



# **$^{209}\text{Po}$ and $^{210}\text{Pb}$ Radioactivity Standards: A REVIEW**

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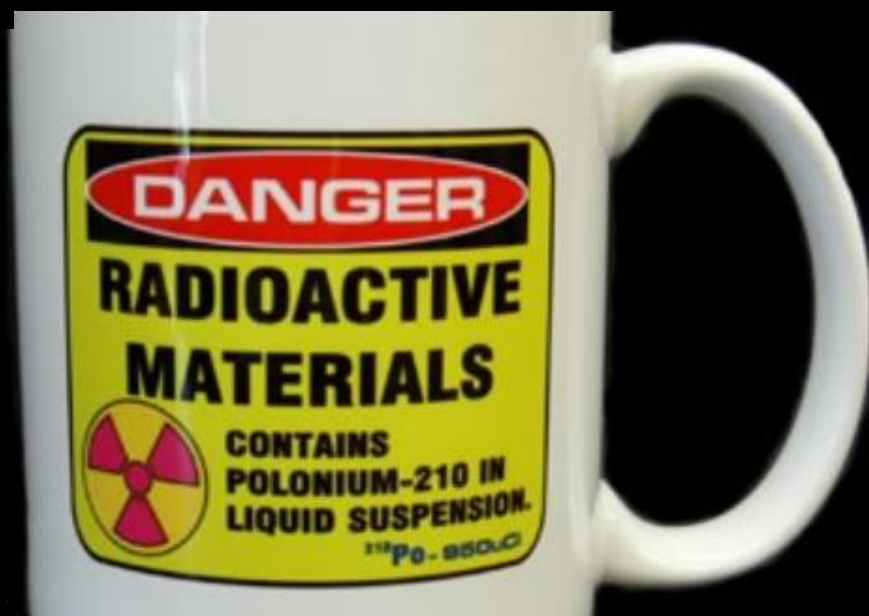


**International Topical Conference on Po and Radioactive Pb Isotopes**

**26 October 2009**

**Sevilla, Spain**

Recent events may have led to increasing interest in Po &  $^{210}\text{Pb}$ ,  
but they have been of interest to our laboratory for many years ....



This is a review of  
about 20 years of  
work by me (and  
colleagues)



# TOPICS

- 
- 1992 \* carrier free Po solution stability
  - 1994 \*  $^{205}\text{Pb}$  ( $^{209}\text{Po}$  decay) confounding of LS measurements
  - 1995 \*  $^{209}\text{Po}$  standard (SRM 4326)
  - 1996 \*  $^{209}\text{Po}$  decay data -- photon emission probabilities ( $\gamma$  &  $K_{\alpha}$  x)
  - 2005 re-standardization of  $^{209}\text{Po}$
  - 2007 \*  $^{210}\text{Pb}$  standard (SRM 4337)
  - 2007 \*  $^{209}\text{Po}$  half-life error
  - 2008 \*  $^{209}\text{Po}$  /  $^{210}\text{Pb}$  links
  - 2007 international comparison(s)  
of UK (NPL) & USA (NIST)  $^{210}\text{Pb}$  standards
  - 2008-9 “
- 

review not given in  
chronological order



\* *publication dates*

# Characteristic Elements of NIST Radioactivity Standardization Program *(R. Collé)*



Choice of nuclides / standards based on identified **user needs**

Standardization based on at least one **primary method**

Validity of primary method supported & confirmed by one or more **independent confirmatory methods**

Standardizations typically utilize **many trials**, with widely varying experimental conditions,  
minimizing type-B uncertainty assessments

Any new standardization **linked back** to all previous ones (when possible)  
through stored solutions or calibration factors for secondary instruments

**Disseminated standards** from primary methods used as SRM transfer standards  
and/or employed as sources for quality assurance, & proficiency testing programs

Primary standardization **uncertainties** ( $k = 2$ ) are typically  $< 1 \%$  (few tenths at  $k = 1$ )

**Comparisons** with others metrology labs to demonstrate & ensure international consistency

# World needs a Po tracer standard !

$^{210}\text{Po}$	0.4 a	5.3 MeV $\alpha$
$^{208}\text{Po}$	2.9 a	5.1 MeV $\alpha$
$^{209}\text{Po}$	102 a	4.9 MeV $\alpha$ + <i>junk</i>

at NIST

< 1980 only  $^{210}\text{Po}$  calibration sources

1984  $^{208}\text{Po}$  standard (SRM 4327)

1994  $^{209}\text{Po}$  standard (SRM 4326)

> 2007 no Po tracer standard available  
(now underway)

# Carrier-free Po solutions & standards are stable

“Long-term stability of carrier-free polonium solution standards”

R. Collé, *Radioact. Radiochem.* **4**, no.2, 20-35 (1993).

## The Periodic Table of the Elements

1 <b>H</b> Hydrogen 1.00794																	2 <b>He</b> Helium 4.003														
3 <b>Li</b> Lithium 6.941	4 <b>Be</b> Beryllium 9.012182											5 <b>B</b> Boron 10.811	6 <b>C</b> Carbon 12.0107	7 <b>N</b> Nitrogen 14.00674	8 <b>O</b> Oxygen 15.9994	9 <b>F</b> Fluorine 18.9984032	10 <b>Ne</b> Neon 20.1797														
11 <b>Na</b> Sodium 22.989770	12 <b>Mg</b> Magnesium 24.3050											13 <b>Al</b> Aluminum 26.981538	14 <b>Si</b> Silicon 28.0855	15 <b>P</b> Phosphorus 30.973761	16 <b>S</b> Sulfur 32.066	17 <b>Cl</b> Chlorine 35.4527	18 <b>Ar</b> Argon 39.948														
19 <b>K</b> Potassium 39.0983	20 <b>Ca</b> Calcium 40.078	21 <b>Sc</b> Scandium 44.955910	22 <b>Ti</b> Titanium 47.867	23 <b>V</b> Vanadium 50.9415	24 <b>Cr</b> Chromium 51.9961	25 <b>Mn</b> Manganese 54.938049	26 <b>Fe</b> Iron 55.845	27 <b>Co</b> Cobalt 58.933200	28 <b>Ni</b> Nickel 58.6934	29 <b>Cu</b> Copper 63.546	30 <b>Zn</b> Zinc 65.39	31 <b>Ga</b> Gallium 69.723	32 <b>Ge</b> Germanium 72.61	33 <b>As</b> Arsenic 74.92160	34 <b>Se</b> Selenium 78.96	35 <b>Br</b> Bromine 79.904	36 <b>Kr</b> Krypton 83.80														
37 <b>Rb</b> Rubidium 85.4678	38 <b>Sr</b> Strontium 87.62	39 <b>Y</b> Yttrium 88.90585	40 <b>Zr</b> Zirconium 91.224	41 <b>Nb</b> Niobium 92.90638	42 <b>Mo</b> Molybdenum 95.94	43 <b>Tc</b> Technetium (98)	44 <b>Ru</b> Ruthenium 101.07	45 <b>Rh</b> Rhodium 102.90550	46 <b>Pd</b> Palladium 106.42	47 <b>Ag</b> Silver 107.8682	48 <b>Cd</b> Cadmium 112.411	49 <b>In</b> Indium 114.818	50 <b>Sn</b> Tin 118.710	51 <b>Sb</b> Antimony 121.76	52 <b>Te</b> Tellurium 127.60	53 <b>I</b> Iodine 126.90447	54 <b>Xe</b> Xenon 131.29														
55 <b>Cs</b> Cesium 132.90545	56 <b>Ba</b> Barium 137.327	57 <b>La</b> Lanthanum 138.9055	72 <b>Hf</b> Hafnium 178.49	73 <b>Ta</b> Tantalum 180.9479	74 <b>W</b> Tungsten 183.84	75 <b>Re</b> Rhenium 186.207	76 <b>Os</b> Osmium 190.23	77 <b>Ir</b> Iridium 192.217	78 <b>Pt</b> Platinum 195.078	79 <b>Au</b> Gold 196.96655	80 <b>Hg</b> Mercury 200.59	81 <b>Tl</b> Thallium 204.3833	82 <b>Pb</b> Lead 207.2	83 <b>Bi</b> Bismuth 208.9804	84 <b>Po</b> Polonium (209)	85 <b>At</b> Astatine (210)	86 <b>Rn</b> Radon (222)														
87 <b>Fr</b> Francium (223)	88 <b>Ra</b> Radium (226)	89 <b>Ac</b> Actinium (227)	104 <b>Rf</b> Rutherfordium (261)	105 <b>Db</b> Dubnium (262)	106 <b>Sg</b> Seaborgium (263)	107 <b>Bh</b> Bohrium (262)	108 <b>Hs</b> Hassium (265)	109 <b>Mt</b> Meitnerium (266)	110 <b>Ds</b> Darmstadtium (269)	111 <b>Rg</b> Roentgenium (272)	112 <b>Cn</b> Copernicium (277)	113 <b>Nh</b> Nihonium (284)	114 <b>Fl</b> Flerovium (289)	115 <b>Mc</b> Moscovium (288)	116 <b>Lv</b> Livermorium (293)	117 <b>Ts</b> Tennessine (294)	118 <b>Og</b> Oganesson (294)														
																		58 <b>Ce</b> Cerium 140.116	59 <b>Pr</b> Praseodymium 140.90765	60 <b>Nd</b> Neodymium 144.24	61 <b>Pm</b> Promethium (145)	62 <b>Sm</b> Samarium 150.36	63 <b>Eu</b> Europium 151.964	64 <b>Gd</b> Gadolinium 157.25	65 <b>Tb</b> Terbium 158.92534	66 <b>Dy</b> Dysprosium 162.50	67 <b>Ho</b> Holmium 164.93032	68 <b>Er</b> Erbium 167.26	69 <b>Tm</b> Thulium 168.93421	70 <b>Yb</b> Ytterbium 173.04	71 <b>Lu</b> Lutetium 174.967
																		90 <b>Th</b> Thorium 232.0381	91 <b>Pa</b> Protactinium 231.03588	92 <b>U</b> Uranium 238.0289	93 <b>Np</b> Neptunium (237)	94 <b>Pu</b> Plutonium (244)	95 <b>Am</b> Americium (243)	96 <b>Cm</b> Curium (247)	97 <b>Bk</b> Berkelium (247)	98 <b>Cf</b> Californium (251)	99 <b>Es</b> Einsteinium (252)	100 <b>Fm</b> Fermium (257)	101 <b>Md</b> Mendelevium (258)	102 <b>No</b> Nobelium (259)	103 <b>Lr</b> Lawrencium (262)

