

LC 18 : Corps purs et mélanges binaires

Niveau : CPGE

Prérequis :

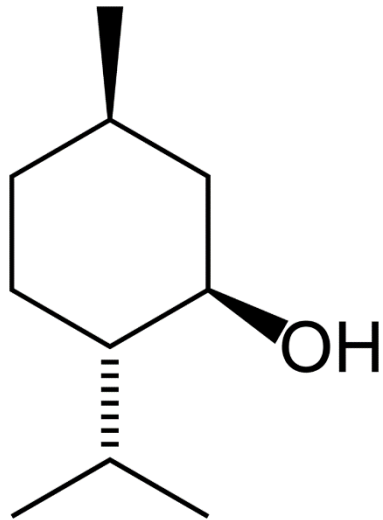
- Fraction massique
- Potentiel chimique
- Changement d'état du corps pur

Mélange de solides

Menthol

1,1 g

$T_{\text{fus}} = 42,45^{\circ}\text{C}$



Acide laurique

0,6 g

$T_{\text{fus}} = 44,33^{\circ}\text{C}$

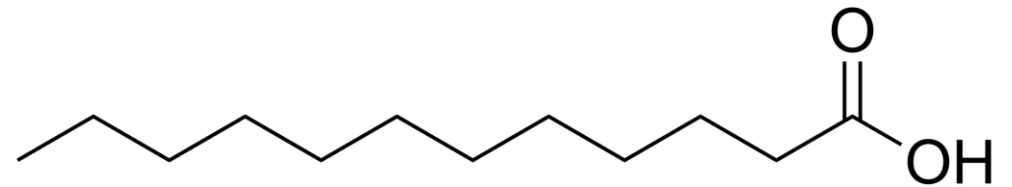


Diagramme de phase d'un corps pur

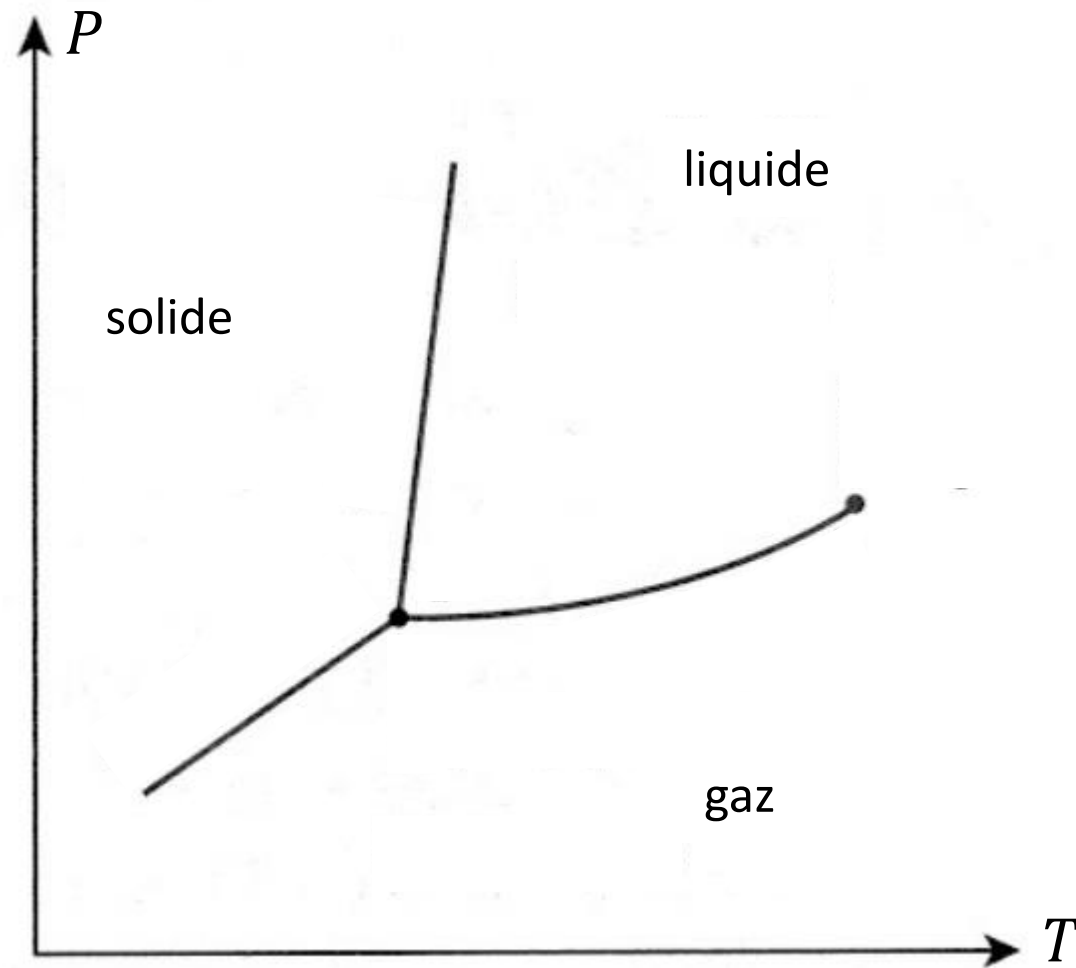
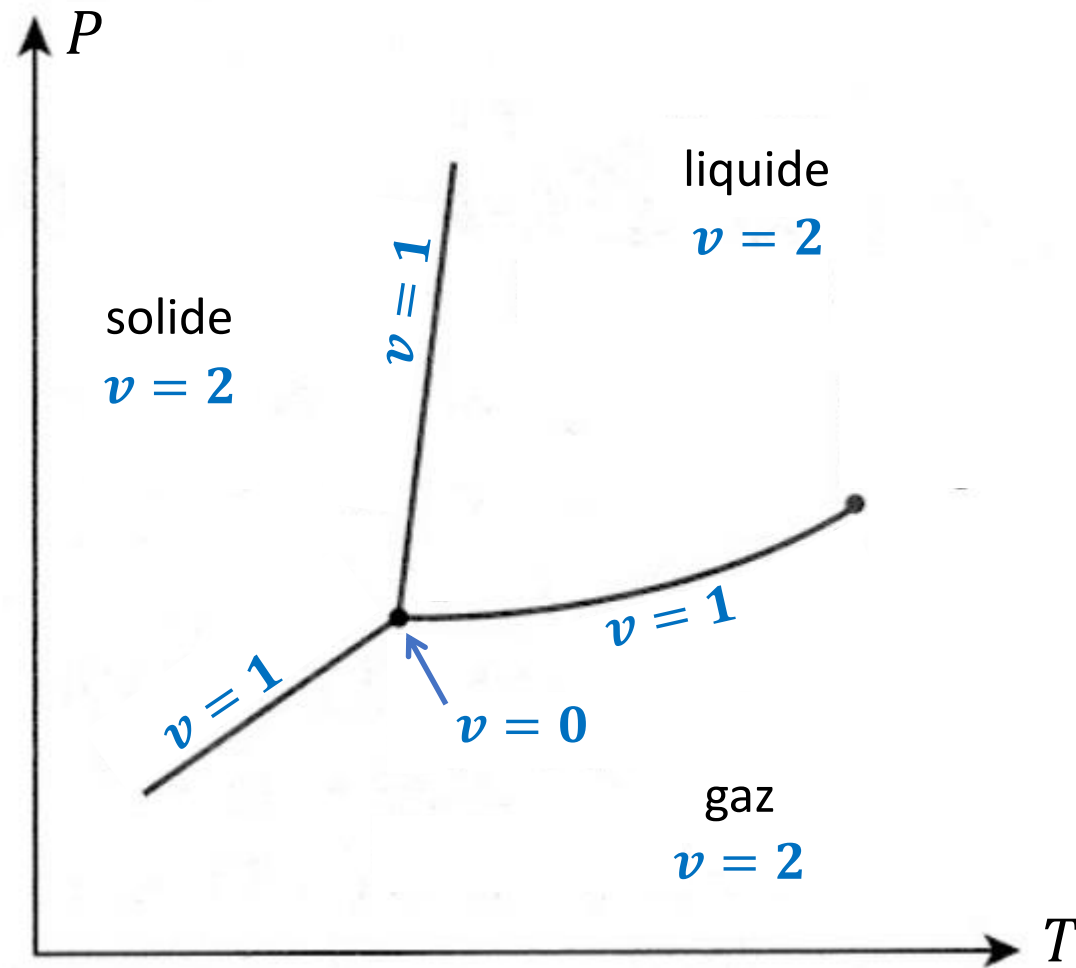
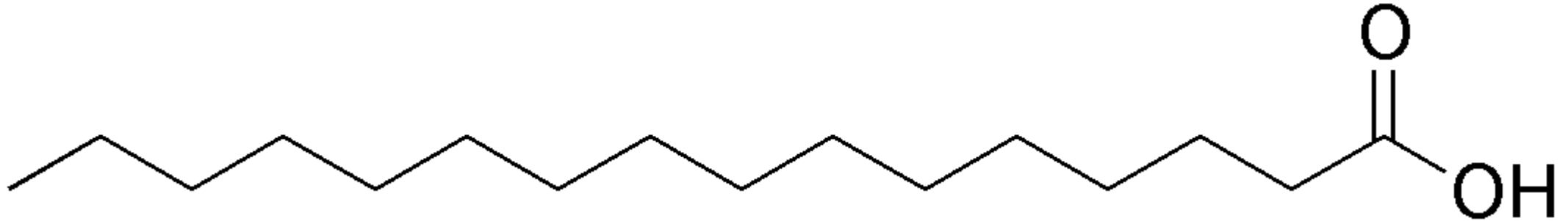


