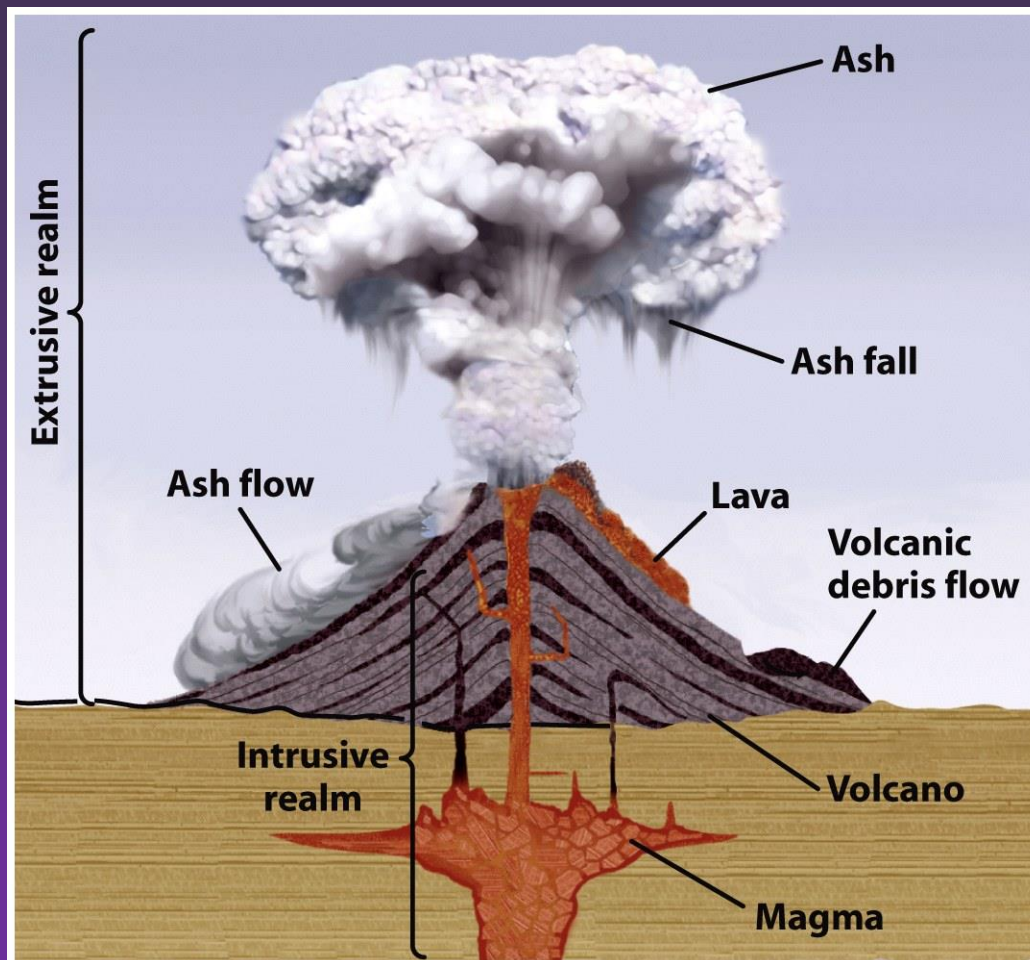


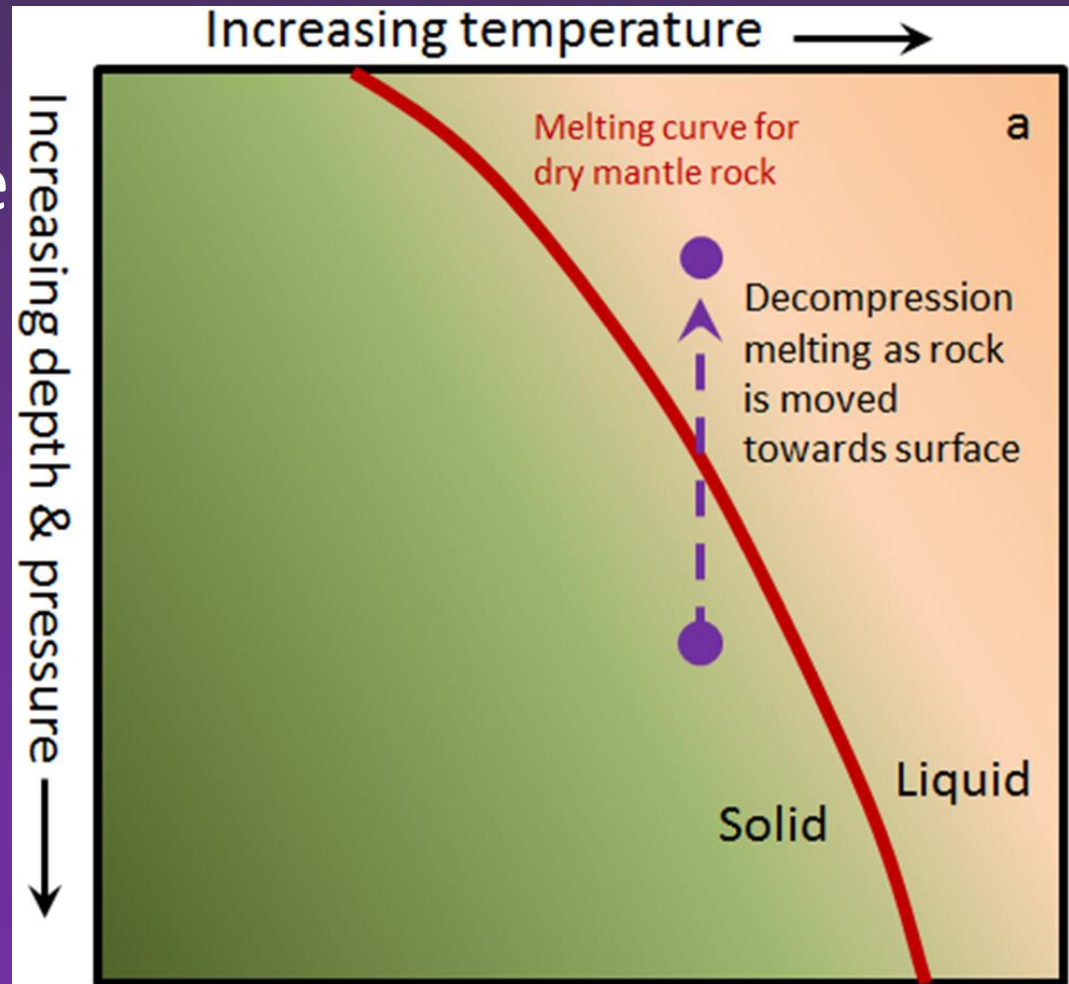
# Igneous Rocks



# How do magmas form?

# How do Magmas Form?

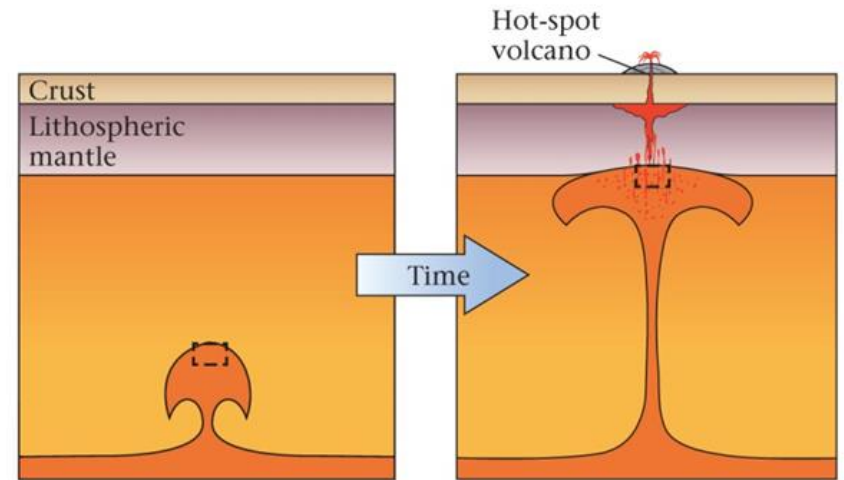
1. Decompression melting  
Decrease in pressure



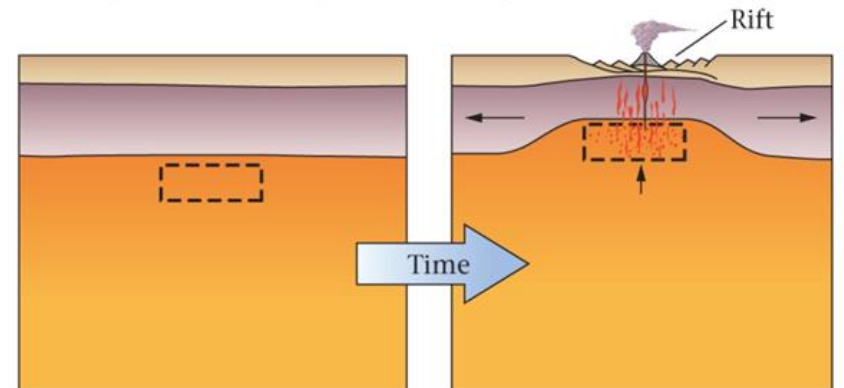
# How do Magmas Form?

## 1. Decompression melting

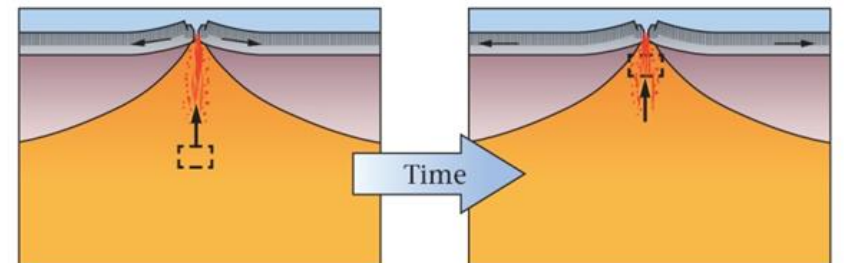
Decrease in pressure



Decompression melting in a mantle plume



Decompression melting beneath a rift



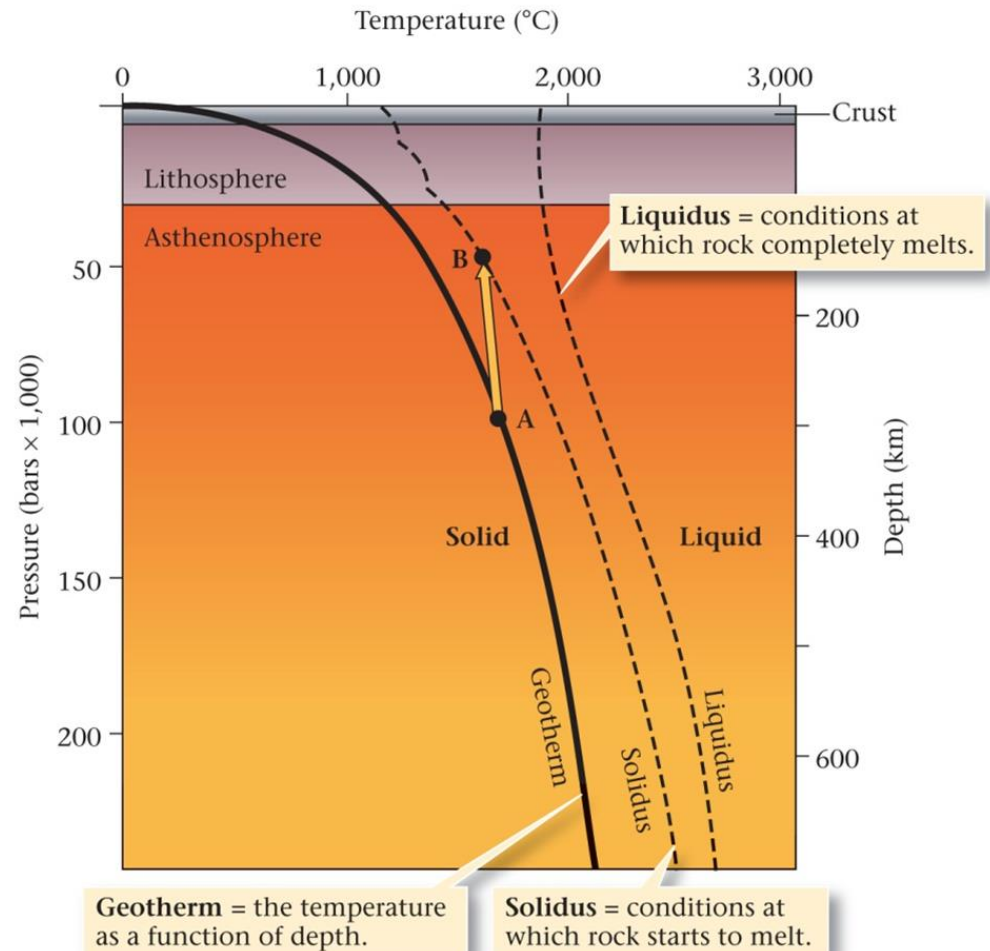
Decompression melting beneath a mid-ocean ridge

(b)

