

Mathematical model for the concentration and electric potential profiles in a solution of electrolytes under a redox reaction

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Chapter 1

Diffusion Of Electrolytes In Solution

1.1 System of equations

The system of equations we need to solve is

$$\frac{\partial C_+}{\partial t} + \nabla \cdot \mathbf{N}_+ = 0 \quad (1.1)$$

$$\frac{\partial C_-}{\partial t} + \nabla \cdot \mathbf{N}_- = 0 \quad (1.2)$$

$$\nabla^2 \phi(x) = -\frac{\rho(x)}{\epsilon} \quad (1.3)$$

Where C_s is the concentration of each species and N_s is the diffusive flux for each species. In our model, the flux is defined as

$$\begin{aligned} \mathbf{N}_+ &= -D_+ \left(\nabla C_+(x) + \frac{z_+ F}{RT} C_+(x) \nabla \phi(x) \right) \\ \mathbf{N}_- &= -D_- \left(\nabla C_-(x) + \frac{z_- F}{RT} C_-(x) \nabla \phi(x) \right) \end{aligned}$$

Therefore, the diffusion equation with electric field interaction incorporated is

$$\frac{\partial C_+}{\partial t} + \nabla \cdot \left[-D_+ \left(\nabla C_+(x) + \frac{z_+ F}{RT} C_+(x) \nabla \phi(x) \right) \right] = 0 \quad (1.4)$$

$$\frac{\partial C_-}{\partial t} + \nabla \cdot \left[-D_- \left(\nabla C_-(x) + \frac{z_- F}{RT} C_-(x) \nabla \phi(x) \right) \right] = 0 \quad (1.5)$$

where $z = |z_+| = |z_-|$ is the valence of the electrolyte, D_{\pm} are the diffusion coefficients for each chemical species and $F = 96485.33 \text{ C mol}^{-1}$ is Faraday's constant.

The reaction we are modeling is the cation reduction. In particular, we are interested in copper reduction at the surface



This chemical reaction yields the following border condition for the flux

$$\begin{aligned} \mathbf{N}_+ \cdot \hat{n}|_{\text{Surface}} &= \frac{I_0}{z_+ F A} \\ \mathbf{N}_- \cdot \hat{n}|_{\text{Surface}} &= 0. \end{aligned} \quad (1.7)$$

with A being the total area of the surface, I_0 the total electric current and \hat{n} the unit normal of the surface.

Chapter 2

Steady State Solution

2.1 Linearization of the Poisson Equation

As a first approach to solving the equilibrium system (that is, system when there is no current at all), we can simplify the Poisson-Boltzmann Equation by linearizing the Boltzmann term in the Debye-Huckel theory to first order in the potential. The Poisson-Boltzmann equation is

$$-\frac{d^2\phi(x)}{dx^2} = \frac{1}{\epsilon} \sum_s z_s F C_{b,s} e^{\frac{z_s F \phi(x)}{RT}},$$

which by expanding the exponential for $\left| \frac{z_s F \phi(x)}{RT} \right| < 1$ we get

$$-\frac{d^2\phi(x)}{dx^2} = \frac{1}{\epsilon} \sum_s z_s F C_s \left(1 - \frac{z_s F \phi(x)}{RT} \right)$$

Due to electro-neutrality in the bulk solution, the first term in the right hand side is zero. Therefore,

$$\frac{d^2\phi(x)}{dx^2} = \kappa^2 \phi(x)$$

where we have defined

$$\kappa = \sqrt{\frac{\sum_s C_{b,s} z_s^2 F^2}{\epsilon RT}}$$

Given the boundary conditions $\phi(0) = V_0$, and considering the reference zero at $\phi(\delta) = \phi_b = 0$, the solution is trivially found to be

$$\phi(x) = V_0 e^{-\kappa x},$$

This gives the potential in a static situation, in which the electrolytes move to an equilibrium position and the configuration as a hole is static.

This results gives a good physical sense of how the potential should look like in a context of electrolyte solutions at equilibrium. We are interested in the dynamics of the system, though, so it makes sense to try to incorporate the current into Eq. 2.1. In the next section we shall incorporate the current flowing through the interface as a border condition of Eq. 2.1.

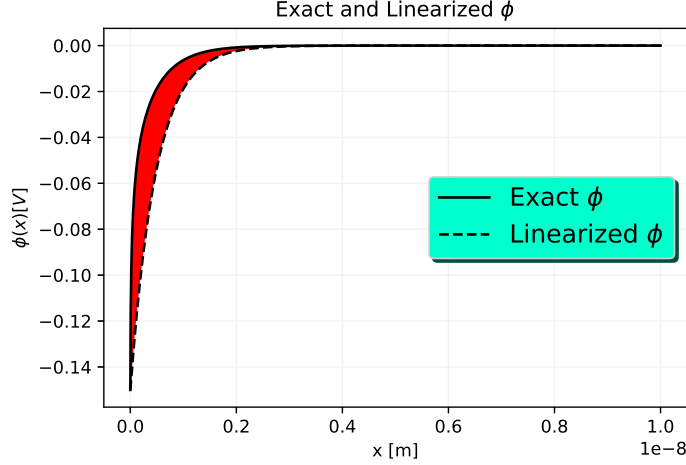


Figure 2.1: Comparison of the linearized PB solution with the analytical solution. The region in red is the error committed in this approximation

2.2 Steady State Solution To The Diffusion-Reaction Problem

In what follows, we shall assume that the system presents concentration gradients and potential gradients along a single dimension x . This is equivalent to considering the interface infinitely large. Therefore, $\mathbf{N}_+ = \hat{x}N_+$ and $\mathbf{N}_- = \hat{x}N_-$.

In the steady state solution the concentration distribution of each electrolyte throughout the system should not change in time, thus

$$\frac{\partial C_+}{\partial t} = \frac{\partial C_-}{\partial t} = 0$$

This yields the following results

$$\nabla \cdot \mathbf{N}_+ = \frac{\partial N_+}{\partial x} = 0$$

$$\nabla \cdot \mathbf{N}_- = \frac{\partial N_-}{\partial x} = 0$$

Therefore, we have

$$N_+ = A_1$$

$$N_- = A_2$$

where A_1 and A_2 are constants determined by border conditions. Since the anion does not interact with the interface, we get after Eq.(1.7) $N_- = 0$. On the other hand, the flux due to the cation reaction with the interface ($Cu^{+2} \rightarrow Cu^0$) gives the boundary condition for N_+ ,

$$N_+ = \frac{I_0}{Az_+F}$$

The complete system of equations to be solved is

$$N_-(x) = 0$$

$$N_+(x) = \frac{I_0}{Aze}$$

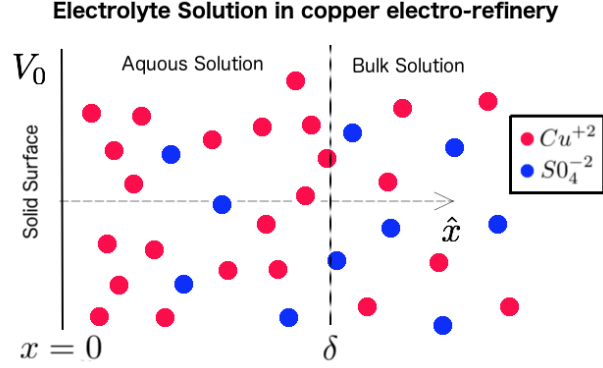


Figure 2.2: Geometry of the problem. $x < \delta$ is the region of laminar. $x > \delta$ is the bulk of the solution. This image is for a value of $V_0 < 0$, such that the positive electrolytes distribute closer to the surface.

