INDIAN INSTITUTE OF TECHNOLOGY KHARAGPUR



Sub. Name: MATERIALS ENGINEERING (Sub. No. MT 30001)

MID Autumn Semester, 2012

Time: 2 Hrs.

Full Marks: 30

3rd Year B. Tech. Students of ME, MF, QE and QM

No. of Students: 176

Dept.: Met. & Mater. Engg.

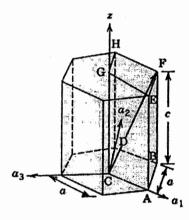
<u>Instructions:</u> Answer <u>ALL</u> the questions. Total number of questions: 4. Time your answers according to the marks allotted.

Where necessary, use schematic diagrams to illustrate your answer. Total number of pages: 3.

1.

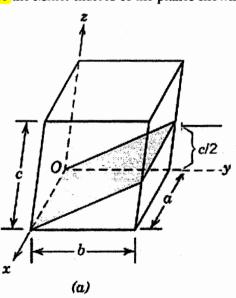
Q1 Total: 8

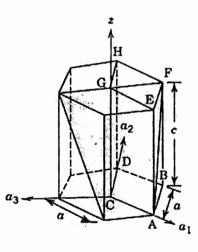
- (a) Which of the following is not a Bravais lattice?
 - b) Base-centered Monoclinic
 - a) Body-centered Orthorhombic,c) Face-centered Tetragonal
- d) Base-centered Orthorhombic
- (b) In four index system what is the crystallographic direction of CF in the following figure?



(c) What are the Miller indices of the planes shown in the figures?

2





	pla	anar defect.	1
(e) What is the relation between equilibrium number fraction of vacancy and temperature?			
	Ca	an there be a crystal with no vacant site at zero degree celcius?	1+1
(f)	~W	hat are the Frenkel and Schottkey defects in ceramic crystal structure?	1
	-		Q2 Total: 8
	(a) ³	What are the usual slip systems of FCC and HCP crystals? What are the	
		slip systems in each of these cases? Which one of these crystals wou	
		ductile? For which of the common metallic crystal structures, the twinning	
	(h)		+1+0.5+0.5
	(D)	Under an applied shear stress, in which directions a screw and an edge will move: parallel or, perpendicular or at an angle to the direction of applie	
		will move, parallel or, perpendicular of at all aligie to the direction of applie	0.5+0.5
	(c)	A single crystal of aluminum is oriented for a tensile test such that its slip p	
	(~)	makes an angle of 28.1° with the tensile axis. Three possible slip direct	ctions make
		angles of 62.4 ⁰ , 72.0 ⁰ , and 81.1 ⁰ with the same tensile axis.	
		(i) Which of these three slip directions is most favored?	1
		(ii) If plastic deformation begins at a tensile stress of 1.95 MPa (280 psi)), determine
		the critical resolved shear stress for aluminum.	1
	(-1\	Management of the state of the second of the	****************
	(a)	If you are given a block of single crystal of a pure metal, how will you strength?	increase its
	(م)	What are the characteristic differences between recovery and recrystallize	ation2 What
	(0)	is the driving force for grain growth?	0.5+0.5
		to the cirring to too is grain grain y	
			O2 Total: 0
(2)	Nic	ckel (Ni) and Copper (Cu) form isomorphous phase diagram. Melting point	Q3 Total: 8
(a)		52 deg C and that of pure Cu is 1083 deg C. Draw a schematic pha	
		dicating the different phase regions.	1
(b)	W	hich of the following statement is true about "Recrystallization":	1
		a. Driving force for it is "Reduction in surface energy"	
		b. Recrystallization temperature is not a function of amount of prior defo	rmation
		c. Driving force for it is "Reduction in volume free energy"	
		d. Recrystallization is associated with increasing the total "grain boundar	ry energy
(c)	Fo	or a 99.65 wt% Fe-0.35 wt% C alloy at a temperature just below the	e eutectoid
(0)		termine the following:	o catootolo,
		The fractions of total ferrite (alpha) and cementite phases.	.2
		The fractions of the proeutectoid ferrite (alpha) and pearlite.	2 2 1
		The fraction of eutectoid ferrite (alpha).	1
	(d)	The degrees of freedom	1
		the entertaid temperature the whole comments are found to be a	U USS 11401
		the eutectoid temperature, the phase compositions are: ferrite (alpha): stenite: 0.76 wt% (eutectoid composition), cementite: 6.7 wt%.	U.UZZ WI%,
		earlite is the eutectoid product (an aggregate of eutectoid ferrite (alpha) and	cementite\
		and is the succious product (an aggregate of succious forms (alpha) and	Jonnonato).

(d) Give examples for each of the following in a crystalline solid: a) Point defect b) Area/

2.

3.

(a) Derive the critical nucleus size for homogeneous nucleation during solidification of a 4. (b) Plot the nucleation rate vs. Temperature for homogeneous nucleation during

(c) Plot the variation in activation energy barrier with degree of undercooling for both homogeneous and heterogeneous nucleation in the same schematic diagram.

(d) Why for pure metal, solidification happens at a particular constant temperature?