# How do magmas form?

## 1. Decompression melting

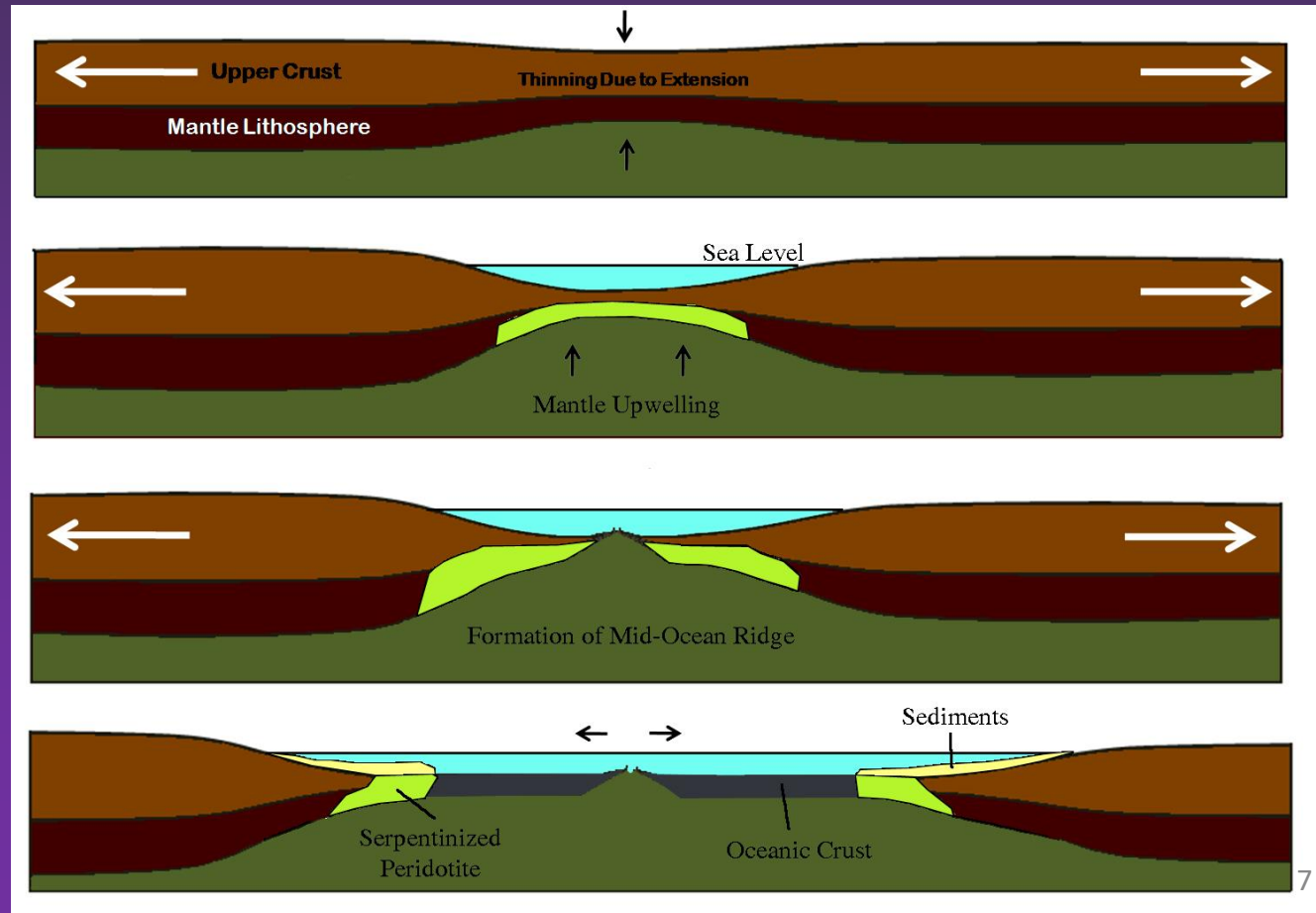
- Geothermal gradient



(a)

# Where Magma Forms

- Decompression
- Divergent margin



# Divergent Margin

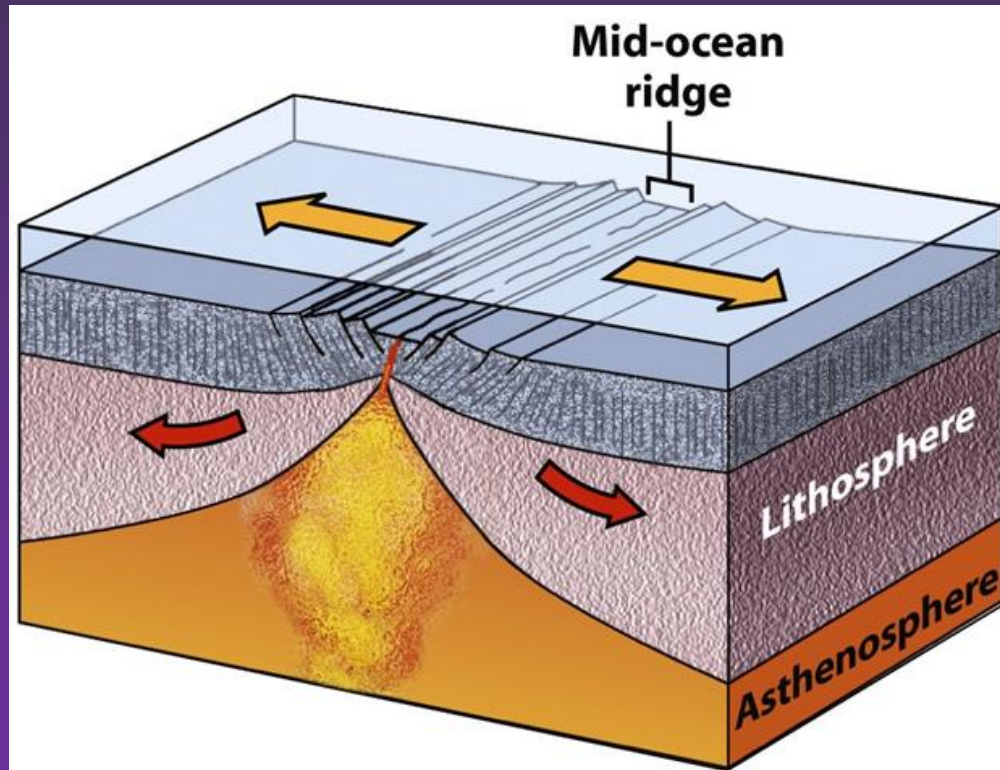
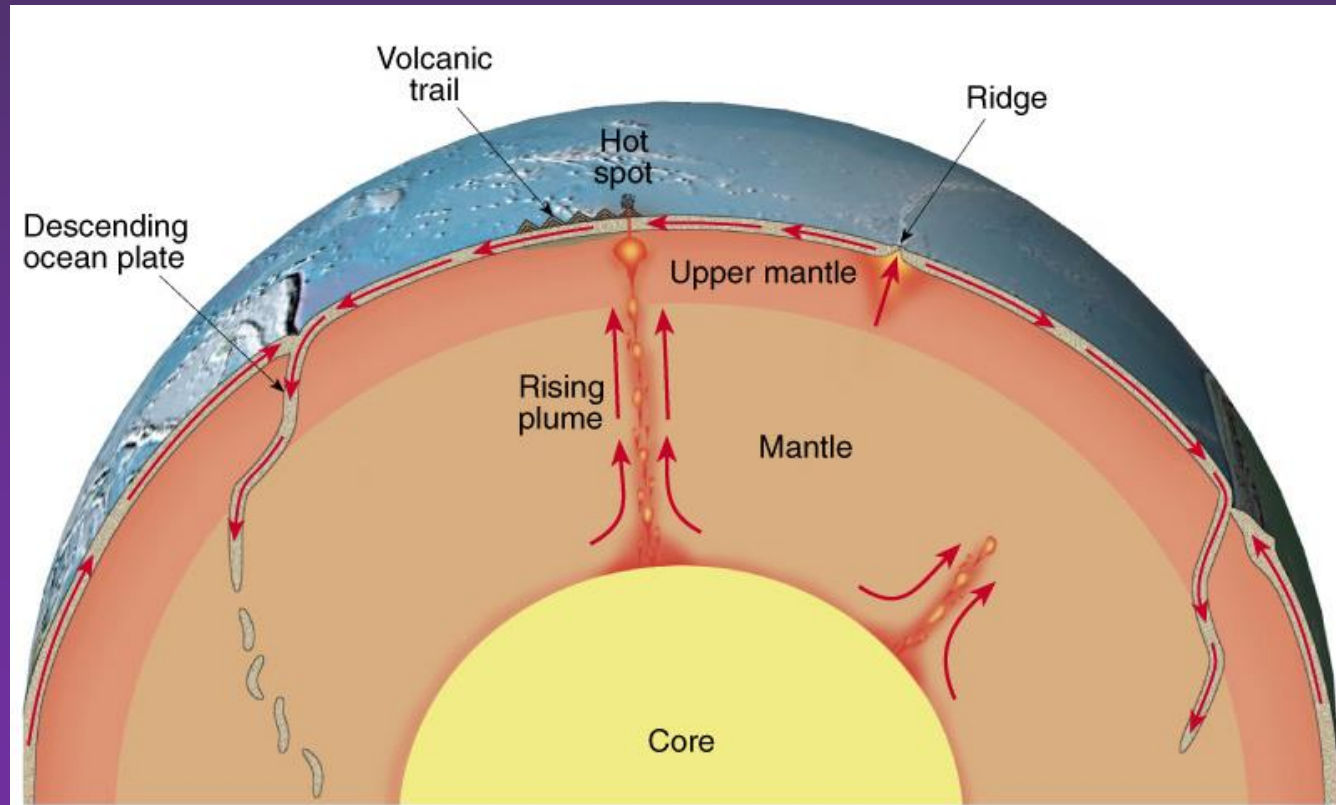


Figure 4-6a Earth: Portrait of a Planet 3/e  
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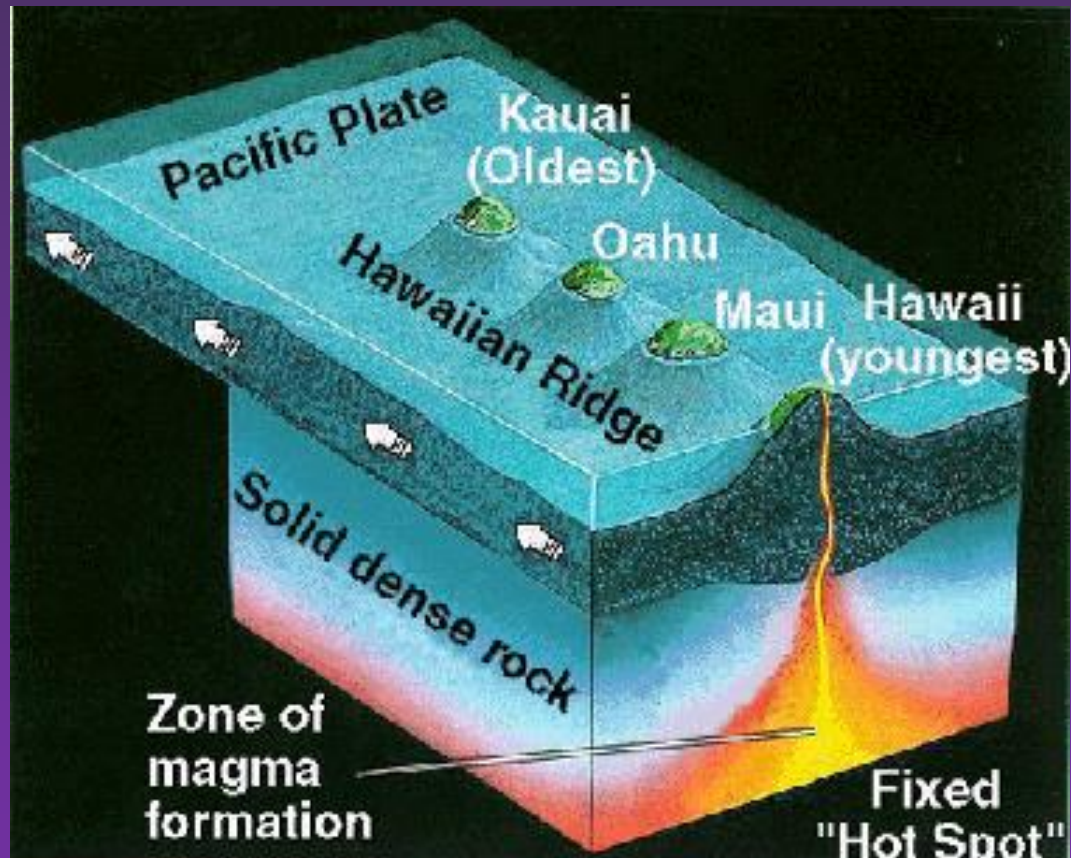
# Where Magma Form

- Mantle Plumes
- “Hot Spots”