Diagramme de phase d'un corps pur



Acide palmitique



$T_{\text{fus}} = 63,69^{\circ}\text{C}$

Variables de composition

$$w_1^l = \frac{m_1^l}{m_1^l + m_2^l}$$

Fraction massique du constituant 1
dans la phase liquide

On définit de même w_2^l , w_1^s et w_2^s

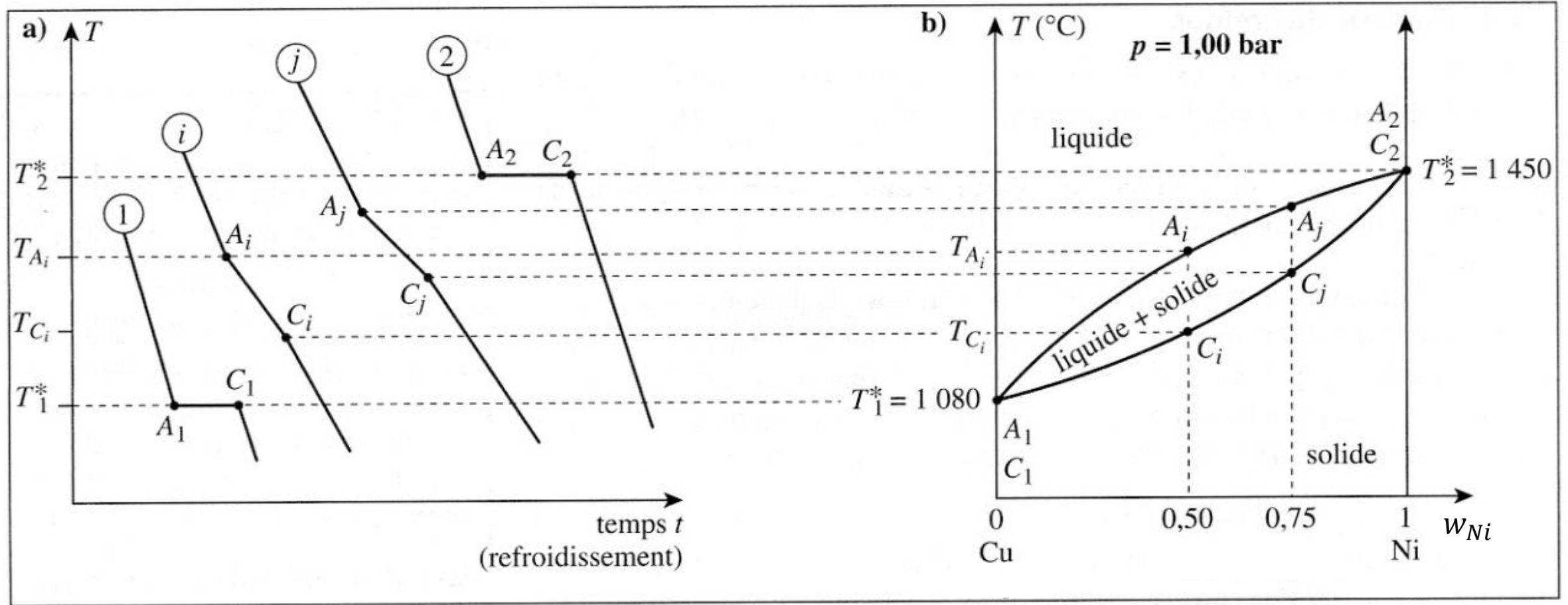
Variables de composition

$$w_1 = \frac{m_1}{m_1 + m_2}$$

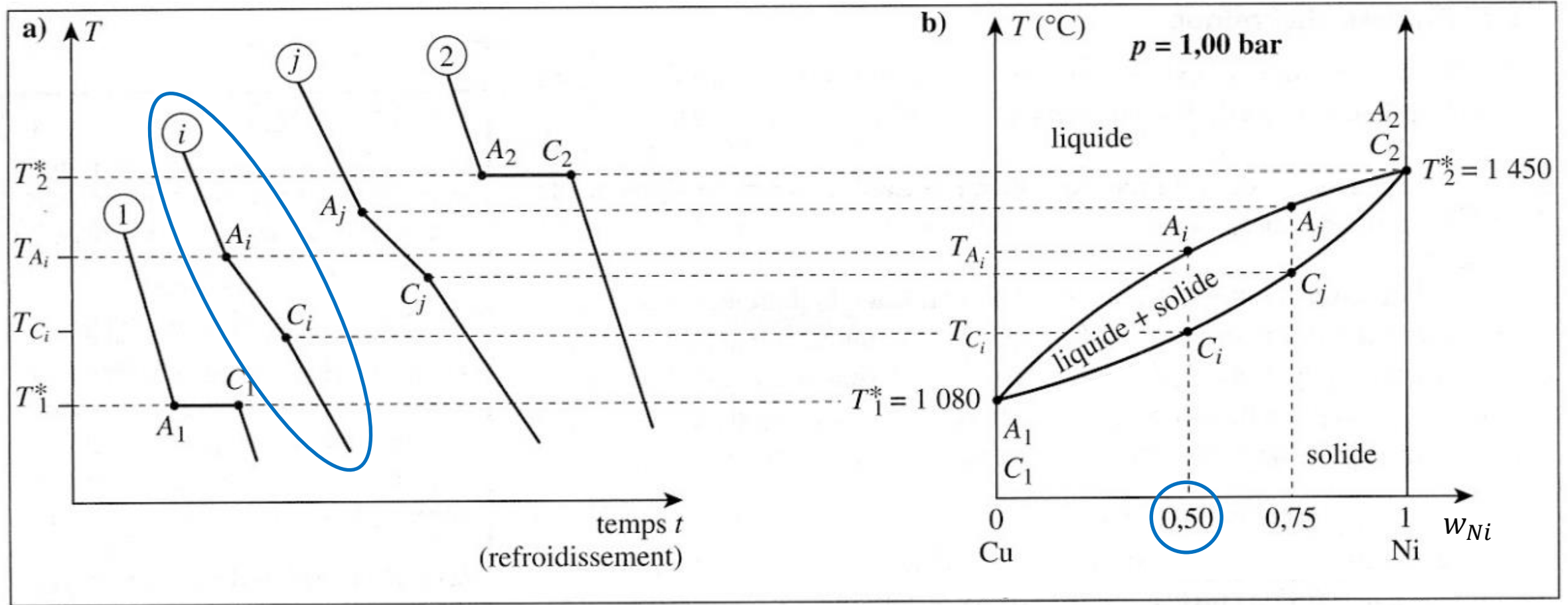
Fraction massique **globale**
du constituant 1

On définit de même w_2

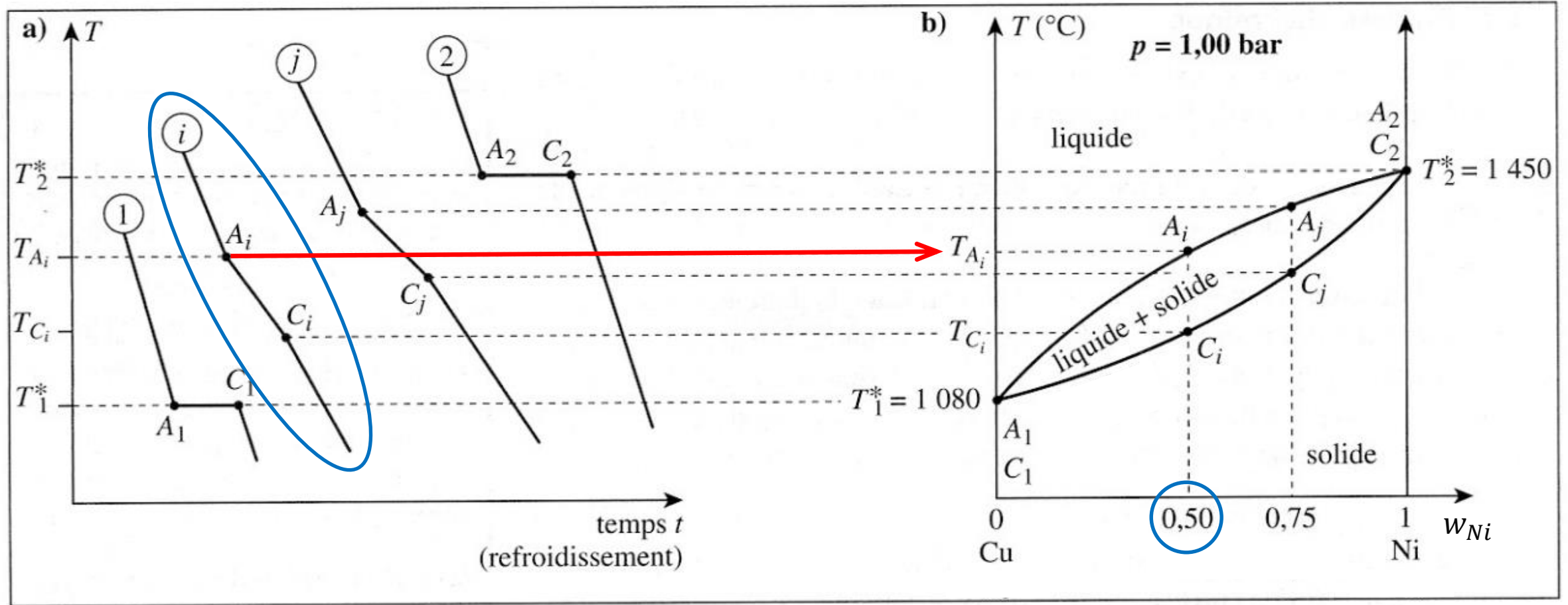
Construction du diagramme binaire Cuivre - Nickel



