

A Primer on Weighing Uncertainties in Radionuclidic Metrology

(not the uncertainty associated with mass calibrations)



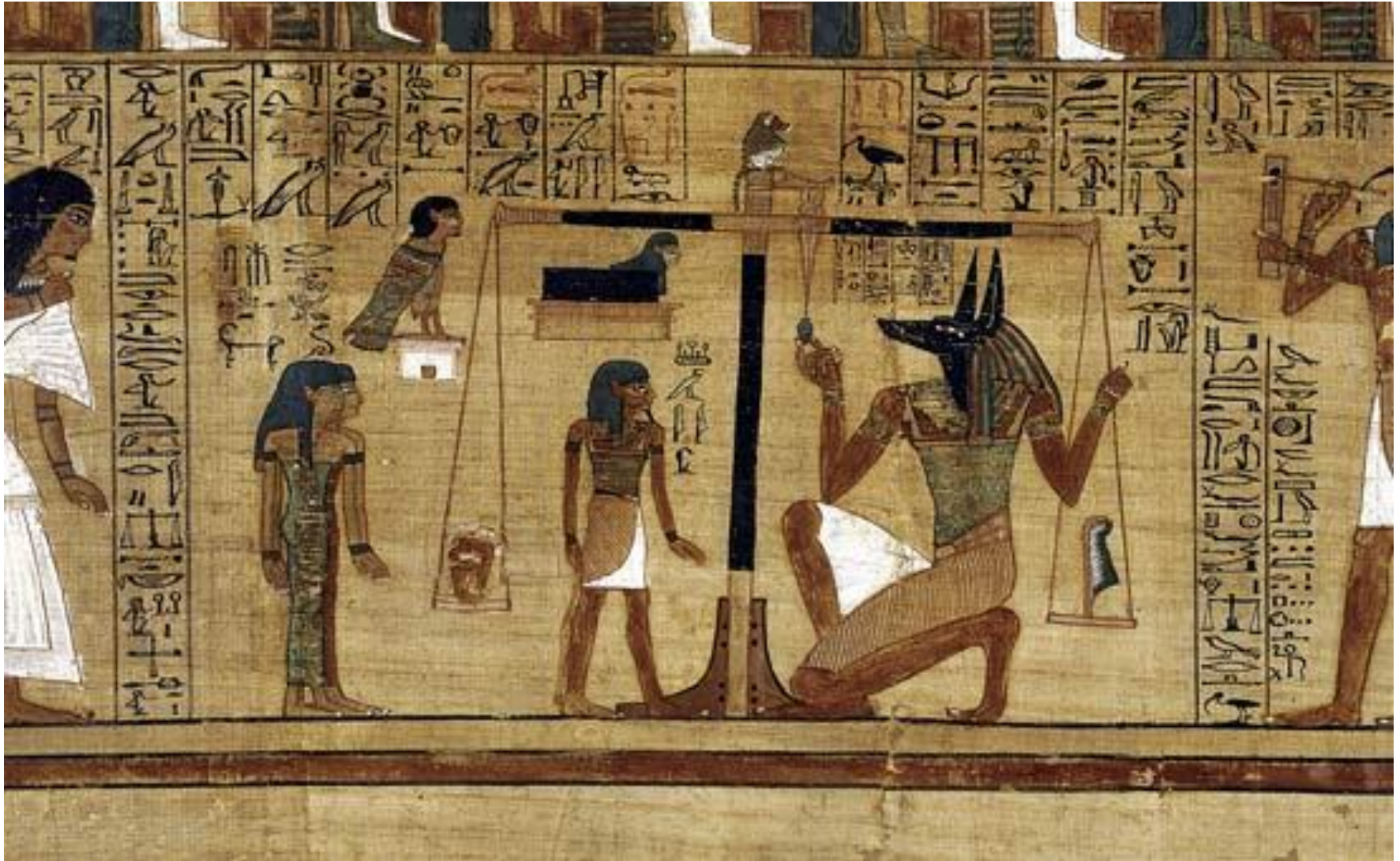
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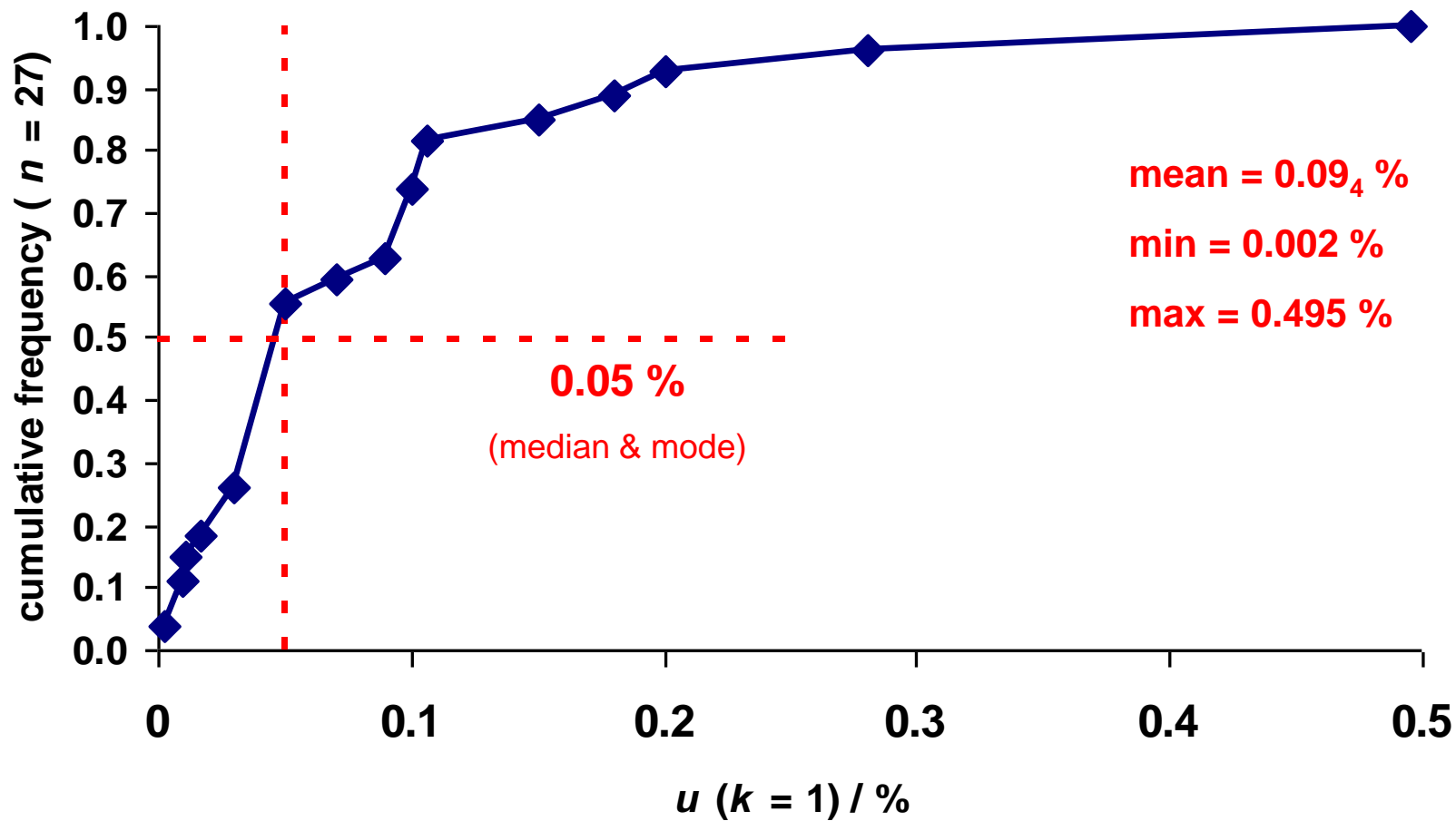


Comparisons and Uncertainties Workshop
*Bureau International des Poids et Mesures
Pavillon de Breteuil Sèvres, FRANCE*
17 september 2008

Anubis weighing the soul of the scribe Ani, from the Egyptian Book of the Dead, c. 1275 BCE.



