

5. Sensors, Types of Sensors

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SENSORS

- Detects and responds to some certain types of inputs from the environment

Input examples: Moisture, molecules, light, heat, gas etc.

TYPES OF SENSORS

S: Semiconductor sensor

Eg: Solar cells

O: Optical sensor

Eg: Ga-As, Ge-Si, photodetective sensor

M: Mass-sensitive sensor

Eg: Piezo device

E: Electrochemical sensor

Eg: O_2 sensor, glucose sensor
[voltage] [current]

C: Conductivity sensor

Eg: Pt electrode

C: Capacitive sensor

Eg: Touch screens, polycarbonate, polyethylene

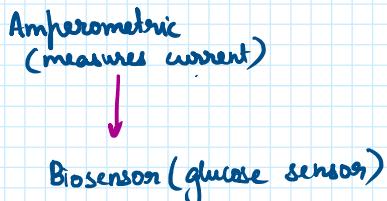
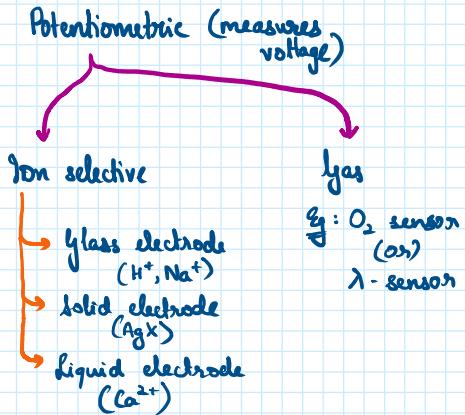
C: Calorific sensor

Eg: Microcalorimeter, Differential Scanning Calorimetry

T: Thermosensitive sensor

Eg: Thermostat

ELECTROCHEMICAL SENSORS → electrode used as transducer; electrode-analyte interaction generates electrical signal

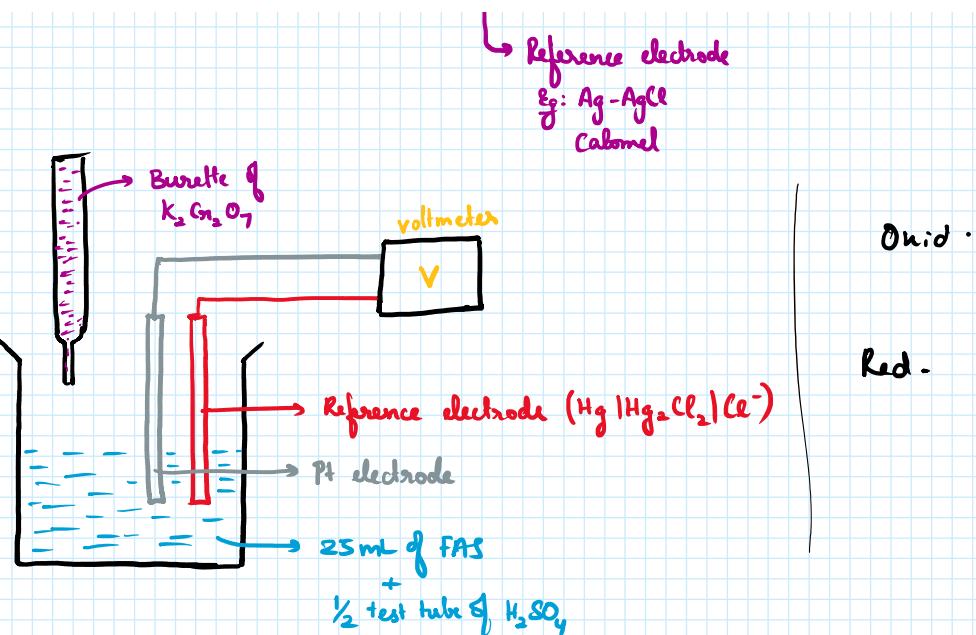


POTENTIOMETRIC SENSOR

- Determines analyte concentration by measuring variation of potential b/w working & reference electrodes
- 2 electrode system

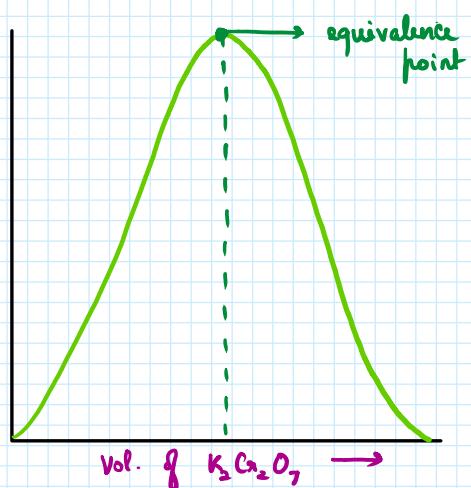
Analyte → Receptor → Transducer → Signal → Computer

→ Pt electrode
→ Reference electrode
Eg: $Ag-AgCl$



Oxid.