Carrier free --

How strong acid ?

In early 1990's not known  
(differing reports)

tests on  $^{208}\text{Po}$  solutions  
over 1.2 to  $\approx 9$  years in  
range 0.1 to 1 mol L<sup>-1</sup> HCl

< 0.3 N unstable

0.3-0.5 N equivocal

> 1 N clearly stable

R. Collé, *Radioact. Radiochem*  
**4**, no. 2, 20 (1993).

electronic homologues

Po [Xe] 4f<sup>14</sup> 5d<sup>10</sup> 6s<sup>2</sup> 6p<sup>4</sup>

Te [Kr] 4d<sup>10</sup> 5s<sup>2</sup> 5p<sup>4</sup>

Solutions & standards – needs  
Te carrier or strong acid

Po at trace concentrations (under  
alkaline, neutral & weak acid  
conditions) known to be unstable:  
can be readily hydrolyzed,  
chemically deposited, volatilized;  
exhibit “radiocolloidal behavior”;  
“plate out & adsorption onto glass  
surfaces.

# $^{209}\text{Po}$

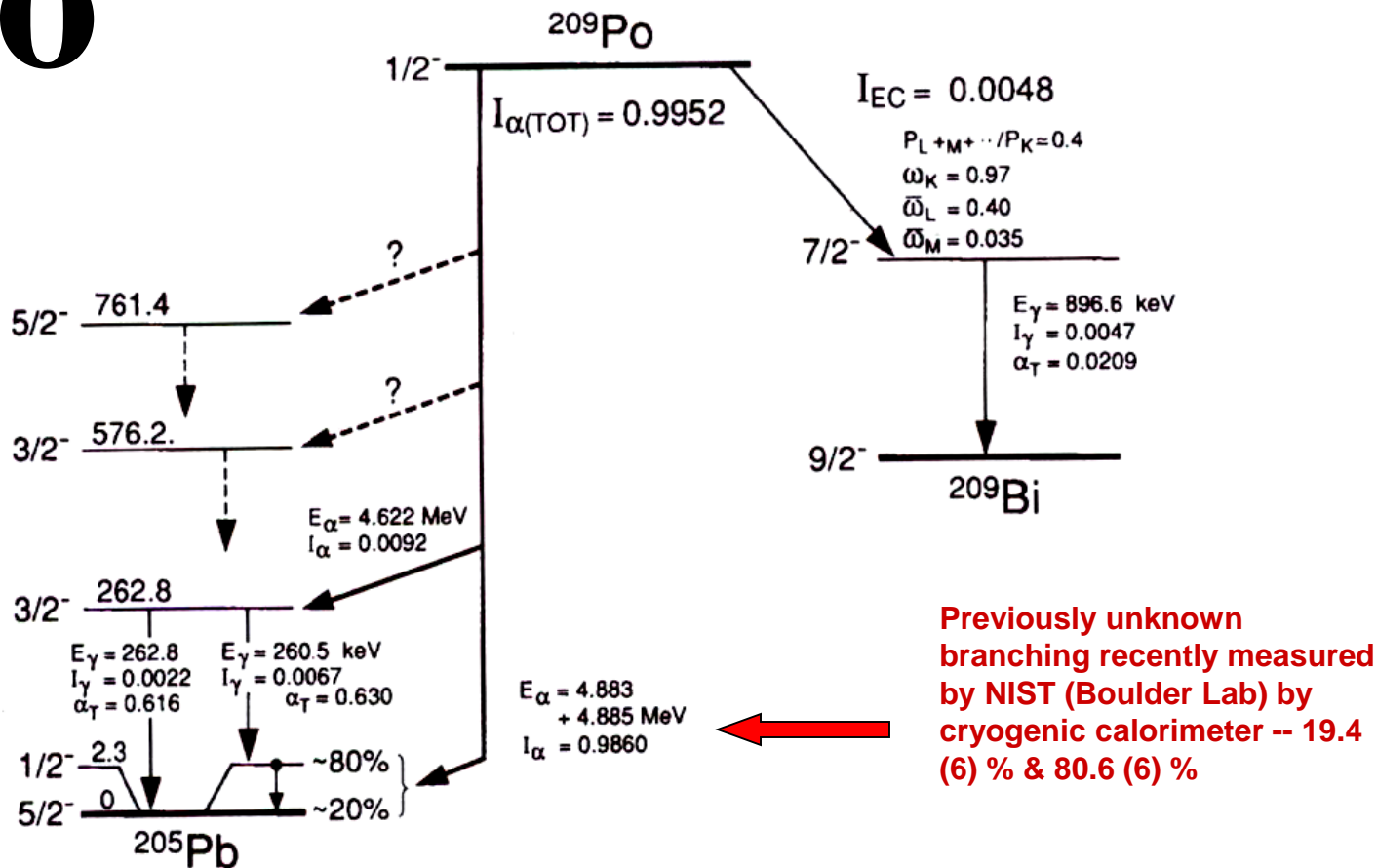


Fig. 2. Partial decay scheme for the  $^{209}\text{Po}$  alpha and electron capture branch decays.



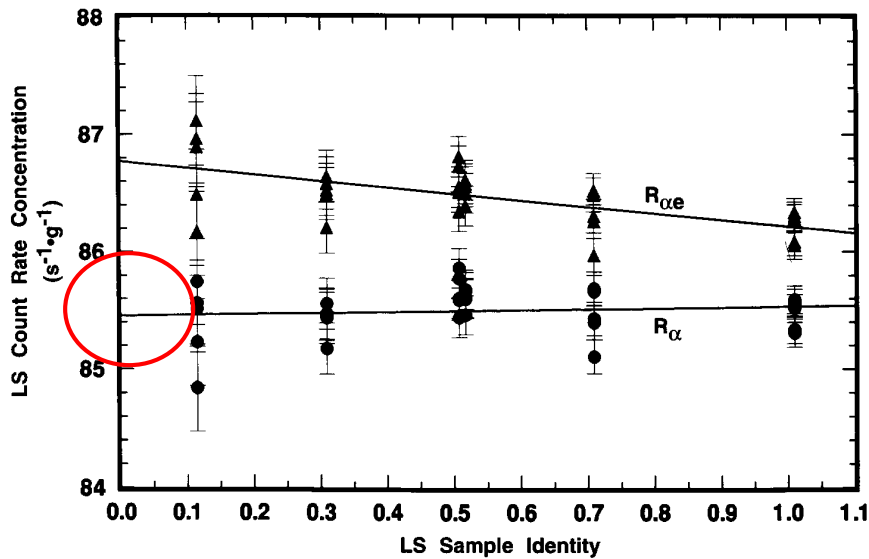


Fig. 11. LS counting rate concentrations  $R_{\alpha e}$  (closed triangles) and  $R_{\alpha}$  (closed circles) obtained with the Beckman instrument for the N series samples as a function of  $m_s$  (and sample quenching). The solid lines are linear regressions fitted to the data.