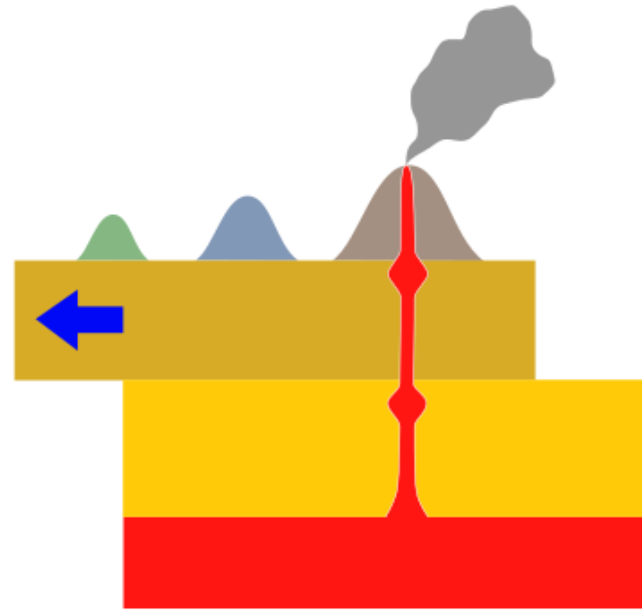
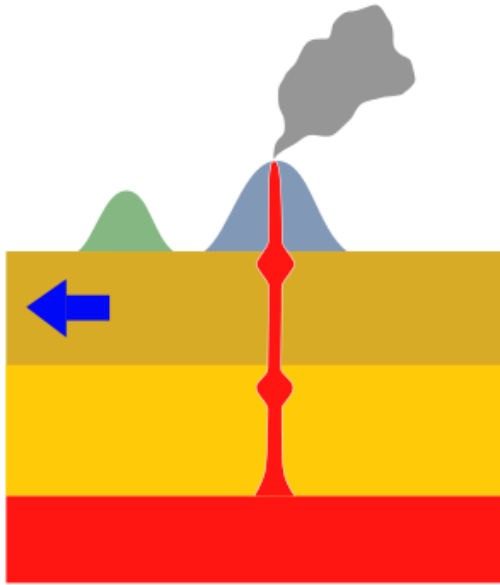


# How Magma Forms

- Mantle Plumes (Hot Spots)
- Decompression



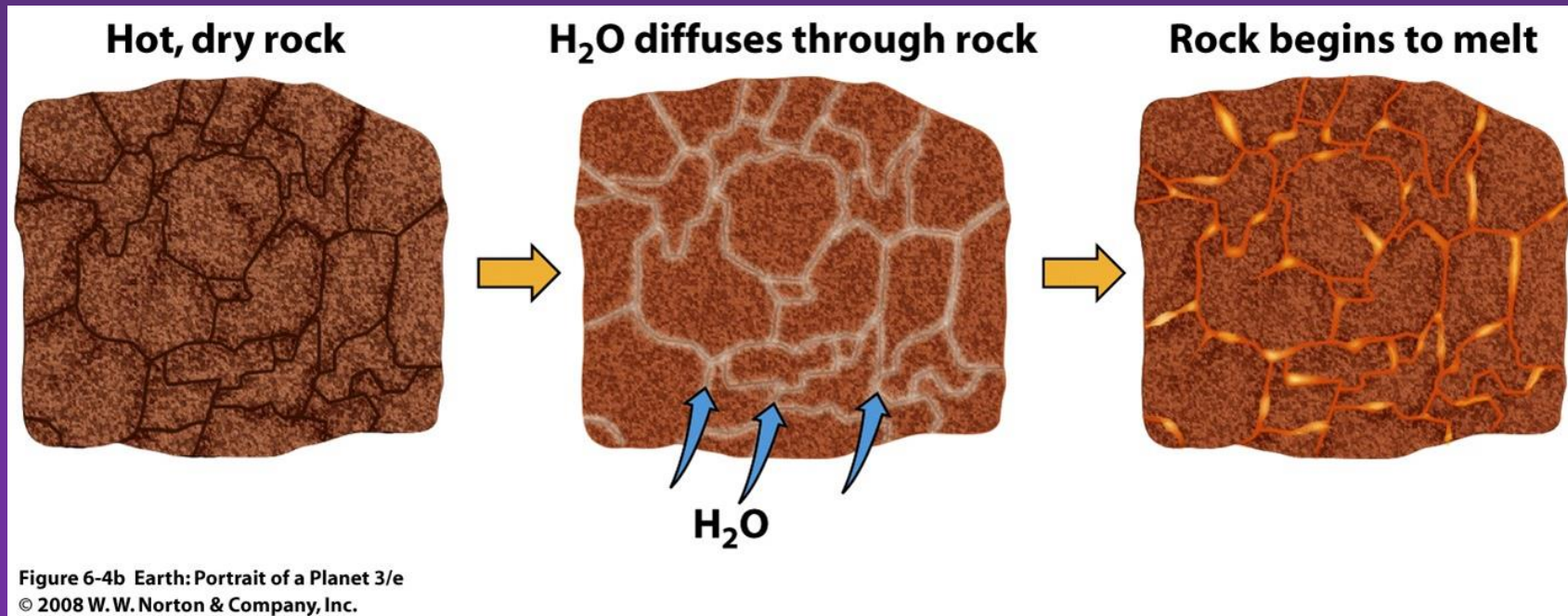
# Hotspots



# How do Magmas Form?

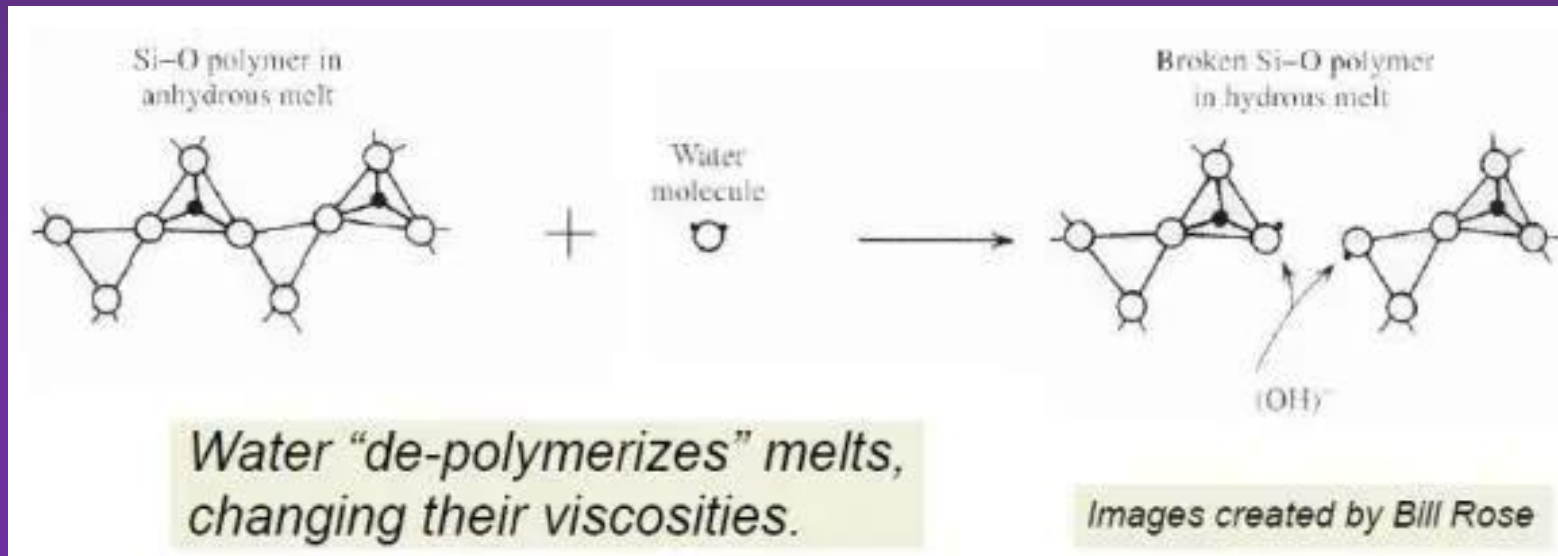
## 2. Addition of volatiles

Water lowers melting temperature at a given pressure (depth)



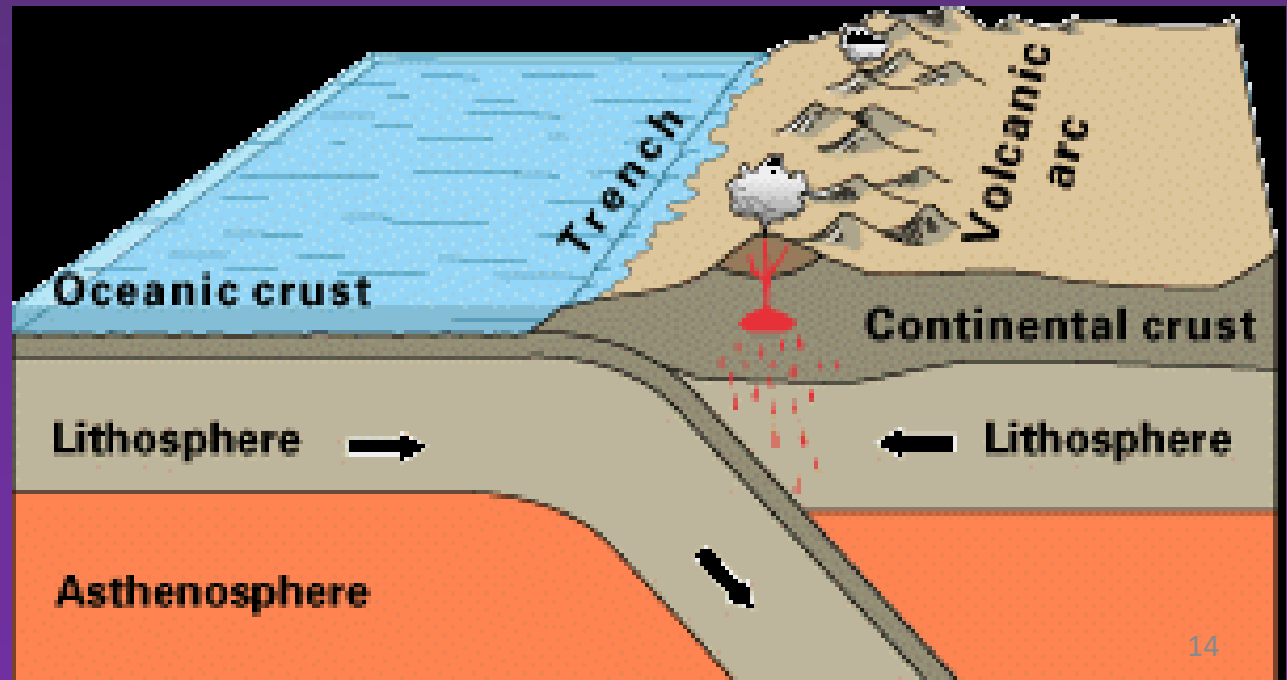
# How Magma Forms

- 2. Addition of volatiles (e.g. water, carbon dioxide)
- Volatiles break silica bonds allowing the material to flow



# How Magma Forms

- 2. Addition of volatiles (e.g. water, carbon dioxide)
- Water comes from rocks subducted
- Convergent margins





