$$n = 3, \text{ mono}, T_A = 300 \text{ K}, \Delta S_{U,\text{cicle}} = 60 \text{ J/K}$$

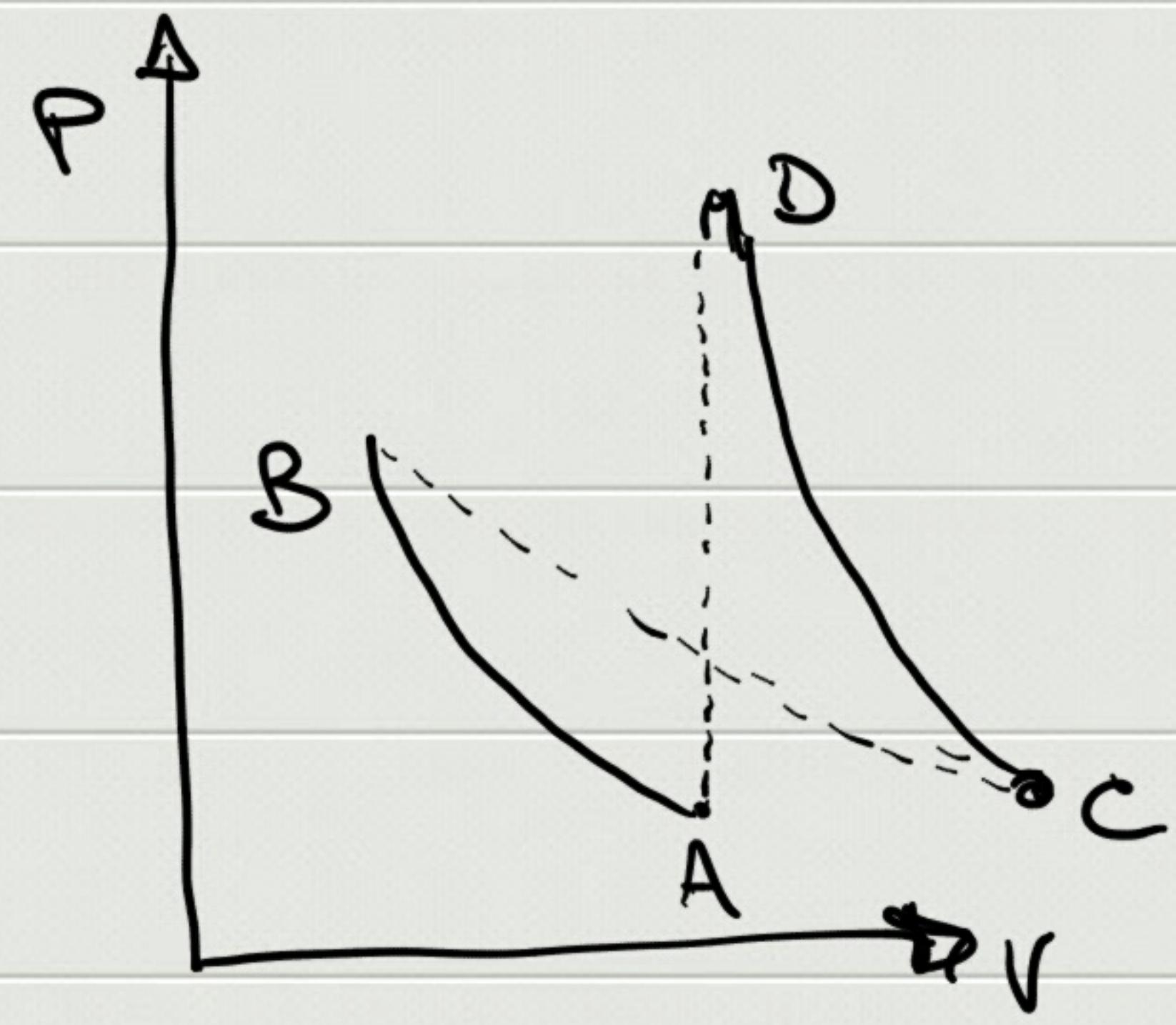
AB: compr. adiab. irr. $V_B = \frac{V_A}{2}$

