

4-5-

0-1

1-2

2-3

isentropic process

$$m_{m} = \frac{V_{1}}{V_{1}} = \frac{V_{2}}{V_{2}} = \frac{V_{3}}{V_{3}} = \frac{V_{4}}{V_{4}} = \frac{V_{7}}{V_{7}}$$

$$P_{4}V_{4} = P_{7}V_{7} = P_{3}V_{3}$$

$$T_{4}P_{4}^{1-7} = T_{7}P_{7}^{2} = T_{3}P_{3}^{3}$$

$$T_{4}V_{4} = T_{7}V_{7}^{2} = T_{3}V_{3}^{3}$$

$$P_{4}V_{4} = T_{7}V_{7}^{2} = T_{3}V_{3}$$

$$P_{5}V_{7}^{2} = T_{7}V_{7}^{2} = T_{7}V_{7}^{2$$

+ matha = (mentona) h

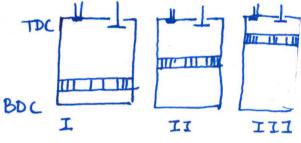
Men
$$T_7 + M_a T_a = \frac{m_a T_1}{m}$$

$$X_7 = \frac{m_{ex}}{m_m},$$

$$x_r = \frac{m_{ex}}{m_m},$$

$$x_r T_7 + (1-x_r)T_a = T_1$$

Vy fictitions volume 47 isentropic expansion process substituting 4-5



$$\begin{array}{ccc}
m_{I} > m_{II} > m_{III} \\
V_{I} > V_{III} > V_{III}$$
with h $v_{I} = \frac{m_{I}}{V_{I}}$
 m_{IM} v_{IM}

VIII U7

Cylinder conditions at the start of compression in an SI engine operating at WOT on an air-standard Otto cycle are 60°C and 98 kPa. The engine has a compression ratio of 9.5:1 and uses gasoline with AF = 15.5. Combustion efficiency is 96% and it can be arruned that there is no exhaust residual Calculate:

(a) temperature at all states in the cycle. [°C]

(6) Premue at all states in the cycle. [Kla]

(e) Specific work done during power stroke [KJ/hg]

(d) Heat added during combustion. [KT/hg]

(e) Net specific work done. [kJ/hg]

(6) Indicated thermal efficiency. [7.]

Take 8=1.35, R= 0.287 RJ/hg-K, Box 44 no kJ/hg.

$$T_1 = 60^{\circ} \text{C} = 60 + 273 = 333 \text{ R}$$
 $P_1 = 98 \text{ kPa}$
 $P_2 = 98 \text{ kPa}$
 $P_3 = P_2 U_2^{\gamma} \Rightarrow P_4 = \left(\frac{U_1}{U_2}\right) = (9.5) = 20.9$
 $P_4 = 98 \times 20.9 = 2048 \text{ kPa}$
 $P_5 = 98 \times 20.9 = 2048 \text{ kPa}$
 $P_7 = P_7 U_2^{\gamma-1} = P_7 U_2^{\gamma-1} \Rightarrow P_7 = \left(\frac{U_1}{U_2}\right) = 9.5 = 9.5 = 9.5$

 $T_1 U_1^{x-1} = \overline{z} U_2^{x-1} \Rightarrow \overline{T_1} = \left(\frac{U_1}{U_2}\right)^{x-1} = 9.5 = 2.2$

Box
$$T_c = \left(\frac{m_a + m_a}{m_f}\right) C_V \left(T_3 - T_2\right)$$

$$\frac{R = 0.287 \text{ kJ/hgK}}{RU = R} = \frac{R}{0.287} = \frac{44000 \times 0.96 = 16.5 \times 0.82 \times (T_5 - T_6)}{3122 = T_3 - T_6}$$

$$T_3 = 3854.6 \text{ K}$$

$$\frac{P_3}{P_2} = \frac{T_3}{T_2}$$
 =) $P_3 = 2048 \times \frac{3854.6}{732.6}$
= 10775.6 Ma

$$P_{3}U_{3}^{\gamma} = P_{4}U_{4}^{\gamma}$$

$$\frac{P_{4}}{P_{3}} = \left(\frac{U_{3}}{U_{4}}\right)^{\gamma} = \left(\frac{1}{r_{c}}\right)^{\gamma} = \left(\frac{1}{9.5}\right) = 0.048$$

$$P_{4} = 517 \text{ kPh}$$

$$T_{3}U_{3}^{\gamma} = T_{4}U_{4}^{\gamma-1}$$

$$\frac{T_{4}}{T_{3}} = \left(\frac{U_{3}}{U_{4}}\right)^{\gamma-1} = \left(\frac{1}{9.5}\right) = 0.455$$

$$T_{4} = 1754 \text{ K}$$

$$T_2 = 732.66 - 273 = 49459.6°C$$
 $P_2 = 2048$ kPa
 $T_3 = 3854.6 - 273 = 3581.6°C$ $P_3 = 10775.6$ kPa
 $T_4 = 1754. - 273 = 1481°C$ $P_4 = 517$ kPa

 $\frac{q_{\text{im}}}{\sqrt{3}} = \frac{5q}{8} - 8\omega = du$ $\frac{3^{1}}{4} = c_{V} (T_{3} - T_{4}) = 0.82 (3854.6 - 1754) = 1722.5 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{3} - T_{2}) = 0.82 (3854.6 - 732.6) = 2560 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{3} - T_{2}) = 0.82 (3854.6 - 732.6) = 2560 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^{1}}{2^{1}} = c_{V} (T_{2} - T_{1}) = 0.82 (732.6 - 333) = 327.7 \text{ kJ/kg}$ $\frac{3^$

The engine is a three-liter V6 engine operating at 2400 RPM. At this speed, the mechanical efficiency is 84%. Calculate: (a) Bruke power. [RW] (6) Torque [N-m]

(c) Brake mean effective premue [RPa]

(d) Friction power lost . [RW]

(e) Brake specific frel consumption [gm/kwh]

(b) Volumetric efficiency [%] (g) Output displacement [RW/L]