According to Fig. 2.2, for an interface area large compared to the laminar flux region we get the following equations

$$\frac{\partial C_+}{\partial x} - \frac{zF}{RT} C_+ \frac{\partial \phi}{\partial x} = -\frac{I_0}{D_+ A z_+ F} \quad (2.1)$$

$$\frac{\partial C_-}{\partial x} + \frac{zF}{RT} C_- \frac{\partial \phi}{\partial x} = 0 \quad (2.2)$$

$$\frac{\partial^2 \phi}{\partial x^2} = -\frac{1}{\epsilon} \sum_{s=\pm} z_s F C_s(x) \quad (2.3)$$

where A is the area of the interface electrode.

To solve this system of partial differential equations, we will expand the concentration and the potential as a series in powers of $r = \frac{I_0}{D_+ A z_+ F}$. We have

$$C_s(x) = \sum_{n=0}^{\infty} r^n C_s^{(n)}(x) \quad (2.4)$$

$$\phi(x) = \sum_{n=0}^{\infty} r^n \phi^{(n)}(x) \quad (2.5)$$

Truncating the series at first order in r , we obtain the following system of equations to

order zero in the current

$$\frac{\partial C_+^{(0)}}{\partial x} - \frac{zF}{RT} C_+^{(0)}(x) \frac{\partial \phi^{(0)}}{\partial x} = 0 \quad (2.6)$$

$$\frac{\partial C_-^{(0)}}{\partial x} + \frac{zF}{RT} C_-^{(0)}(x) \frac{\partial \phi^{(0)}}{\partial x} = 0 \quad (2.7)$$

$$\frac{\partial^2 \phi^{(0)}}{\partial x^2} = -\frac{1}{\epsilon} \sum_{s=\pm} szFC_s^{(0)}(x) \quad (2.8)$$

and to first order in the current

$$\frac{\partial C_+^{(1)}}{\partial x} - \frac{zF}{RT} \left(C_+^{(1)}(x) \frac{\partial \phi^{(0)}}{\partial x} + C_+^{(0)}(x) \frac{\partial \phi^{(1)}}{\partial x} \right) = -1, \quad (2.9)$$

$$\frac{\partial C_-^{(1)}}{\partial x} + \frac{zF}{RT} \left(C_-^{(1)}(x) \frac{\partial \phi^{(0)}}{\partial x} + C_-^{(0)}(x) \frac{\partial \phi^{(1)}}{\partial x} \right) = 0, \quad (2.10)$$

$$\frac{\partial^2 \phi^{(1)}}{\partial x^2} = -\frac{1}{\epsilon} \sum_{s=\pm} szFC_s^{(1)}(x). \quad (2.11)$$

2.3 Zero order solution to Poisson's equation for the electrolyte solution

We want to work with the dimensionless potential

$$\Phi(x) = \frac{zF}{RT}\phi(x).$$

The zero order system can thus be written as

$$C_+^{(0)}(x)' - C_+^{(0)}(x)\Phi^{(0)}(x)' = 0 \quad (2.12)$$

$$C_-^{(0)}(x)' + C_-^{(0)}(x)\Phi^{(0)}(x)' = 0 \quad (2.13)$$

$$\Phi^{(0)}(x)'' = -\frac{(zF)^2}{RT\epsilon}(C_+^{(0)}(x) - C_-^{(0)}(x)) \quad (2.14)$$

In this section we will solve equations 2.10 and 2.11.

From Eq. 2.12,

$$\frac{\partial C_s^{(0)}}{\partial x} - sC_s^{(0)}\Phi^{(0)}(x) = 0$$

which yields

$$\int_{C_{b,s}}^{C_s^{(0)}(x)} \frac{dC_s^{(0)}}{C_s^{(0)}} = s \int_{\phi_b}^{\phi^{(0)}} d\Phi^{(0)}$$

where $\Phi_b = \Phi(\delta)$ is the potential at the bulk, which we will consider as the reference zero, $\phi_b = 0$. $C_{b,s}$ is the bulk concentration of each species in solution.

$$C_s^{(0)}(x) = C_{b,s}e^{s\Phi^{(0)}(x)}$$

Notice that

$$C_s^{(0)}(\delta) = C_{b,s}e^{\frac{zF}{RT}\phi_b} = C_{b,s}. \quad (2.15)$$

Consider the bulk values of the concentration, $C_{s,b}$. In the bulk, the solution should be electrically neutral due to conservation of charge. Therefore, for a two electrolyte salt at the bulk

$$\sum_{s=\pm} C_{s,b}szF = 0 \quad (2.16)$$

where z is the valence of the electrolytes and $s = \pm$ is the sign of the charge of the electrolyte. $F = eN_A$ is the Faraday constant. From 2.16 we get,

$$\begin{aligned} C_{b,+}Fz - C_{b,-}Fz &= 0 \\ \rightarrow \frac{C_{b,+}}{C_{b,-}} &= 1 \end{aligned} \quad (2.17)$$

$$C_{b,+}Fz - C_{b,-}Fz = 0$$

$$\rightarrow \frac{C_{b,+}}{C_{b,-}} = 1$$

In the case of a symmetric salt, $q_s = -q_{-s}$ (since in our notation $s = \pm$). Poisson's equation can be written (to zero order in the current) as

$$\frac{\partial^2 \Phi^{(0)}}{\partial x^2} = -\frac{zF}{\epsilon} \left(C_{b,+} e^{\Phi^{(0)}(x)} - C_{b,-} e^{-\Phi^{(0)}(x)} \right) \quad (2.18)$$

where $z = |z_-| = |z_+|$. From equations 2.15 and 2.17 we have $C_{b,+} = C_{b,-} = C_b$. Equation 2.18 can be written in terms of the hyperbolic sine,

$$\frac{\partial^2 \Phi^{(0)}}{\partial x^2} = 2 \frac{(zF)^2 C_b}{RT\epsilon} \sinh(\Phi^{(0)}(x)) \quad (2.19)$$