# Convergent Margin

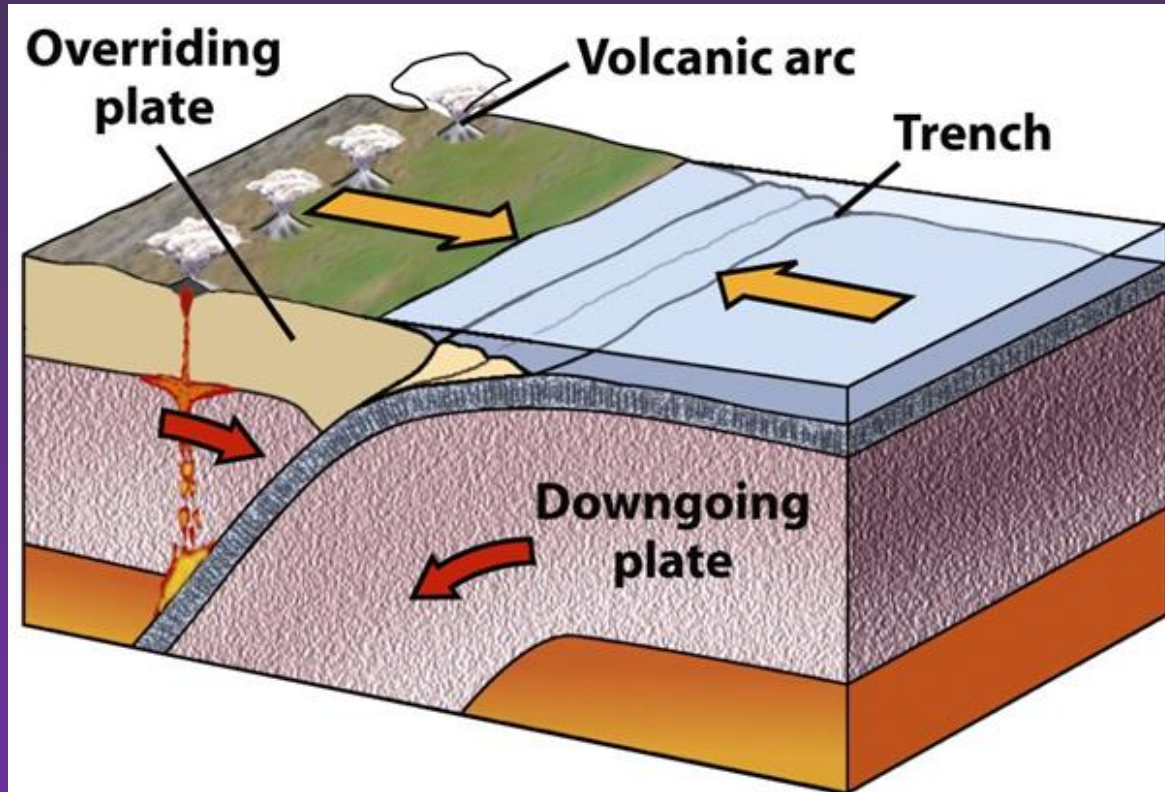
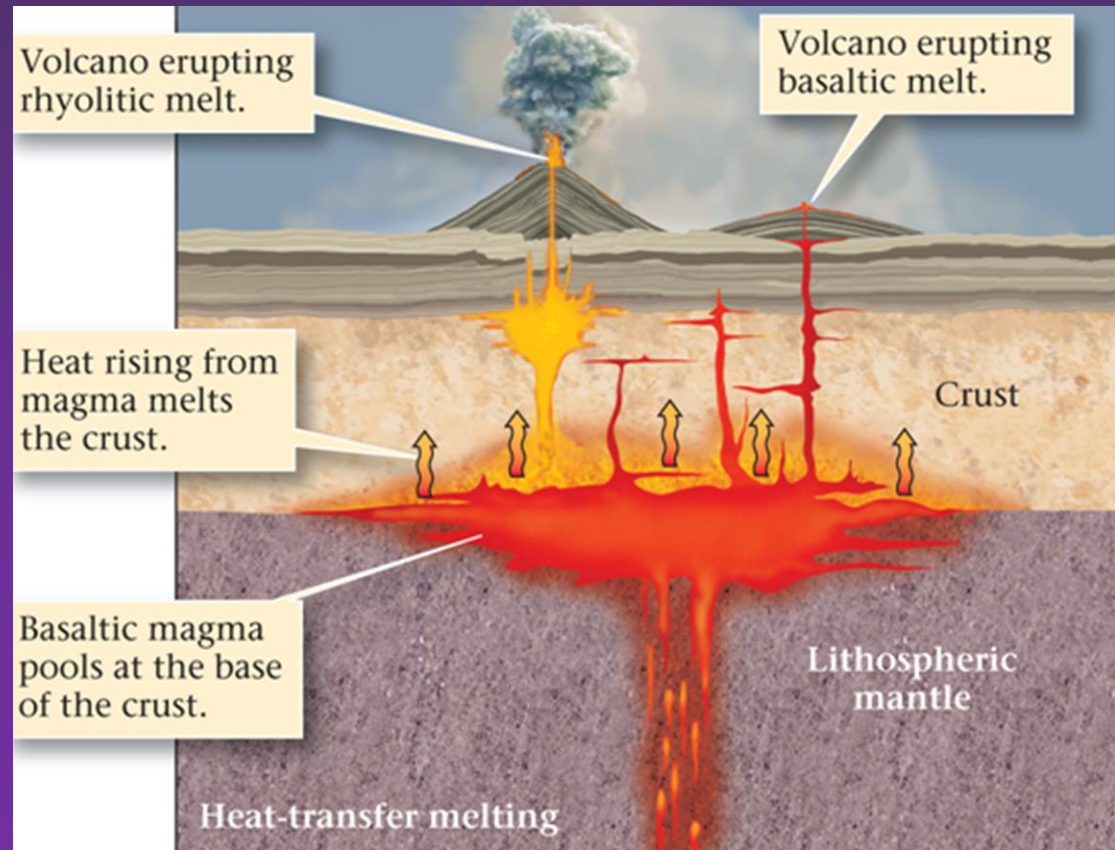


Figure 4-6b Earth: Portrait of a Planet 3/e  
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# How do Magmas Form?

## 3. Heat transfer from rising magma





# Igneous Rocks

- Cooling magma similar to freezing process
- Different magmas “freeze” at different temperatures (650° – 1100° C)

# Magma Composition

- Expressed as oxides
  - Silicon ( $\text{SiO}_2$ )
  - Aluminium ( $\text{Al}_2\text{O}_3$ )
  - Iron ( $\text{FeO}/\text{Fe}_2\text{O}_3$ )
  - Magnesium ( $\text{MgO}$ )
  - Calcium ( $\text{CaO}$ )
  - Potassium ( $\text{K}_2\text{O}$ ) and sodium ( $\text{Na}_2\text{O}$ )
  - And “volatiles” ( $\text{H}_2\text{O}$ ,  $\text{CO}_2$ )
- Other elements in minor amounts

# Classifying Igneous Rocks

- Amount of Silica

– Felsic	66-76% SiO <sub>2</sub>
– Intermediate	52-66% SiO <sub>2</sub>
– Mafic	45-52% SiO <sub>2</sub>
– Ultramafic	38-45% SiO <sub>2</sub>

# Igneous Minerals

## Mafic Minerals

- Olivine  $(\text{Fe, Mg})_2\text{SiO}_4$
- Pyroxene  $(\text{Fe, Mg})\text{SiO}_3$
- Amphibole  $(\text{Fe, Mg})\text{Al SiO}$
- Biotite  $(\text{FeMg})\text{KAlSiO}$

## Felsic Minerals

- Plagioclase feldspar  $(\text{Ca, Na})(\text{Al, Si})_4\text{O}_8$
- Quartz  $\text{SiO}_2$
- K-feldspar  $\text{KAlSiO}_3$
- Muscovite

# Bowen's Reaction Series

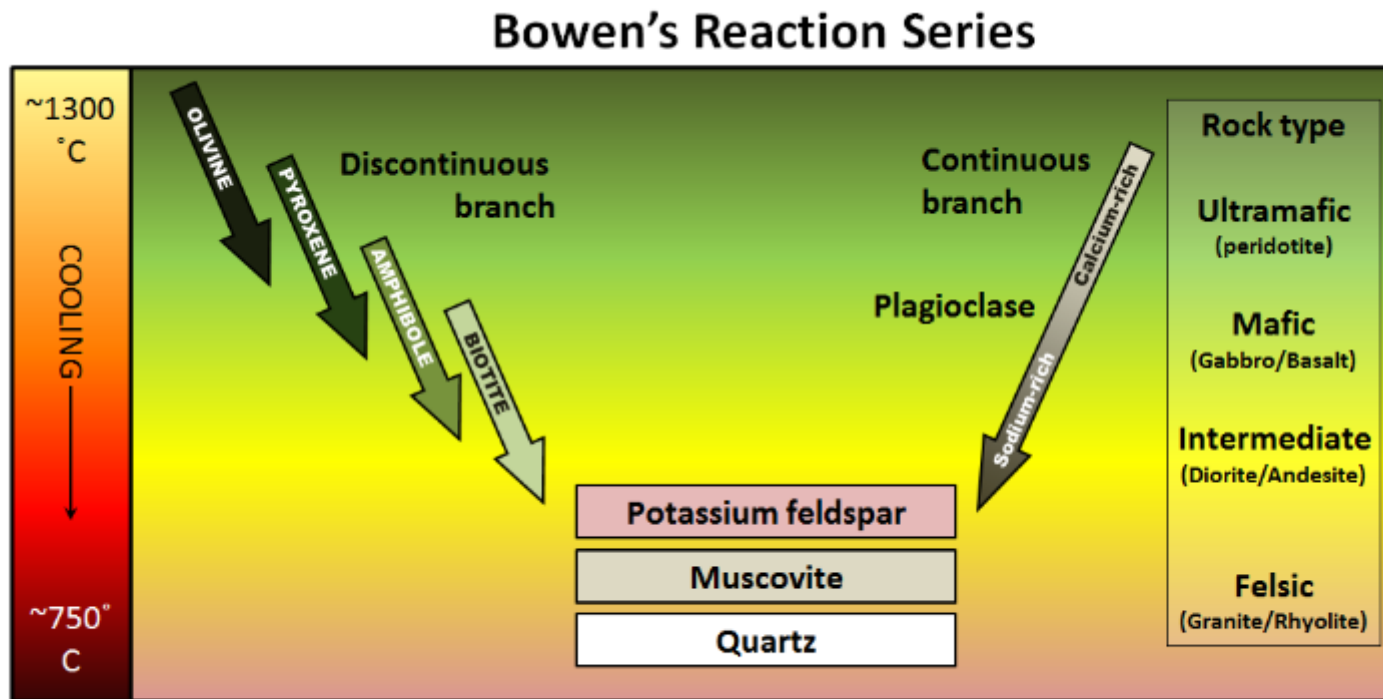
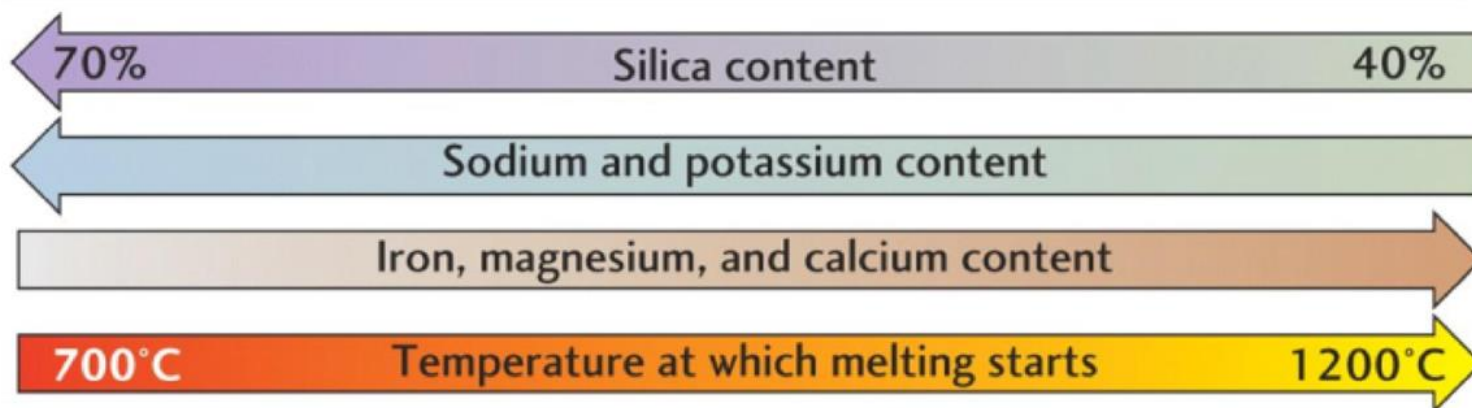
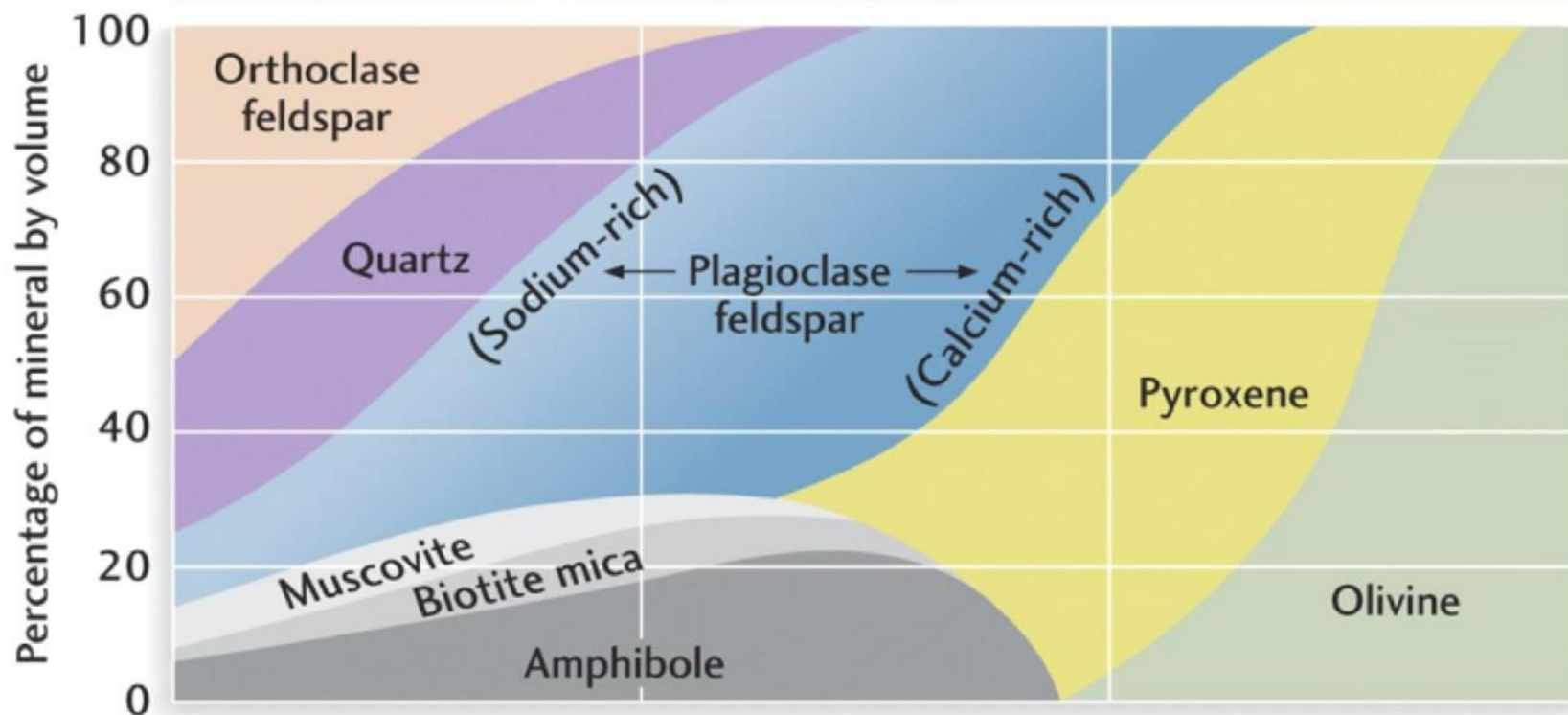


Figure 3.3.1 The Bowen reaction series describes the process of magma crystallization.