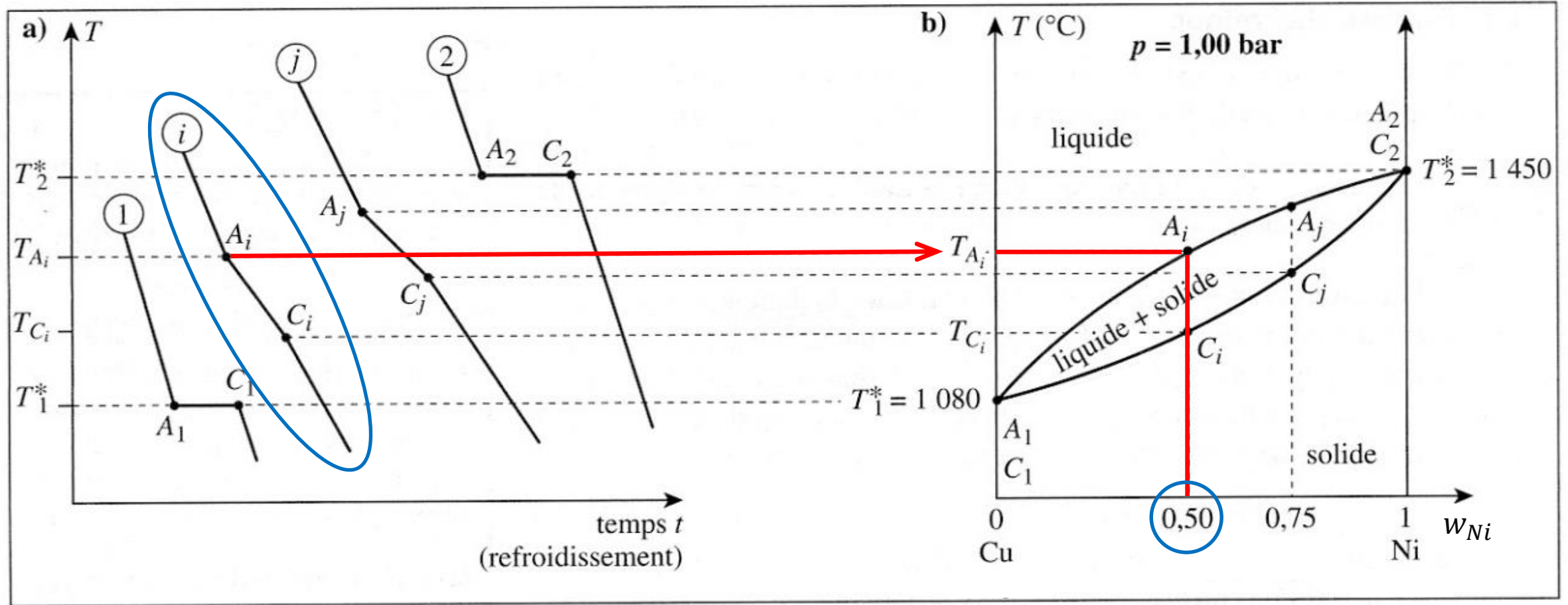
Construction du diagramme binaire Cuivre - Nickel



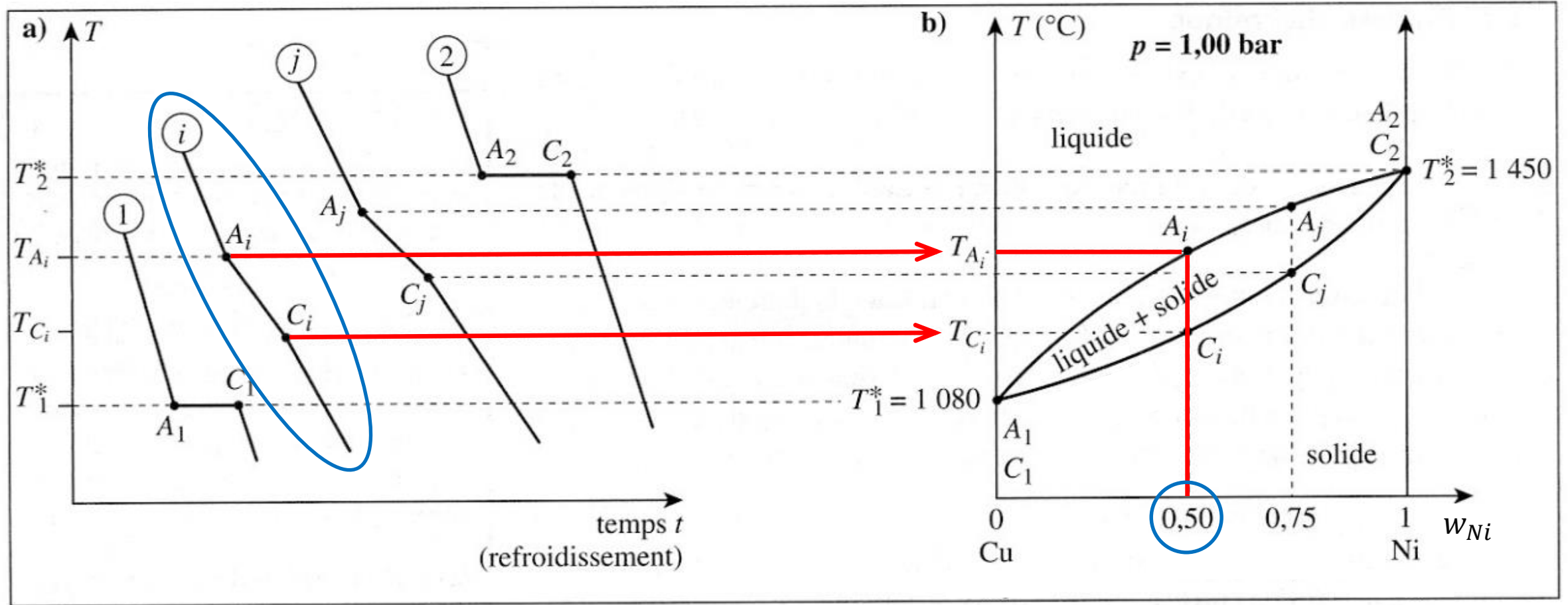
Construction du diagramme binaire Cuivre - Nickel



