Tribhuvan University

Institute of Science and Technology

4 Years Bachelor in Computer Science and Information Technology

Entrance Examination

2071

Mathematics

d) 1

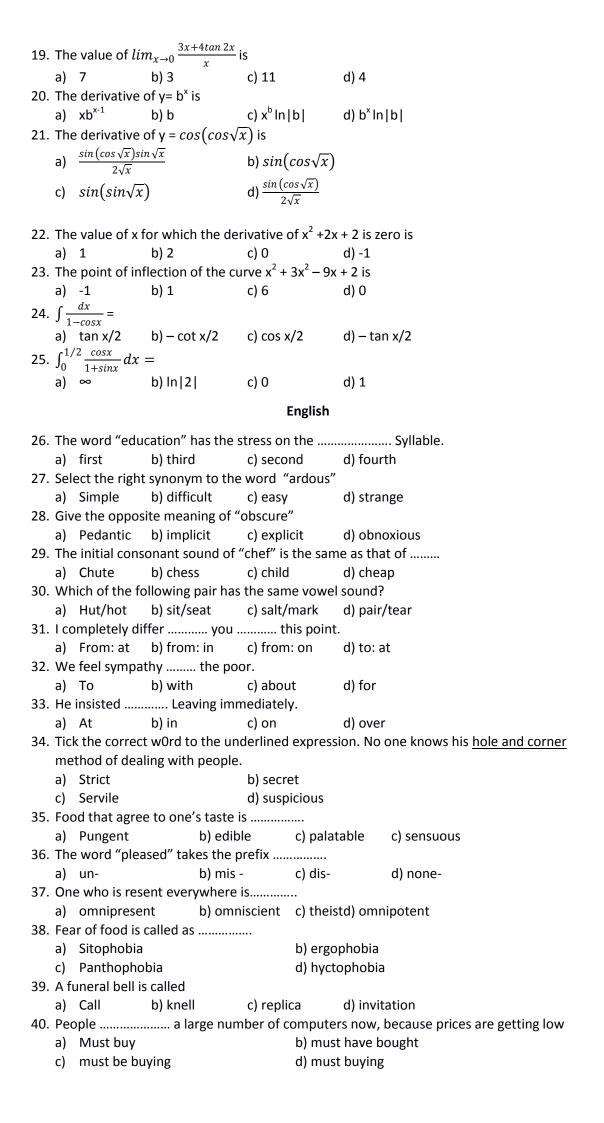
c) 7

1. If (X-3) = 4 then X is a) 7 and -1 b) 7

2. If $\{X-X^2=9 \text{ and } 2X=6\}$

b) 7 and 1

	(
	a) Null set b) {3}		d) {4,3}					
3.	If: R \rightarrow R defined by f(X)=X ² + 3	B then f ⁻¹ (4) is						
	a) Null set b) {-1,1}	c) {1,0}	d) {4,3}					
4.	The value of $\log_2 \sqrt{2} + \log_3 \sqrt{3}$	$\overline{3} + log_{\Lambda}\sqrt{4}$ is						
			d) $\pi/6$					
5.	If $\cos^2 \theta = 1/2$, then θ is	-,, -	2,, 2					
٥.	a) $n\pi \pm \pi/4$ b) $n\pi + \pi/4$	c) nπ + π/3	d) 2nπ					
6	Cos ($\sin^{-1}3/5$) =	c) 11/1 ± 11/3	a) zinc					
0.		c) 0/2E	d) 4/2E					
7	a) $3/5$ b) $4/5$	C) 9/25	u) 4/23					
/.	If a^2 , b^2 , c^2 are in H.P. then a) $2 a^2 c^2 = b^2 (a^2 + c^2)$	L \ -2-2 L-2 /-	22\					
	a) $2ac = b(a + c)$	b) a c = b (a	$\begin{array}{cccccccccccccccccccccccccccccccccccc$					
	c) $2b^2c^2 = a^2(b^2 + c^2)$	d) c a = 2 b	(a + c)					
8.	If $A = \begin{bmatrix} 1 & 2 \\ 2 & 3 \end{bmatrix}$ and $A^2 - nA - I =$							
	a) 0 b) 4	c) 1	d) -4					
	If $A = \begin{bmatrix} 1 & 0 & 1 \\ 0 & 1 & 1 \\ 1 & 0 & 0 \end{bmatrix}$, then A is							
9.	If $A = \begin{bmatrix} 0 & 1 & 1 \end{bmatrix}$, then A is							
		1.3.1						
	a) Symmetric	b) skew-symr						
	c) Singular	d) Non singul						
10	The system of equations 2x +		3 is					
	a) Consistent and dependen							
	b) Inconsistent and independent							
	c) Inconsistent and depende							
	d) Consistent and independe	ent						
11	. The value of i ⁷ + 1/i ⁴ is							
	a) 1 b)-1	c) —i	d) –i +1					
12	. The value of $\frac{x-wy+w^2z}{y+wz+w^2x}$, w is t	he cube root of	unity, is					
	a) w b) $x + y + z$		d) 1					
13	If the roots of $x^2 + 4x + k = 0$, a		is					
		c) -4						
14	The two lines $a_1x + b_1y + c_1 = 0$	•	•					
	The two lines $2x^2 - 2xy + ky^2 =$							
13	a) 2 b) -2	c) 0	d) 1					
16	,	•	ve x- axis and y- axis with radius 2 units is					
10	a) $x^2 + y^2 - 4x - 4y + 4 = 0$	ching the positiv	re x- axis and y- axis with radius 2 dilits is					