WEIGHING UNCERTAINTIES REPORTED FOR ^{55}Fe COMPARISON



SOME PERSONAL OBSERVATIONS / OPINIONS

- ➡ weighing uncertainty not generally a major or dominant uncertainty component
- ➡ weighing uncertainty does not appear to be assessed very rigorously
- ➡ assignment of weighing uncertainty appears to be largely guessed at – or based on past practice & basis (but conditions have changed)
- ➡ weighing uncertainty (particularly for small masses – microweighing) is probably underestimated
- ➡ guidance on weighing uncertainty has not kept abreast of changes in uncertainty treatments (GUM)

1970's radionuclidic metrology -- ICRM

MICROWEIGHING for source preparation

Some excellent work

Janet Merritt (AECL)

Y. Le Gallic (LMRI)

NPL, NBS/NIST, et alia



emphasis on ≈ 20 mg

M-5 mechanical balance w/ built-in weights

Systematic evaluations of balance performance

Great guidance on practices

Even comparisons to test weighing

20 mg

best precision 2 μg

best 5 – 10 μg

typical ≥ 20 μg

can reach 100 μg

NOW ??

more applications (larger mass range)

+ bigger sources ! – LS

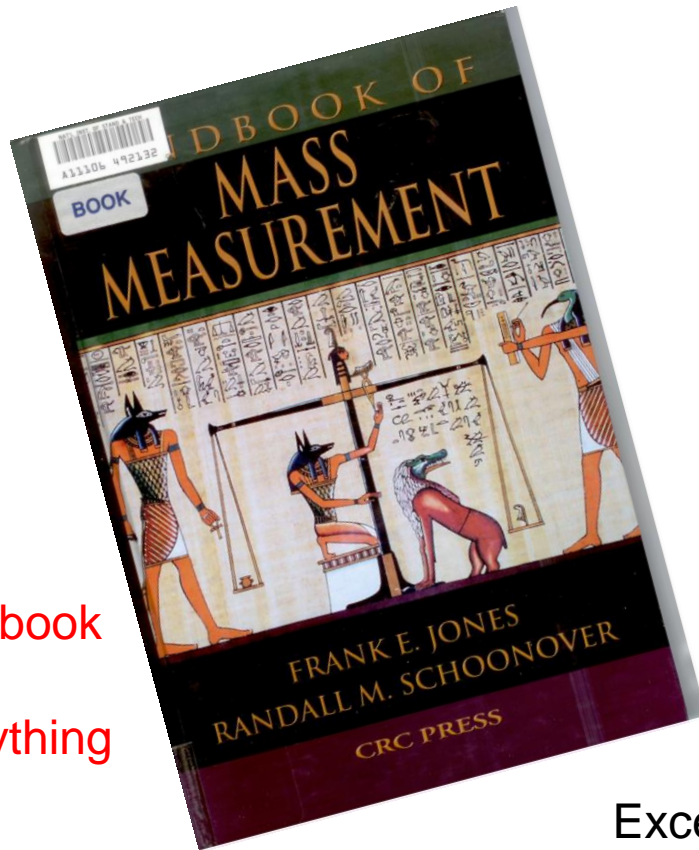
“electronic” EM force compensation balances

less “care” in labs ?

**some preliminary stuff
before starting**

Many, many references available from the mass people in **YOUR** laboratories

particularly for mass calibrations



this
handbook
has
everything

Principles,
Good practices,
Conditions
Weighing procedures
Corrections
Uncertainty treatments

Excellent documents from NPL, PTB, NIST, others

Mass is a physical quantity

Weighing is a measurement process

(used to obtain mass)

We measuring (certifying) massic activity, not mass ...

**So we want the uncertainty associated with the
weighing process (includes mass standard)**

We measure

Conventional Mass

(Mass in air versus reference density of 8.0 g/cm³)

NOT

True Mass *(Mass in vacuum)*

Weight *(gravitational force acting on mass)*

*The **conventional mass value** of a body is equal to the mass of a standard, of conventionally chosen density, at a conventionally chosen temperature, which balances this body at this reference temperature in air of conventionally chosen density.*

The conventionally chosen conditions are:

reference density = $8.0 \text{ g}\cdot\text{cm}^{-3}$;

reference temperature = $20 \text{ }^{\circ}\text{C}$;

normal air density = $1.2 \text{ mg}\cdot\text{cm}^{-3}$

formal
definition

The unit of the quantity “conventional mass” is the kilogram.

The conventional mass has the same unit as mass, because its values are defined through the multiplication of a mass by a dimensionless quantity.

Conventional mass was formerly called “apparent mass versus $8.0 \text{ g}\cdot\text{cm}^{-3}$ ” in USA

Conventional value of the result of weighing in air

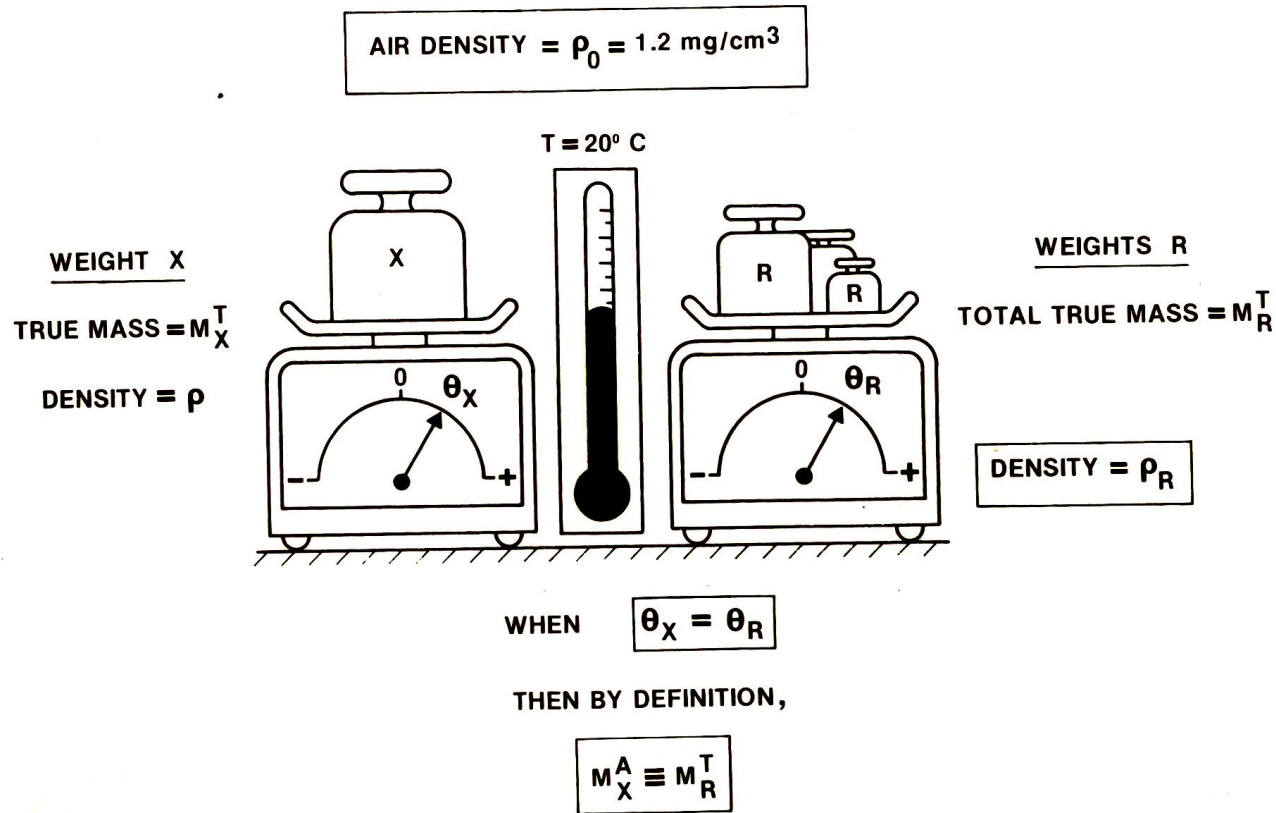
(Valeur conventionnelle du résultat des pesées dans l'air)