Composition	FELSIC	INTERMEDIATE	MAFIC	ULTRAMAFIC
Rock types	Granite Rhyolite	Diorite Andesite	Gabbro Basalt	Peridotite



# Igneous Rocks

- Intrusive
  - Form from magma
  - Cool underground
  - Coarser grained
- Extrusive (“volcanic”)
  - Form from lava
  - Cool above ground
  - Fine-grained or aphanitic

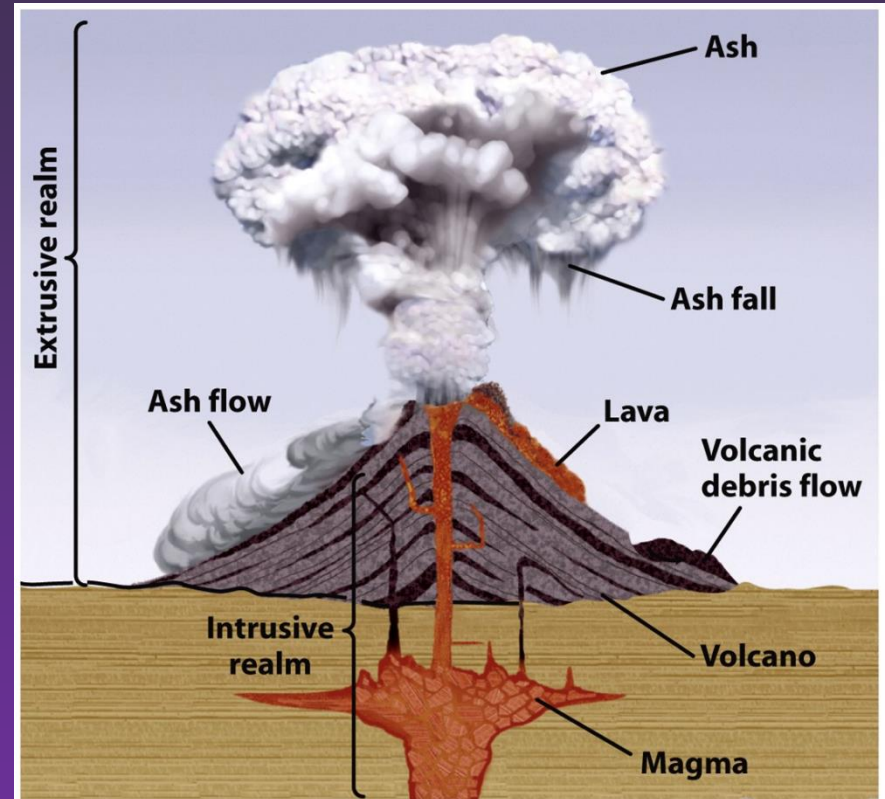


Figure 6-2 Earth: Portrait of a Planet 3/e  
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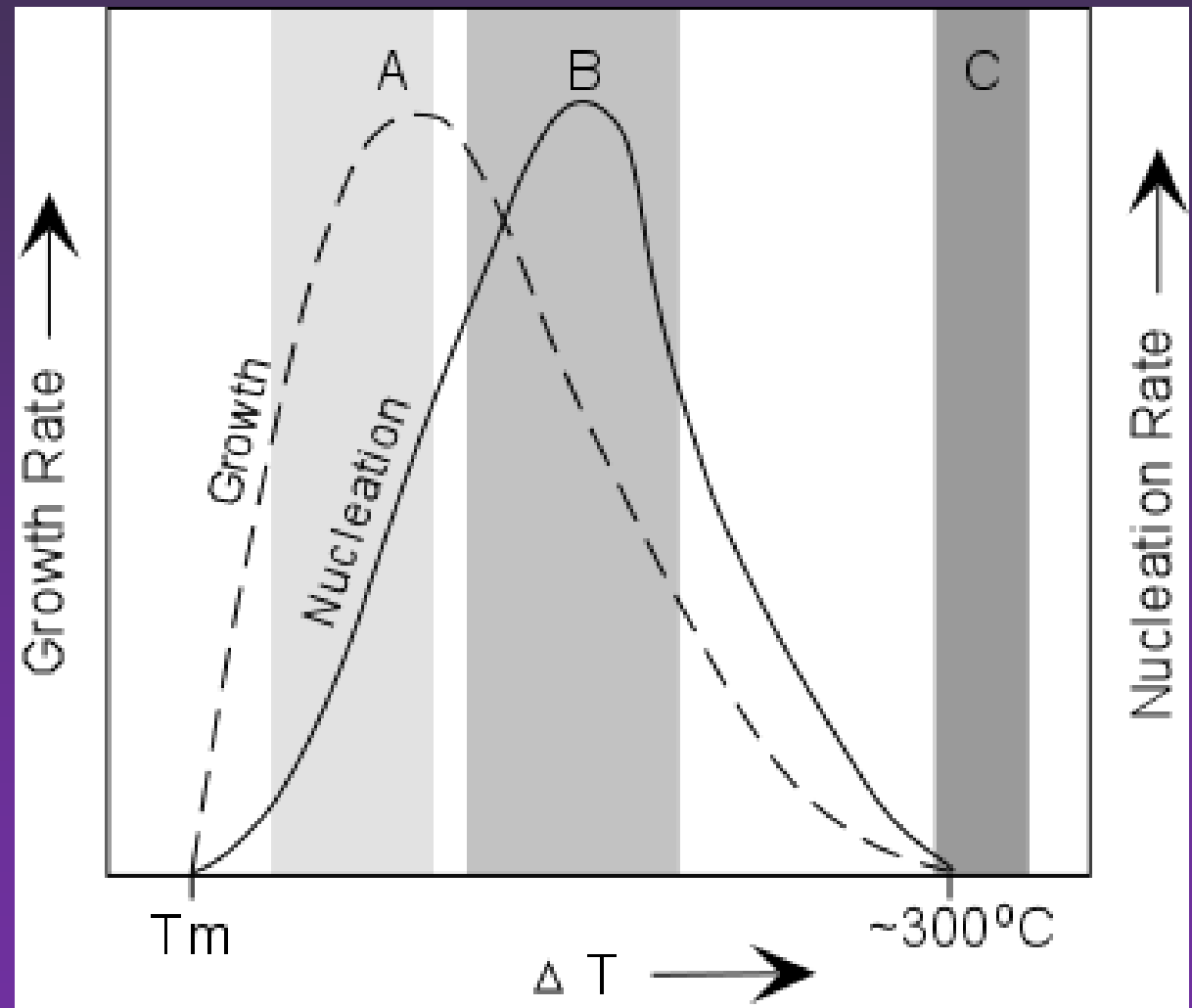
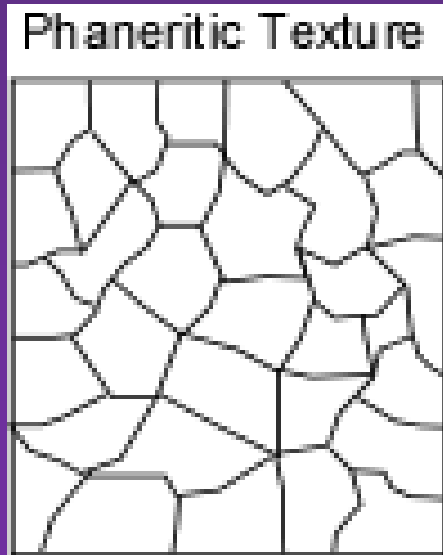
# Factors Controlling Texture

- **Cooling rate ( $\Delta T/\Delta t$ )**
- Diffusion rate
- Rate of nucleation of new crystals
- Rate of growth of crystals



# Phaneritic Textures

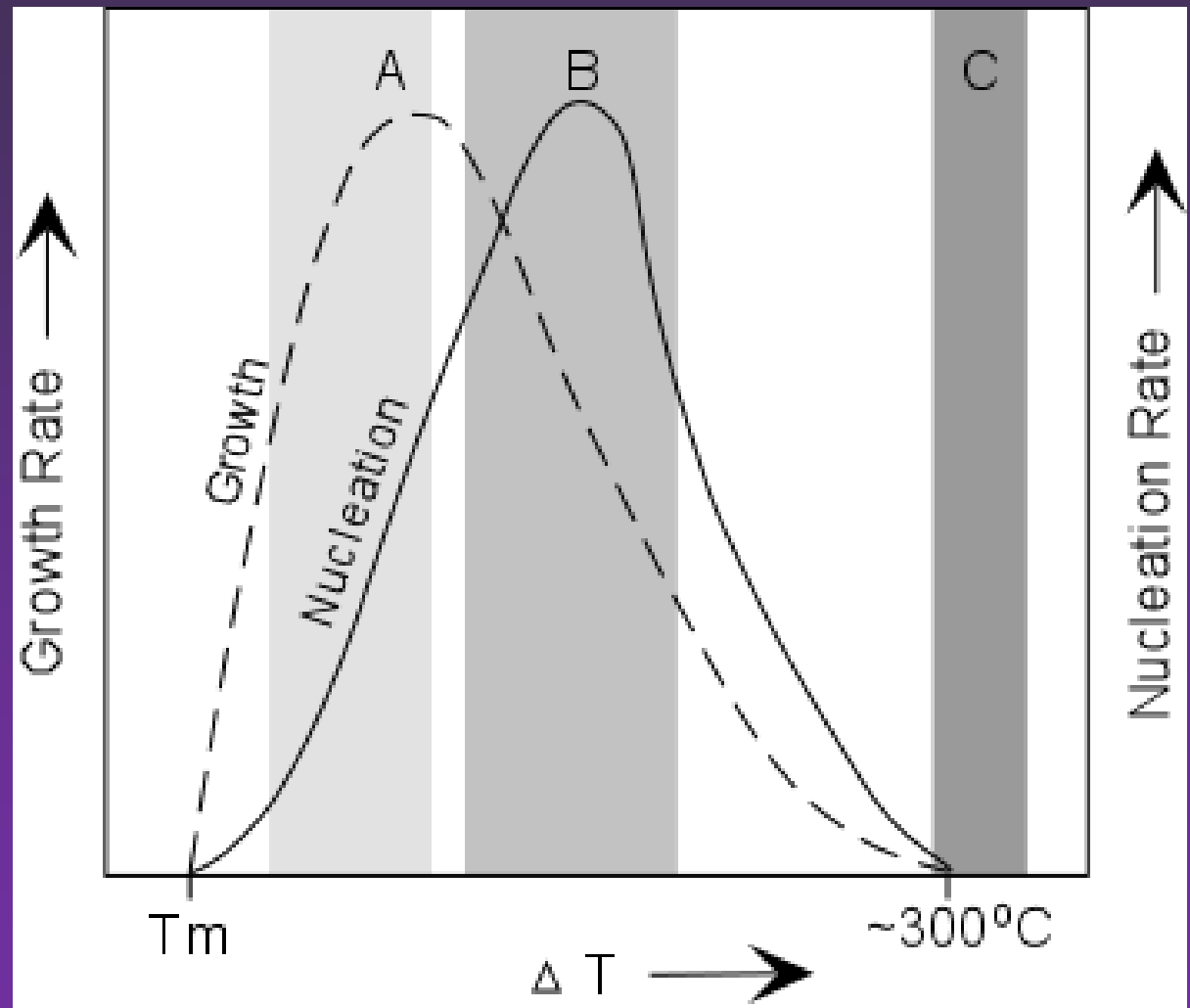
- A.