We define the quantity

$$\kappa^2 = \frac{(zF)^2 C_b}{RT\epsilon}$$

$$\frac{\partial^2 \Phi^{(0)}}{\partial x^2} = 2\kappa \sinh(\Phi^{(0)}(x)) \quad (2.20)$$

We want to change variables to the dimensionless quantity κx ,

$$x \rightarrow \xi = \kappa x$$

$$d\xi = \kappa dx$$

Thus, we write,

$$\Phi^{(0)}(\xi)'' = 2 \sinh(\Phi^{(0)}(\xi)) \quad (2.21)$$

Multiplying by $\Phi'^{(0)}$ and integrating the equation over the interval $[\xi, \delta]$ we get,

$$\Phi^{(0)}(\kappa\delta)^2 - \Phi^{(0)}(\xi)^2 = 2(\cosh(\phi(\kappa\delta)) - \cosh(\phi(\xi))) \quad (2.22)$$

The border condition for the potential yields,

$$\Phi^{(0)}(\xi)' \rightarrow 0.$$

Therefore

$$\Phi^{(0)}(\xi)^2 = 2(\cosh(\phi(\xi)) - 1), \quad (2.23)$$

$$= 4\sinh^2(\phi(\xi)), \quad (2.24)$$

$$\rightarrow \Phi^{(0)} = \pm 2 \sinh(\phi(\xi)). \quad (2.25)$$

The border conditions for the potential are $\Phi^{(0)}(0) = \bar{V}_0 < 0$ and $\Phi^{(0)}(\kappa\delta) = 0$, the slope of $\Phi^{(0)}$ must be positive and decreasing. This yields the positive solution to equation 2.25

$$\Phi^{(0)} = 2\sqrt{\sinh(\phi(\xi))}. \quad (2.26)$$

This is a separable equation which can be integrated directly, yielding

$$\log \left(\frac{1 - e^{\Phi^{(0)}/2}}{1 + e^{\Phi^{(0)}/2}} \right) = -\xi + C, \quad (2.27)$$

$$\rightarrow \frac{1 - e^{\Phi^{(0)}/2}}{1 + e^{\Phi^{(0)}/2}} = Ae^{-\xi}. \quad (2.28)$$

It can be found using the border condition $\Phi^{(0)}(0) = \bar{V}_0$

$$A = \tanh(\bar{V}_0/4) \quad (2.29)$$

Solving for $\Phi^{(0)}$,

$$\Phi^{(0)}(\xi) = 2 \log \left(\tanh \left(\frac{\xi - \xi_0}{2} \right) \right), \quad (2.30)$$

where we have defined

The minus sign in the previous definition is due to the fact that $\bar{V}_0 < 0$ for our case.

2.4 Charge Density At The Interface

From equation 2.31 we can obtain the electric field, which in terms of x has the form

$$E(x) = \frac{2\kappa}{\beta q} \csc(\kappa(x - x_0)), \quad (2.31)$$

where

$$e^{\xi_0} = -\tanh \left(\frac{\bar{V}_0}{4} \right). \quad (2.32)$$

From electrostatics we know that the border condition for the electric field at a conductors interface is

$$E(x)|_{interface} = \frac{\sigma}{\epsilon}. \quad (2.33)$$

We thus obtain the surface charge in terms of the voltage at the interface

$$\sigma = -\frac{2\epsilon\kappa}{\beta q} \csc(kz_0), \quad (2.34)$$

which using Eq. 2.32 can be written as

$$\sigma(V_0) = -\frac{2\epsilon\kappa}{\beta q} \frac{\tanh\left(\frac{q\beta V_0}{4}\right)}{\tanh\left(\frac{q\beta V_0}{4}\right) - 1} \quad (2.35)$$

This equation can be solved for V_0 in terms of the surface charge density σ

$$V_0 = \tanh^{-1} \left(-\frac{8\epsilon\kappa}{(q\beta)^2\sigma} \left(1 \pm \frac{1}{2} \sqrt{1 + \frac{\beta^2 q^2 \sigma^2}{\epsilon^2 \kappa^2}} \right) \right) \quad (2.36)$$

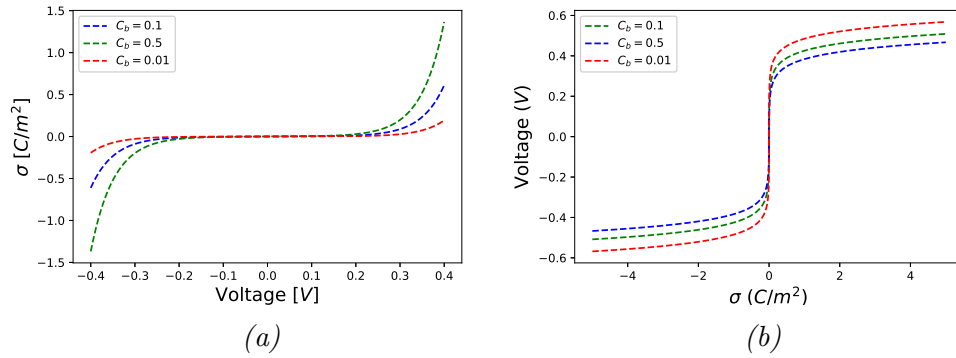


Figure 2.3: **(a)** Surface charge density in terms of the potential border condition. **(b)** Inverse of (a)

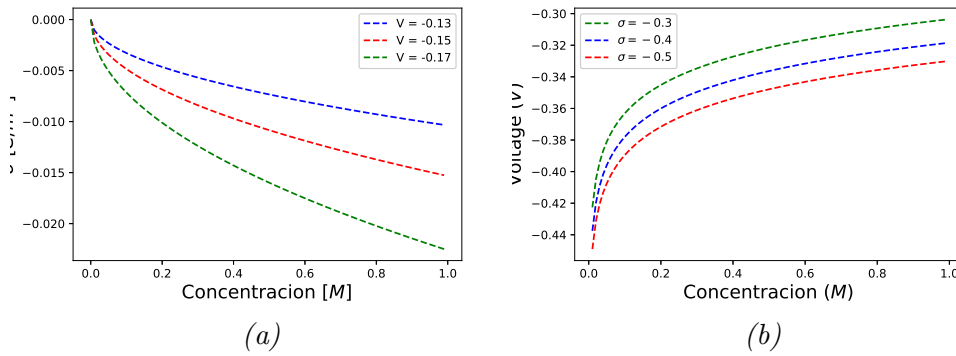


Figure 2.4: **(a)** Surface charge density in terms of the bulk concentration. **(b)** Voltage at the interface in terms of the bulk concentration (a)