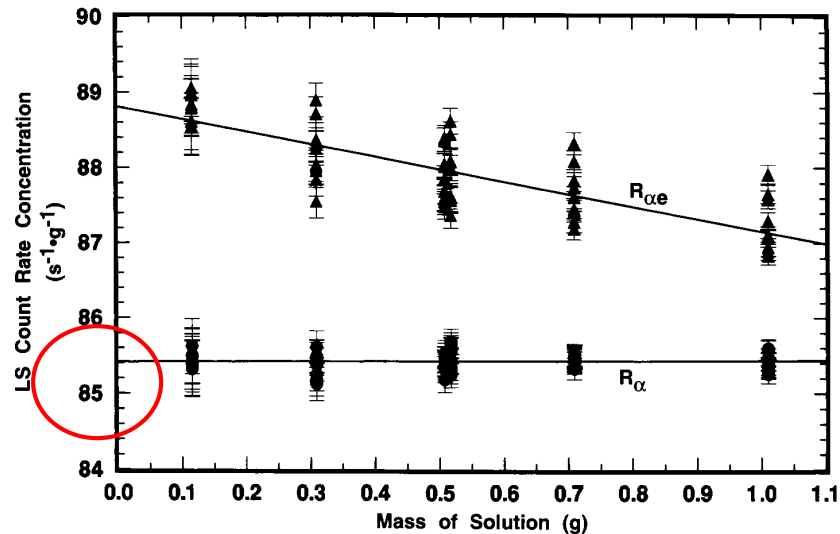


Fig. 12. LS counting rate concentrations  $R_{\alpha e}$  and  $R_{\alpha}$  as a function of  $m_s$  (analogous to that of Fig. 11) as obtained with the Packard instrument.

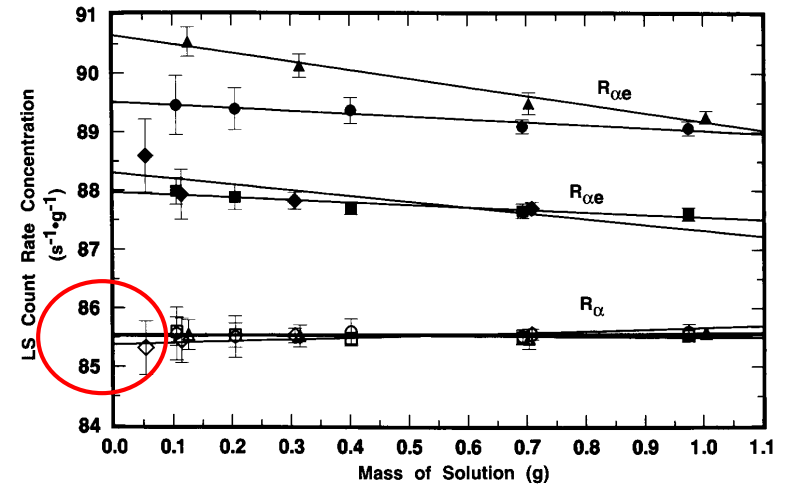


Fig. 13. LS counting rate concentrations  $R_{\alpha e}$  and  $R_{\alpha}$  obtained with the two LS systems for the P and Q series samples in 1994. Closed squares ( $R_{\alpha e}$ ) and open squares ( $R_{\alpha}$ ) represent the mean values for samples Q5 through Q8 with the Packard; closed and open triangles represent  $R_{\alpha e}$  and  $R_{\alpha}$ , respectively, for samples P1 through P5 with the Packard; closed and open circles ( $R_{\alpha e}$  and  $R_{\alpha}$ ) are for samples Q1 through Q4 with the Beckman; and closed and open circles ( $R_{\alpha e}$  and  $R_{\alpha}$ ) are for samples P1 through P5 with the Beckman. Each plotted value corresponds to the mean of 5 to 18 replicate measurements on each sample. The error bars represent standard deviation uncertainty intervals on the means. The solid lines are unweighted linear fits to the data. Although the  $R_{\alpha e}$  values vary with the instrument used to perform the measurements (Packard or Beckman) and with sample compositions, all of the  $R_{\alpha}$  values are statistically equivalent and invariant.

In 1994

LS result confirmed by  
 $2\pi\alpha$  proportional  
counting

from

Collé, et al., *J. Res. Natl. Inst. Stds. Tech.*  
**100**, 1 (1995).

SAME IN 2005

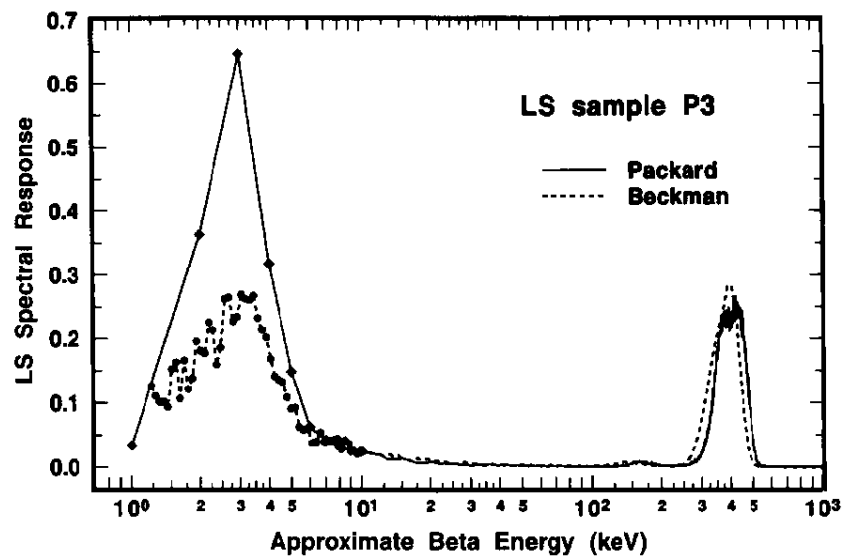


Fig. 6. Comparison of the  $^{209}\text{Po}$  LS spectra obtained with the Beckman and Packard instruments.

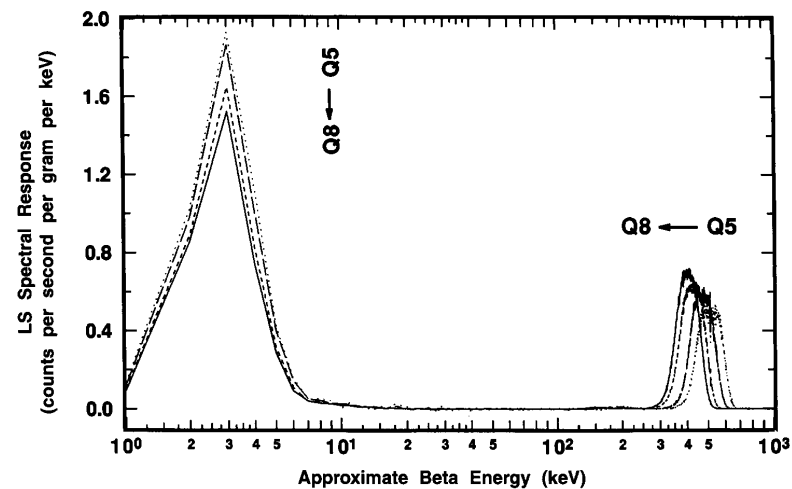


Fig. 14. LS spectra of increasingly quenched samples Q5 through Q8 obtained with the Packard counting system.

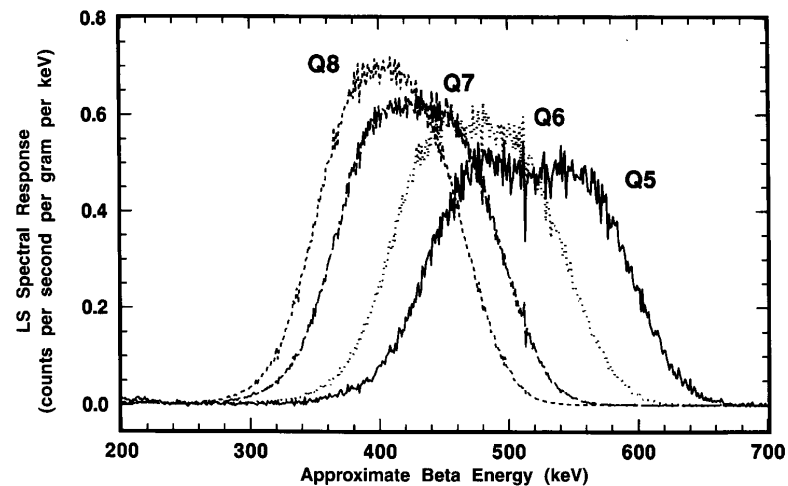


Fig. 15. Details of the broad alpha peaks shown in the full spectra of Fig. 14. The peak widths (FWHM) on a relative basis and peak areas are approximately equal in all four samples Q5 through Q8.

