

Lab 3: Spectroscopy

September 25, 2024

1 Introduction: The Rainbow Connection

Spectroscopy is the study of the light that objects emit or absorb. Last lab, we saw that the Universe is filled with a myriad flavors of light, spanning orders of magnitude in wavelength. Spectroscopy hones in on a specific object and holistically studies the range of wavelengths emitted or absorbed by that object – the set of light emitted or absorbed by an object is called the object’s **spectrum**. An object’s spectrum can provide a remarkable amount of information; the spectrum tells us about the chemical makeup of the object, the motion and speed of the object, the temperature of the object, the distance to the object (in certain contexts), and much more. Thus, spectroscopy is an extremely popular technique for studying planets, stars, and galaxies. Much of what we know about these objects is *because* of spectroscopy.

In the last lab, I posed the question: “If we were to shine *starlight* through a prism, what could the resulting light pattern tell us about the star?” Today, we will work to answer this question.

2 Blackbody Radiation

Before we dive into the spectra of real objects, we’ll consider the simplified, idealized model of a **blackbody**. A blackbody is an object that perfectly *absorbs* all wavelengths of light, with no reflection. In addition to absorbing all wavelengths of light, a perfect blackbody will also *emit* light at all wavelengths. **Answer the following questions in your lab write-up:**

1. Consider the processes of *emission*, *absorption*, and *reflection* (or “*scattering*”) of light. A and B are only graded for correctness.
 - (a) Define these three terms in **your own words**. If you use online sources, be sure to **paraphrase** and **cite**.
 - (b) What determines the perceived color of an object? (Hint: this is review from the last lab; if you need a reminder, check Figure 1 of Lab 2).
 - (c) Why do you think a blackbody is called a “blackbody”? Hint: think about what it’s like when you wear black clothing in the summer.

Plotting the intensity (or brightness) of light at each wavelength emitted by a blackbody yields the *blackbody spectrum*. Although blackbody spectra are smooth, while stars and other astronomical objects are not, we can usually approximate them as blackbodies. We’ll explore the ways in which real spectra deviate from blackbody spectra in later sections.

For now, though, let's examine the blackbody spectrum hands-on. Navigate to https://phet.colorado.edu/sims/html/blackbody-spectrum/latest/blackbody-spectrum_en.html. The plot to the left of the screen shows a blackbody spectrum, with spectral power density (the amount of light emitted) on the vertical axis and wavelength on the horizontal axis; don't worry too much about these units. Make sure to check the boxes for "Graph Values" and "Labels" in the box to the upper right; after turning on these features, you should see a pair of yellow numbers labeling the *peak* of the blackbody curve.

To the right of the screen is a thermometer showing the temperature of the blackbody. You can click and drag the slider on the thermometer to change the temperature of the object. If you need to adjust the size of your graph to see the full curve, you can use the zoom buttons (the magnifying glasses) to the upper left and bottom right of the plot.

Make a table with three columns: 1) temperature (in units of K); 2) the wavelength (in units of μm) corresponding to the peak of the spectrum (i.e., the value labeled by the yellow number on the horizontal axis); and 3) the *reciprocal* of the peak wavelength ($1/\lambda_{\text{peak}}$, in units of $1/\mu\text{m}$). Record the following in your lab write-up:

2. Test **at least 5** different temperatures across the full range of the thermometer and fill out the columns in your table. Report your numbers using *two* significant figures. **Three of these temperatures** should be that of a **light bulb**, the **Sun**, and **Sirius A**. Use the markings on thermometer to easily get these three.
3. What do you notice about the relationship between temperature and peak wavelength? Do temperature and peak wavelength vary *directly* with one another (temperature increases when peak wavelength increases), or do they vary *inversely* (temperature increases when peak wavelength decreases, or vice versa)?
4. Make a plot of your data with the *reciprocal* of the peak wavelength ($1/\lambda_{\text{peak}}$) on the vertical axis and the temperature on the horizontal axis. Make sure to label your axes with the correct units! What do you notice about this plot?
5. Make a new column in your table and calculate the value of the peak wavelength *multiplied* by the temperature (using two significant figures). What do you notice about these values?