OIML D 28 Edition 2004 (E)

ORGANISATION INTERNATIONALE DE MÉTROLOGIE LÉGALE

Conceptually,

conventional mass of object X is equal to true mass of just enough reference material R to produce equal balance reading to that produced by X if measurements are done at 20 °C and with $\rho_0 = 1.2 \text{ mg}\cdot\text{cm}^{-3}$



CONVENTIONAL MASS OF X = TRUE MASS OF R

Boxed information is specified in definition

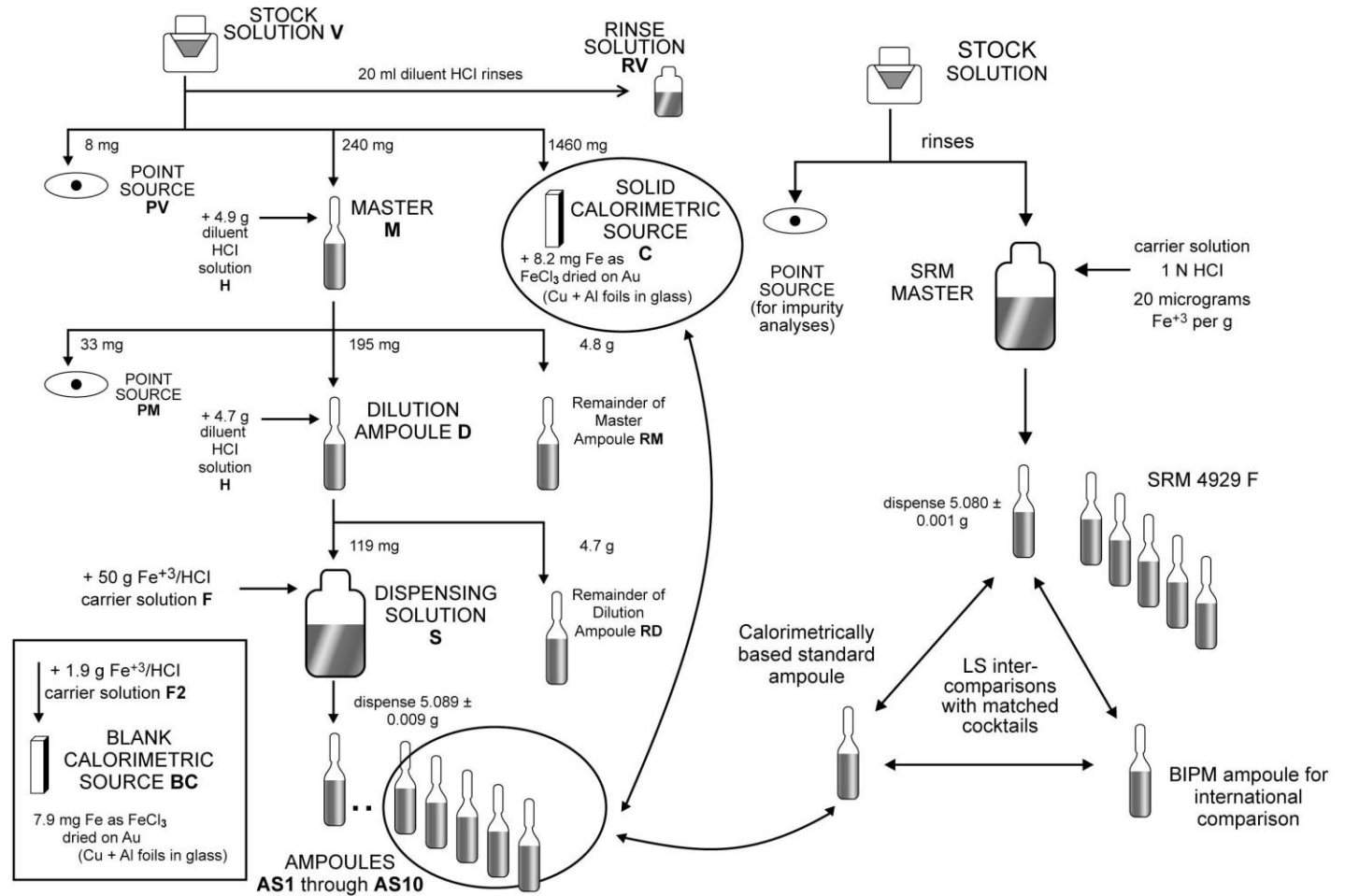
**Let's first consider
balances needed
and used**

need mass measurements over wide range of precision & capacity

precise microbalances for source preparations,

analytical balances for carrier solutions,

to large capacity balances for master solutions



NIST Physics Lab -- Radioactivity Group Balances & Scales

Balance	Type	Room	Number	Load (g)	Readability (mg)
Microbalances					
Mettler M5	Dial Weights & Optical scale	B46 E106	3	20	0.001
Mettler AX26	Electronic	B152	1	22	0.001
Mettler AT20	Electronic	B46	1	22	0.002
Analytical					
Mettler B5	Dial Weights and Optical scale	E106 B152	2	200	0.1
Mettler XP205	Dial Weights and Optical scale	B152	1	220	0.01
Mettler H311	Dial Weights and Optical scale	B46	1	240	0.1
Denver Instr. M220D	Electronic	E103	1	220 31	0.1 0.01
Mettler AE240	electronic	B152 B156 C135	3	240 40	0.1 0.01
Laboratory					
Mettler B6	Dial Weights and Optical scale	C135	1	100	1
Mettler P120	Scale, optical	B46	1	120	10
Mettler H6T	Dial Weights and Optical scale	B48	1	160	1
Large capacity					
Mettler B4C1000	Dial Weights and Optical scale	C135	1	1 000	10
Voland Jupiter 3000	Dial Weights and Optical scale	B152	1	3000	0.01
Mettler PB3002	Electronic	C135	1	3 100	10
Mettler AX12004	weights + electronic (100 g)	B46	1	12 111	0.1
Adams CBC-15	Scale, electronic	B46	1	15 000	100
Mettler ID1 Plus	Scale, electronic	C138	1	150 000	1000

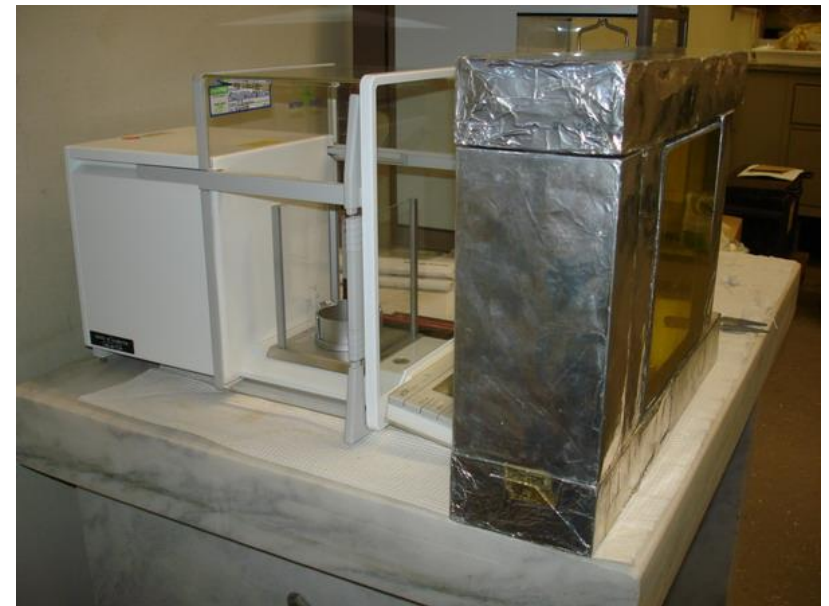




Balances with
built-in dial
weights &
optical scale



“Electronic” (EMF compensation)





others





Mass
comparator

**Now, let's talk about
reference weights
& density**

Want balance to measure conventional mass vs. $\rho_R = 8.0 \text{ g}\cdot\text{cm}^{-3}$

For either single pan balances using built-in dial weights or “electronic” EMF compensation

Manufacturers adjust the dial weights or EM force so that conventional masses equal the value of the dial position or the electronic reading.