BC: espansione libera

CD: compr. adiab. irr. $V_D = V_A$

DA: inverso controllo termico T_A

$$T_C, T_D, W_{\text{cicle}}, \Delta S_{U,\text{BC}}$$



$$T_A V_A^{\gamma-1} = T_B V_B^{\gamma-1}$$

$$\Rightarrow T_B = T_A \left(\frac{V_A}{V_B} \right)^{\gamma-1} = 476 \text{ K}$$

$$\frac{T_C}{T_B}$$

$$T_C V_C^{\gamma-1} = T_D V_D^{\gamma-1}$$

$$\Delta S_{U,\text{cicle}} = - \frac{m_C (T_A - T_D)}{T_A} = \frac{Q_{\text{verb}, DA}}{T_A} *$$

$$T_D = T_C$$

$$\Delta S_{U,\text{cicle}} = m_R \ln \frac{V_C}{V_B} + m_C \ln \frac{V_A}{V_D} - \frac{m_C (T_A - T_D)}{T_A} *$$

$\underbrace{\Delta S_{\text{ges}, BC}}_{\Delta S_{U,\text{BC}}} \quad \underbrace{\Delta S_{U, DA}}_{\Delta S_{U,\text{DA}}}$

$$T_D = T_A + \frac{T_A \Delta S_{U, \text{cycle}}}{m_C} = 781 \text{ K}$$

$$W_{AB} = -\Delta U_{AB} = -m_C(T_B - T_A)$$

$$W_{CD} = -\Delta U_{CD} = -m_C(T_D - T_C)$$

$$W_{\text{tot}} = W_{AB} + W_{CD} = -1.8 \cdot 10^4 \text{ J}$$

$$T_C V_C^{\gamma-1} = T_D V_D^{\gamma-1} \Rightarrow V_C = V_D \left(\frac{T_D}{T_C} \right)^{\frac{1}{\gamma-1}}$$

$$\Rightarrow \Delta S_{U, BC} = \Delta S_{\text{gas}, BC} = m_C R \ln \frac{V_C}{V_B} = 35.8 \text{ J/K}$$

$$\Delta S_{\text{gas, circ}} = 0 = \Delta S_{\text{gas, BC+DA}}$$

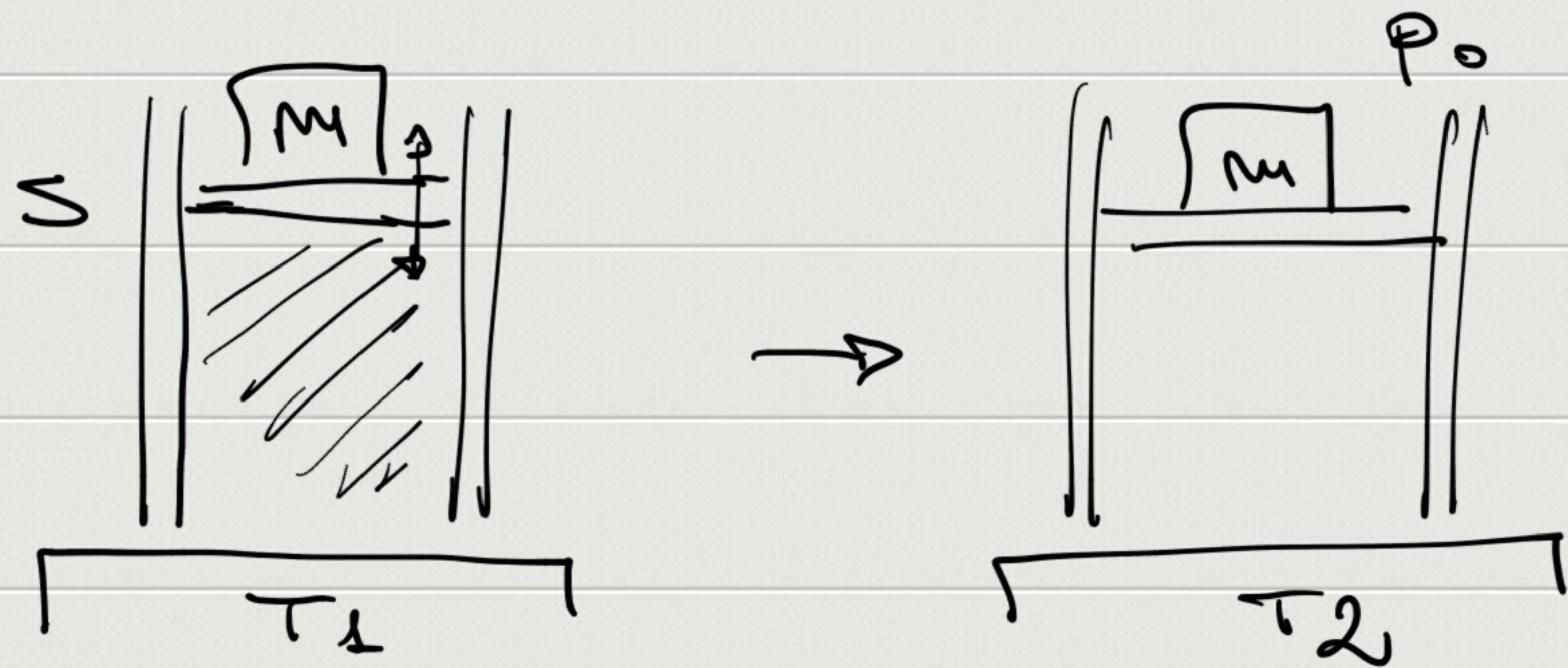
$$\Rightarrow \Delta S_{U, BC} = -\Delta S_{\text{gas, DA}} = -m_C R \ln \frac{T_A}{T_D}$$