2.5 Solution to the concentration at first order in the current I_0

Now we solve Equation 2.1 at first order in the current I_0 . From Eq. 2.11, we consider the term proportional to $\frac{\partial \phi^{(1)}}{\partial x}$ such that

$$\left| \frac{\partial \phi^{(1)}}{\partial x} \right| \ll \frac{\kappa V_0}{r}, \quad (2.37)$$

since the gradient of the correction to the potential should be negligible compared to the gradient of the zero order contribution. This will be shown later in a numerical analysis. With this approximation, the system to first order in r becomes

$$\frac{\partial C_+^{(1)}}{\partial \xi} - C_+^{(1)}(\xi) \frac{\partial \Phi^{(0)}}{\partial \xi} = -\frac{1}{\kappa} \quad (2.38)$$

$$\frac{\partial C_-^{(1)}}{\partial \xi} + C_-^{(1)}(\xi) \frac{\partial \Phi^{(0)}}{\partial \xi} = 0 \quad (2.39)$$

$$\frac{\partial^2 \phi^{(1)}}{\partial \xi^2} = -(C_+^{(1)}(\xi) - C_-^{(1)}(\xi)) \quad (2.40)$$

In Eq. 2.39, we obtain

$$\frac{\partial C_+^{(1)}}{\partial \xi} - C_+^{(1)} \frac{\partial \Phi^{(0)}}{\partial \xi} = -\frac{1}{\kappa}$$

Using an integrating factor of the form

$$\mu(\xi) = e^{-\int_{\kappa\delta}^{\xi} \frac{zF}{RT} \frac{d\phi^{(0)}}{d\xi'} d\xi'} = e^{-(\Phi^{(0)}(\xi) - \phi_b)} = e^{-\Phi^{(0)}(\xi)}$$

We can write 2.39 as

$$\frac{d}{d\xi} \left(C_+^{(1)}(\xi) \mu(\xi) \right) = -\frac{\mu(\xi)}{\kappa},$$

where $z = |z_{\pm}|$. Integrating over x and considering that $C_+^{(1)}(\xi) \mu(\xi) \big|_{\xi \rightarrow \infty} = C_{+,b}^{(1)} = 0$ due to border conditions, we get

$$C_+^{(1)}(\xi) = \frac{1}{\kappa \mu(\xi)} \int_0^{\xi} \mu(\xi') d\xi'$$

Eq. 2.40 can be integrated by separation of variables, and the solution is

$$C_-^{(1)}(\xi) = C_{b,-}^{(1)} e^{\phi^{(0)}(\xi)}.$$

Since our border condition yields $C_{b,-}^{(1)} = 0$, the contribution to first order of the negative ion concentration is zero.

Computing the integral, we find to first order in the current the following solutions for the concentrations.

Analytic concentration to zero and first order in r

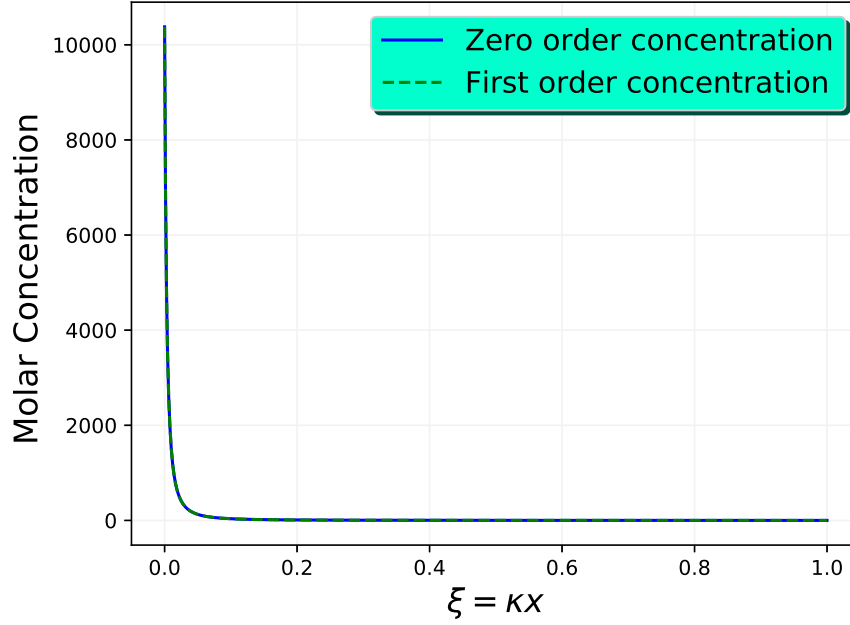


Figure 2.5: (b) Comparison of the equilibrium concentration and the concentration modified by the reaction with the interface

$$\begin{aligned}
 C_+^{(1)}(\xi) &= -\frac{1}{\kappa} e^{-\Phi^{(0)}(\xi)} \left(\xi - 2 \left(\tanh \left(\frac{\xi - \xi_0}{2} \right) + \tanh \left(\frac{\xi_0}{2} \right) \right) \right) \\
 C_-^{(1)}(\xi) &= 0
 \end{aligned} \tag{2.41}$$

In Figure 2.5, we show the concentration for $\bar{V}_0 = -\frac{zF}{RT}0.15V$, with $T = 300K$. (add Reference)

2.6 Potential to first order in the current

Now we need to solve Eq. 2.40

$$\frac{\partial^2 \Phi^{(1)}}{\partial \xi^2} = -(C_+^{(1)}(\xi) - C_-^{(1)}(\xi)). \quad (2.42)$$

Expanding, we have

$$\begin{aligned} \frac{\partial^2 \Phi^{(1)}}{\partial \xi^2} &= -C_+^{(1)}(\xi) \\ &= \frac{1}{\kappa} e^{-\Phi^{(0)}(\xi)} \left(\xi - 2 \left(\tanh \left(\frac{\xi - \xi_0}{2} \right) + \tanh \left(\frac{\xi_0}{2} \right) \right) \right) \end{aligned} \quad (2.43)$$

$$= \frac{1}{\kappa} \coth^2 \left(\frac{\xi - \xi_0}{2} \right) \left(\xi - 2 \left(\tanh \left(\frac{\xi - \xi_0}{2} \right) + \tanh \left(\frac{\xi_0}{2} \right) \right) \right) \quad (2.44)$$

Notice that

$$\int d\xi \coth \left(\frac{\xi - \xi_0}{2} \right) = \xi - 2 \coth \left(\frac{\xi - \xi_0}{2} \right) \quad (2.45)$$

$$\int d\xi \xi \coth^2 \left(\frac{\xi - \xi_0}{2} \right) = \frac{1}{2} \xi^2 - 2\xi \coth \left(\frac{\xi - \xi_0}{2} \right) - 4G(\xi) \quad (2.46)$$

$$\int d\xi \xi \coth \left(\frac{\xi - \xi_0}{2} \right) = 2 \log \left| \sinh \frac{\xi - \xi_0}{2} \right| = 2G(\xi) \quad (2.47)$$