No matter what density the balance weights ρ_{BW} actually are, user generally assumes that uncorrected conventional mass is equal to the reading (either dial positions or digital value) and that $\rho_{BW} = \rho_R$

Equivalent force for $\rho_{BW} = \rho_R$ is easy for “electronic” balances

But $\rho_{BW} = \rho_R$ is not the case for many older balances

Extant NIST balances have $\rho_{BW} = 7.76, 7.8, 8.0$ -- even $8.4 \text{ g}\cdot\text{cm}^{-3}$ in very old
error about $10^{-4} \%$

Similarly the standard weight set used for calibrations and/or substitution weighing may also differ from $\rho_R = 8.0 \text{ g}\cdot\text{cm}^{-3}$

The OIML weight classes are defined as follows:

Class E1: Weights intended to ensure traceability between national mass standards (with values derived from the International Prototype of the kilogram) and weights of class E2 and lower. Class E1 weights or weight sets **shall be accompanied by a calibration certificate** (see 15.2.2.1).

Class E2: Weights intended for use in the verification or calibration of class F1 weights and for use with weighing instruments of special accuracy class I. Class E2 weights or weight sets **shall be accompanied by a calibration certificate** (see 15.2.2.2). They may be used as class E1 weights if they comply with the requirements for surface roughness, magnetic susceptibility and magnetization for class E1 weights, and if their calibration certificate gives the appropriate data as specified in 15.2.2.1.

Class F1: Weights intended for use in the verification or calibration of class F2 weights and for use with weighing instruments of special accuracy class I and high accuracy class II.

Class F2: Weights intended for use in the verification or calibration of class M1 and possibly class M2 weights. Also intended for use in important commercial transactions (e.g. precious metals and stones) on weighing instruments of high accuracy class II.

Class M1: Weights intended for use in the verification or calibration of class M2 weights, and for use with weighing instruments of medium accuracy class III.

Class M2: Weights intended for use in the verification or calibration of class M3 weights and for use in general commercial transactions and with weighing instruments of medium accuracy class III.

Class M3: Weights intended for use with weighing instruments of medium accuracy class III and ordinary accuracy class IIII.

Classes M1–2 Weights from 50 kg to 5 000 kg of lower accuracy intended for use with weighing instruments of **and M2–3:** medium accuracy class III.

Note: The error in a weight used for the verification of a weighing instrument shall not exceed 1/3 of the maximum permissible error for an instrument. These values are listed in section 3.7.1 of OIML R 76 *Nonautomatic Weighing Instruments* (1992).





1 mg to 5 kg
standard weight
set is a thing of
beauty to behold

Get pretty
certificate too



International Bureau of Standards

Mass and density inserted into equation of definition of the conventional mass (OIML D28:2004)

Nominal value	Conventional Mass deviation mg	Uncertainty mg	Tolerance Limit (OIML R111-1:2004) mg	Within Limit?
1 mg	0.00137	0.00080	0.003	✓
2 mg	0.00225	0.00080	0.003	✓
2 mg	0.00108	0.00080	0.003	✓
5 mg	0.00158	0.00080	0.003	✓
10 mg	0.00146	0.00080	0.003	✓
20 mg	0.0011	0.0010	0.003	✓
20 mg	0.0017	0.0010	0.003	✓
50 mg	0.0010	0.0012	0.004	✓
100 mg	0.0005	0.0015	0.005	✓
200 mg	0.0021	0.0020	0.006	✓
200 mg	0.0037	0.0020	0.006	✓
500 mg	0.0043	0.0025	0.008	✓
1 g	0.0033	0.0030	0.01	✓
2 g	0.0036	0.0040	0.012	✓
2 g	0.0040	0.0040	0.012	✓
5 g	0.0088	0.0050	0.016	✓
10 g	0.0114	0.0060	0.02	✓
20 g	0.0168	0.0080	0.025	✓
20 g	0.0157	0.0080	0.025	✓
50 g	0.011	0.010	0.03	✓
100 g	0.009	0.015	0.05	✓
200 g	0.054	0.030	0.1	✓
200 g	0.047	0.030	0.1	✓
500 g	0.060	0.080	0.25	✓
1 kg	0.12	0.10	0.5	✓

Calibration Certificate

CONVENTIONAL MASS

(Apparent mass versus 8.00 g/cm³)

**NIST
Radioactivity
Group**

**Class E1
weight set**

**Calibrated
June 2008**

m/g	r/mg	u/mg	d/(kg/m ³)	mass (g)	% unc
0.001	0.00137	0.0008		0.00100137	0.080
0.002	0.00025	0.0008		0.00200025	0.040
0.002	* 0.00108	0.0008		0.00200108	0.040
0.005	0.00158	0.0008		0.00500158	0.016
0.01	0.00146	0.0008		0.01000146	0.0080
0.02	0.0011	0.001		0.02000110	0.0050
0.02	* 0.0017	0.001		0.02000170	0.0050
0.05	0.001	0.0012		0.05000100	0.0024
0.1	0.0005	0.0015		0.10000050	0.0015
0.2	0.0021	0.002		0.20000210	0.0010
0.2	* 0.0037	0.002		0.20000370	0.0010
0.5	0.0043	0.0025		0.50000430	0.00050
1	0.0033	0.003	7998	1.00000330	0.00030
2	0.0036	0.004	8008	2.00000360	0.00020
2	* 0.004	0.004	8008	2.00000400	0.00020
5	0.0088	0.005	8007	5.00000880	0.00010
10	0.0114	0.006	8007.2	10.00001140	0.000060
20	0.0166	0.008	8012.7	20.00001660	0.000040
20	* 0.0157	0.008	8012.9	20.00001570	0.000040
50	0.011	0.01	8010.91	50.00001100	0.000020
100	0.009	0.015	8026.94	100.00000900	0.000015
200	0.054	0.03	8009.8	200.00005400	0.000015
200	* 0.047	0.03	8010.82	200.00004700	0.000015
500	0.06	0.08	8013.623	500.00006000	0.000016
1000	0.12	0.1	8012.199	1000.00012000	0.000010
2000	0.29	0.1	8010.25	2000.00029000	0.000005
2000	* 0.08	0.1	8011.4	2000.00008000	0.000005
5000	0	0.1	8010.9	5000.00000000	0.000002

**Let's next look at the
air bouyancy correction**

$$C = \left[\begin{array}{c} 1 - \frac{\rho_a}{\rho_R} \\ 1 - \frac{\rho_a}{\rho_S} \end{array} \right]$$

$$C \approx 1 + \rho_a \left[\frac{1}{\rho_S} - \frac{1}{\rho_R} \right]$$

Large 0.1 % effect for typical ambient ρ_a and ρ_x

Everyone knows about it and does it !

Formulas for air density and correction readily available (CIPM recommended)

Air density depends on temperature, pressure, humidity (not too sensitive)

Correction depends on air density, density of object being weighed, balance weight (or reference) density

OIML R111 calls for ρ_a range $1.2 \text{ mg}\cdot\text{cm}^{-3} \pm 10 \%$

So typical condition requirements are:

$$T = 20 \text{ }^{\circ}\text{C} \pm 5 \text{ }^{\circ}\text{C}$$

$$P = 101 \text{ kPa} \pm 6 \text{ kPa (760 mm Hg mbar)}$$

$$\text{RH} = 50 \% \pm 30 \%$$

Typical T-controlled lab has ρ_a range of $1.2 \text{ mg}\cdot\text{cm}^{-3} \pm 5 \%$

Typical routine lab measurement capability is

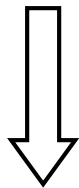
T to $0.1 \text{ }^{\circ}\text{C}$

P to $0.05 \text{ kPa (0.4 mm Hg)}$

RH to 5% (at dewpoint of $0.25 \text{ }^{\circ}\text{C}$)

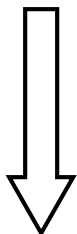
} NOT stability

air T, P, RH measurements



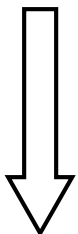
Air density equation

ρ_a



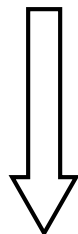
Assumed
reference
density

ρ_R



corrected
density of
solution

ρ_x



BUOYANCY CORRECTION

Uncertainty components on correction:

ρ_a (T, P, RH meas. unc. including variability in T, P, RH)

ρ_R (may depend on ρ_{BW} or ρ_{WR} differences)

ρ_x (density measurement, with unc. in mass volume, temp., etc.)