# <sup>209</sup>Po half-life in error by 25 % !!

Result supported by work  
on <sup>210</sup>Pb – previous story

15 march 1994

$$R_{\alpha} = (85.42 \pm 0.18) \text{ s}^{-1} \text{ g}^{-1}$$

15 November 2005

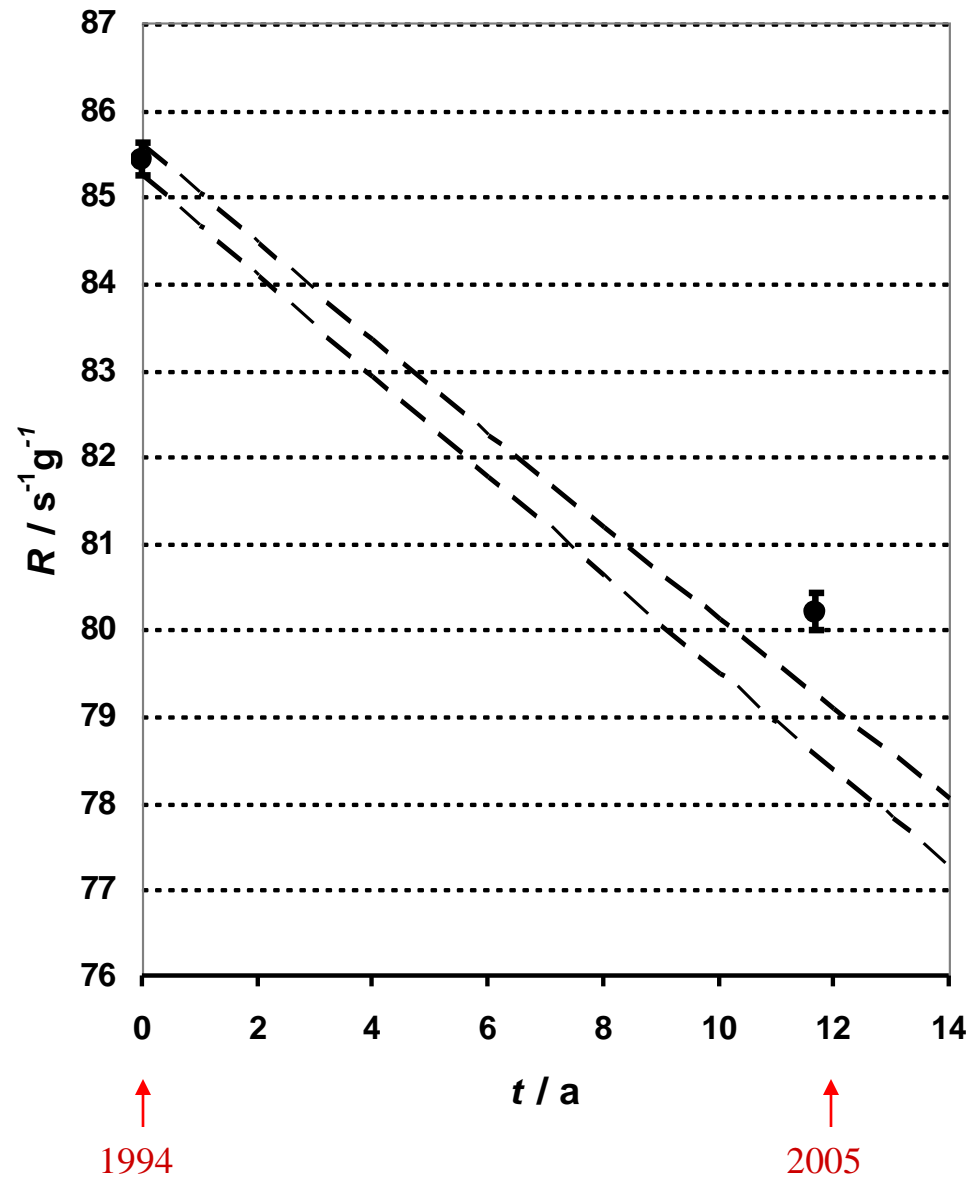
$$R_{\alpha} = (80.20 \pm 0.22) \text{ s}^{-1} \text{ g}^{-1}$$

2 point fit gives

$$T_{1/2} = 128 \text{ a}$$

U = 5.5 % (7 a)