[https://www2.tulane.edu/~sanelson/eens212/textures\\_igneous\\_rocks.htm](https://www2.tulane.edu/~sanelson/eens212/textures_igneous_rocks.htm)

# Aphanitic Texture

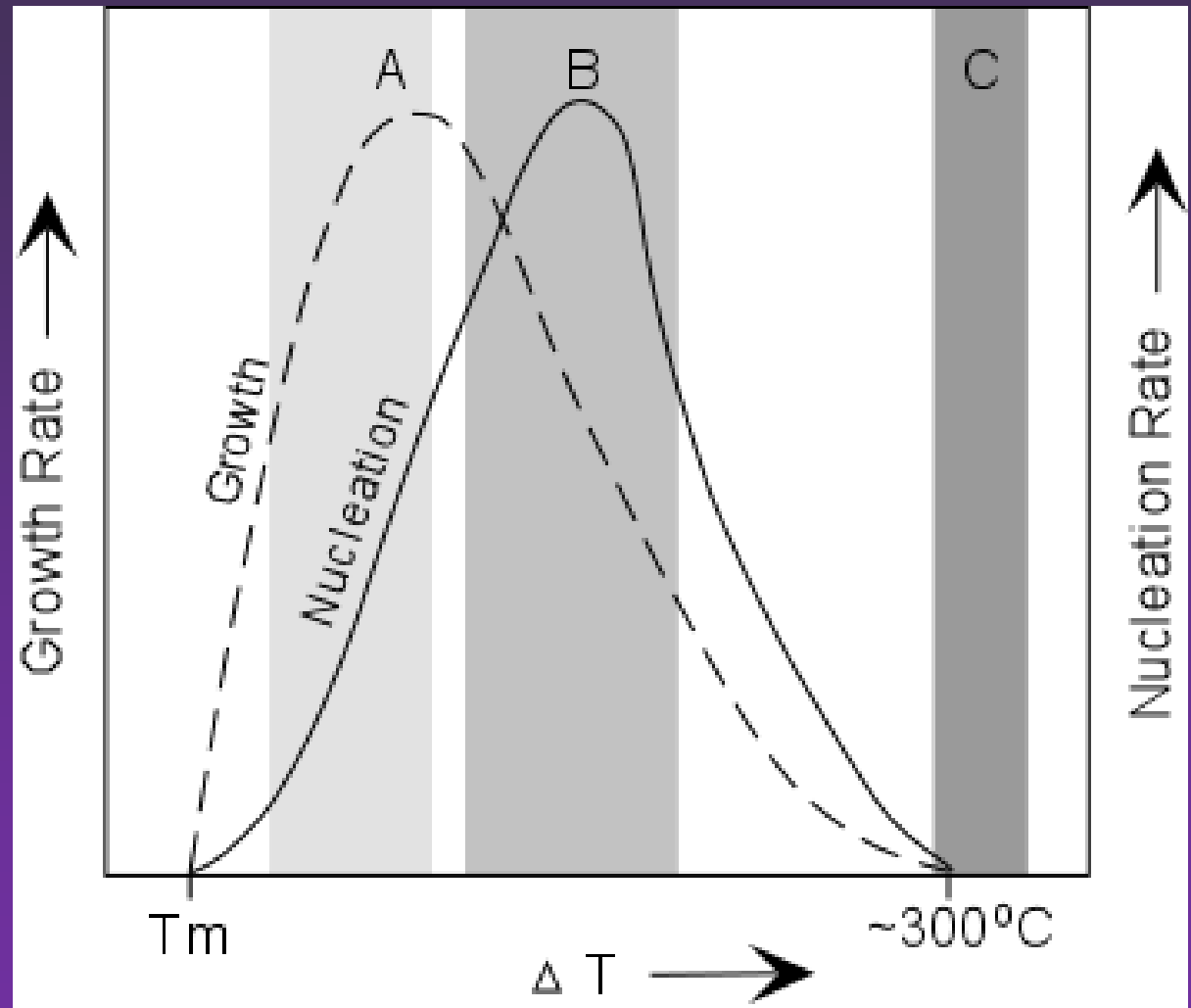
- B



[https://www2.tulane.edu/~sanelson/eens212/textures\\_igneous\\_rocks.htm](https://www2.tulane.edu/~sanelson/eens212/textures_igneous_rocks.htm)

# Holohyaline (glassy) Texture

- C



[https://www2.tulane.edu/~sanelson/eens212/textures\\_igneous\\_rocks.htm](https://www2.tulane.edu/~sanelson/eens212/textures_igneous_rocks.htm)

# Intrusive Rocks

- Slow cooling magmas because of insulation by country rocks
- Relatively coarse-grain size ( $>1\text{mm}$ )



# Classification by Texture

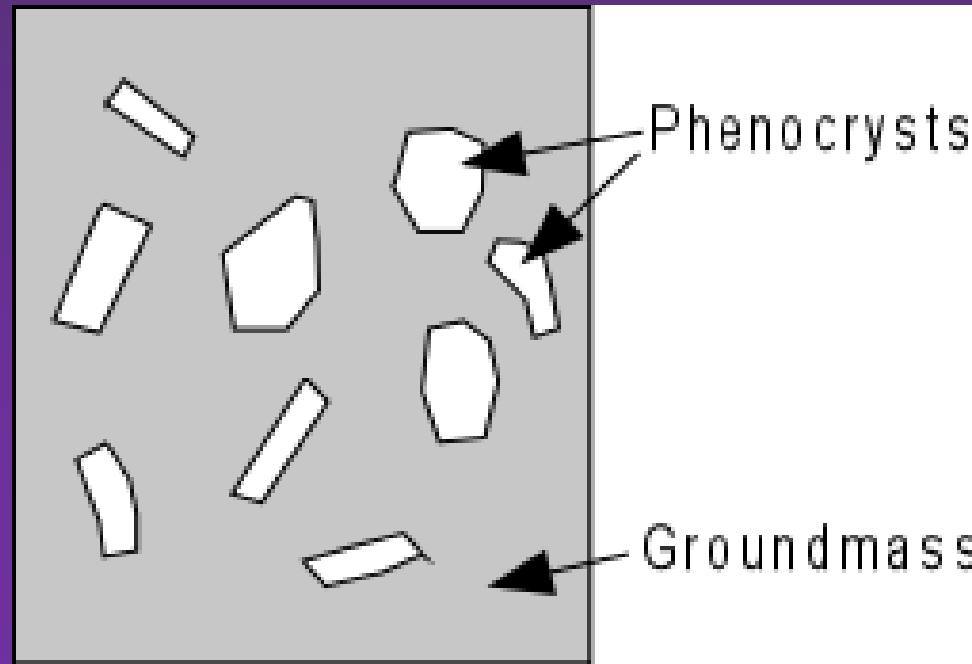
- Intrusive textures (slow cooling)
  - Crystalline
    - Grain size – are grains large or small?
      - Fine-grained (<1mm)
      - Medium-grained
      - Coarse-grained (>5mm)
    - Equigranular – grains are the same size

# Classification by Texture

- Extrusive (or rapidly cooled intrusive rocks)
  - Aphanitic
    - crystals cannot be distinguished with a hand lens)
  - Holohyaline
    - glassy

# Textures

- Porphyritic (both intrusive and extrusive)
  - mix of large, euhedral grains, and fine-grained or aphanitic groundmass



# Why do magmas move?

- Buoyancy
- Weight of overlying rocks



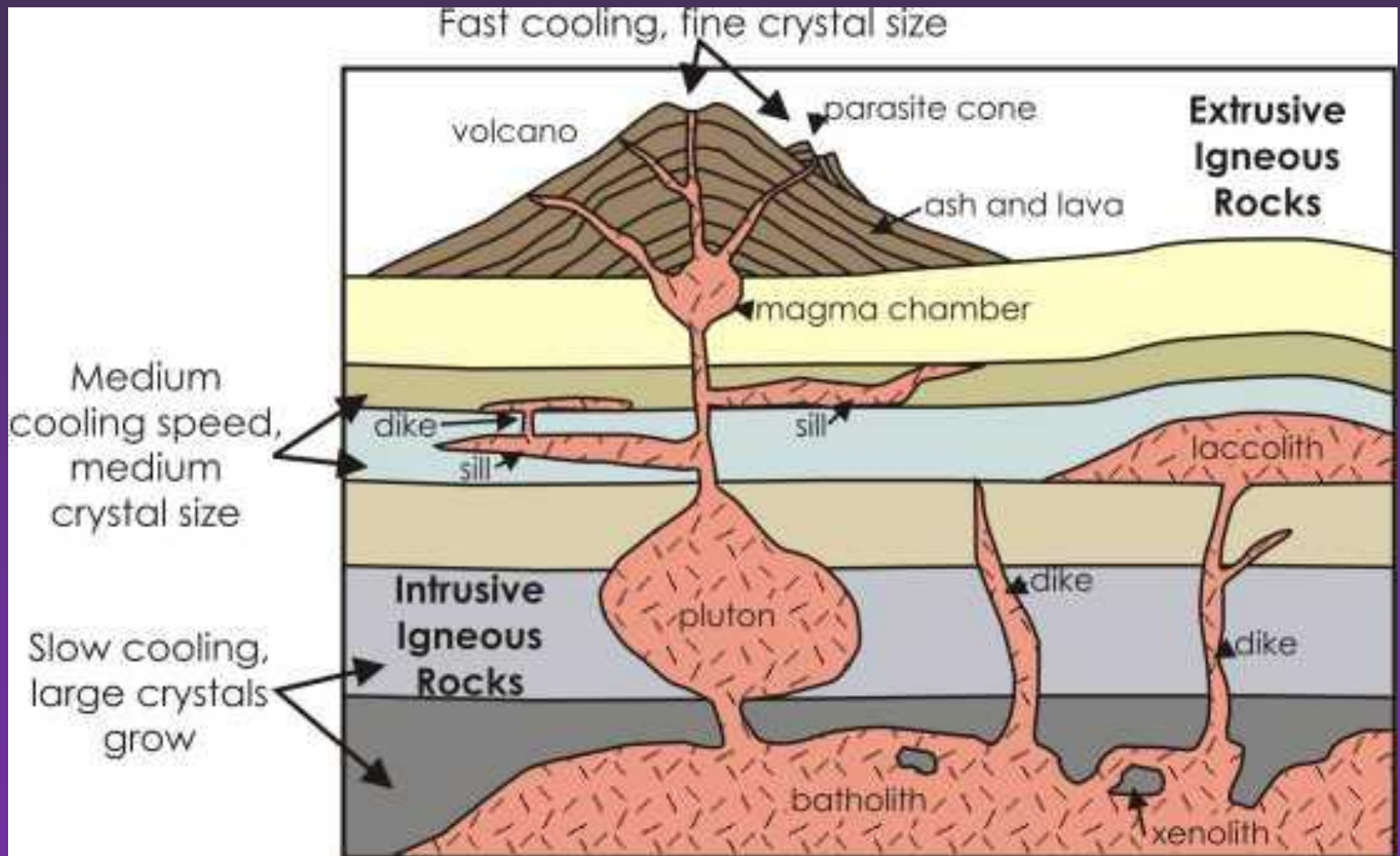
# Speed of magma?

- Determined by viscosity
- Controlled by
  - Temperature
  - Silica content
  - Volatiles

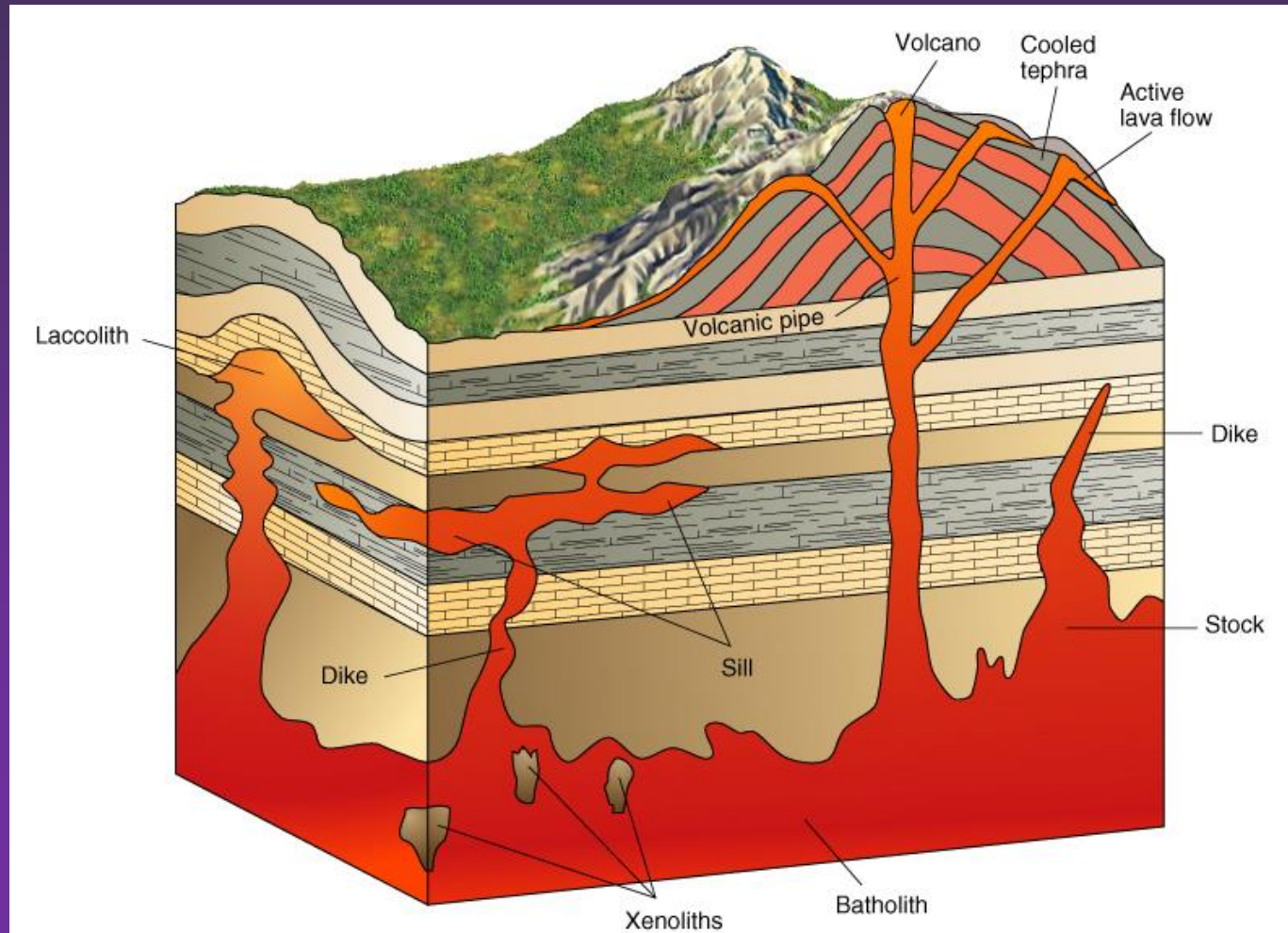
# “Freezing”

- Occurs due to
  - Cooling
  - Loses volatiles
- Rate of cooling
  - Depth
  - Shape and size
  - Groundwater