$$\Delta S_{U, \text{cycle}} = \Delta S_{U, BC} + \Delta S_{U, DA}$$

$$m_{\text{mono}} = 290 \text{ kg}, V_1 = 0.02 \text{ m}^3, S = 0.1 \text{ m}^2$$

$$m = 500 \text{ kg}, P_0 = P_{\text{amb}} = 10^5 \text{ Pa}, T_2 = 410 \text{ K}$$

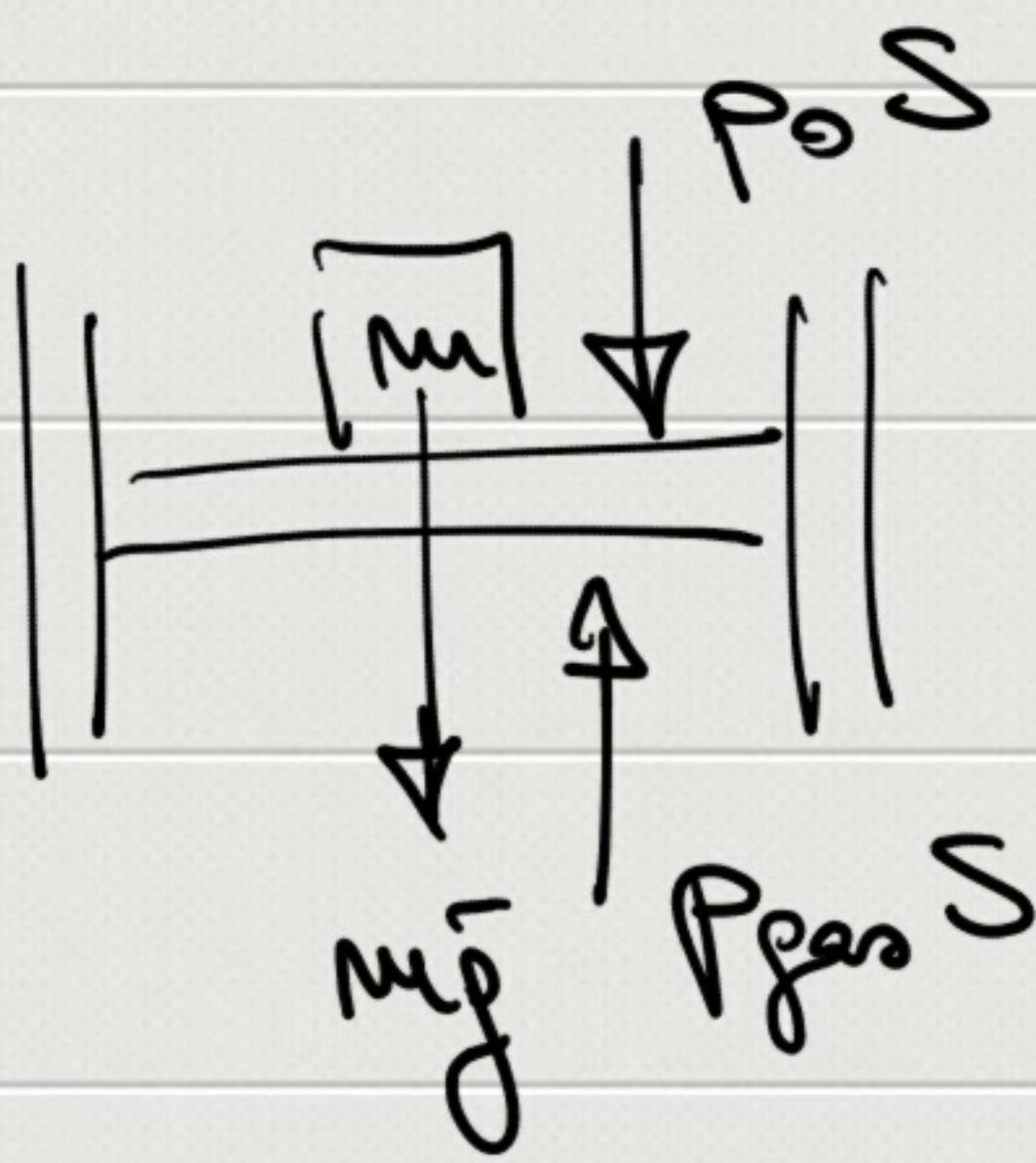


W_{gas} , ΔS_C

$$\begin{aligned} W_{1,2} &= P_0 (V_2 - V_1) \\ &= \left(P_0 + \frac{mg}{S} \right) (V_2 - V_1) \quad * \\ &= mR(T_2 - T_1) \quad * \\ &= mC_p(T_2 - T_1) + mC_v(T_2 - T_1) \end{aligned}$$

$$W_{\text{gas}} = -W_{\text{amb}} = -P_{\text{amb}} \Delta V_{\text{amb}} = P_{\text{amb}} \Delta V_{\text{gas}}$$

\uparrow



$$(F = pS)$$

$$mg + p_0 S - p_{gas} S = 0$$

$$\Rightarrow \boxed{p_{gas} = p_0 + \frac{mg}{S}} = 1,49 \cdot 10^5 \text{ Pa}$$