**Not considered a new  
determination**



# $^{210}\text{Pb}$

Difficult case

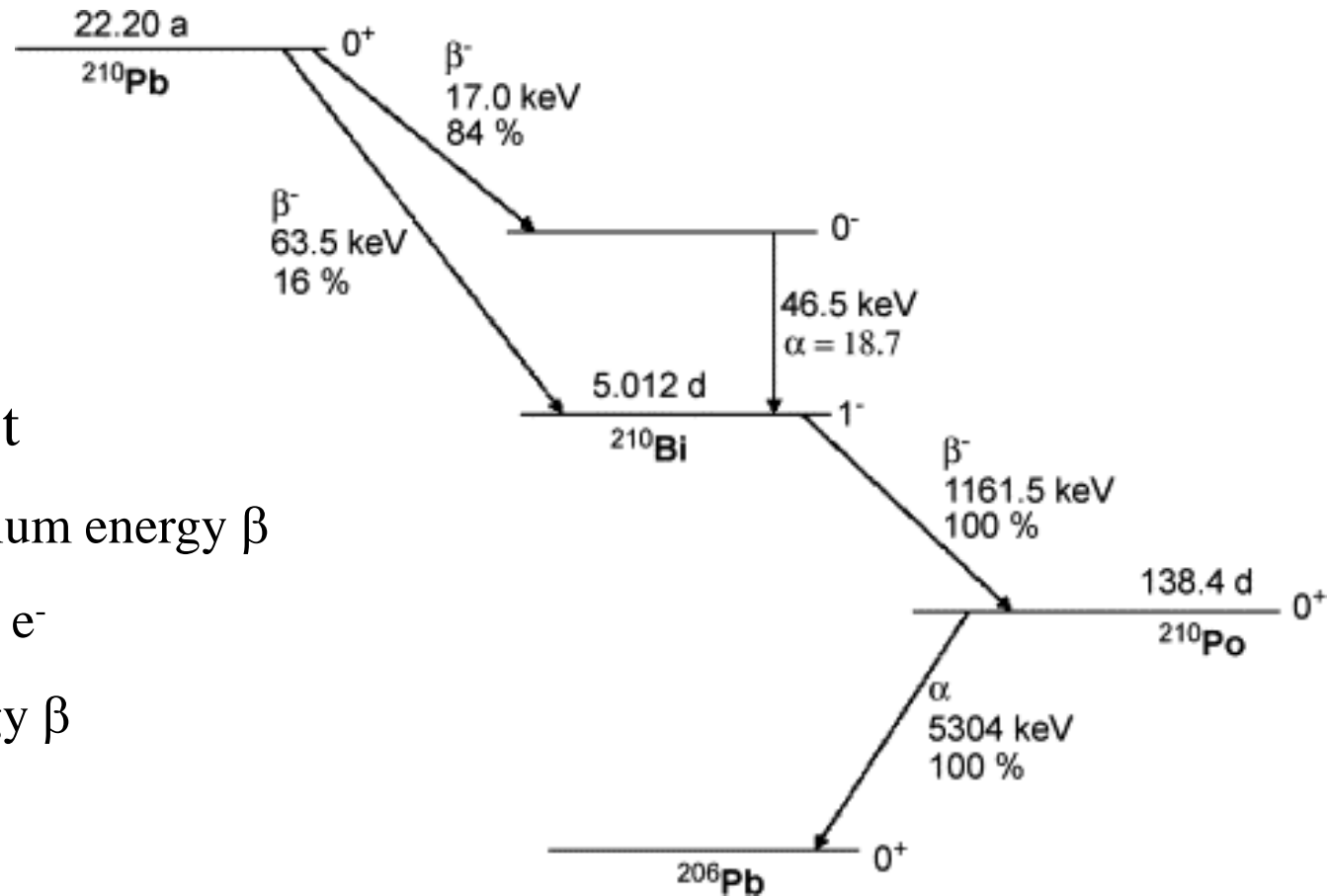
rarely done by metrology labs

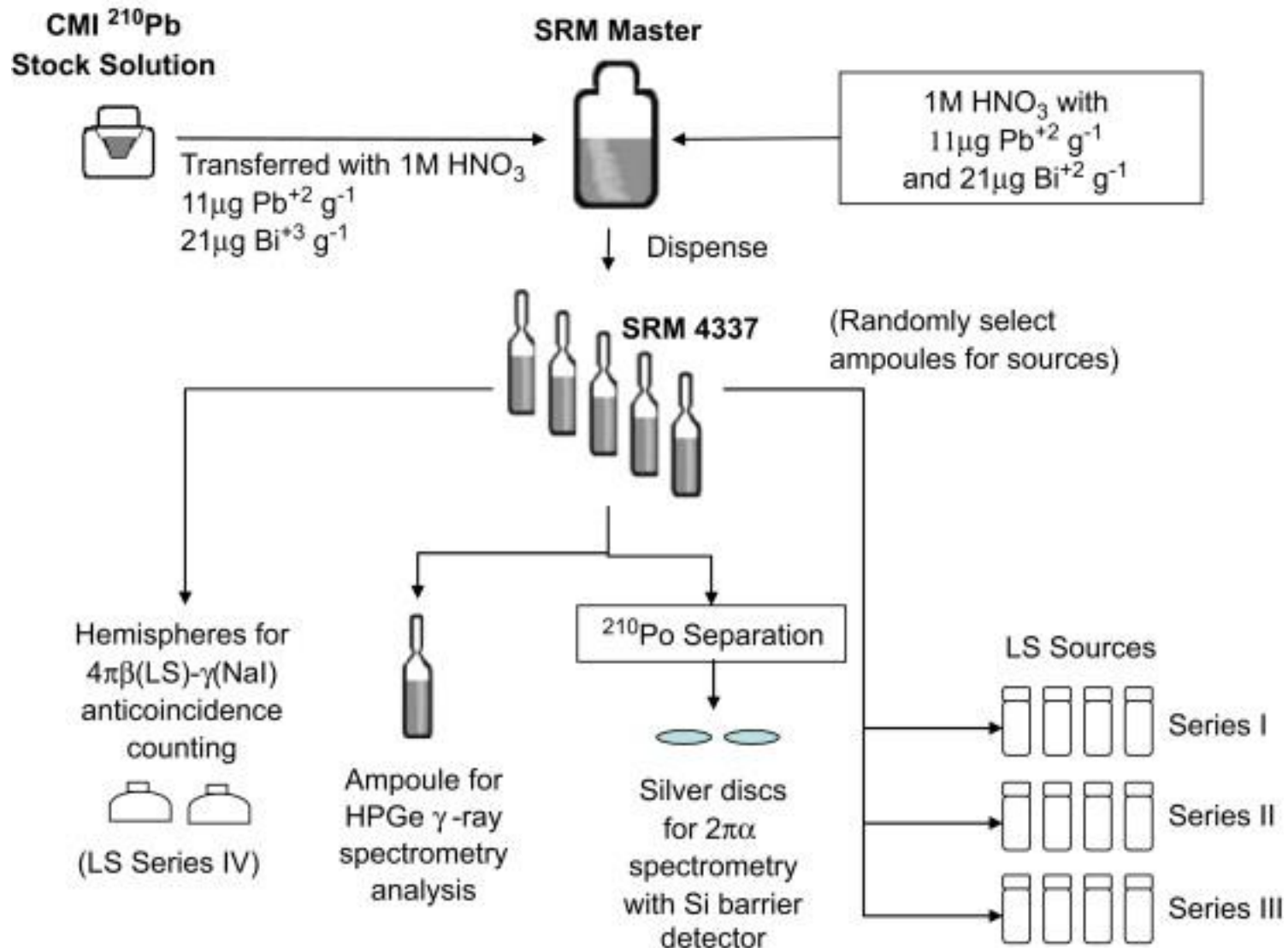
different methods used

Comparisons with NPL standard (via PTB too)

Need to detect

- low – medium energy  $\beta$
- conversion  $e^-$
- High energy  $\beta$
- $\alpha$





**LTAC**

**$\gamma$  spect**

**$\alpha$  spect**

**CNET**

INADEQUATE TIME TO COVER

$^{210}\text{Pb}$  standardization methods & data are published

L. Laureano-Perez, R. Collé, et al., *Appl. Radiat. Isot.* 65, 1368 (2007).

Ask or see me if interested -- I have 6 back-up slides of information !



**NIST**

# $^{210}\text{Pb}$ standardization results (SRM 4337)

	primary method	relative standard uncertainty	confirmatory measurements	percent differences	
$^{210}\text{Pb}$	$4\pi\alpha\beta$ LS CNET	1.2 %	$4\pi\alpha\beta(\text{LS})-\gamma(\text{NaI})$ anticon. Counting ( $\pm 1.7\%$ )	+ 0.7	← many assumptions, big extrapolation
			$^{210}\text{Po}$ $\alpha$ spect. (102 a $^{209}\text{Po}$ tracer) ( $\pm 1\%$ )	- 3.0	} $^{209}\text{Po}$ half-life link
			$^{210}\text{Po}$ $\alpha$ spect. (128 a $^{209}\text{Po}$ tracer) ( $\pm 1\%$ )	- 1.3	
			HPGe photon spect. ( $\pm 4.2\%$ )	+ 4.7	←* big unc. if don't use some other $^{210}\text{Pb}$ std

uncertainty atypically large compared to  
most NIST radioactivity standardizations

\* probable error due to  $^{210}\text{Bi}$  bremsstrahlung

Relatively large 2.4 % uncertainty ( $k = 2$ ) because of

- (1) LS cocktail composition effects
- (2) tracing code differences & assumptions,
- (3) lack of good confirmatory measurements,



**$^{209}\text{Po}$**

**$^{209}\text{Po}$**

**$^{210}\text{Pb}$**

**LINKED**

**In applications**

**– use of  $^{209}\text{Po}$  tracer for  $^{210}\text{Po}$  ( $^{210}\text{Pb}$  assays)**

**In our standardization measurements**

# $^{210}\text{Pb}$ standardization results (SRM 4337)

	primary method	relative standard uncertainty	confirmatory measurements	Percent differences
$^{210}\text{Pb}$	$4\pi\alpha\beta$ LS CNET	1.2 %	$4\pi\alpha\beta(\text{LS})-\gamma(\text{NaI})$ anticoin. Counting	+ 0.7
			$^{210}\text{Po}$ $\alpha$ spect. (102 a $^{209}\text{Po}$ tracer)	- 3.0
			$^{210}\text{Po}$ $\alpha$ spect. (128 a $^{209}\text{Po}$ tracer)	- 1.3
			HPGe photon spect.	+ 4.7

Needs longer half-life for consistency

\*

\* probable error due to  $^{210}\text{Bi}$  bremsstrahlung

uncertainty atypically large compared to most NIST radioactivity standardizations

NPL  
A051087



10 g

5 g



NPL#1

5 g



NPL#2

5 g



NIST#1

5 g



NIST#2



NIST  
4337-47

Dilute and adjust volume  
+ carrier concentrations  
to match NPL ampoules

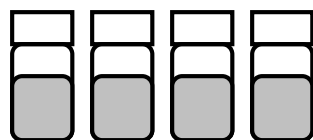
**Compare with  
NPL (PTB)**

Comparative measurements:

$4\pi\gamma$  NaI sandwich detector

HPGe  $\gamma$  spectrometry

NPL#1



$4\pi\alpha\beta$  LS



windowless Si(Li) detector

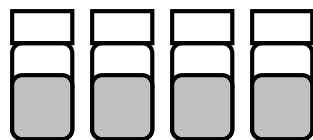


Po chemistry



$2\pi\alpha$  Si surface  
barrier detector

NIST#1



LS vials



thin point  
sources



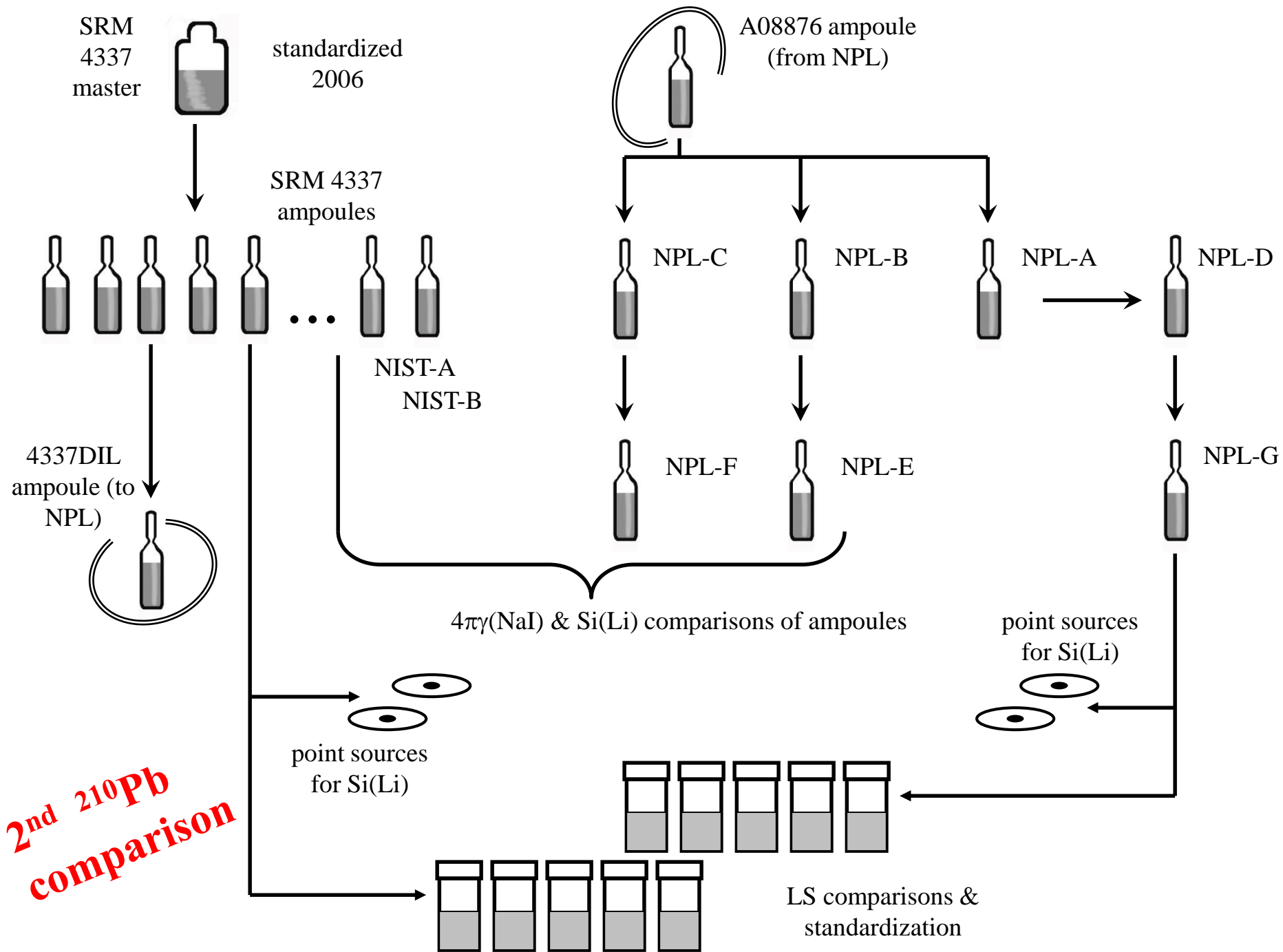
Ag disks

# 2007 Comparison of the NIST and NPL $^{210}\text{Pb}$ standards by five measurement methods.

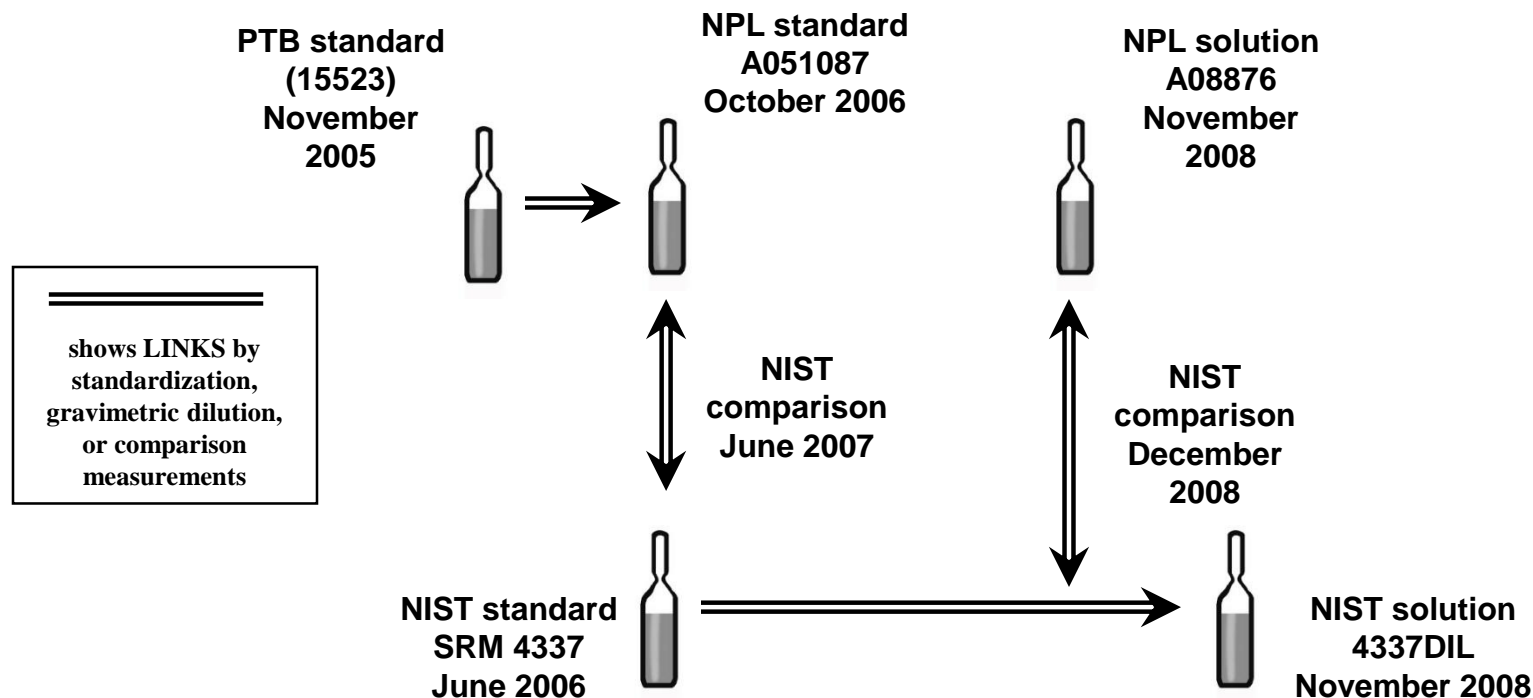
method	NPL / NIST ratio	relative standard uncertainty	difference
NPL and NIST certified values from primary standardizations	<b>0.037484</b>	1.5 %	--
$4\pi\gamma(\text{NaI})$	0.037373	0.56 %	- 0.30 %
HPGe spectrometry.	0.036542	0.71 %	- 2.6 %
$4\pi\alpha\beta(\text{LS})$	0.037249	0.17 %	- 0.63 %
$^{210}\text{Po}$ assay ( $2\pi\alpha$ spect.)	0.03736	0.75 %	- 0.33 %
Si(Li) low-energy spectrometry	0.0381	1.9 %	+ 1.6 %

← \*

\* probable error due to  $^{210}\text{Bi}$  bremsstrahlung



# $^{210}\text{Pb}$ NPL / NIST comparisons



similar 0.6 % NPL / NIST agreement – consistent with previous comparison  
(for  $\cong 1$  % combined standard uncertainty in two standards)

For data, see **Lena Johansson** poster – *this conference*

# Summary important points:

- carrier-free Po solutions,  $\geq 1 \text{ mol L}^{-1} \text{ HCl}$
- $^{205}\text{Pb}$  isomeric state may confound  $^{209}\text{Po}$  standardization / measurements
- $^{209}\text{Po}$  standard will be limited by half-life uncertainty
- $^{209}\text{Po}$  &  $^{210}\text{Pb}$  linked through standards
- $^{210}\text{Pb}$  standard uncertainty may be over-estimated
- reasonable international agreement on  $^{210}\text{Pb}$  standards  $^{210}\text{Pb}$



**Thank you.**

# Acknowledgements & Special Thanks to

Lizbeth Laureano-Perez

Ryan Fitzgerald

Lena Johansson (NPL)

Iisa Outola

Leticia Pibida

Arzu Arinc (NPL)

Chris Gilligan (NPL)

Philippe Cassette (LNHB)