PROPAGATE
UNC. BY

$$U^2(p_a) = \sum \left(\frac{\partial p_a}{\partial v_i} \right)^2 U^2(v_i)$$

$$\nabla U^2(C) = \sum \left(\frac{\partial C}{\partial p_i} \right)^2 U^2(p_i)$$

REAL
DATA

$$T = (23.0 \pm 0.4)^\circ \text{C} \quad (\text{variability})$$

$$P = 29.8 \pm 1.7 \text{ in Hg} \quad (\text{storm})$$

$$(101 \pm 6) \text{ kPa}$$

$$RH = 45\% \pm 5\%$$

$$\rightarrow \rho_a = 1.1815 \text{ mg/cm}^3 \pm 5.6\%$$

$$\rho_{BW} = 7.76$$

$$\rho_s = (1.015 \pm 0.002) \text{ g} \cdot \text{cm}^{-3}$$

$$C = 1.001030 \pm 0.000057 \quad (\text{or } 0.006\%)$$

$$\text{effect of unc. on } \rho_s < 0.0002\%$$

$$\rho_R \text{ negligible}$$

Obviously, buoyancy correction is important,

but once made - uncertainty due it is fairly small

Weighing uncertainty must consider other things

Like

the weighing PROCEDURE used

And

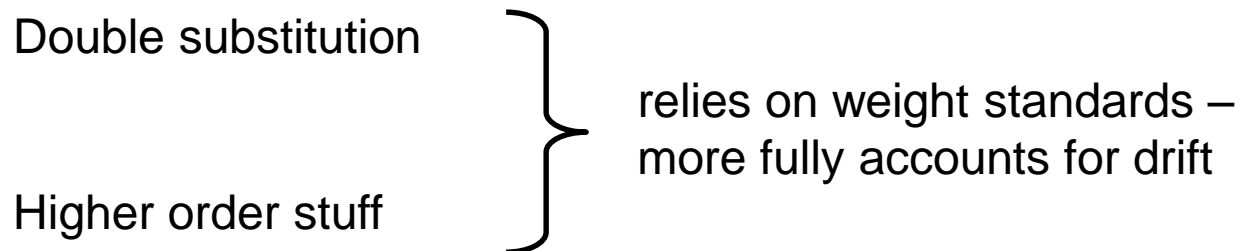
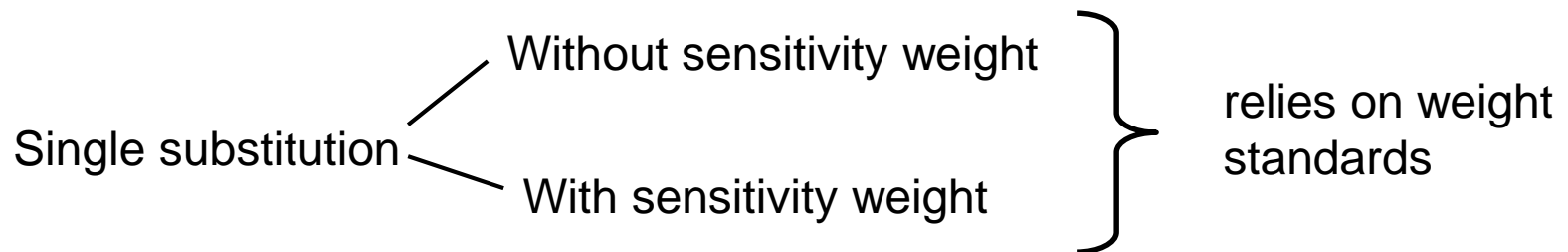
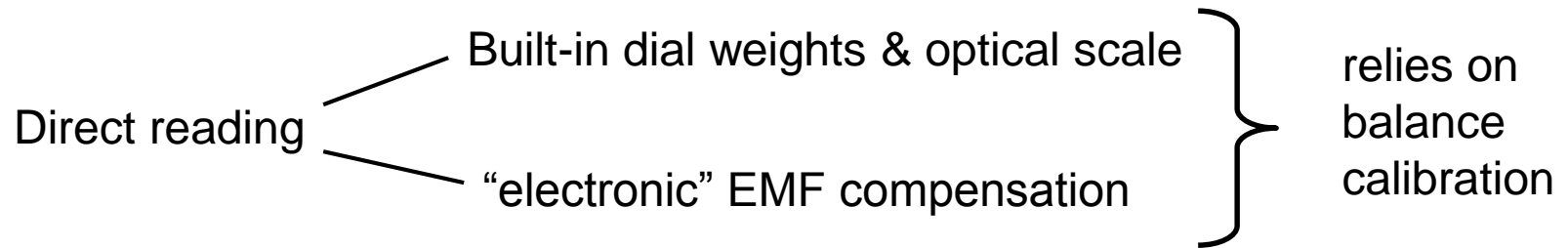
the procedure's calibration

or basis of the weight standards

**So, we now need to specifically address
different weighing procedures
& how they affect uncertainty**

Lots of weighing schemes for single-pan balance

Many available references to them



Adequacy of balance calibration or calibration of weight set usually not the biggest uncertainty concern

BUT, assumes that balances & standards handled appropriately

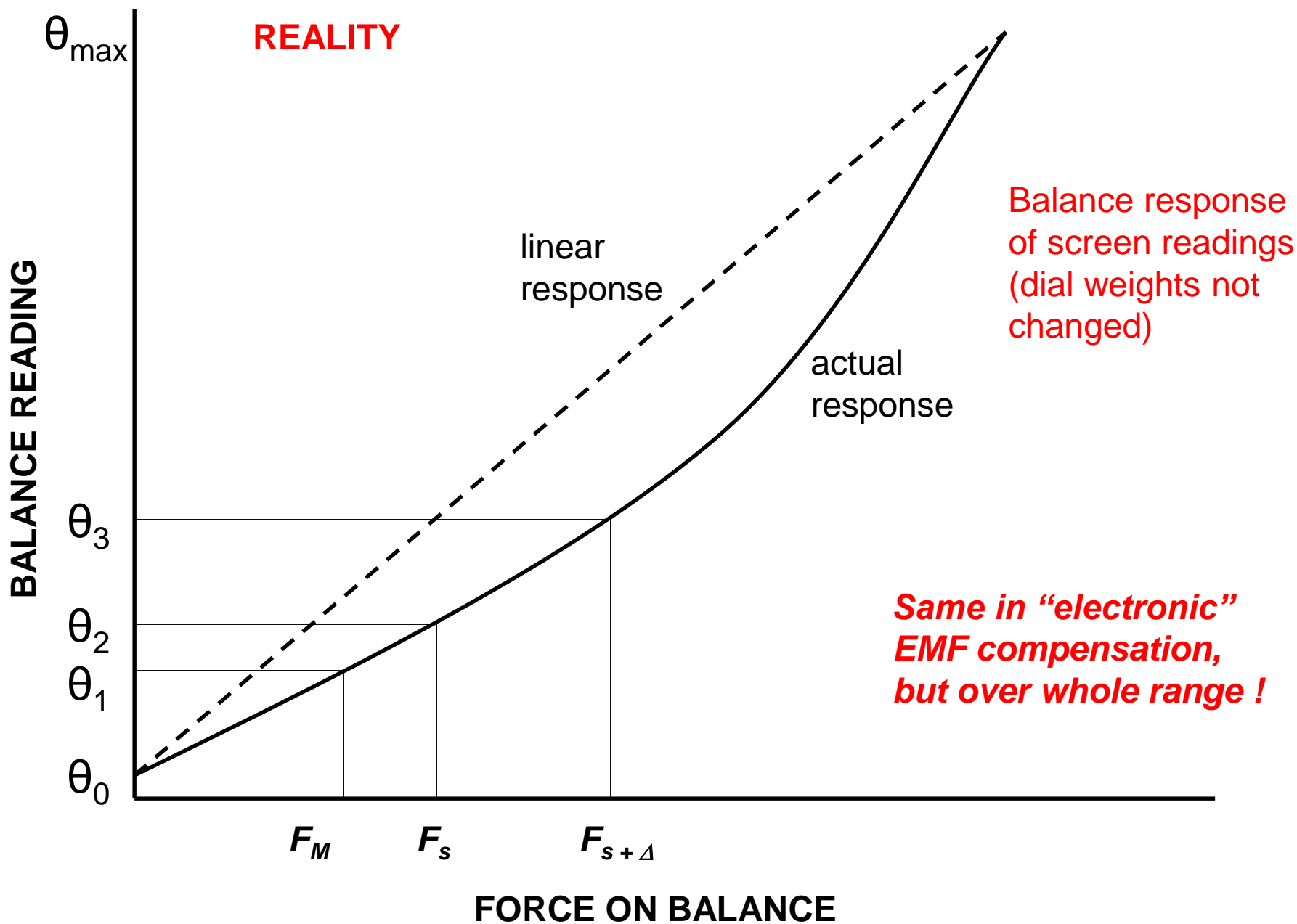
Built-in balance weights in high quality balances were **ORIGINALLY** as good as E2 weights. Probably still ok, if not F....ED with.