Red.



Advantages

- Accurately measure voltage
- Simple
- Qualitative and quantitative measurements

Disadvantages

- Calibration
- Variation in temperature causes change in value of voltage measured

O₂ SENSOR (solid oxide sensor) (potentiometric sensor) → measures proportion of O₂ in analyte

Applications

- I.C. engines
- Medical applications
 - respiration
 - anaesthesia
- Deep sea diving

- Medical applications
 - Deep sea divers
- anaesthesia

Anode: Pt

Cathode: Pt

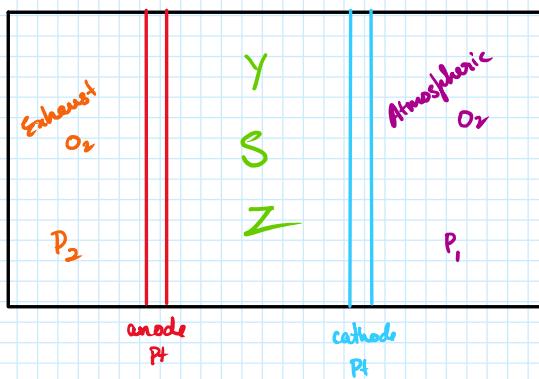
Electrolyte: Ceramic material : **YSZ** → releases O^{2-} ions

Yttria Stabilised Zirconia

ZrO_2 doped with Y_2O_3

Operating temp.: $360^\circ C$ (minimum)

- Air: fuel = 14.7:1
 - Ideal voltage = $0.45 V$
 - Must use lead-free gasoline as lead poisons Pt electrodes
- For ideal combustion

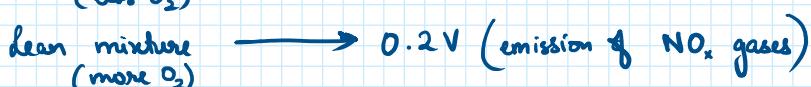
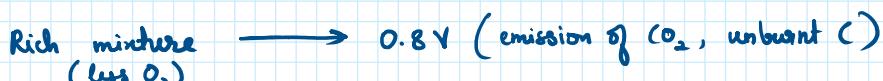
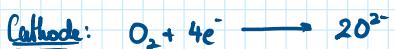


$$E_{cell} = \frac{2.303RT}{nF} \log \frac{P_1}{P_2} \quad \begin{matrix} \xrightarrow{\text{atmospheric } O_2} \\ \xrightarrow{\text{exhaust } O_2} \end{matrix}$$

At $25^\circ C$,

$$E_{cell} = \frac{0.0591}{n} \log \frac{P_1}{P_2} \quad \begin{matrix} \xrightarrow{\text{atmospheric } O_2} \\ \xrightarrow{\text{exhaust } O_2} \end{matrix}$$

Equations



Signs of failure

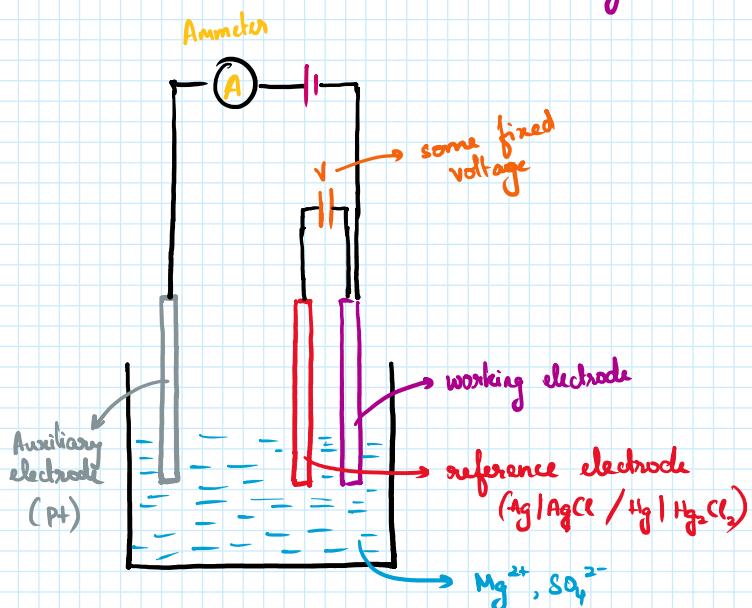
- Jagged emission
- Increased fuel consumption
- Hesitation on acceleration

