Lecture 16 Problems Problem 1

(i): 1/2 H2(9) + 1/2 M (12(9) -> HEI (9)

(11): Att = - 92.31 KJ/mol

(111):AS, = 10.03 KJ/not 10.03 J/(K mol)

(1111): AG? = -95,3 KJ/mol

Problem 2 Nosony above 257K.

Problem3 69.5J/(K mol)

Problem