$$p = \text{const} \Rightarrow \frac{V}{T} = \text{const} \Rightarrow \frac{V_1}{T_1} = \frac{V_2}{T_2} \Rightarrow V_2 = V_1 \frac{T_2}{T_1} = 0,0283 \text{ m}^3$$

$$\Rightarrow W_{12} = 1237 \text{ J}$$

$$n = \frac{P_{gas} V_1}{RT_1} = 1,236 \text{ mol}$$

$$\Delta S_U = 1,37 \text{ J/K}$$

$$\Delta S_U = \underbrace{nC_p \ln \frac{T_2}{T_1}}_{= nC_p \ln \frac{T_2}{T_1}} + \cancel{\frac{nC_p (T_2 - T_1)}{T_2}}$$

*

$$= nC_p \ln \frac{T_2}{T_1} - \frac{nC_p (T_2 - T_1)}{T_2}$$

$$= \cancel{nC_p \ln \frac{T_2}{T_1}} + MR \ln \frac{V_2}{V_1} - \frac{nC_p (T_2 - T_1)}{T_2}$$

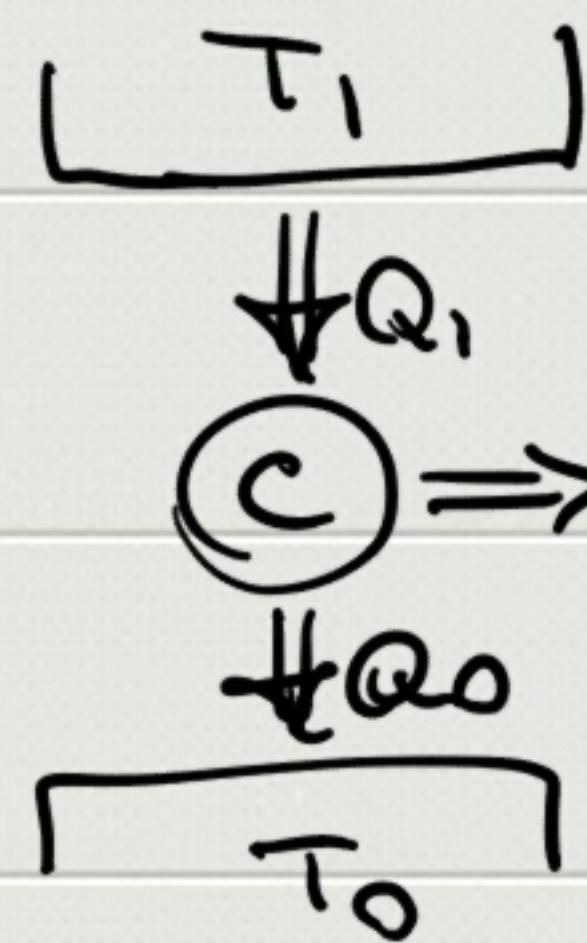
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$$= nC_p \ln \frac{T_2}{T_1} + \cancel{\frac{nC_p (T_2 - T_1)}{T_1}}$$

