Table 1: Contributions to the interaction second virial coefficient for $Ar\text{-}CO_2$. $B^{(0)}$ is the classical result; $B_{tr}^{(1)}$ is translational quantum correction; $B_{rot}^{(2)}$ is the rotational quantum correction. $B = B^{(0)} + B_{tr}^{(1)} + B_{rot}^{(1)}$ is the final estimate. Units are cm^3/mol .

T(K)	$B^{(0)}$	$B_{tr}^{(1)}$	$B_{rot}^{(1)}$	В	B_{exp}
213.0	-83.79	0.72	0.14	-82.93	$\begin{cases} -86.3 \pm 5.0^{\mathbf{a}} \\ -94.0 \pm 7.0^{\mathbf{b}} \end{cases}$
223.0	-76.09	0.65	0.13	-75.13	-75.5 ± 5.0^{a}
233.0					$-74.8 \pm 1.0^{\mathbf{c}}$
242.0	-63.68	0.54	0.11	-63.03	$\begin{cases} -62.9 \pm 5.0^{\mathbf{a}} \\ -70.0 \pm 7.0^{\mathbf{b}} \end{cases}$
248.2	-60.16	0.51	0.10	-59.55	$-58.4 \pm 1.0^{\mathbf{c}}$
262.0	-53.06	0.45	0.09	-52.52	$-50.8 \pm 5.0^{\mathbf{a}}$
273.2	-47.96	0.41	0.08	-47.47	$-50.6 \pm 1.0^{\mathbf{c}}$
276.0	-46.75	0.40	0.08	-46.27	$\begin{cases} -43.4 \pm 5.0^{\mathbf{a}} \\ -51.0 \pm 6.0^{\mathbf{b}} \end{cases}$
288.2	-41.89	0.37	0.07	-41.45	$-40.3 \pm 2.0^{\mathbf{d}}$
290.0	-41.21	0.36	0.07	-40.78	$\begin{cases} -45.2 \pm 1.4^{\mathbf{e}} \\ -46.4 \pm 4.0^{\mathbf{f}} \end{cases}$
295.0	-39.37	0.35	0.07	-38.95	$\begin{cases} -37.2 \pm 5.0^{\mathbf{a}} \\ -44.0 \pm 6.0^{\mathbf{b}} \end{cases}$
296.0	-39.02	0.35	0.07	-38.60	$-37.0 \pm 2.0^{\mathbf{d}}$
296.15	-38.97	0.35	0.07	-38.55	$-44.1 \pm 5.0^{\mathbf{b}}$
300.0	-37.63	0.34	0.07	-37.22	$\begin{cases} -40.8 \pm 1.3^{\mathbf{e}} \\ -41.7 \pm 4.0^{\mathbf{f}} \end{cases}$
303.15	-36.55	0.33	0.06	-36.16	$-31.8 \pm 4.6^{\mathbf{g}}$
303.2	-36.55	0.33	0.06	-36.16	$-34.2 \pm 2.0^{\mathbf{d}}$
310.0	-34.34	0.32	0.06	-33.96	$-38.6 \pm 4.0^{\mathbf{f}}$
313.2	-33.34	0.31	0.06	-32.97	$-31.2 \pm 2.0^{\mathbf{d}}$
320.0	-31.30	0.30	0.06	-30.94	$-35.3 \pm 1.3^{\mathbf{e}}$
322.85	-30.48	0.29	0.06	-30.13	$-30.1 \pm 2.0^{\mathbf{h}}$
323.1	-30.40	0.29	0.06	-30.05	$-28.3 \pm 2.0^{\mathbf{d}}$
330.0	-28.48	0.28	0.05	-28.15	$\begin{cases} -27.3 \pm 5.0^{\mathbf{a}} \\ -35.0 \pm 5.0^{\mathbf{b}} \end{cases}$
333.15	-27.63	0.27	0.05	-27.95	$-25.8 \pm 4.2^{\mathbf{g}}$
363.15	-20.46	0.23	0.04	-20.19	$-19.6 \pm 4.2^{\mathbf{g}}$
365.0	-20.06	0.23	0.04	-19.79	$-16.2 \pm 5.0^{\mathbf{a}}$
400.0	-13.39	0.19	0.04	-13.16	$\begin{cases} -6.0 \pm 4.0^{\mathbf{a}} \\ -13.0 \pm 3.0^{\mathbf{b}} \end{cases}$
425.0	-9.42	0.17	0.03	-9.22	-3.1 ± 4.0^{a}
450.0	-5.97	0.16	0.03	-5.78	$\begin{cases} 0.5 \pm 4.0^{\mathbf{a}} \\ -7.0 \pm 2.0^{\mathbf{b}} \end{cases}$
475.0	-2.94	0.14	0.03	-2.77	$1.7 \pm 4.0^{\mathbf{a}}$

^aSchmiedel, H.; Gehrmann, R.; Schramm, B.; Ber. Bunsen-Gest. Phys. Chem. 84 (1980) 721.

^bSchramm, B.; Mueller, W.; Ber. Bunsen-Ges. Phys. Chem. **86** (1982) 110.

^cBrewer, J.; Air Force Off.Sci. Res., [Tech. Rep.] AFOSR-TR 67-2795, (1967).

^dLichtenthaler, R. N.; Schaefer, K.; Ber. Bunsen-Ges. Phys. Chem. **73** (1969) 42.

^eMartin, M.L.; Trengove, R.D.; Harris, J.R.; Dunlop, P.J.; Aust. J. Chem. **35** (1982) 1525.

^fBell, T. N.; Bignell, C.M.l Dunlop, P.J.; Physica A: (Amsterdam) 181 (1992) 221.

gCottrell, T.L.; Hamilton R. A.; Taubinger, R.P.; Trans. Faraday Soc. 52 (1956) 1310.

^hBose, T.K.; Cole, R.H.; J. Chem. Phys. **52** (1970) 140.