# Intrusive Rocks



# Intrusive Rocks



- Amount Of Silica: (Intrusive / Extrusive)
- Felsic - 66-76% silica (granite / rhyolite)
  - high viscosity - less dense
  - Light colour - solid @ 650C +/-
- Intermediate - 52-66% (diorite / andesite)
  - interm viscosity - mod density
  - interm colour - 900C+/-
- Mafic - 45-52% (gabbro / basalt)
  - low viscosity - dense
  - dark colour - >1000C
- Ultramafic - 38-45% (peridotite/komatiite)
  - (1300C+/-)
  - very low viscosity - very dense - dark coloured

# Volcanic (Extrusive) Rocks

- Two main groups of volcanic rocks
  - Lava flows
  - Pyroclastic rocks



# Classifications

- Textures

- Obsidian- volcanic glass- dark, sharp
- Pumice - felsic, “frothy”, vesicles, light colour, floats
- Scoria -mafic, vesicles, dark colour, dense

- Fragmented – Pyroclastic – extrusive

- Depends on size of fragments
- Tuff – fine (ash)
- Volcanic breccia – fragments of volcanic debris (coast fabrics)



# Bowen's Reaction Series

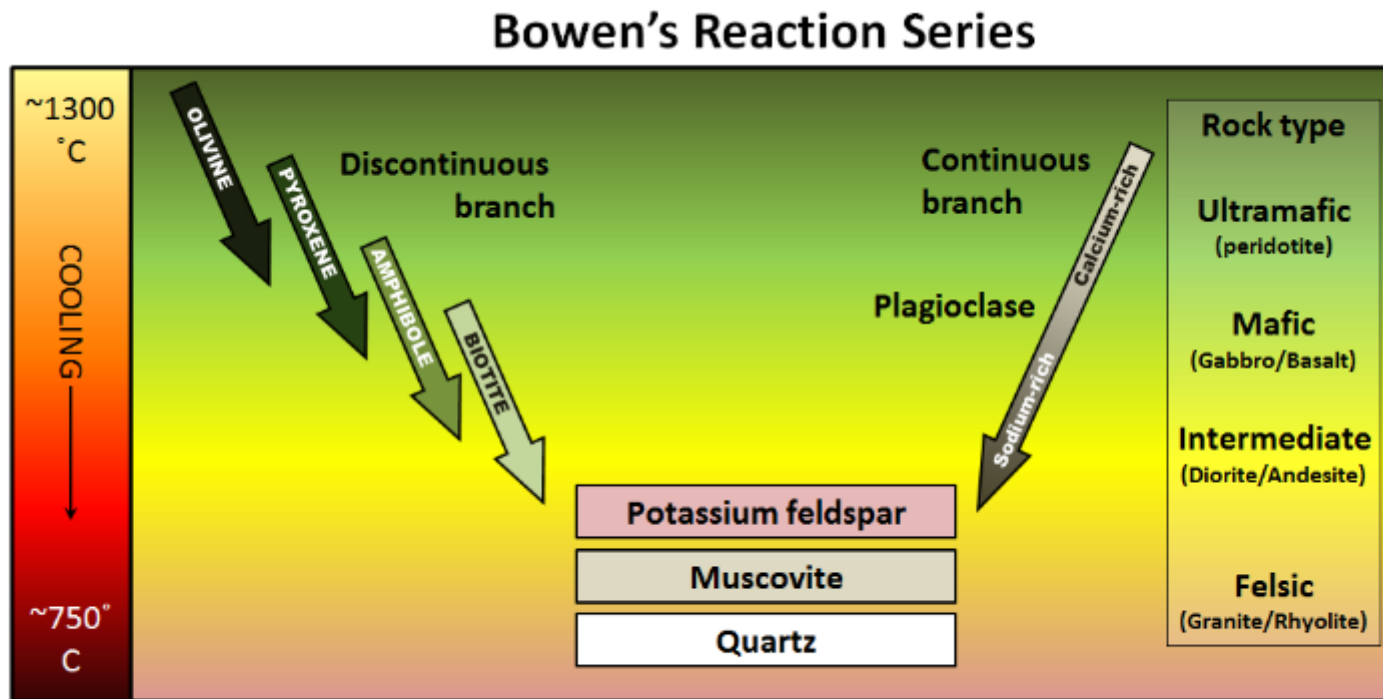
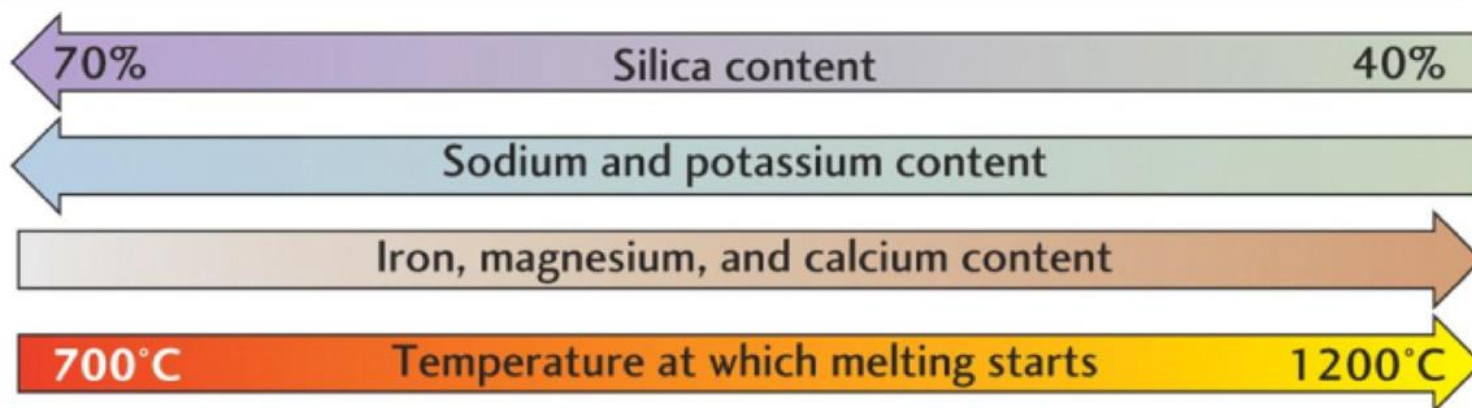
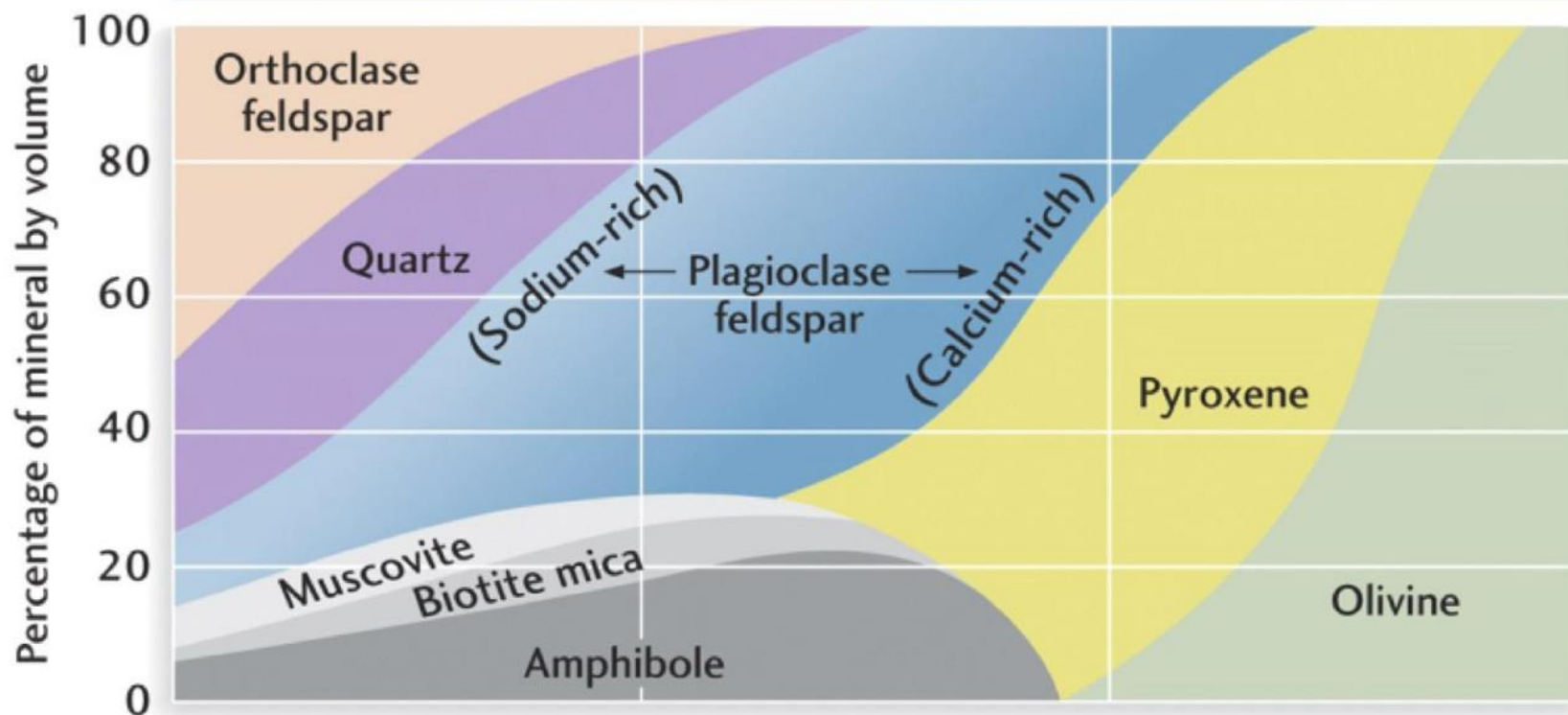


Figure 3.3.1 The Bowen reaction series describes the process of magma crystallization.

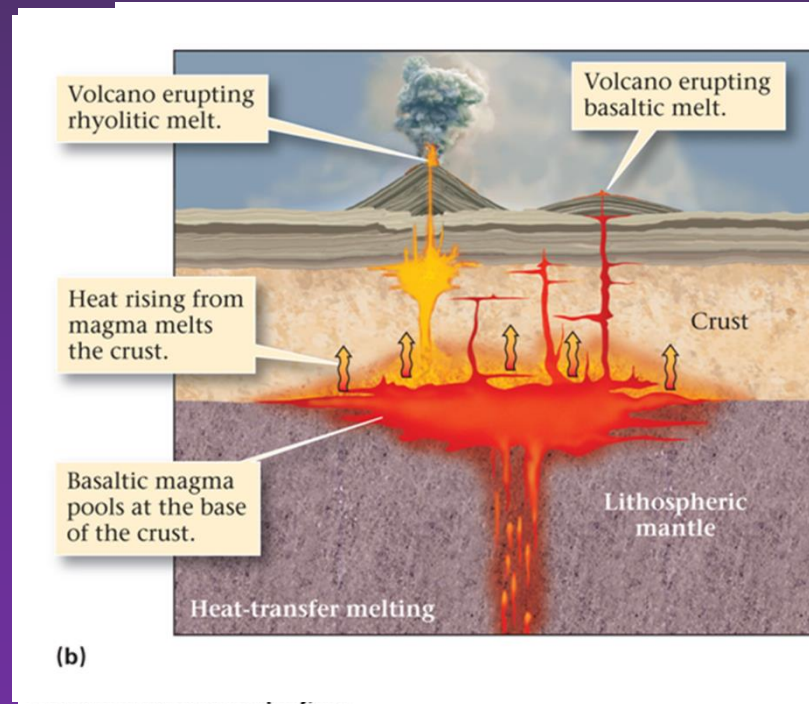


Composition	FELSIC	INTERMEDIATE	MAFIC	ULTRAMAFIC
Rock types	Granite Rhyolite	Diorite Andesite	Gabbro Basalt	Peridotite



