

B.A. / B.Sc. SEMESTER 2 EXAMINATION, 2021
FAKIR CHAND COLLEGE CENTRE (551)

INSTRUCTIONS FOR CANDIDATES

READ ALL THE INSTRUCTIONS CAREFULLY BEFORE WRITING ANSWERS

1. Total **TIME OF EXAMINATION: 1 HOUR**
2. Question Paper Comprises Of Two Separate Questions – CC3 (10 Marks) And CC4 (10 Marks). Candidates Must Have To Answer The Two Question papers Separately And Have To Prepare Two pdf Files.
3. In The Answer Script Only Question Number & Correct Option To Be Mentioned. No Need Of Writing The Whole Answer.
4. ANYONE PREVIOUS SEMESTER ADMIT CARD As The Last Page Of Each pdf File
5. University Roll Number must be there in each pdf file name
6. Use Only **WHITE PLAIN A4 PAPERS** For Writing Answers
7. Use **ONLY BLACK INK** For Writing Your Answers
8. Give **A TOP PAGE** With Clear Mention Of University **REGISTRATION NO. AND UNIVERSITY ROLL NO.** Of Anyone Previous Semester
9. Give **AT LEAST 1CM MARGINS** In All The Four Sides Of Each Page

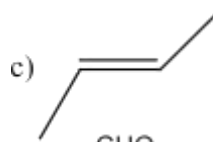
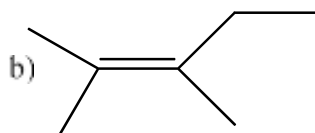
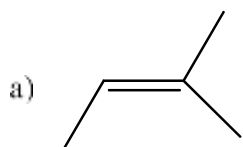
2021
B.A. /B.Sc. Semester 2 Examination
University of Calcutta
CHEMISTRY – HONOURS
INTERNAL EXAMINATION
Paper : CC3
F.M. 10

FAKIR CHAND COLLEGE CENTRE (551)

Choose the correct option in each case and report (no need to write the answer in sentence):

1x10

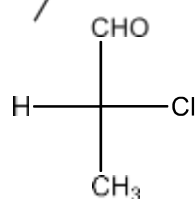
1. Most stable conformation of 1,2-difluoroethylene is
a) Staggered form b) gauche form c) eclipsed form d) none of these
2. S_N1 reaction takes place with
a) Inversion of configuration b) retention of configuration c) racemisation d) resolution
3. Which reacts faster in S_N1 reaction
a) Tertiary butyl chloride b) Isopropyl chloride c) Ethyl chloride d) Methyl chloride
4. Base catalysed dehydration of aldol undergoes the path
a) E_2 b) E_1 c) E_{1cb} d) None of these
5. Which reacts faster in S_N2 Path ?
a) CH_3Br b) CH_3Cl c) CH_3F d) CH_3I
6. Which is the correct topic face of acetone?
a) diastereotopic b) homotopic c) enantiotopic d) none of these
7. Which molecule shows geometrical isomer?



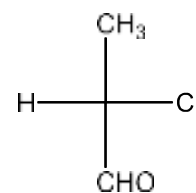
d) none of these

8. Relationship between the molecules

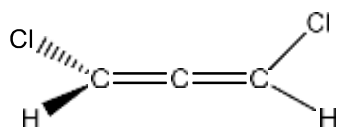
- a) Homomer b) Enantimer c) Diastereomer
d) None of these



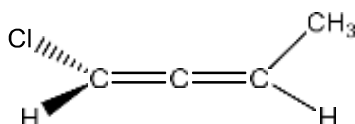
and



9. The following molecule is an example of a) Symmetry allene b) Asymmetry allene
c) Dissymmetry allene d) None of these



10. Absolute configuration of the molecule is a) R b) S c) E d) Z



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CHEMISTRY – HONOURS
INTERNAL EXAMINATION
Paper : CC4
F.M. 10

FAKIR CHAND COLLEGE CENTRE (551)

Choose the correct option in each case and report (no need to write the answer in sentence):

1x10

- Q.1) The shape of XeF_4 molecule is
a) square planar b) tetrahedral c) octahedral
- Q.2) When Ga is added to Ge as an impurity
a) a n-type semiconductor is formed b) a p-type semiconductor is formed
c) an insulator is formed
- Q.3) Crystals of AgCl exhibit
a) Schottky defect b) Metal excess defect c) Frenkel defect
- Q.4) The solubility trend among the following sulphates is
a) $\text{MgSO}_4 > \text{CaSO}_4 > \text{SrSO}_4 > \text{BaSO}_4$
b) $\text{MgSO}_4 < \text{CaSO}_4 < \text{SrSO}_4 < \text{BaSO}_4$
c) $\text{CaSO}_4 < \text{BaSO}_4 < \text{SrSO}_4 < \text{MgSO}_4$
- Q.5) Which is not a magic number?
a) 2 b) 12 c) 50
- Q.6) The mesons have masses nearly
a) 200 times the mass of an electron
b) 250 times the mass of an electron
c) 100 times the mass of an electron
- Q.7) In nuclear reactors , the fissionable material is usually
a) Uranium-255 b) Uranium-238 c) Uranium-235
- Q.8) Bond order of a molecule is
a) directly proportional to its bond length
b) inversely proportional to its bond length
c) not related to its bond length
- Q.9) Which structure is derived from cubic-close packing of ions?
a) Sodium chloride b) Wurtzite c) Rutile
- Q.10) Madelung constant depends on
a) electronic configuration of the ions b) geometry of the crystal
c) lattice energy of the crystal