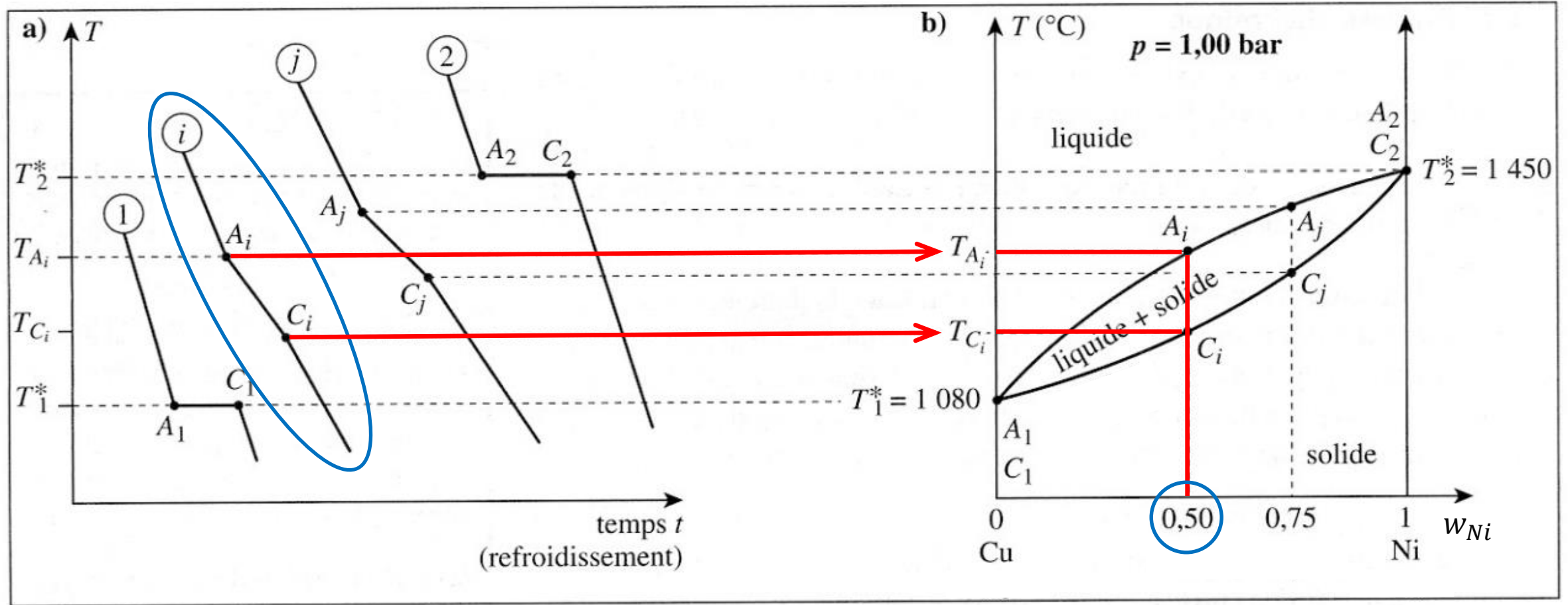
Construction du diagramme binaire Cuivre - Nickel



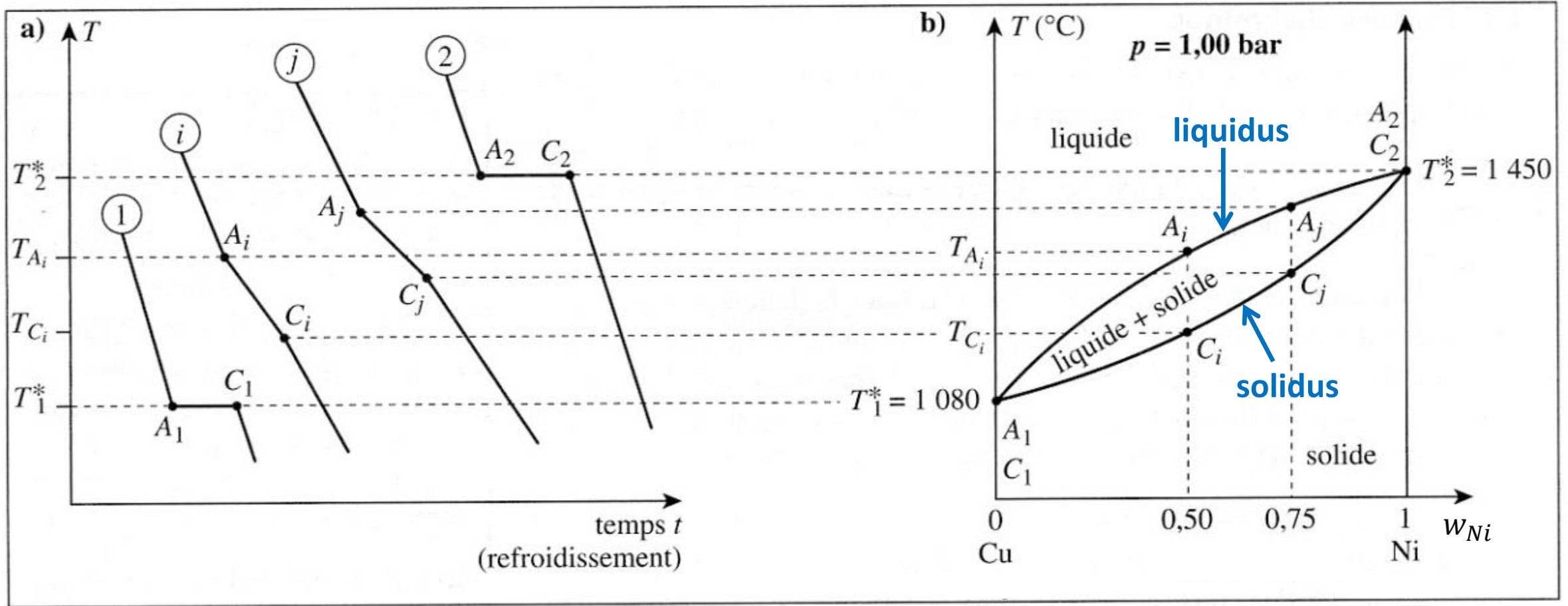
Construction du diagramme binaire Cuivre - Nickel