# Why are there different types of Igneous Rocks?

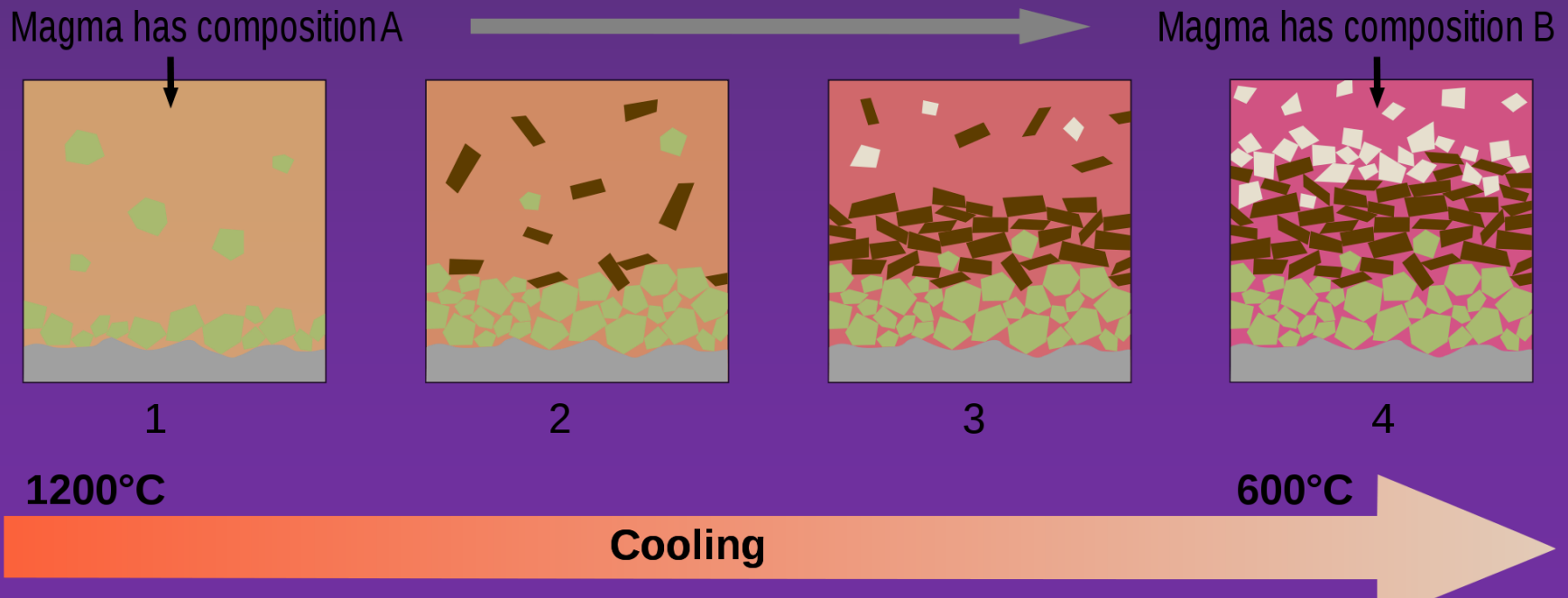
- Source rock
- Crystal Fractionation
- Partial Melting
- Assimilation
- Magma mixing



**Conve**  
**also c**  
**Conve**  
**Subdu**  
**Consu**  
**Trench**

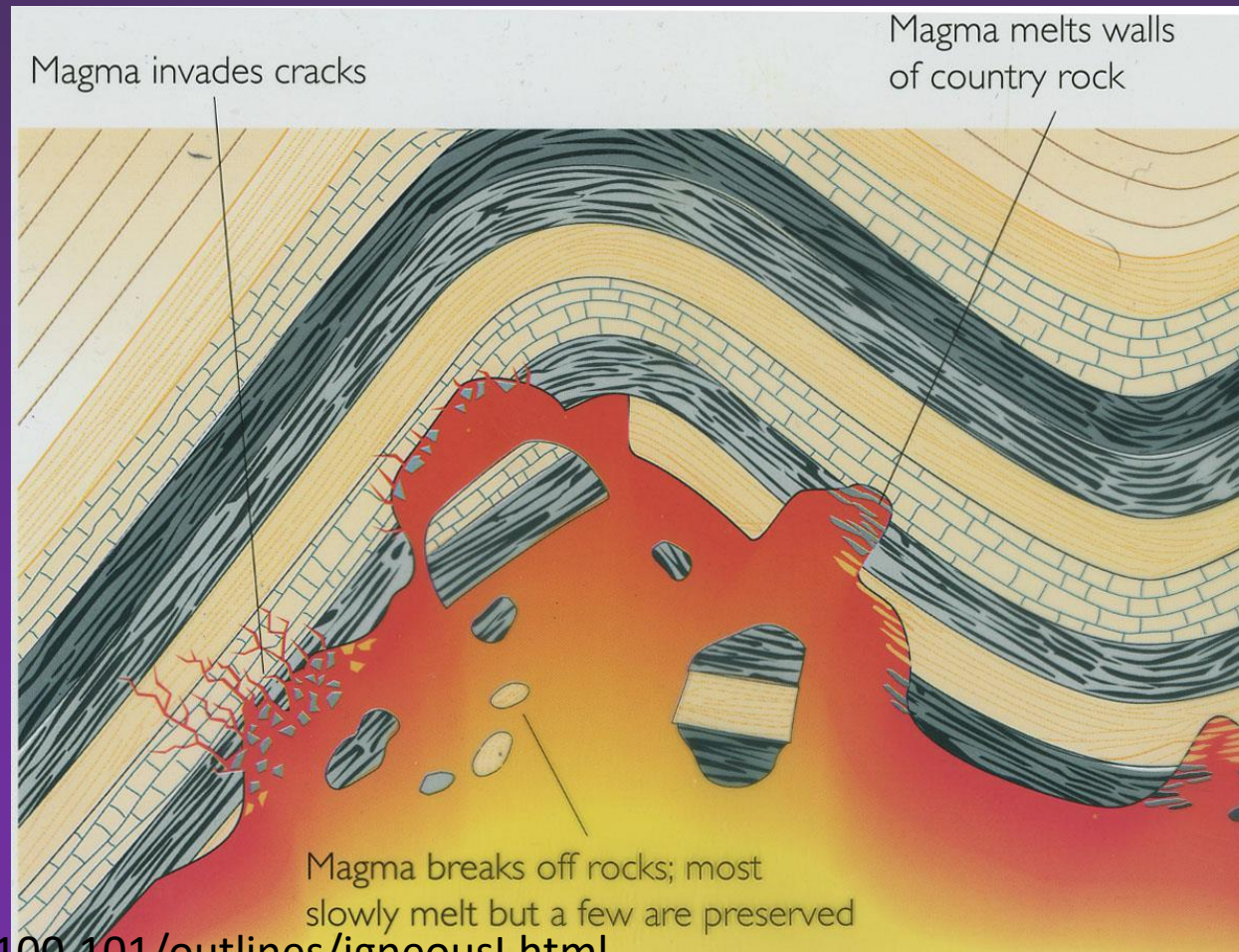
# Crystal Fractionation

- Mafic minerals crystallize first
- More dense crystals fall out, leaving a more silica-rich magma



# Assimilation

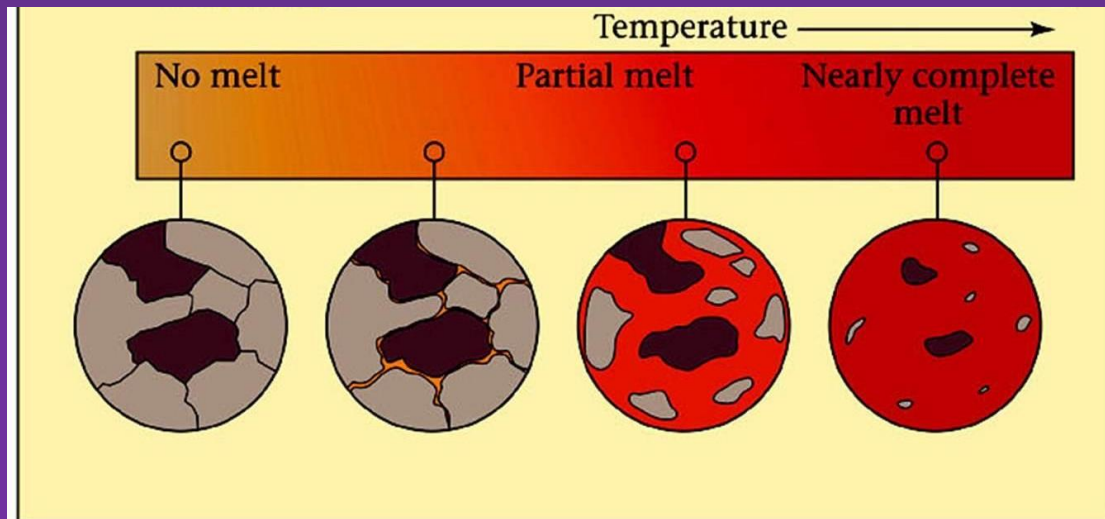
- Add elements from the surrounding rocks





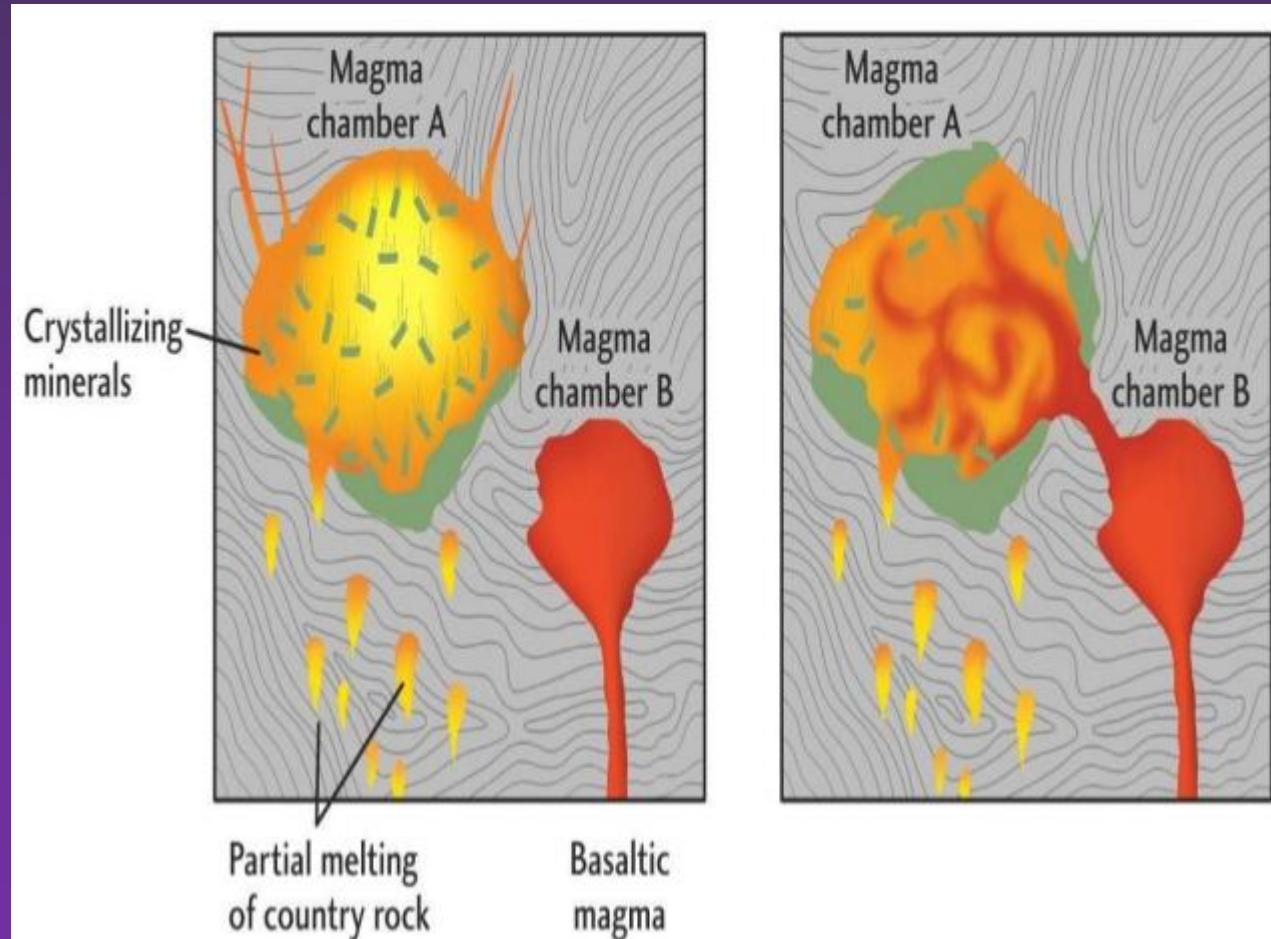
# Partial Melting

- Silica-rich minerals melt first
- Partial melting produces melts with more silica
- Remove melt:
  - Silica-rich melt
  - Mafic residue



# Magma Mixing

- Magmas of different compositions mix together, producing a magma of intermediate composition



# Magma Composition

- Silicon (Si)
- Oxygen (O)
- Al
- Mg, Fe
- Ca, K, Na
- And water! (H<sub>2</sub>O)
- Other elements in minor amounts

# Classifying Igneous Rocks

- Amount of Silica

— Felsic	66-76% $\text{SiO}_2$
— Intermediate	52-66% $\text{SiO}_2$
— Mafic	45-52% $\text{SiO}_2$
— Ultramafic	38-45% $\text{SiO}_2$



# Classifying Igneous Rocks

- | <u>Amount of silica</u>   | <u>Intrusive/Extrusive</u> |
|---|----------------------------|
| • Felsic <ul style="list-style-type: none"><li>– High viscosity – less dense</li><li>– Light colour – solid @ 650°C</li></ul>                             | Granite/ Rhyolite          |
| • Intermediate <ul style="list-style-type: none"><li>– Intermediate viscosity and moderate density</li><li>– Intermediate colour – solid @900°C</li></ul> | Diorite/ Andesite          |
| • Mafic <ul style="list-style-type: none"><li>– Low Viscosity – dense</li><li>– Dark colour - &gt;1000°C</li></ul>  | Gabbro/ Basalt             |
| • Ultramafic <ul style="list-style-type: none"><li>– Very low viscosity – very dense – dark colour 1300°C</li></ul>                                       | (intrusive only)           |

# Geochemistry of Rocks

## GRANITE CHEMISTRY

SiO <sub>2</sub>	70.24
TiO <sub>2</sub>	1.21
Al <sub>2</sub> O <sub>3</sub>	13.23
FeOt	2.21
MgO	1.49
CaO	1.11
Na <sub>2</sub> O	3.50
K <sub>2</sub> O	5.43
P <sub>2</sub> O <sub>5</sub>	0.13
LOI	2.60

LOI = Loss on Ignition

FeOt = total iron