Congratulations! You've just derived a key equation in astronomy, **Wien's law**. Because the product of the peak wavelength and the temperature is always the same number, we can determine the surface temperature of an object (like a star) by measuring the wavelength at which the object emits most of its light. In math, Wien's law reads

$$\lambda_{\text{peak}} \times T = (2.9 \times 10^3) \mu\text{m K}, \quad (1)$$

where λ_{peak} is the peak wavelength (in μm) and T is the temperature (in K). Let's use Wien's law to answer a few questions. **Record your answers in your lab write-up.**

6. Are redder objects hotter or cooler than bluer objects? (note that this applies to both the flames on our kitchen stoves and to the stars illuminating the night sky) This answer can be a few words.
7. The surface temperature of the Sun is around 5780 K.
 - (a) At which wavelength does the Sun emit most of its light (give your answer in μm to two sig figs)? **Show your work.**
 - (b) What color of light does the Sun emit the most? (Note: you can use the tool from the previous section, too; make a good approximation and **justify** your answer). This answer should **NOT** come from the Internet.
8. The *cosmic microwave background* (CMB) is the oldest perceptible light in the Universe. Since the CMB permeates the entirety of outer space, the temperature of the CMB is often referred to as simply the temperature of the Universe. The CMB forms a remarkably perfect blackbody spectrum, with a peak wavelength around 1.06 mm (see Figure 1).
 - (a) What is the temperature of the Universe (in K, to two sig figs)? Be careful of units!
 - (b) Why is “cosmic *microwave* background” an appropriate name for this light? Refer back to the last lab if you need to.

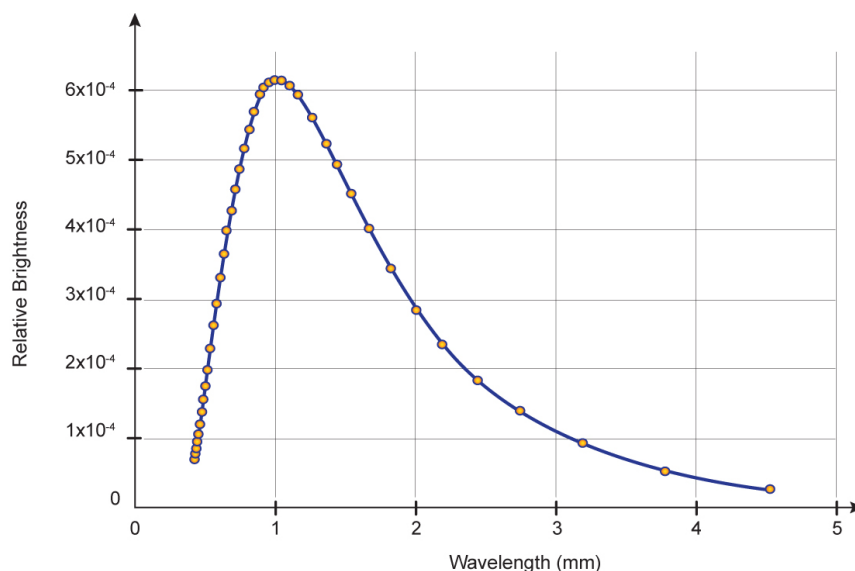


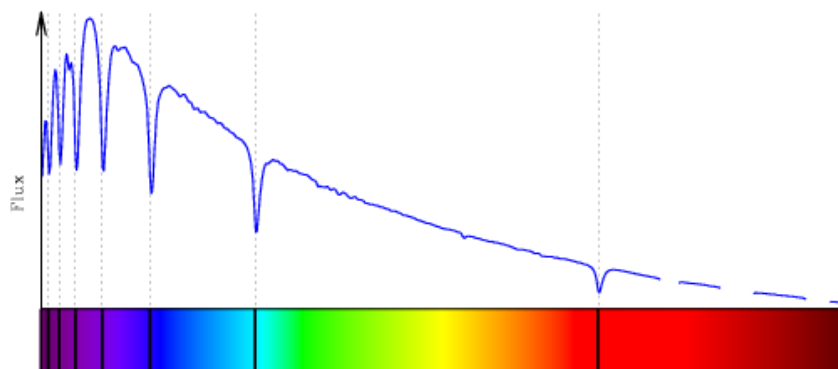
Figure 1: Spectrum of the cosmic microwave background (CMB).

