Mass Spectrometry (MS)

Measure mass of molecule



How?



Mass Spectrometry (MS)

Light molecule



Large deflection

Heavy molecule



Small deflection

Measure mass of molecule

Measure the degree of deflection



Cocaine

Molecular formula



$$C_{17}H_{21}NO_4$$

DNA

Cocaine

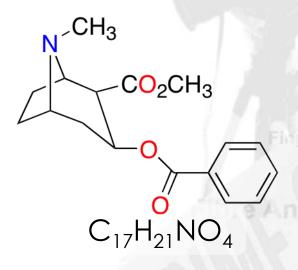
Molecular weight

$$\rightarrow$$
 12 x 17 = 204

$$\rightarrow$$
 1 x 21 = 21

$$16 \times 4 = 64$$

Total =
$$303$$





Cocaine

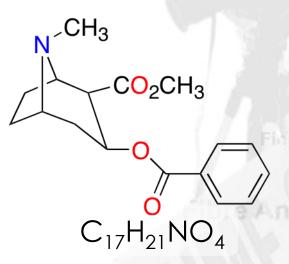
Molecular weight = 303

Cocaine?



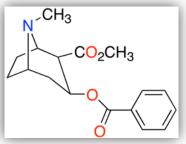
No

Same atoms can be arranged in different ways

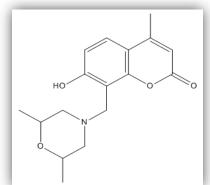


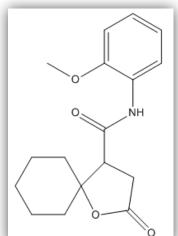


C₁₇H₂₁NO₄ Isomers



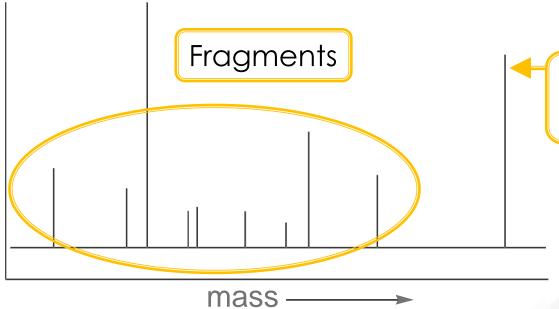
Cocaine





Pattern of fragments at detector





amount of deflection

Molecular ion (M+)

bre Analysis

Pattern of fragments at detector

- Highest weight
- Molecular ion
- Lower weights Fragment ions
- Always the same pattern



Cocaine?

Authentic sample of cocaine

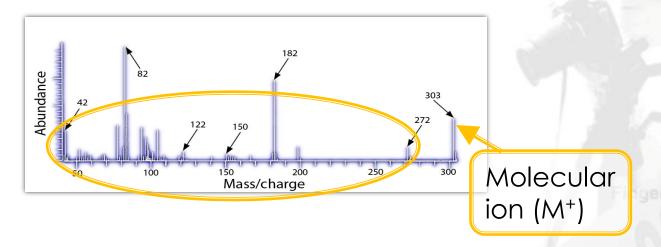
Compare the 2 fragmentation patterns

Same → Cocaine

DNA

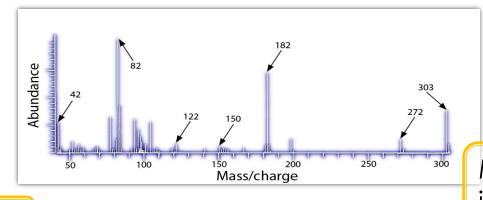
Mass Spectrum

Cocaine

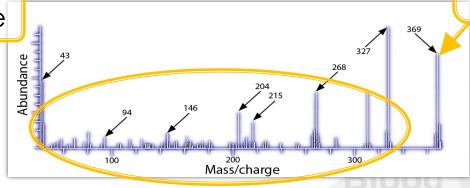




Cocaine



Diamorphine



Molecular ion (M+)

Techniques

- 1 Chromatography
 - Separation of mixtures
- Spectroscopy and spectrometry
 - ldentification of components

Techniques

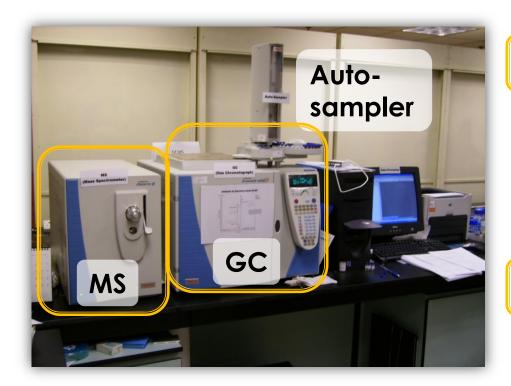
Combining separation and identification

Gas chromatography - Mass Spectrometry (GC-MS)



Hyphenated technique

GC-MS



GC output

Goes to mass spectrometer

MS



Detector









Retention time of components



Quantify amount

MS output



Identification

Combines separation and identification

Summary

- 1 Chromatography
 - TLC, GC and HPLC
 - Separation of mixtures

Summary

- Chromatography
- 2 Infrared spectroscopy
 - Identification by molecular vibrations
 - Molecular stretching and bending

Summary

- Chromatography
- Infrared spectroscopy
- 3 Mass spectrometry
 - Identification by fragmentation