Frank Schima

Zhichao Lin

Pamela Hodge

J.M.R. (Robin) Hutchinson

Joy Thomas

Bert Coursey

and probably others I inadvertently forgot

**for their contributions over the years**



# FIN



**Backup slides only follow**

**$^{210}\text{Pb}$  standardization data**

# $^{210}\text{Pb}$ LS results (CN2003 code)

Pb-210

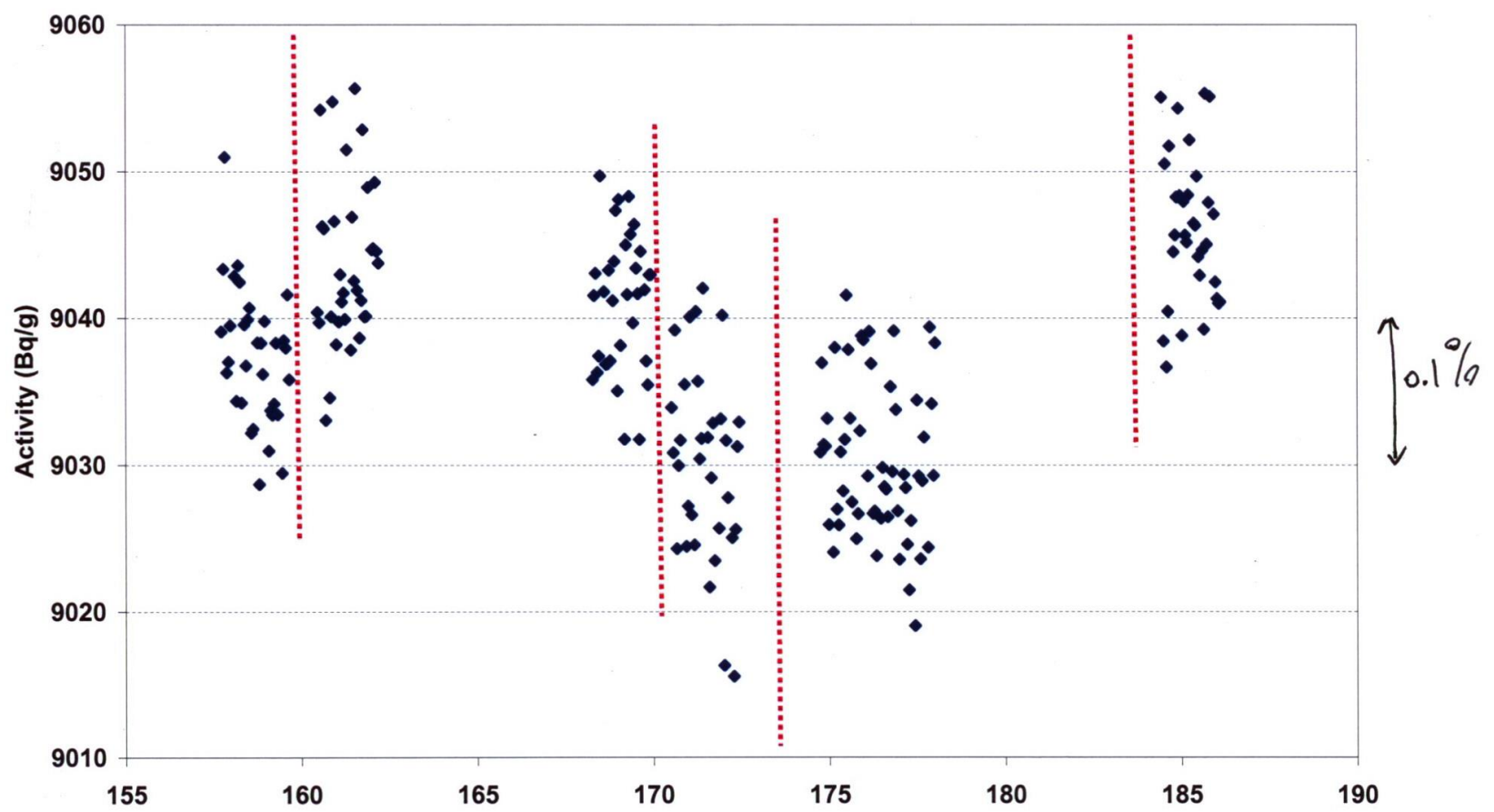
Series	Average	sd (%)	Normal	nc	ns	Counter	Scint	Age start	Age end	f <sub>H2O</sub>	ε <sub>H-3</sub>
1	9037.397	0.029	Y	3	11	Packard	HS	0.21	2.12	0.1	0.36-0.30
	9043.779	0.008	Y	3	11	Beckman	HS	2.95	4.65		
	9041.030	0.014	Y	3	11	Wallac	HS	10.76	12.4		
	9030.169	0.021	Y*	3	11	Packard	HS	13	14.9		
	9030.377	0.017	Y*	5	11	Packard	HS	17.22	20.46		
	9046.129	0.007	Y*	3	11	Wallac	HS	26.93	28.57		
2	9034.269	0.031	N	5	7	Packard	PCS	0.11	4.06	0.01	0.40-0.22
	9035.597	0.035	Y	5	7	Packard	PCS	0.11	4.06	0.04	
	9039.466	0.027	N	3	7	Wallac	PCS	4.78	6.91	0.01	
	9044.048	0.014	Y	3	7	Wallac	PCS	4.78	6.91	0.04	
	9040.539	0.026	no	3	7	Beckman	PCS	10.74	12.83	0.01	
	9041.935	0.026	yes	3	7	Beckman	PCS	10.74	12.83	0.04	
	9032.072	0.056	no	5	7	Packard	PCS	14.17	18.6	0.01	
	9026.263	0.034	yes	5	7	Packard	PCS	14.17	18.6	0.04	

\* Data normal after removing sample with unstable cocktail

436 determinations

Series	Average	SD	SD (%)	Normal
1	9038.147	6.7577	0.07477	Yes
2	9036.774	5.8702	0.06496	Yes
Total	9037.362	6.0511	0.06696	Yes

Pb-210: all counters; Composition 1 *f* 3



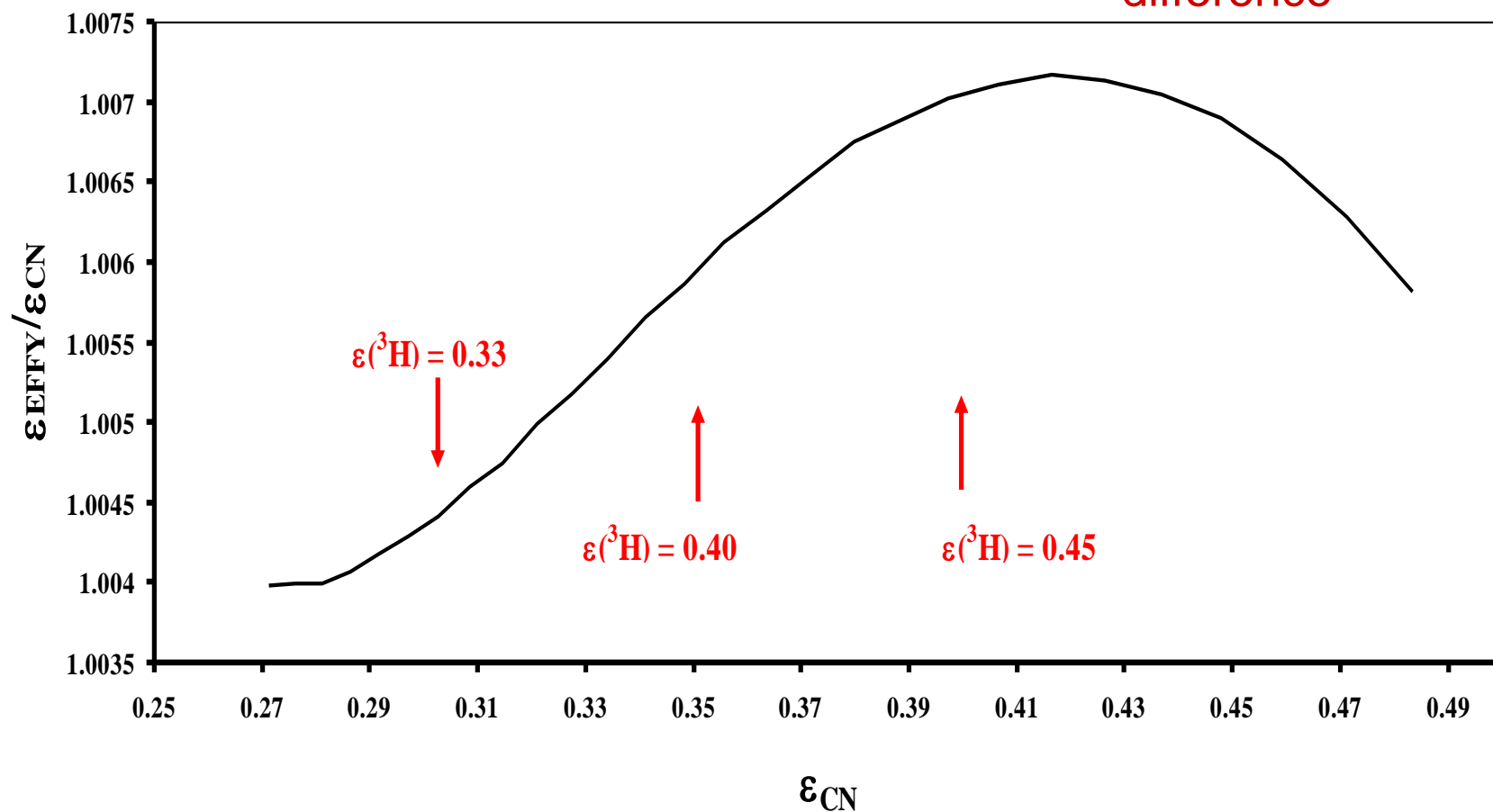
	P	B		W	P	P		W
	3	3		3	3	5		3
cycles								
x 12 samples.	36	36		36	36	60		36