AMPEROMETRIC SENSOR

- Measures current
- 3 electrode system

Analyte → Receptor → Transducer → Signal → Electronic display

- Working electrode
- Reference electrode [Ag|AgCl, calomel]
- Auxiliary electrode



Advantages

- Simple to construct
- Qualitatively and quantitatively measure

Disadvantages

- -ve voltage leads to errors due to liberation of H₂

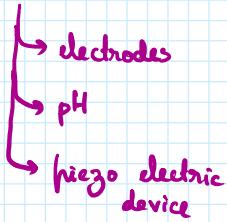
BIOSENSOR → determines presence, concentration of specific substance in biological analyte

Analyte → Biorceptor → Transducer → Signals → Electronic display

↓
electrodes

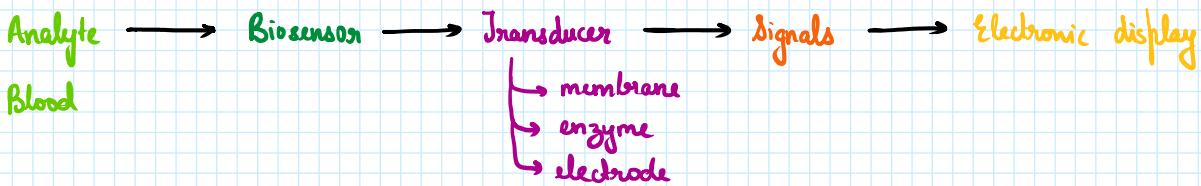
Blood
...
...

Blood
RNA
DNA
cells

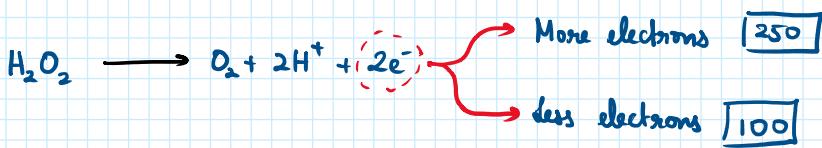
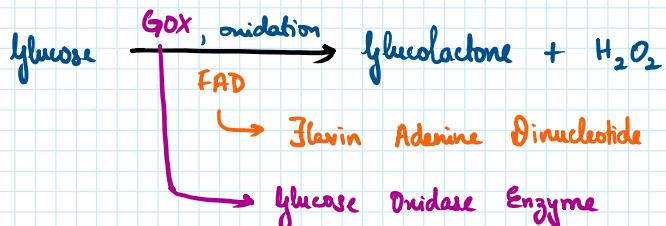


Glucose Sensor

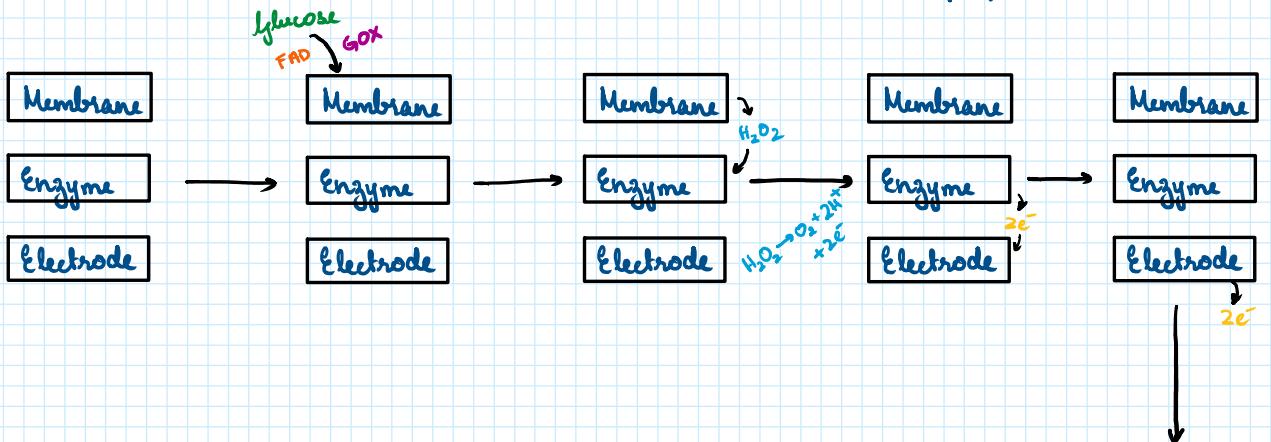
- Amperometric sensor



Blood → HIGH glucose → hyperglycemia
Blood → LOW glucose → hypoglycemia



No. of electron transfers at electrode surface → directly proportional to no. of glucose molecules



Advantages

- Simple method

Disadvantages

- Treatment errors

Advantages

- Simple method
- Checking blood sugar

Disadvantages

- Treatment errors
(glucose levels)