Using integrals 2.47, we obtain

$$\begin{aligned} \Phi'^{(1)}(\xi) &= \frac{1}{\kappa} \left(\frac{1}{2} \xi^2 - 2\gamma \xi + 2(2\gamma - \xi) \coth \left(\frac{\xi - \xi_0}{2} \right) \right) + C \\ C &= -\frac{1}{\kappa} \left(\frac{1}{2} \xi_\delta^2 - 2\gamma \xi_\delta + 2(2\gamma - \xi_l) \coth \left(\frac{\xi_l - \xi_0}{2} \right) \right) \end{aligned} \quad (2.48)$$

Here, we have used the fact that $\lim \Phi'^{(1)}(\xi) = 0$, and we defined $\xi_\delta = \kappa \delta$ and

$$\gamma = \tanh \left(\frac{\xi_0}{2} \right).$$

Integrating again, we obtain

$$\phi^{(1)}(\xi) = \frac{1}{\kappa} \left(\frac{1}{6} \xi^3 - \gamma \xi^2 + C\xi + 8\gamma (G(\xi) - G(0)) - 4\xi G(\xi) + 4I(\xi) \right), \quad (2.49)$$

where

$$I(\xi) = \int_0^\xi d\xi G(\xi). \quad (2.50)$$

This integral must be evaluated numerically for each value of ξ . For this, we use the *integrate* module in Python's scientific library SciPy. The evaluation is made using the quadrature method. Fig. 2.6 shows the potential to first order in the current alongside with the zero order potential.

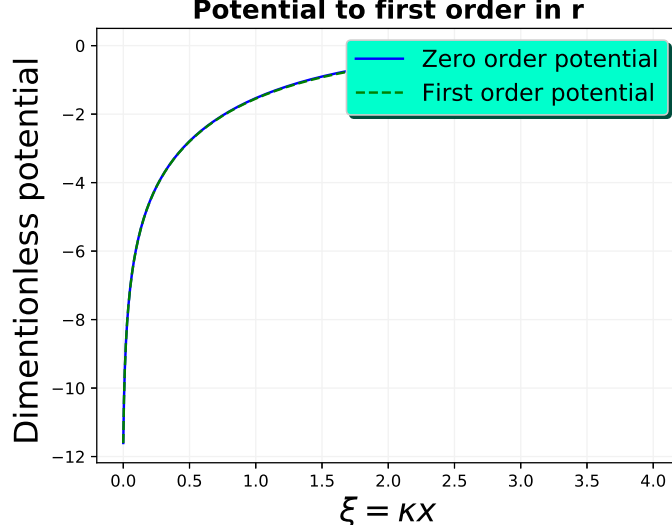


Figure 2.6: First order potential is plotted for a value of $r = k \times 10^{-5}$

2.7 Numerical Analysis

In order to validate the approximations made in previous sections, we solve system 2.1 numerically. In order to do so, we use the Runge-Kutta Method of order 4 along with the Shooting Method to solve the two point boundary value (reference to Numerical Methods in python).

The complexity of the problem lies exactly in the two point boundary feature of the problem, where the concentrations are known in the bulk, but the potential is known at the interface (see Fig. 2.2).

The approach used in these type of problems is transforming the second order system 2.1 into a first order one,

$$C'_+(x) - \frac{zF}{RT} E(x) C_+(x) = -r, \quad (2.51)$$

$$C'_-(x) + \frac{zF}{RT} E(x) C_-(x) = 0, \quad (2.52)$$

$$E'(x) = \frac{zF}{\epsilon} (C_+(x) - C_-(x)), \quad (2.53)$$

$$\phi'(x) = -E(x). \quad (2.54)$$

subject to the border conditions

$$C_+(\delta) = C_b \quad (2.55)$$

$$C_-(\delta) = C_b \quad (2.56)$$

$$\phi(\delta) = 0 \quad (2.57)$$

$$\phi(0) = V_0 \quad (2.58)$$

Here the primes denote derivatives with respect to x . Once the system is in the form of Eqn. 2.58, we can apply the Runge-Kutta Method with the shooting method to obtain the numerical solutions.

2.7.1 Runge-Kutta Method

The Runge-Kutta Method is used to determine the numerical value of a system of differential equations to which we know the boundary conditions. Let

$$\vec{y}' = \vec{F}(x, \vec{y})$$

$$\vec{y}(x_0) = \vec{y}_0$$

be the system of equations we want to solve with the method. We approximate the derivative by

$$\frac{\vec{y}(x+h) - \vec{y}(x)}{h} = \vec{F}(x+h, \vec{y} + \vec{y}(x+h))$$

We want to approximate the left hand side with a series expansion to arbitrary order in h . We obtain thus an expression for $\vec{y}(x+h)$,

$$\vec{y}(x+h) = \vec{y}(x) + h\vec{F}(x+h, \vec{y} + \vec{y}(x+h)).$$

In this work we use the Runge-Kutta of fifth order in the integration step h . The expression for the increment at this order of precision is

$$\vec{y}_5(x+h) = \vec{y}(x) + \sum_{i=0}^6 C_i K_i,$$

and

$$K_0 = hF(x, y),$$

$$K_i = hF(x + A_i h, y + \sum_{j=0}^{i-1} B_{ij} K_j)$$

where $i = 1, 2, 3, \dots, 6$.

We also need the fourth order Runge-Kutta method in order to obtain the error in the integration. This is given by

$$\vec{y}_4(x+h) = \vec{y}(x) + \sum_{i=0}^6 D_i K_i,$$

The error is given by

$$\vec{E}(h) = \vec{y}_5(x+h) - \vec{y}_4(x+h). \quad (2.59)$$

Of this expression, we can take the root mean square of the components of $\vec{E}(h)$,

$$e(h) = \sqrt{\frac{1}{n} \sum_{i=0}^{n-1} E_i^2(h)} \quad (2.60)$$

This way we obtain a scalar measure of the error.

If h_1 and h_2 are two consecutive steps in the integration of the differential equation, then it can be shown that

$$\frac{e(h_1)}{e(h_2)} \approx \left(\frac{h_1}{h_2} \right)^5 \quad (2.61)$$

We don't know what the error of the next step will be, but we can define a tolerance $\epsilon > 0$ such that the error of the next step is at most ϵ . Thus, the following step can be computed from Eq. 2.62 by letting $e(h_2) = \epsilon$

$$h_2 = h_1 \left(\frac{\epsilon}{e(h_1)} \right)^{1/5} \quad (2.62)$$

This method is good for our purposes since the curves we are trying to analyze are flat at the begining but have sharp slopes towards the interface (see Fig. 2.5 and 2.6).