3 Fingerprints of the Elements

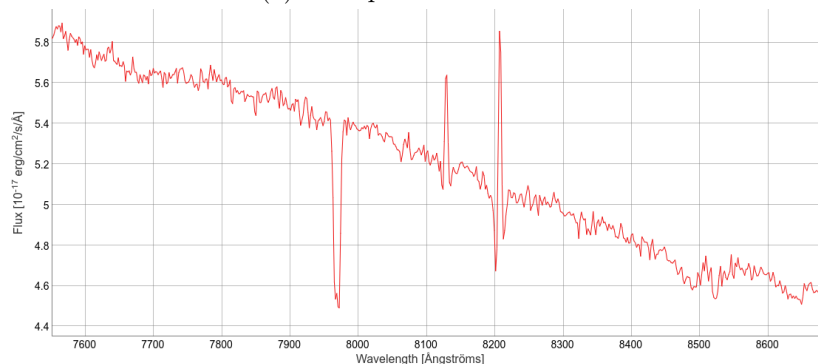
The blackbody spectrum is a perfectly smooth spectrum. In reality, however, objects (like stars and galaxies) possess many “features” – or *lines* – in their spectra that can tell us much more than

just the temperature of the object. Let's explore some real spectra now.

Figure 2 shows examples of two different real spectra. On the horizontal axis, we have the wavelength of light (as was also the case in the blackbody spectrum); in general, spectra can cover wavelengths of light spanning the entire electromagnetic spectrum. On the vertical axis, we have *flux*, another measure for how much light is being emitted and detected (in particular, flux measures the amount of light passing through a given area in a given amount of time).



(a) The spectrum of a star.



(b) The spectrum of a galaxy.

Figure 2

Answer the following in your lab write-up:

9. The color bar below Figure 2a (the stellar spectrum) is an alternative representation of the spectrum. How do the features in the Flux vs. Wavelength plot correspond to the features on the color bar?
10. What differences do you notice between the stellar spectrum and the galaxy spectrum? What are some similarities? Hypothesize as to *why* the spectrum of a star differs from that of a galaxy. This will be graded on completion/quality of answers rather than correctness.

3.1 Lamp Activity

If you do not already have a *diffraction grating*, come get one from me. **Please keep track of your diffraction grating and ensure it is returned at the end of the lab.** A diffraction grating splits light into its component colors, much like a prism. There are several lamps set up in the room; each lamp is actually heated gas of a specific element. Go up to each lamp and look at it through your diffraction grating. **Please do not touch the lamps, they are hot!**

Answer the following questions in your write-up:

11. Record the name and draw the spectrum you see for each gas - give it your best shot. Using color is nice, but not necessary; label features or use words if you think you need to.
12. Note the similarities and differences between the spectra.
13. Craft a hypothesis as to *why* the lamps look different from each other when viewed through a diffraction grating.
14. Now look at one of the light bulbs (a source of white light) in the room with your diffraction grating. How is the spectrum of the light bulb different from the element lamps?

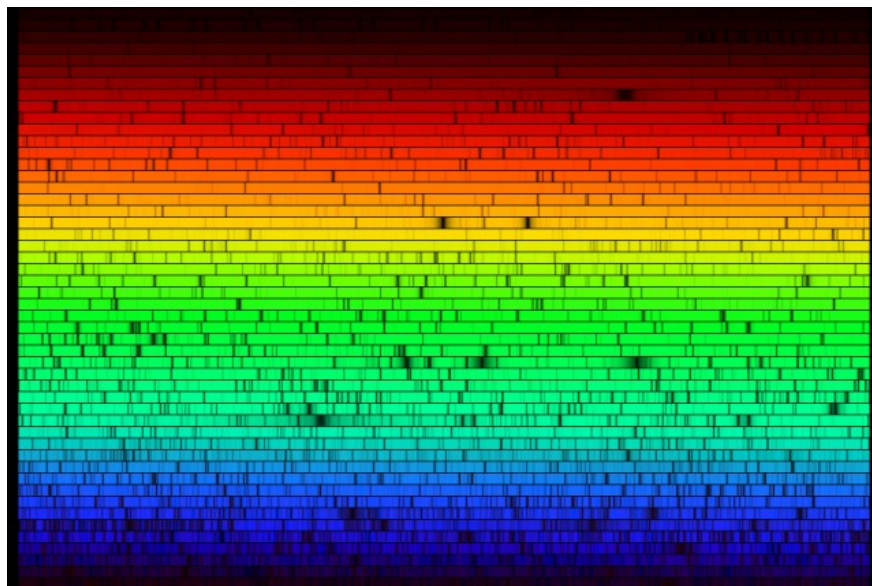


Figure 3: The spectrum of the Sun. Note that the actual spectrum should be one long, horizontal color gradient (like the ones seen through your diffraction gratings) – this spectrum has been broken up into rows to fit onto the page and to show more detail.