Bigger concern might be for small masses, linearity of optical scale

Important to realize that “calibrations” as part of routine balance maintenance (yearly) only do two point checks.

Linearity has same concerns in “electronic” balances – internal calibrations rely on reference voltage in precision resistor

REALITY



All deviations from linearity can depend on temperature.

All weight sets + balances (“electronic”) have a temperature coefficient (TC) – indicates “sensitivity” change in mass reading per °C

Example Weigh 5 g on analytical balance with resolution of 0.1 mg. Tech specs for balance has $TC \leq 5(10^{-6}) \text{ g/}^\circ\text{C}$. With 2°C change (from calib temp. or lab variability), get change of 0.05 mg (or half resolution from TC alone)

Temperature also affects “drift” of ideal response curve – in both optical scale & EMF compensation

Consider conversion steps in typical EMF compensation balance

mass



weight (force) on pan



force on coil



electrical current



voltage



digital readout

misalignment, vibrations, electrostatics, air currents,...

forces on lever, alignment ...

*aging, external magnetic fields, temp. coefficient,
position of coil ...*

*change in precision resistor because of temp or aging,
thermoelectric voltages ...*

*change in reference voltage in ADC (aging & temp),
offset voltage amplification changes ...*

RETURN

Lots of weighing schemes for single-pan balance

Many available references to them

Direct reading

useful for routine work

Preparation of many sources

Can have dial weight calibrations

Substitution
weighing

better for high accuracy source preparation

takes time – certainly better for weighing
metal standards – is it, for an evaporating
liquid ?

Re-visit “electronic” EMF compensation balances

SOME BENEFITS

- tare control over whole range
- dual capacity * & precision
- filters noise
- computer compatible
- selectable sampling of reading (on some)

SOME PROBLEMS

- static effects (plastic pycnometers)
- weigh ferromagnetic materials
- EM radiation fields
- dust in magnetic poles

Biggest concerns

- Linearity errors
- Force to mass calibration
- Precision (probably dominant)

To identify uncertainty components in weighing need to look at what affects the determination

TWO BIG COMPONENTS

➔ the basis of the mass standardization

- balance or weight set calibration
(condition & handling)
- linearity of screen or EMF reading
(optical scale or electronic)

*probably smaller
easier to handle*

**Some stop
here !**

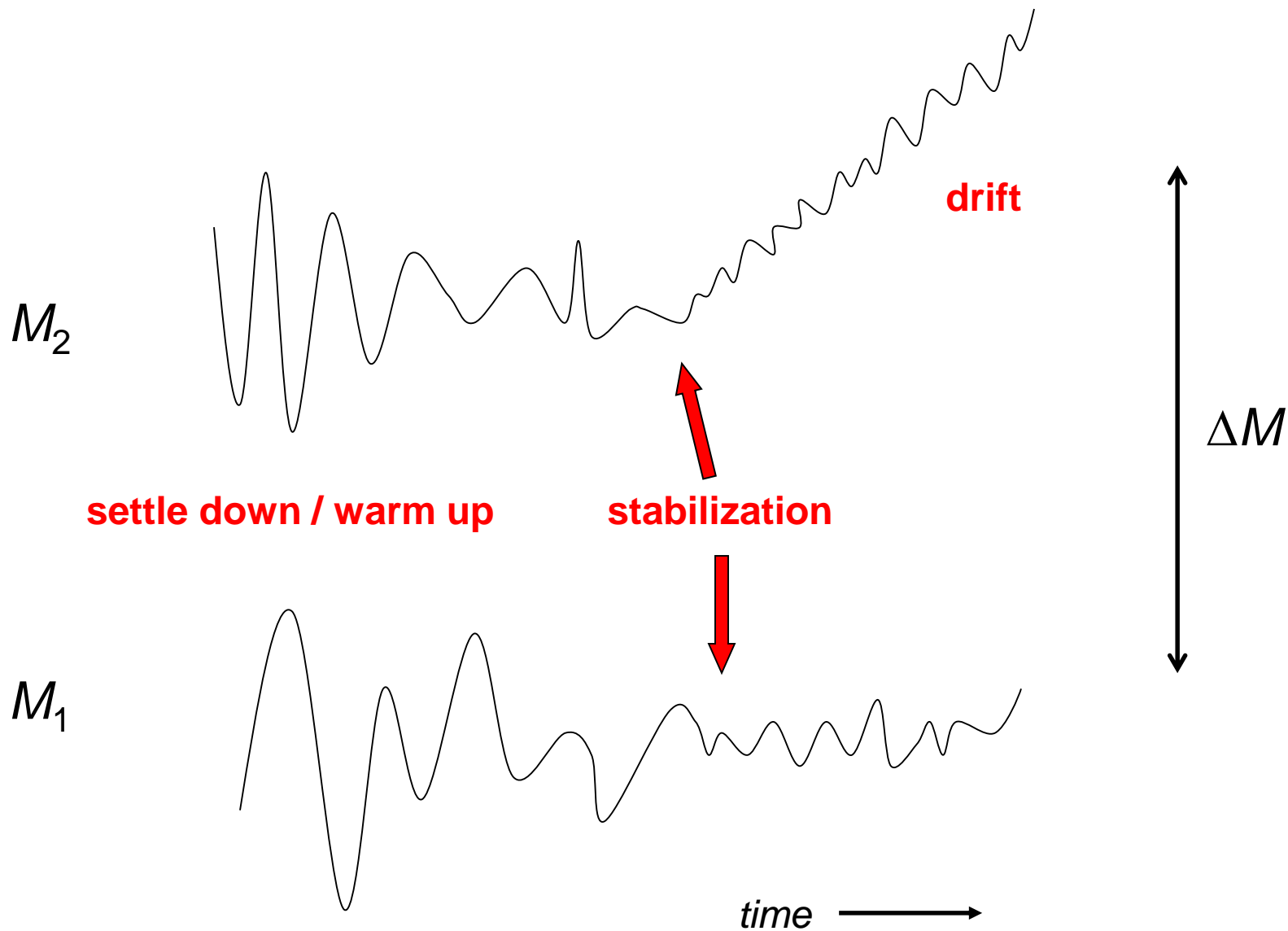
➔ environmental / operator effects

- ambient lab controls (mainly temp.)
- electrostatics
- vibrations
- air currents
- evaporation losses (really tough)
- practice / habits of operator
- Many, many more ...

usually dominant

*must evaluate for given
conditions & operator*

*Some tests obvious (dispensed &
contained mass differences;
replications; etc.), but evaluation is
hard*