2.7.2 The Shooting Method

The method described in section 2.7.1 can solve the system of equations if the border conditions are all known at a given value of $x = x_0$. Nevertheless, as it can be seen from ??, that we have boundary conditions at $x = 0$ (the interface) and at $x = \delta$ the bulk solution (see Fig. 2.2). This is a complication which can be overcome by using the shooting method. The idea is to transform the problem with boundary conditions

$$\begin{aligned} C_+(\delta) &= C_-(\delta) = C_b \\ \phi(\delta) &= 0 \\ \phi(0) &= V_0 \end{aligned}$$

into the problem

$$\begin{aligned} C_+(\delta) &= C_-(\delta) = C_b \\ \phi(\delta) &= 0 \\ E(\delta) &= u, \end{aligned}$$

where u is a constant yet to be determined. To determine u , we define a residual function

$$r(u) = \phi(0) - V_0.$$

which is zero when the value of $\phi(0)$ (given an appropriate u) is the original boundary condition. Therefore, the two-point boundary problem is transformed into a root finding problem. First, we need to find two values of u , u_1 and u_2 such that $r(u_1) < 0$ and $r(u_2) > 0$. Once we have that, we can use Ridder's method to find the root.

2.7.3 Ridder's Method

In this method it is assumed that the root is bracketed within to values. We define an auxiliary function

$$G(x) = F(x)e^{(x-x_1)Q}$$

where Q is determined by imposing that the points $(x_1, G(x_1))$, $(x_2, G(x_2))$ and $(x_3, G(x_3))$ be in a straight line. That is, if $x_1 - x_2 = 2h$ and $x_1 - x_3 = h = x_3 - x_2$ we should get

$$G(x_3) = \frac{G(x_1) - G(x_2)}{2}. \quad (2.63)$$

Using equation 2.63, we obtain that

$$e^{hQ} = \frac{f(x_1) + f(x_2)e^{2hQ}}{2} \quad (2.64)$$

Which is a quadratic equation on e^{hQ} . Solving for this quantity we obtain

$$e^{hQ} = \frac{f(x_3) \pm \sqrt{f(x_3)^2 - f(x_2)f(x_1)}}{f(x_2)} \quad (2.65)$$

Now that we have determined completely our function $G(x)$, we can do linear interpolation on it to get a better estimation of the root of $f(x)$. Let x_r be it. Then

$$x_r^\pm = x_3 \pm (x_3 - x_1) \frac{f(x_3)}{\sqrt{f^2(x_3) - f(x_1)f(x_2)}}$$

It can be show that

- If $f(x_1) - f(x_2) < 0$, we should pick the x_r^- solution
- If $f(x_1) - f(x_2) > 0$, we should pick the x_r^+ solution

When we have our solution, we make $x_3 = x_r^s$ ($s+$, $-$, for every case) and find a new correction to x_r . We stop when $x_r^+ - x_r^-$ is below the define tolerance.

To test convergence, our method assumes the function we are passing on is of the form

$$f(x) = g(x) - g_0,$$

where we are trying to find x_0 such that $g(x_0) = g_0$. This means that when the root is found, we should get

$$f(x_0) \in [-\epsilon, +\epsilon],$$

where $\epsilon > 0$ is a predefined tolerance.

2.7.4 Results

Figures 2.7 and 2.12 show the results obtained. The tendency looks correct, since the concentration of SO_4^{-2} drops rapidly close to the interface and the concentration of Cu^{+2} explodes at the same location. The potential drops close to the value of $\bar{V}_0 \approx -0.15V$.

Dealing with the equations as presented in Eq. 2.1 is extremely difficult when doing a numerical analysis. This is due to the fact that the natural units of the system (x which is measured in meters) are too small for the computer to handle. Also, non-linearity gives extreme fluctuations of the concentration of the positive ion near the surface. Since we were using the shooting method, we needed to change the values of the electric field at the bulk, such that the boundary condition for the electric potential is correct, but this induced such strong concentrations at the interface sometimes, that the computer could not handle the numbers resulting from Runge-Kutta method. We had to make a little adaptation in order to be able to find the correct boundary condition for the electric field with the shooting method.

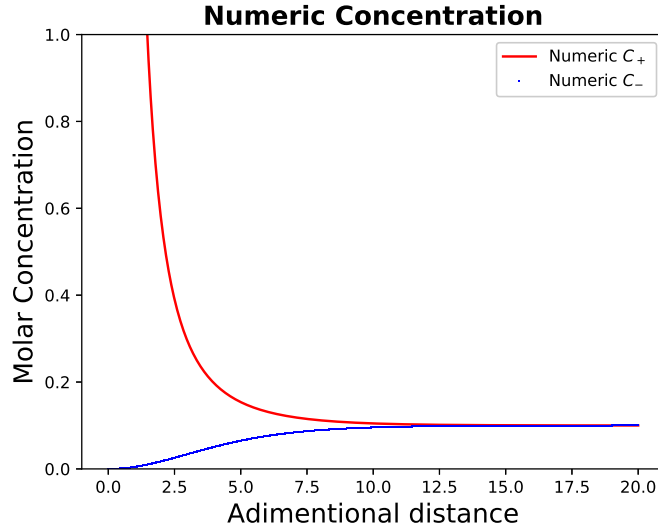


Figure 2.7: This figure shows the concentration found analytically and through numerical analysis.

As it can be seen in figures 2.12, 2.8, the electric field and potential approach zero when the start moving into the bulk of the solution, as expected. The further away from the interface we go, the better the border conditions for the electric field are met. For our particular case, we cut the integration range when we reached a tolerance of $E_{bulk} < 1 \times 10^{-3}$, where E_{bulk} is the border condition of the dimensionless electric field at the bulk.

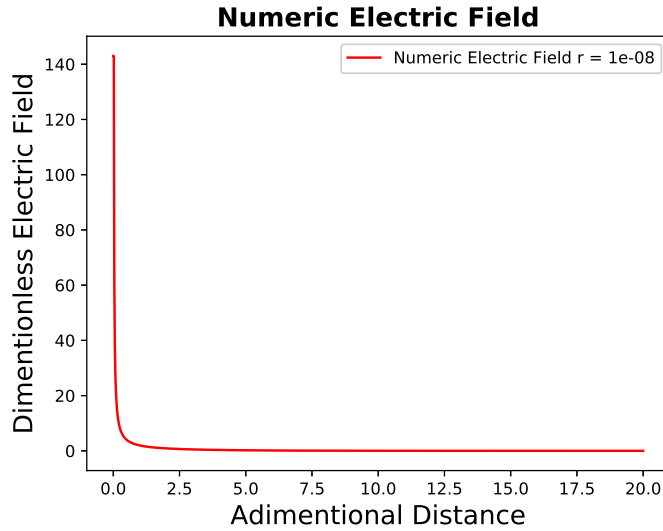


Figure 2.8: The numerical and the analytic electric field found with Runge-Kutta Method.