Now check out Figure 3, the spectrum of the Sun. **Answer the following in your lab write-up:**

15. How is the Sun's spectrum different from the lamp spectra and the light bulb spectrum?

Figure 4 shows three different types of spectra and how they're formed. Blackbodies emit across all wavelengths and thus have a **continuous spectrum**. A sample of gas is usually a conglomerate of multiple different chemical elements, so when the atoms of a gas are heated, they emit light at certain wavelengths, thus forming an **emission spectrum**. By contrast, if light from a blackbody or a continuous spectrum passes *through* a gas, the atoms of the gas absorb some light, yielding an **absorption spectrum**. Due to the quantum nature of atoms, each element and molecule absorbs or emits at a unique set of wavelengths; each element has its own spectral “fingerprint” that can be used to uniquely identify the chemical makeup of an object.

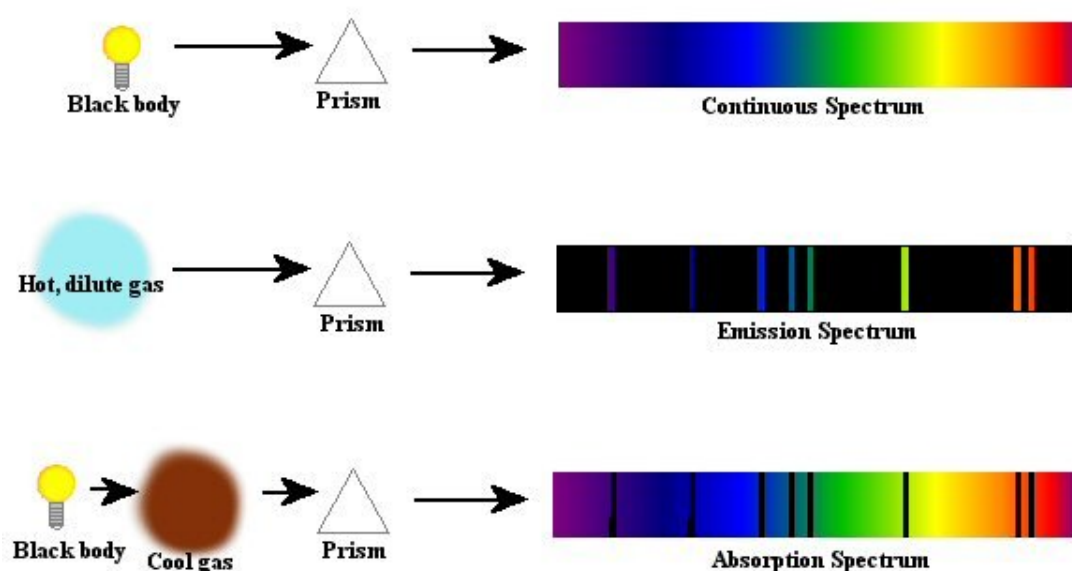


Figure 4: Types of spectra.

In your lab write-up, answer the following, based on Figure 4:

16. What type of spectrum matches the Sun's spectrum? Why do you think the Sun has this particular type of spectrum? (Hint: we observe light coming from the Sun's photosphere, but the Sun's corona lies between the photosphere and us)
17. What type of spectrum matches the light bulb spectrum? This answer can be 1-2 words.
18. What type of spectrum matches the lamp spectra? This answer can be 1-2 words.

3.2 Decomposing the Stars

The spectral fingerprints we just explored allow astronomers to probe the chemical compositions of stars, galaxies, and planets. If we see an absorption line that we know is from carbon, then we know that there must be carbon in the star. The deeper the carbon absorption line in the spectrum, the more carbon there must be.

Check Courseworks for the "Stellar Spectra Activity" document in the Lab 3 folder. This should be a PDF document that has about 10 pages. Within this document, there are two real spectra from stars, labelled in orange, and some example spectra of seven specific elements, like the ones you just looked at with the lamps, labelled in blue. To make things easier, I have already lined up the stellar spectra with each element for you. Each page has different elements on it that are lined up with the stellar spectra. **In your lab write-up, complete the following:**

19. Make a table with a column for each element (seven total) and a row for each star (two total). Record which elements you find in each star by lining up the spectra of each element with the spectra of each star - **I have already done this for you**. Mark the elements you do *and* don't see in each star.
20. Which elements are in both stars?