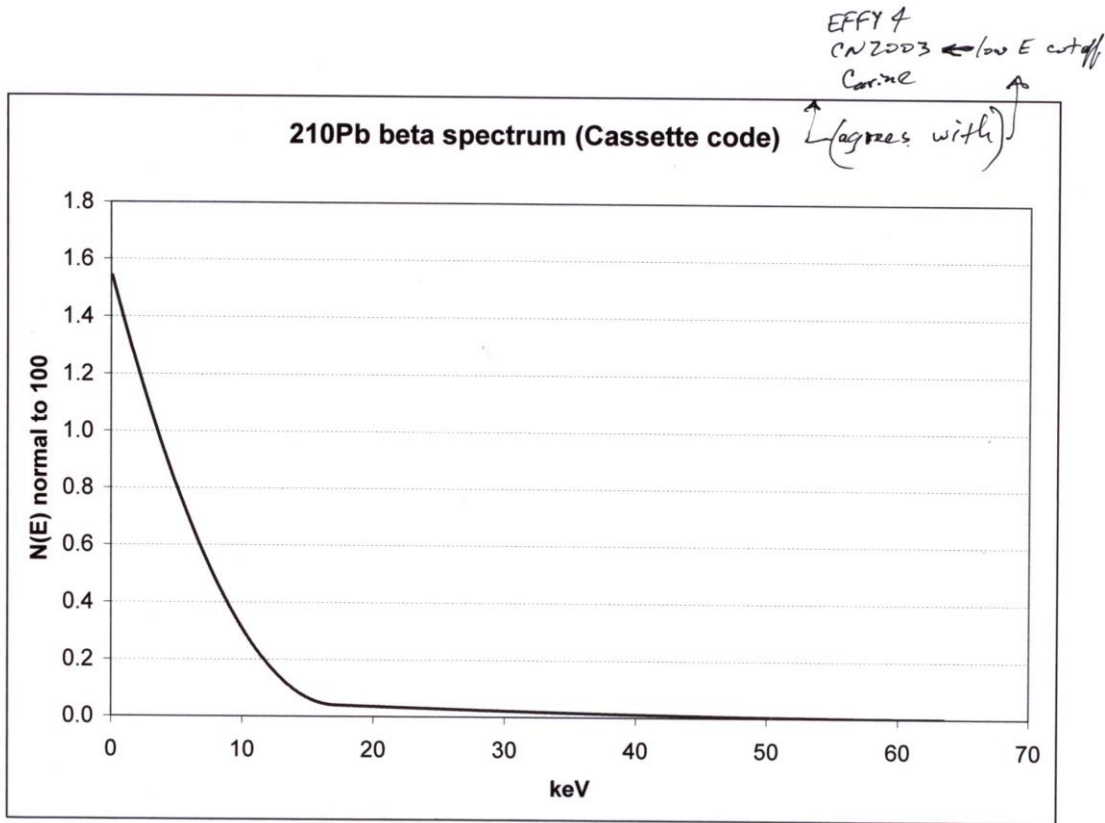
# CN2003 vs EFFY4 code differences

(just Beta efficiency part)

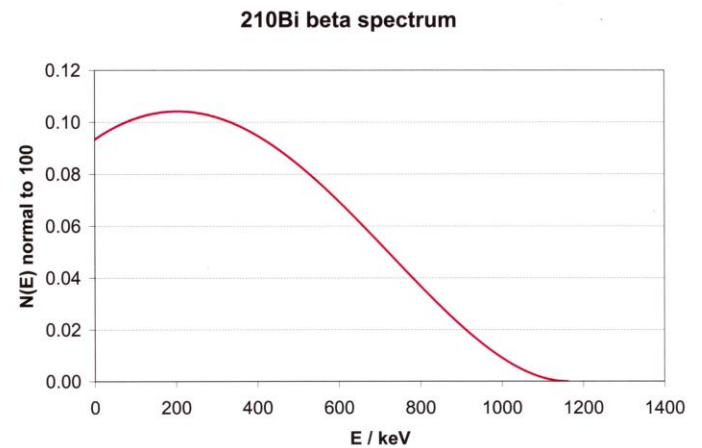
$^{210}\text{Pb}$

about 0.6 %  
difference





differences not due  
to spectra



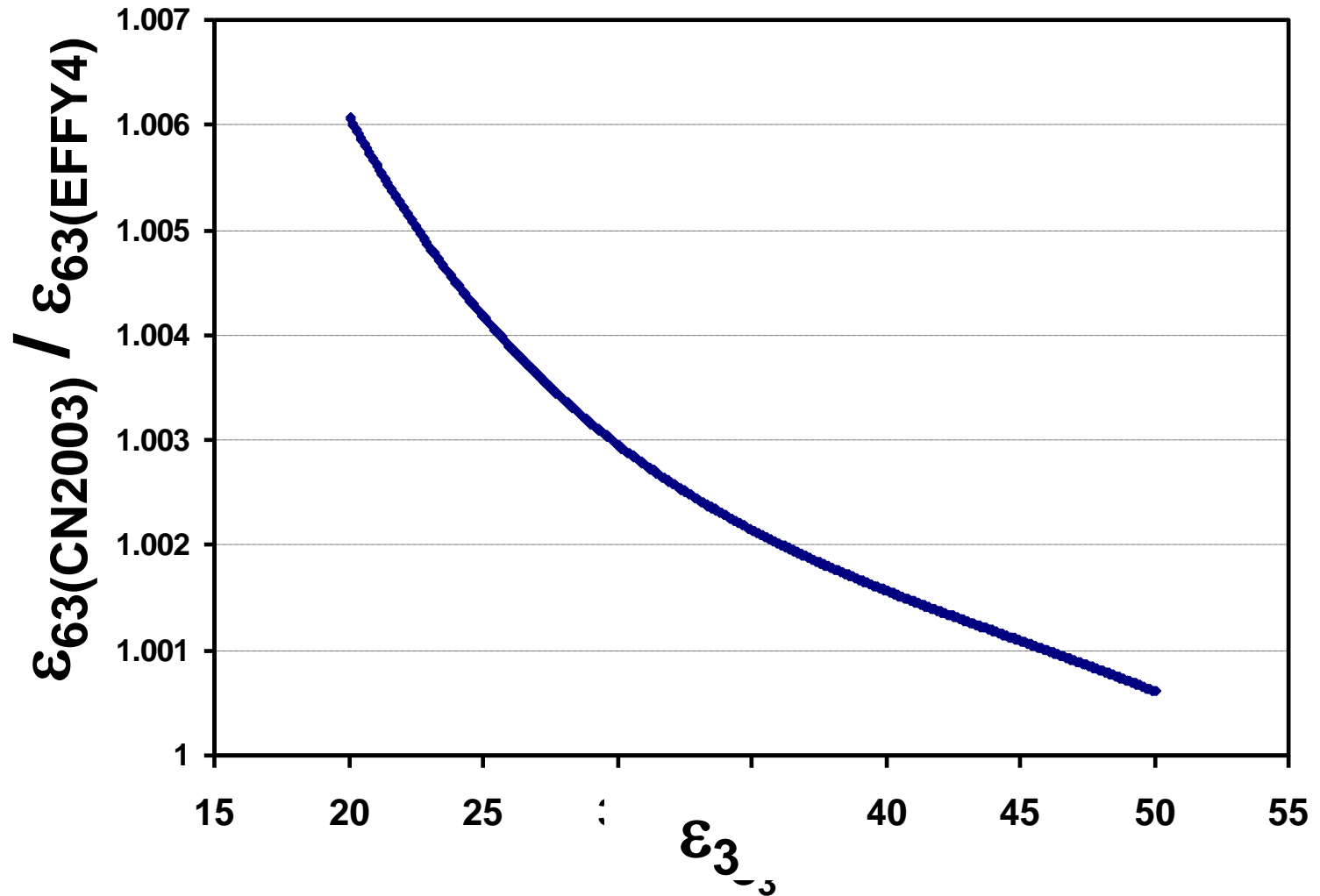
## CN2003 vs EFFY4 code differences – due to assumed Quench function

**EFFY4** – for toluene cocktails

**CN2003** – for DIN (like HS3 used)  
but also used xylene based (PCS)

Effect seen for other nuclides, like

$^{63}\text{Ni}$  -- 17 keV  $E_{\beta(\text{ave})}$  allowed



# $^{210}\text{Pb}$ massic activity results

	kBq/g ( $k = 2$ )	diff from LS	
LS (CN2003)	$9.037 \pm 2.4 \%$	----	
$\gamma$ -spect (HPGe)	$9.46 \pm 8.3 \%$	+ 4.7 %	big unc. if don't use $^{210}\text{Pb}$ $\gamma$ std
$4\pi\beta(\text{LS})$ - $\gamma(\text{NaI})$ anticoincidence ( <i>attempt</i> )	$9.10 \pm 3.3 \%$	+ 0.7 %	lots of assumptions, big extrapolation
$\alpha$ -spect (Po tracer)	8.77	- 3.0 %	$T_{1/2} = 102 \text{ a}$
	$\pm 1 \%$ 8.92	- 1.3 %	$T_{1/2} = 128 \text{ a}$

**Relatively large 2.4 % uncertainty because of**

- (1) LS cocktail composition effects**
- (2) tracing code differences & assumptions,**
- (3) lack of good confirmatory measurements,**