Macchina di Carnot $T_1 = 300 \text{ K}$ $T_0 = 279 \text{ K}$

$$n=2 \quad \text{esp. isot (ur)} \quad V_f/V_i = 3$$

$$W_C = ?$$



$$W_{AB} = nRT_1 \ln \frac{V_f}{V_i} = 5480 \text{ J}$$

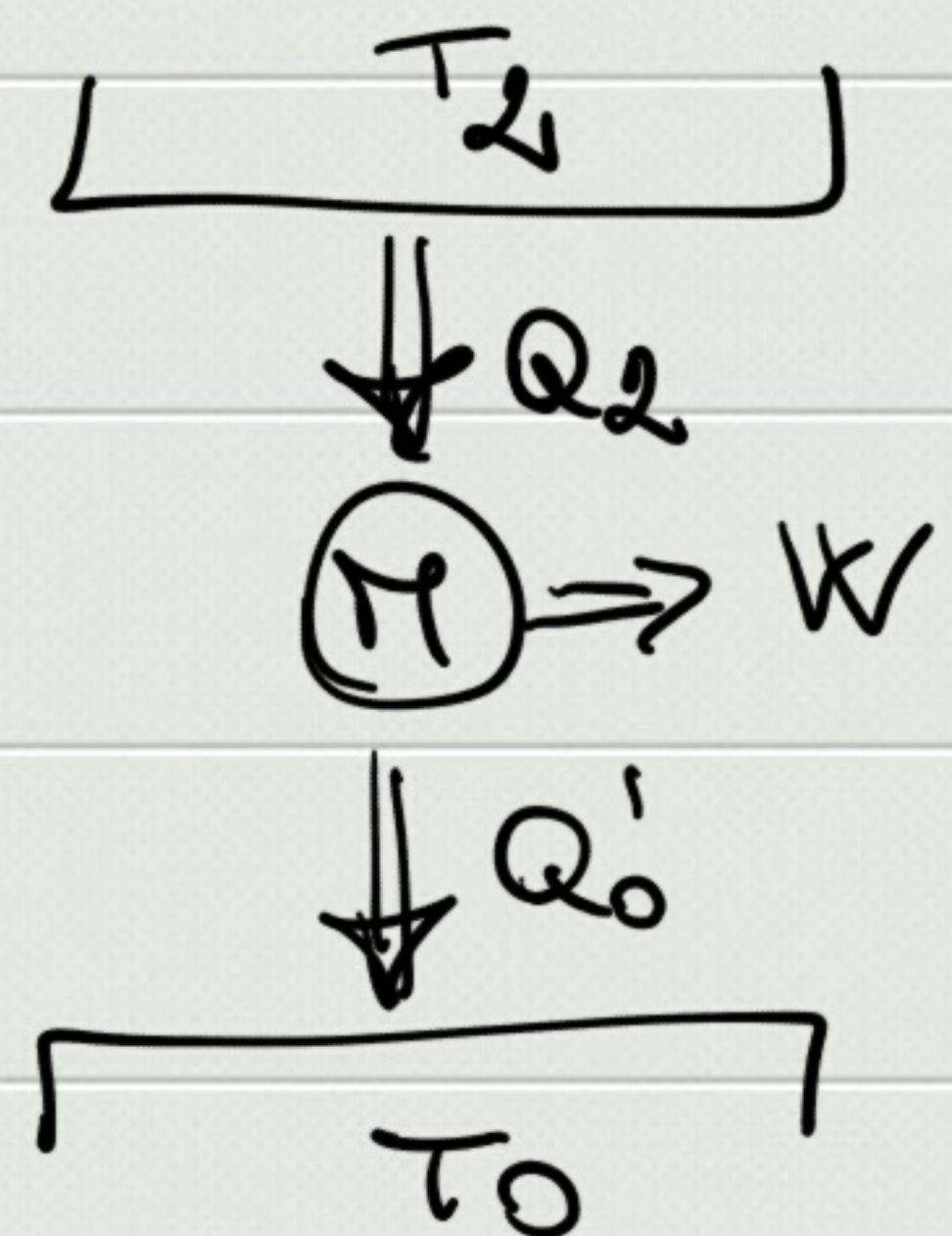
$$\stackrel{!}{=} Q_{AB} = Q_{Ass} (\leftarrow Q_1)$$