A similar analysis can be done for the concentrations (Fig. 2.7), but this time the value of both concentrations at the bulk is C_b , as defined by the border conditions for the system 2.1.

Another difficulty is that, since we used the adaptive Runge-Kutta method, the concentrations change so abruptly that the adaptive step h becomes incredibly small. The problem with this is that the number of iterations needed with a step of the order of $h \approx 10^{-29}$ to

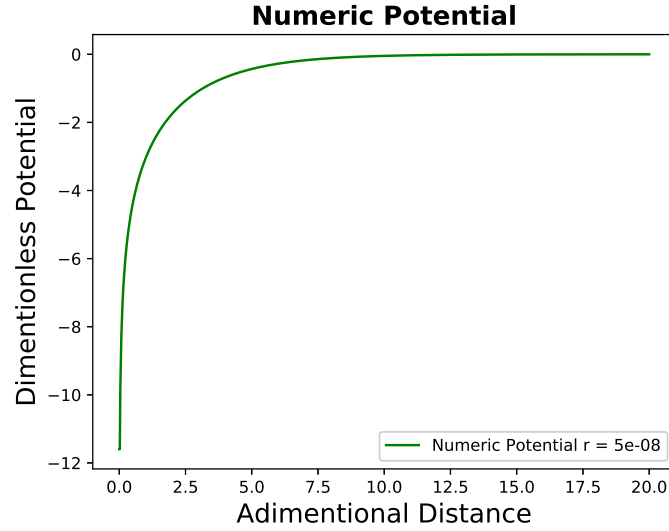


Figure 2.9: The numerical and the analytic solution to the potential to first order in the current.

reach the full length of integration is too big and therefore, we obtain only a part of the integration interval and not as close to the interface as we should like.

To avoid such difficulties, we have worked with the adimensional potential in a scale of adimensional length $\xi = \kappa x$. We have integrated on the interval $[0, 20\kappa\delta]$.

Comparing numeric an analytic results

Fig. 2.12 shows the comparison between the numeric and analytic results for the potential. In figures 2.10 to 2.12 there is a clear overestimation in the concentration of the positive ion, as well as an over estimation of the electric potential. This is due to the fact that we neglected the term proportional to

$$C_s \frac{\partial \phi^{(1)}}{\partial x}.$$

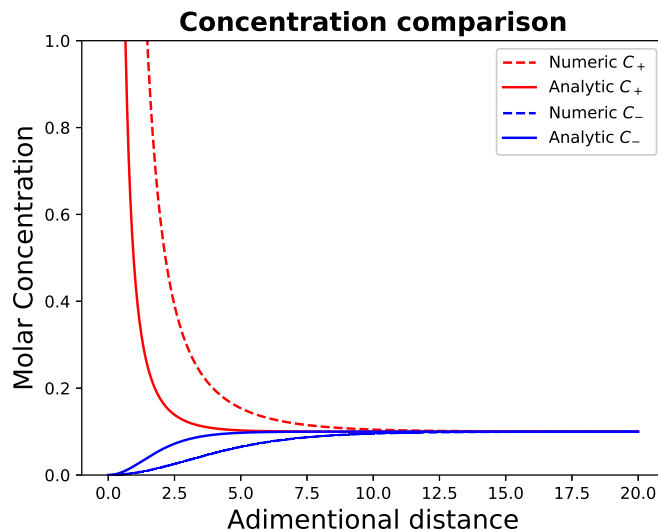


Figure 2.10: Blue curves represent the concentration of the positive ion. Red curves represent the concentration of the negative ion. Dashed lines are numerical results and solid lines represent analytical results.

This is a good approximation when calculating the C_- concentration, since it drops to zero near the interface. Since the C_+ concentration explodes near the interface, the approximation should be valid only to about 8κ , as can be seen from Fig. 2.10. Nevertheless, since $\phi^{(1)}$ is only a small contribution to the potential and by condition ??, since r is small, the approximation might be consider for even closer distances to the interface.

The estimation for the Electric field, however, does look promising (Fig. 2.12) . The analytical and numerical solutions of the model fit together particularly well the closer they get to the interface, even though both results explode to really high values around this point.

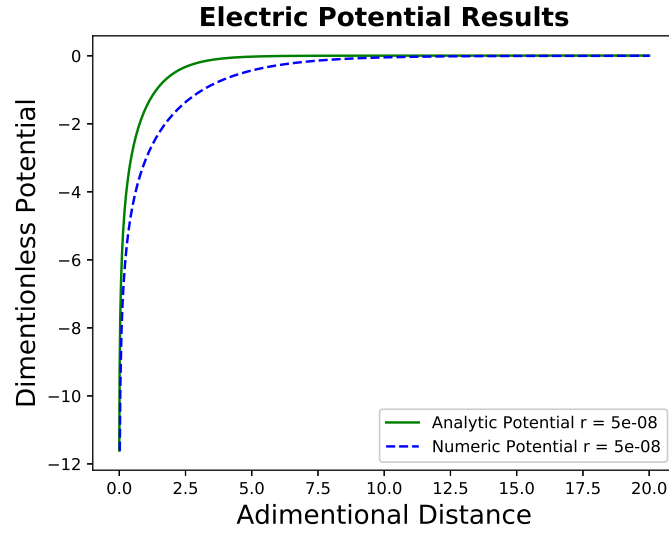


Figure 2.11: Electric potential of the diffusion-reaction system subject to a non zero border condition for the potential. Dashed curve represents the numerical (exact) solution to system 2.1, whereas the solid curve is the analytical solution obtained on section ???. Screening due to positive ions gathering close to the interface is appreciated.

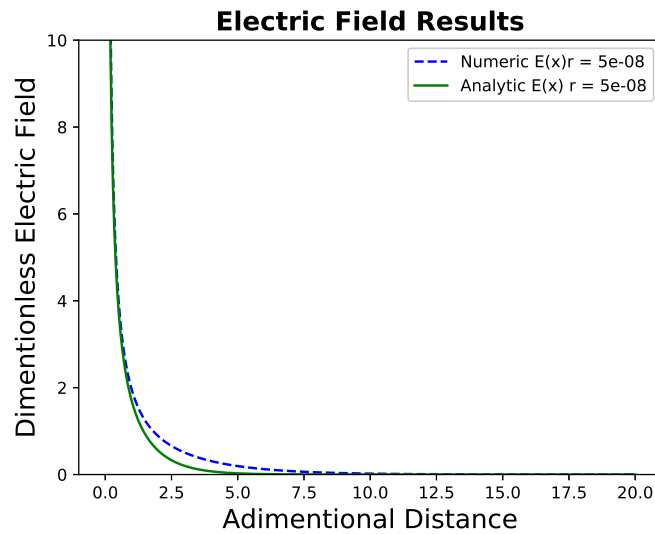


Figure 2.12: The electric field is plotted comparing analytical (solid line) to numeric (dashed line) results.