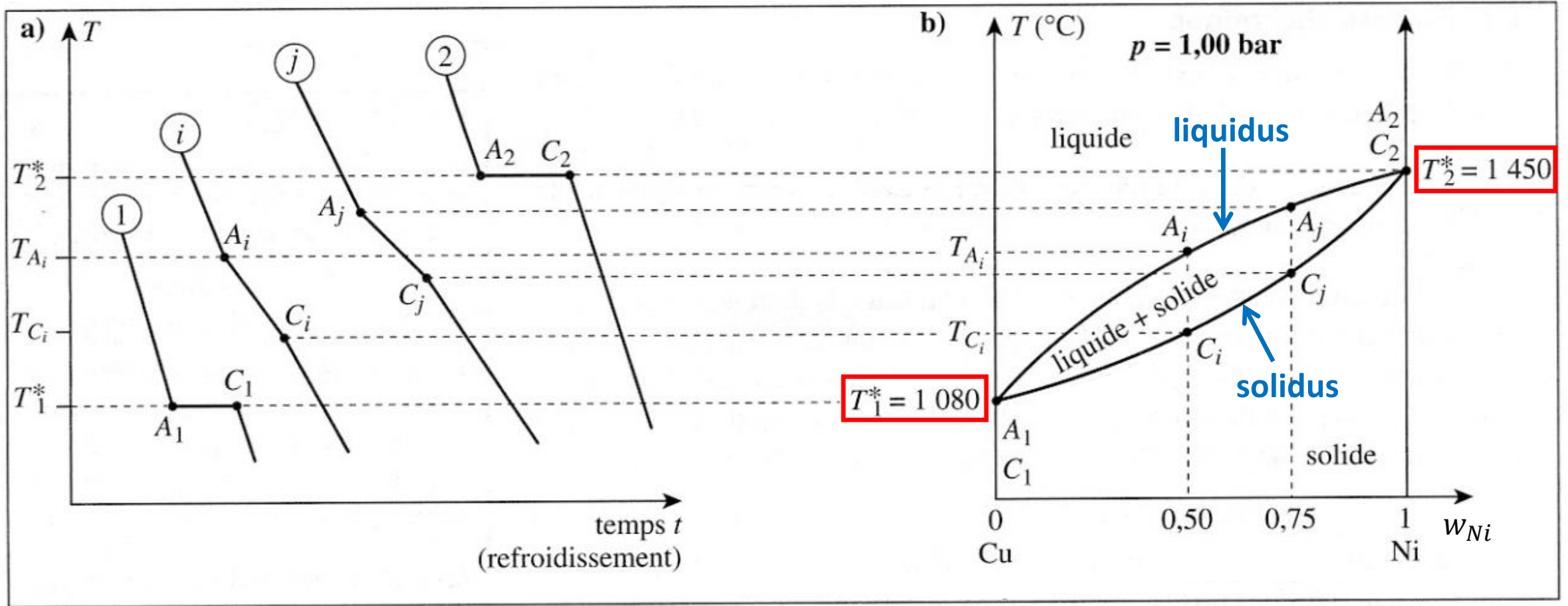
Construction du diagramme binaire Cuivre - Nickel



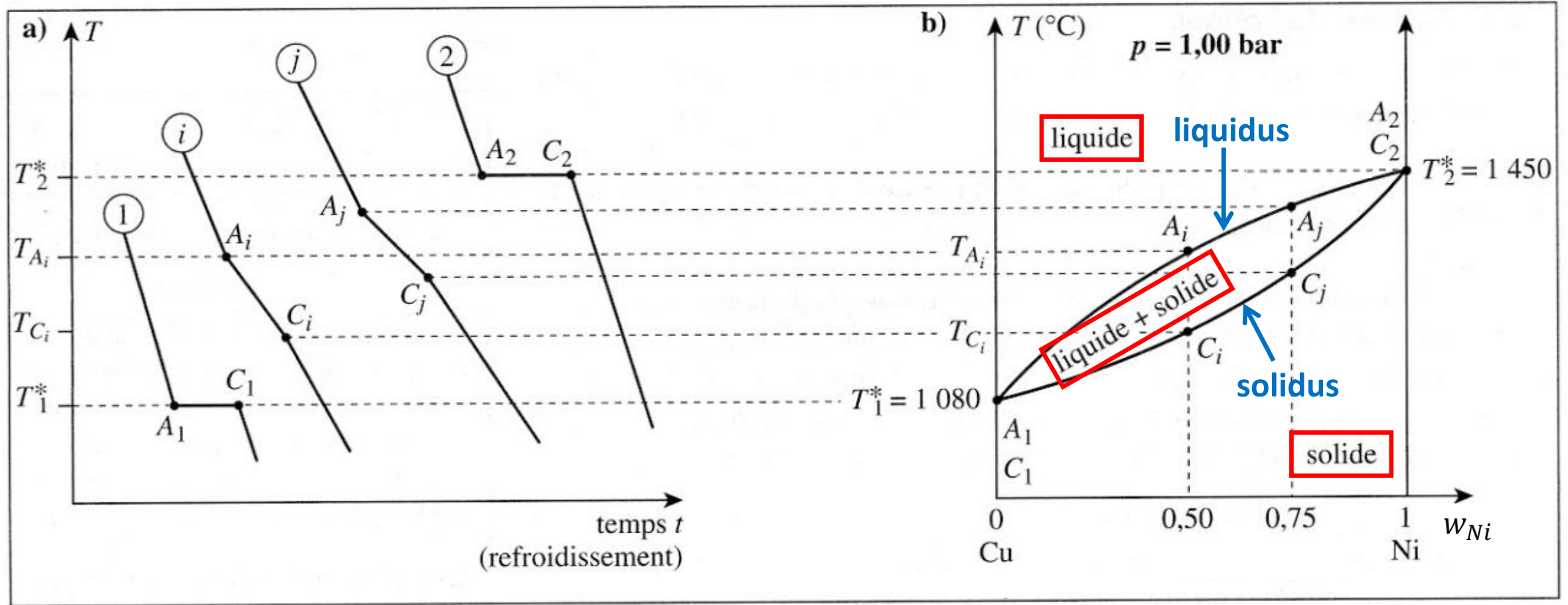
Construction du diagramme binaire Cuivre - Nickel



Construction du diagramme binaire Cuivre - Nickel

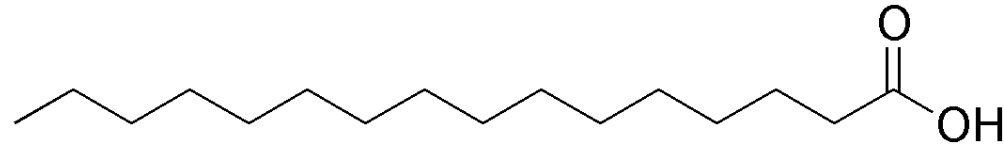


Construction du diagramme binaire Cuivre - Nickel



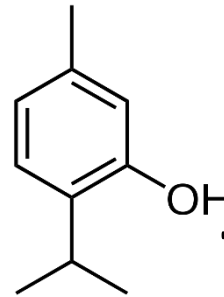
Expérience : diagramme binaire acide sébacique – acide adipique