4 Spectra and Motion: Oneshift, Twoshift, Redshift, Blueshift

Spectra are not only useful for chemistry – we can also use spectroscopy to study the *motions* of astronomical bodies. Much like how the pitch of an ambulance changes as it moves towards or away from you, the wavelength of light changes depending on the motion of the light source – this is a manifestation of the *Doppler effect*. As a source of light moves towards you, the emitted light waves are *compressed*, yielding shorter, bluer wavelengths: this is called *blueshift*. Similarly, if a source of light is moving away from you, the light waves are *stretched*, yielding longer, redder wavelengths: this is called *redshift*. For fun, you can check out https://javablab.org/en/doppler_effect_and_redshift_en/ to see blueshift and redshift in action.

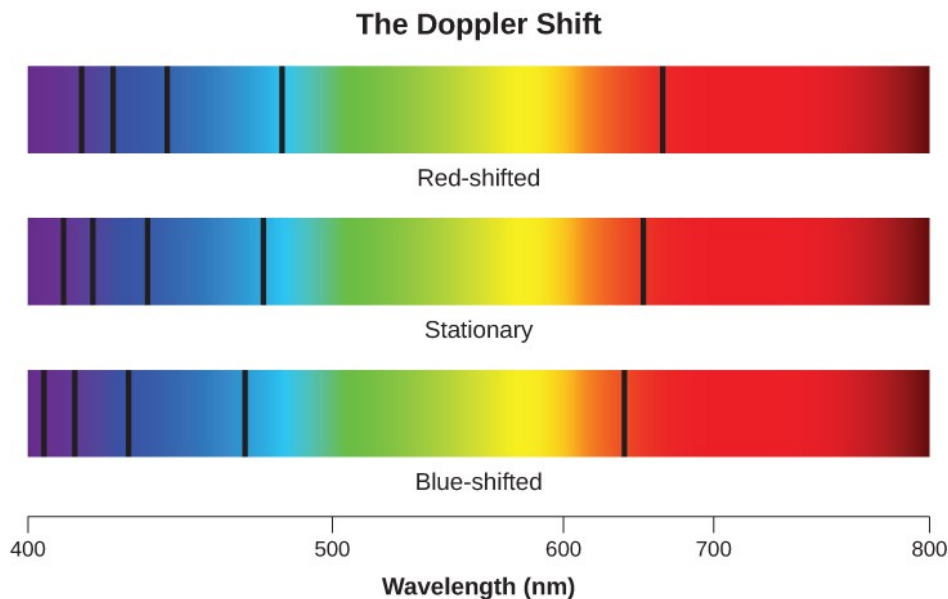


Figure 5: Redshift vs. Blueshift

This redshift or blueshift manifests itself in the spectra of moving objects, as shown in Figure 5. The middle row of Figure 5 shows the spectrum in the *rest-frame* of the object – that is, this is the spectrum we would observe if the object were perfectly stationary. **Answer the following in your lab write-up, and be sure to answer all parts of the question.:**

21. The spectrum on the top row of Figure 5 corresponds to an object moving *away* from us. How do the positions of the spectral absorption lines differ from those in the stationary spectrum? Are these features shifted towards bluer or redder wavelengths? **Both answers to this question be a few words.**
22. The spectrum on the bottom row of Figure 5 corresponds to an object moving *towards* us. How do the positions of the spectral absorption lines differ from those in the stationary spectrum? Are these features shifted towards bluer or redder wavelengths? **Both answers to this question be a few words.**
23. Because the Universe is expanding, most objects are moving *away* from us. Consider a galaxy moving with the expansion of the Universe: how would the galactic spectrum we *observe* differ from the spectrum in the rest-frame of the galaxy?

We can quantify the shift in a spectrum via the quantity z , often just referred to as *the* redshift (even though it quantifies both blue-shifts and red-shifts). If an object at rest emits light of wavelength λ_{rest} , but we observe that light at a redder or bluer wavelength, λ_{obs} , due to the motion of the object, then the redshift, z , of the object is given by

$$z = \frac{\lambda_{\text{obs}} - \lambda_{\text{rest}}}{\lambda_{\text{rest}}} \quad (2)$$

5 Wrapping things up

In your lab write-up, reflect on the following:

24. Qualitatively describe what the *spectrum* of an object is.
25. What can we learn by studying the spectra of astronomical objects (i.e., what information does *spectroscopy* give us)?
26. Look back on what you've learned in lab so far and describe three different ways astronomers use light to learn about the Universe.
27. How do you think one could use spectroscopy to determine the rotation speed of a star? (Hint: You can take individual spectra of different parts of the star. You should also think about how the material on the surface of the star is moving relative to the observer) This question will not be graded for correctness. Do not hesitate to ask for help if you aren't sure!