Mass standards / balance calibration

Operator & practice

Weighing method

Balance / sample sources of inaccuracy

sensitivity (slope) – slope $\neq 1$

linearity – sensitivity not constant

hysteresis - different up & down

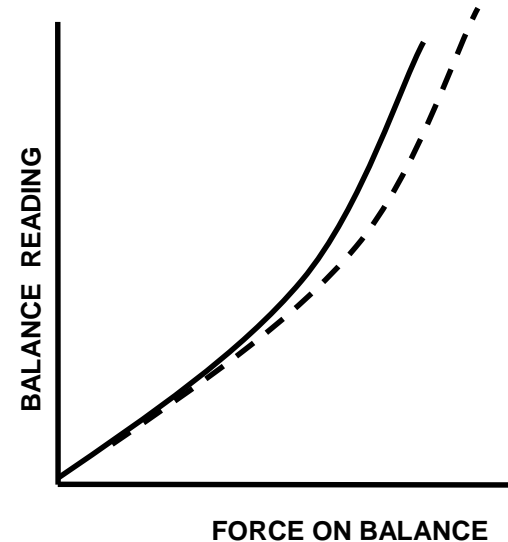
drift – instability

temperature effects

buoyancy effects

off-load (eccentricity) effects

evaporation / condensation



Main environmental problems

Floor & Bench / vibrational instability

Temperature

Humidity

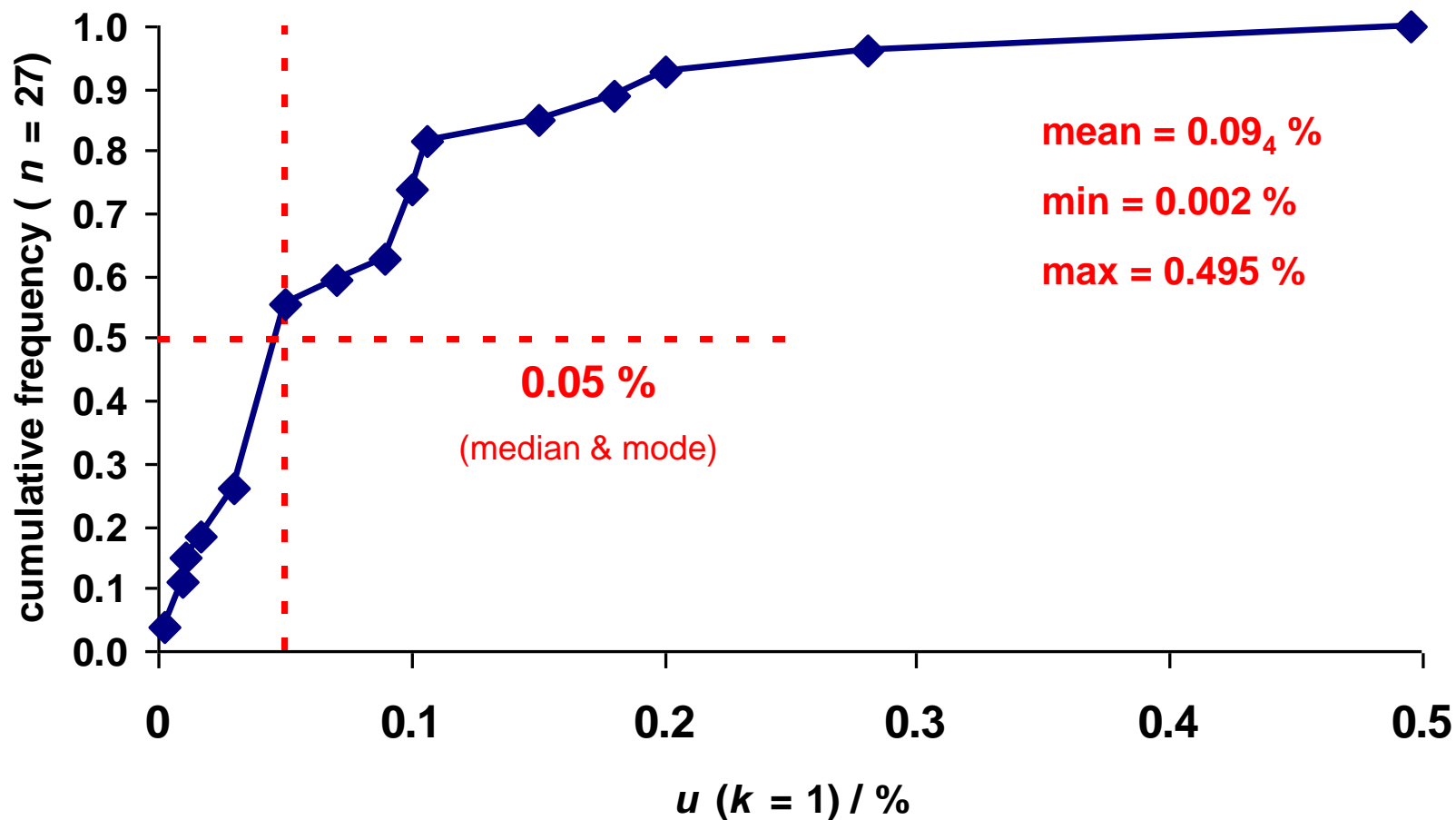
Light

Air

Many evaluation tests available
– but lots of work to do

RECALL THIS

WEIGHING UNCERTAINTIES REPORTED FOR ^{55}Fe COMPARISON



What is a realistic weighing uncertainty ?

How many weighings enter into final value & how combine ?

Is 0.05 % realistic ?

➡ For a single 20 mg source, this is 10 μg .

Not likely for direct reading measurement – many reasons

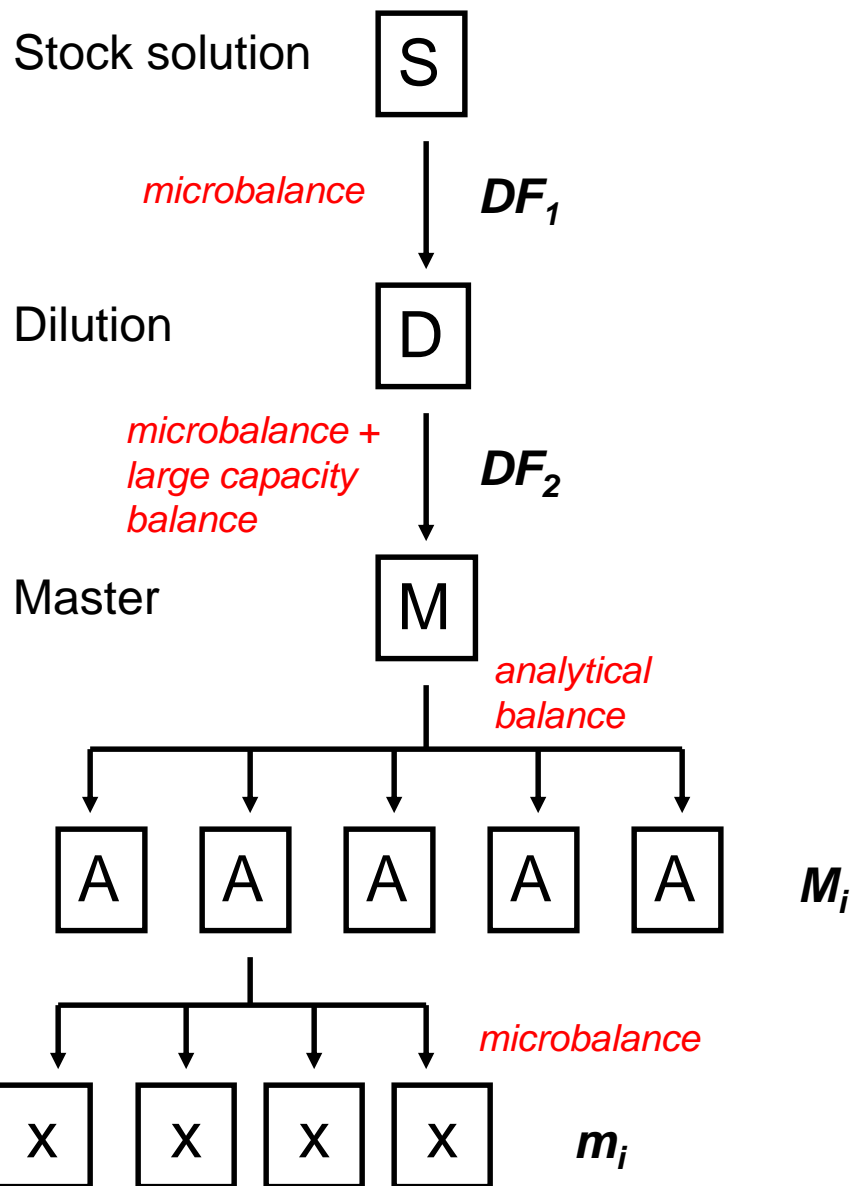
For substitution weighing, the E1 mass standard tolerance limit is 3 μg alone ! Maybe for a very good operator with excellent ambient controls. But probably not, unless using multiple sources to get precision.