Chapter 3

Dynamic Solution

3.1 Analytic Solution

Consider now the dynamical model in which time dependence is not neglected

$$\frac{\partial C_+}{\partial t} = D_+ \left(\frac{\partial^2 C_p}{\partial x^2} - \frac{\partial}{\partial x} \left(C_+ \frac{\partial \Psi}{\partial x} \right) \right) \quad (3.1)$$

$$\frac{\partial C_+}{\partial t} = D_+ \left(\frac{\partial^2 C_p}{\partial x^2} - \frac{\partial}{\partial x} \left(C_+ \frac{\partial \Psi}{\partial x} \right) \right) \quad (3.2)$$

$$\nabla^2 \Psi = -(C_+ - C_-) \quad (3.3)$$

By making the following change of variables

$$\xi = \frac{\kappa x}{\sqrt{4Dt}} \quad (3.4)$$

$$\frac{\partial \xi}{\partial x} = \frac{\kappa}{\sqrt{4Dt}} \quad (3.5)$$

$$\frac{\partial \xi}{\partial x} = -\frac{\kappa x}{2\sqrt{4Dt^3}} \quad (3.6)$$

we can re-write the system of equations as

$$\frac{\partial^2 C_+}{\partial \xi^2} + \frac{2\xi}{\kappa} \frac{\partial C_p}{\partial \xi} - \frac{\partial C_+}{\partial \xi} \frac{\partial \Psi}{\partial x} - C_+ \frac{\partial^2 \Psi}{\partial \xi^2} \quad (3.7)$$

$$\frac{\partial^2 C_+}{\partial \xi^2} + \frac{2\xi}{\kappa} \frac{\partial C_p}{\partial \xi} + \frac{\partial C_+}{\partial \xi} \frac{\partial \Psi}{\partial \xi} + C_+ \frac{\partial^2 \Psi}{\partial \xi^2} \quad (3.8)$$

$$\nabla^2 \Psi + 4Dt(C_+ - C_-) = 0 \quad (3.9)$$

In order to solve the previous system, we will integrate equations 3.7 and 3.7. Lets write the two equations as one in terms of the charge of the electrolyte $s = \pm$.

$$\frac{\partial^2 C_s}{\partial \xi^2} + \frac{2\xi}{\kappa} \frac{\partial C_s}{\partial \xi} - s \frac{\partial C_s}{\partial \xi} \frac{\partial \Psi}{\partial \xi} - s C_s \frac{\partial^2 \Psi}{\partial \xi^2} \quad (3.10)$$

Since,

$$\int_0^\xi \frac{\partial C_s}{\partial \xi} = \xi C_s(\xi) - \int_0^\xi C_s(\xi) d\xi \quad (3.11)$$

$$\int_0^\xi \frac{\partial C_s}{\partial \xi} \frac{\partial \Psi}{\partial \xi} = \frac{\partial \Psi}{\partial \xi} C_s - \int_0^\xi C_s(\xi) \frac{\partial^2 \Psi}{\partial \xi^2} d\xi \quad (3.12)$$

We get,

$$\frac{\partial C_s}{\partial \xi} + \frac{2\xi}{\kappa} - \frac{2}{\kappa} \int_0^\xi C_s(\xi) - s \frac{\partial C_s}{\partial \xi} \frac{\partial \Psi}{\partial \xi} = 0 \quad (3.13)$$

Dividing the equation by C_s , which should be always possible, we get

$$\frac{1}{C_s} \frac{dC_s}{d\xi} = \frac{-2}{\kappa} \xi + \frac{2C_s}{\kappa} \int_0^\xi C_s(\xi) d\xi + s \frac{\partial \Psi}{\partial \xi} \quad (3.14)$$

Integrating we get

$$\log \left(\frac{C_s}{C_s(0)} \right) = -\frac{\xi^2}{\kappa} + \Psi + \frac{2}{\kappa} \int_0^\xi C_s(z) \int_0^z C_s(z') dz' dz \quad (3.15)$$

From the Poisson Equation 3.9 we get

$$C_s(\xi) = C_{-s} - \frac{s}{4Dt} \frac{\partial^2 \Psi}{\partial \xi^2}$$

We make the approximation that $C_+ \ll C_-$ near the interface. Therefore

$$\int_0^\xi C_s(z) \int_0^z C_s(z') dz' dz \approx \int_0^\xi dz \frac{1}{(4Dt)^2} \Psi' \Psi'' = \frac{1}{(4Dt)^2} \left(\frac{\partial \Psi}{\partial \xi} \right)^2 \quad (3.16)$$

Therefore, the concentration for each species in terms of the potential and the electric field is

$$C_s = C_b e^{-\frac{1}{\kappa} \left(\xi^2 - \frac{E^2}{8D^2 t^2} \right)} e^{s\Psi} \quad (3.17)$$

3.2 Potential

Now we solve for the electric potential. The Poisson equation in terms of the previously found concentration profiles is

$$\frac{\partial^2 \Psi}{\partial \xi^2} = -2C_b e^{\frac{1}{2}\xi^2 - \frac{E^2}{8D^2 t^2}} \sinh \Psi \quad (3.18)$$

This can be written as

$$e^{-\frac{E^2}{8D^2 t^2 \kappa}} \Psi'' = -2C_b e^{\frac{\xi^2}{\kappa}} \sinh \Psi \quad (3.19)$$

Integrating over ξ we get

$$\frac{\sqrt{\pi}}{2} \operatorname{erf}(\sqrt{\alpha}\Psi') \left(-\frac{4Cb}{\sqrt{\pi}} \int_0^\xi e^{-\frac{s^2}{\kappa}} \sinh \Psi(s) ds \right) \quad (3.20)$$

$$\Psi' = \frac{1}{\sqrt{\alpha}} \operatorname{erf}^{-1} \left(-\frac{4Cb}{\sqrt{\pi}} \int_0^\xi e^{-\frac{s^2}{\kappa}} \sinh \Psi(s) ds \right) \quad (3.21)$$

Integrating and using the properties of the erf^{-1} function we get

$$\Psi' = \frac{1}{\sqrt{\pi\alpha}} \operatorname{erf}^{-1} \left(-\frac{4Cb}{\sqrt{\pi}} \int_0^\xi e^{-\frac{s^2}{\kappa}} \sinh \Psi(s) ds \right) \quad (3.22)$$

Bibliography