Acide palmitique



$T_{\text{fus}} = 63,69^{\circ}\text{C}$

Thymol

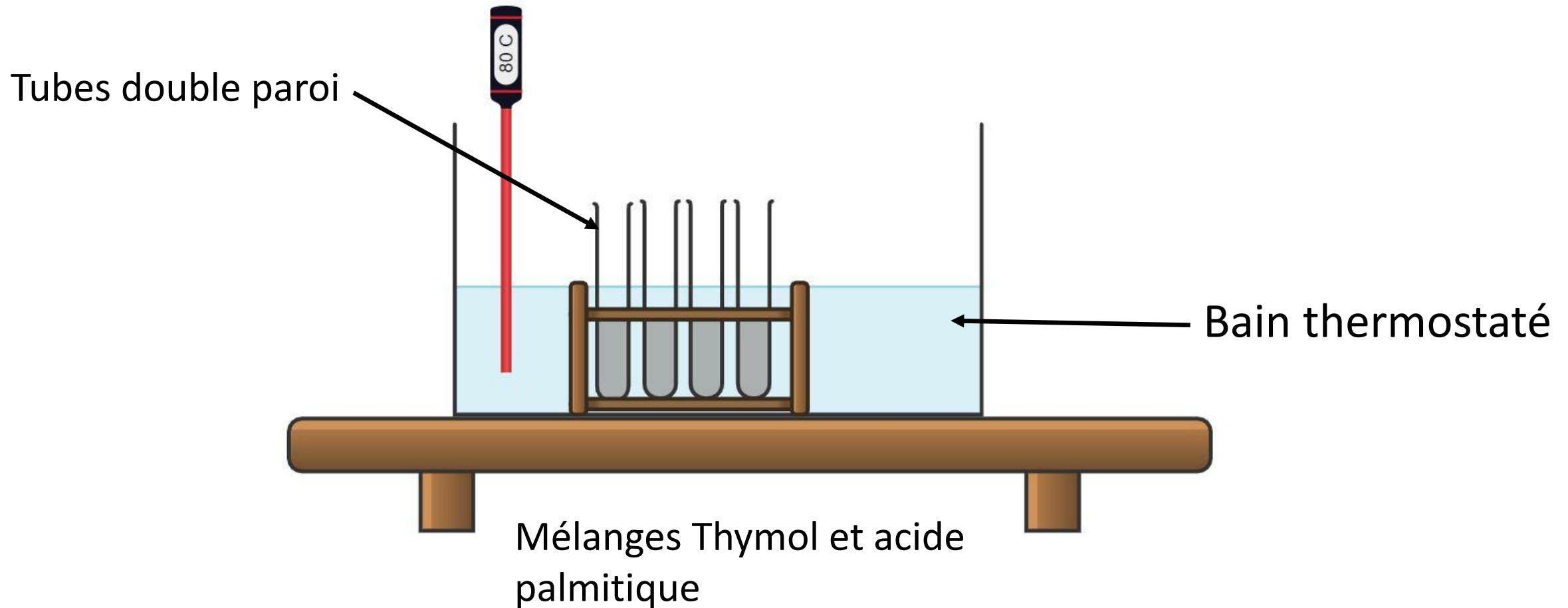


$T_{\text{fus}} = 49,95^{\circ}\text{C}$

Mélanges réalisés

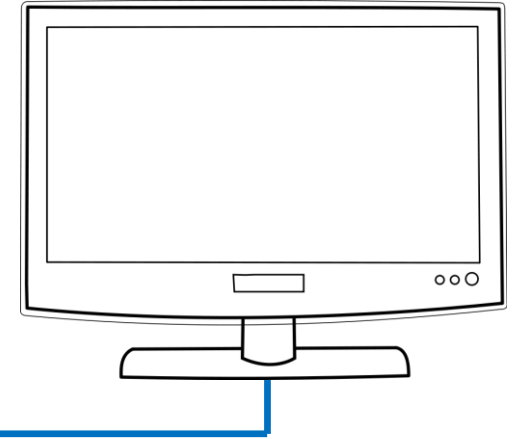
Fraction molaire d'acide palmitique	0	0,2	0,24	0,5				1
-------------------------------------	---	-----	------	-----	--	--	--	---