➡ For a single 5 g mass, this is 2.5 mg.

Probably, but must have good analytical balance or better, good ambient controls, and reasonably good operator. May depend on what measuring.

➡ For a single 2 kg mass, this is 1 g.

Definitely, assuming have reasonable calibrated large capacity balance or weight set, and adequate ambient conditions and operator. Again, may depend on what measuring.



What is weighing uncertainty ?

Report one value in uncertainty budget ?

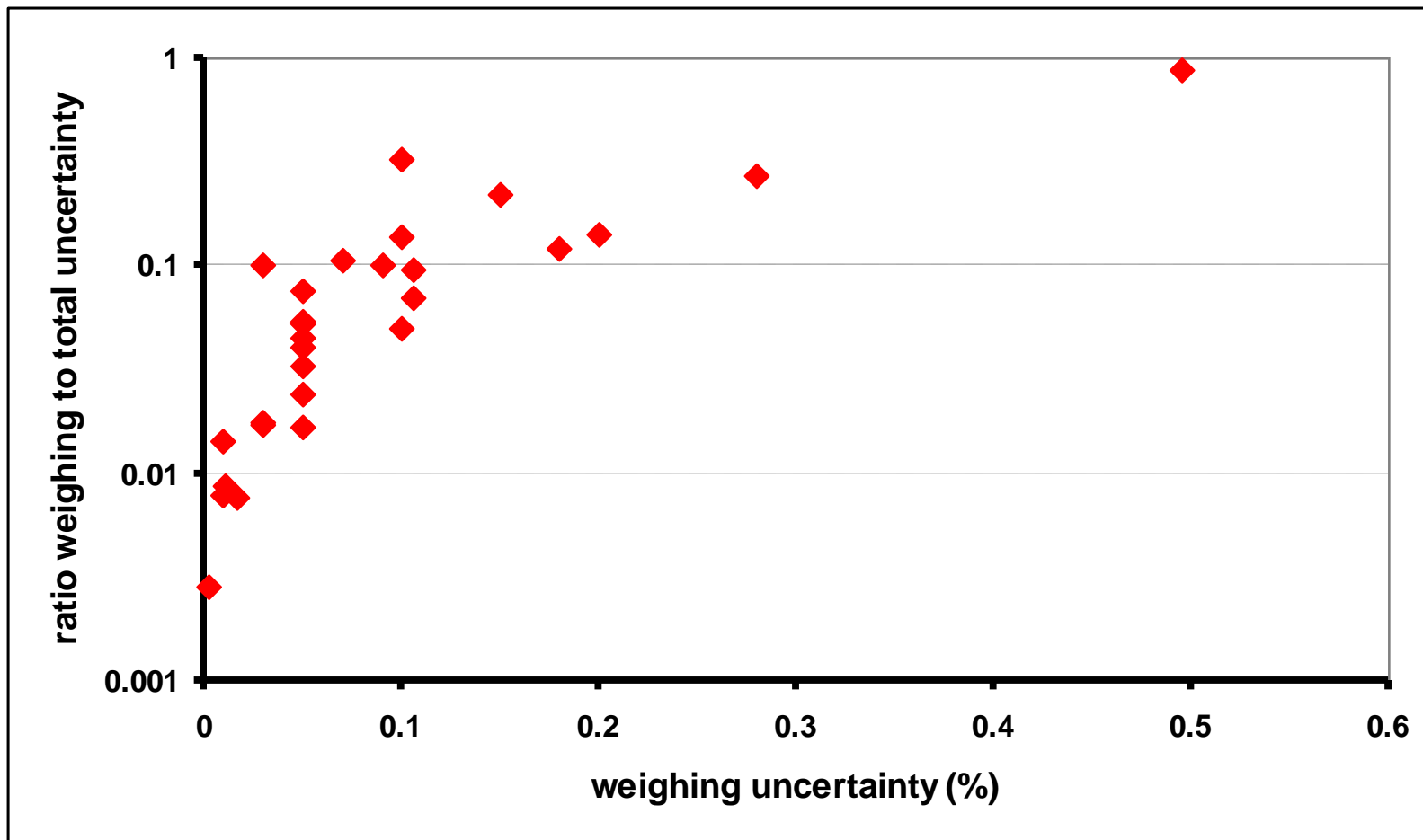
How combine ?

How handle evident correlations ?

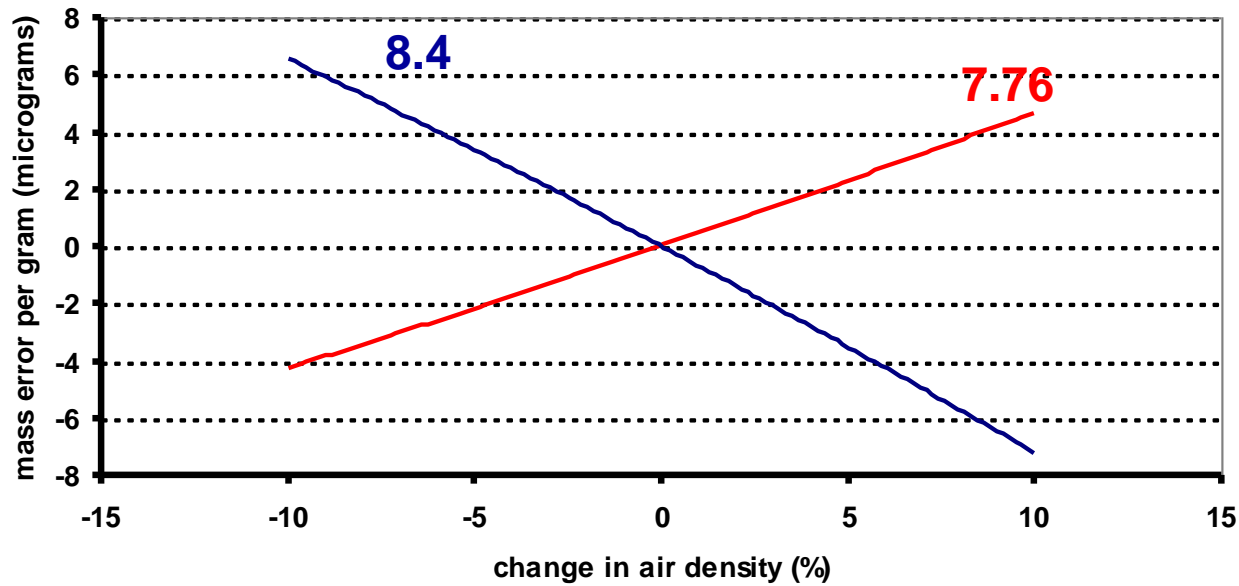


Some support slides follow, if needed

uncertainties reported for ^{55}Fe comparison



effect of density of built-in dial weights



$$\rho_0 = 0.00120 \text{ g}\cdot\text{cm}^{-3}$$

$$\rho_R = 8.0 \text{ g}\cdot\text{cm}^{-3}$$

For

$$\rho_{BW} = 7.76 \text{ g}\cdot\text{cm}^{-3}$$

$$\text{and } 8.4 \text{ g}\cdot\text{cm}^{-3}$$

$$\rho_a = 0.0012 \text{ g}\cdot\text{cm}^{-3} \pm 10 \%$$

$$\Delta M = W(\rho_0 - \rho_a)[1/\rho_R - 1/\rho_{BW}]$$

↑
1 g