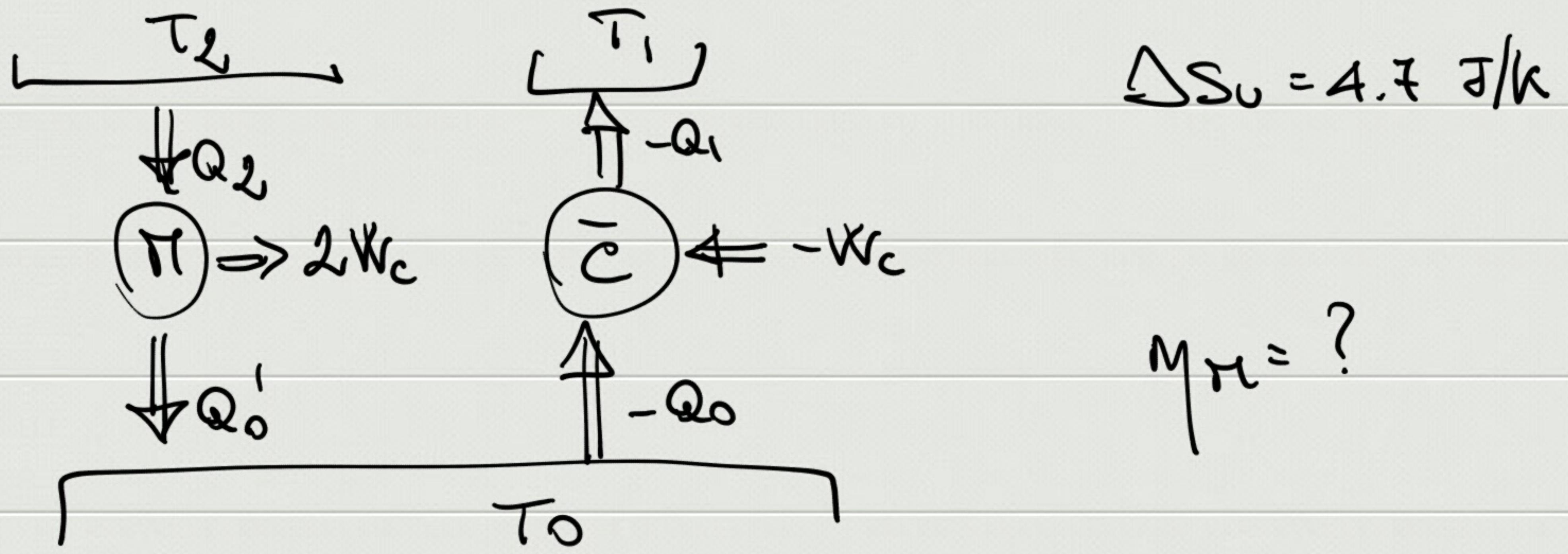
$$\eta_C = 1 - \frac{T_0}{T_1} = \frac{W_C}{Q_{Ass}} \Rightarrow$$

$$\Rightarrow W_C = Q_{Ass} \left(1 - \frac{T_0}{T_1} \right) = 383,6 \text{ J}$$

$$W_C = Q_1 + Q_0 \Rightarrow Q_0 = W_C - Q_1 = -5096,7 \text{ J}$$

$$T_2 = 400 \text{ K}, W = 2W_C$$





$$\eta = \frac{2W_C}{Q_2}$$

$$\left\{ \begin{array}{l} \Delta S_U = \Delta S_{U,\pi} + \cancel{\Delta S_{U,C}} = \Delta S_{amb,\pi} = \frac{-Q_2}{T_2} + \frac{-Q_0'}{T_0} \\ Q_2 + Q_0' = 2W_C \end{array} \right.$$

$$\Rightarrow Q_2 = \frac{\Delta S_U + 2W_C/T_0}{1/T_0 - 1/T_2} = 687 \text{ J}$$

$$\eta = 0.112$$