Tracé de courbes d'analyse thermique : protocole

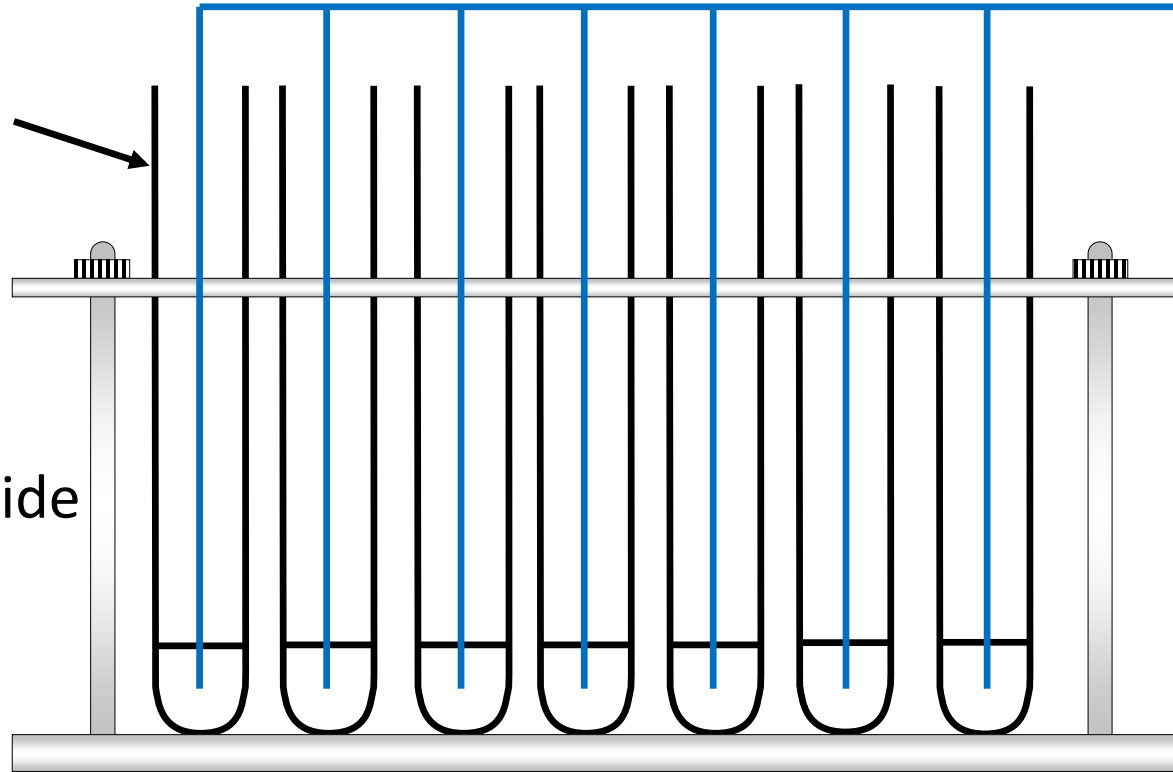


Acquisition de courbes de refroidissement

thermocouples

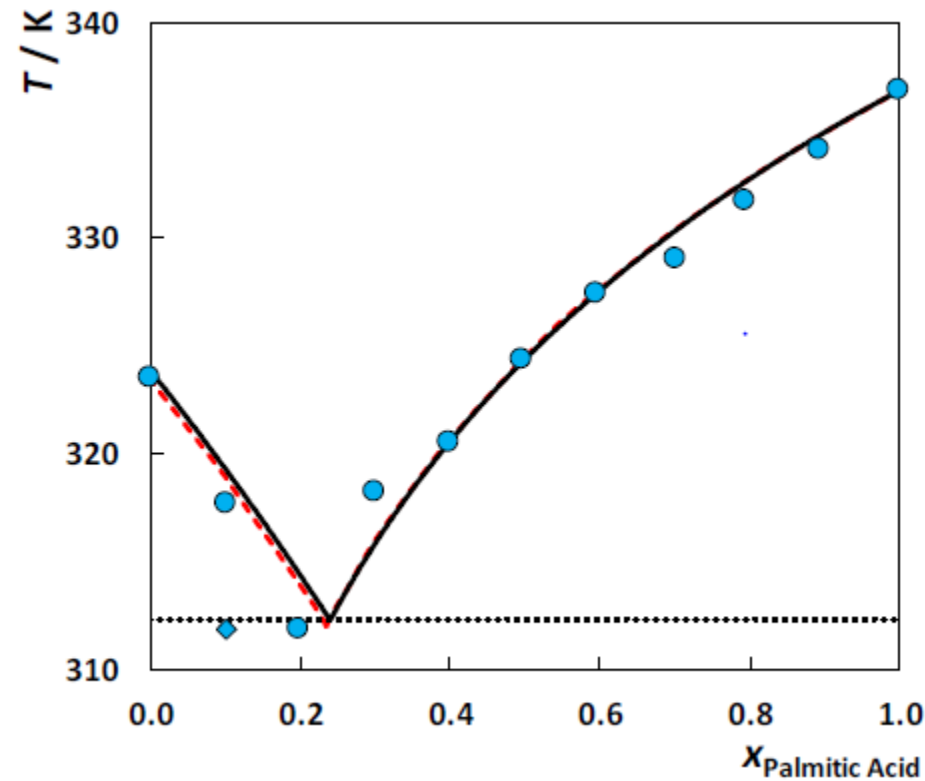


Tubes double paroi



Mélanges Thymol et acide
palmitique

Diagramme binaire isobare Thymol-Acide palmitique

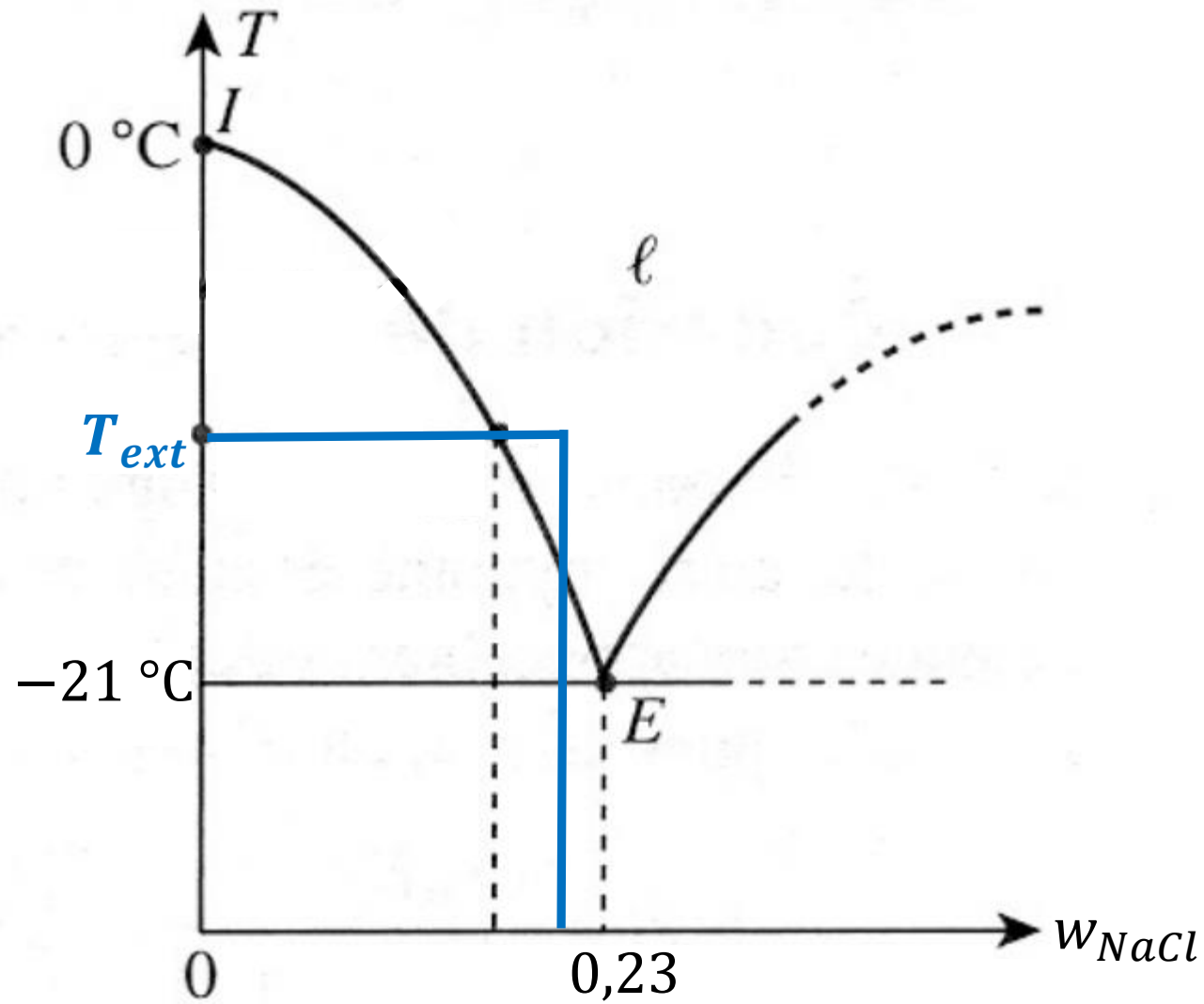


Tunable hydrophobic eutectic solvents based on terpenes and monocarboxylic acids, Mónia Andreia Rodrigues Martins et al. *ACS Sustainable Chem. Eng*

Salage des routes en hiver



Diagramme partiel eau – chlorure de sodium



Application des diagrammes binaires à eutectique



Minerai de bauxite,
contient de l'alumine Al_2O_3



Aluminium

Mélanges	Liquide à partir d'une température de
Alumine pure	2050°C
Alumine + cryolithe Na_3AlF_6 (80% environ)	960°C