	b) $x^2 + y^2 + 4x + 4y + 4 = 0$							
	c) $x^2 + y^2 - 4x + 4y + 4 = 0$							
	d) $x^2 + y^2 - 4x - 4y - 4 = 0$		2 2 2 2 3 4 2 3 4 2 4 3					
17	If the line $4x + 3y = k$ is tanger							
	a) 25 b) -25	c) ±25	d) ±11					
18	. The value of $\lim_{x\to 2} \frac{x^5-2^5}{x-2}$ is							
	a) 80 b) 40 ²	c) 10	d) 0					
	-							



41. Although they are similar, o	hipmunks are	most other ground squirrels.
a) Alike b) like as	c) like to	d) like
42. She the whole day	long.	
a) Has been working	b) wo	rks
c) had worked	d) wo	
43. Now a days Shyam is	•	
a) Very b) to	c) so	d) extremely
44. She demandedhe		N
- · · · · · · · · · · · · · · · · · · ·	c) to know	,
45. The mail comes at ten O'clo		
	and c) also	o d) but
46. In my office, salary is paid be a) The b) a	-	d) some
a) The b) a	a c) an	d) some
47. Every boy and every girl	very respons	ihle
a) Are b) i		
48. Which of the following is ac	•	a, nave
a) Some of rice is still left		
b) Time and tide wait for i	none	
c) She absents herself fro	m the college today	/
d) None of the squares wa	as properly drawn	
49. Which of the following is a	complex sentence?	
a) However, hard she wor	ked, she failed in t	he exams.
b) It rained heavily, howe	ver he came to sch	ool
c) In spite of heavy rain, h		
d) Neither she nor he will		
50. Which of the following is an	•	
a) Sympathize b)	lighten c) per	ception d) moral
	Dh	
51. The dimensions of (h/e) are		ysics
a) Magnetic field	b) electric fiel	d
c) magnetic flux		
52. The number of significant f	<u>-</u>	
a) 5 b) 4	c) 11	d) 2
•	•	ictions between two bodies. Then
a) $\mu_s \geq \mu_k$	b) $\mu_s < \mu_k$	
b) $\mu_s = 0$ and $\mu_k > 0$		ı _k < 0
	body, the change in	n angular momentum () of the body is
a) Perpendicular to $ec{L}$	b) parallel to	$ec{L}$
c) always zero	d) $180^{\circ} \vec{L}$	
55. The Poisson's ratio of mate	•	
a) 0.10 b) 0.20	c) 0.40	d) 0.65
56. The depth of water at whic	h air-bubble of rad	ius 0.4 mm may remain in equilibrium (surface
tension of water = 72×10^{-5}	3 N/m, g= 9.8 m/s 2	is given by
a) 7.348 cm b) 0.918 cm	n c) 3.674 cm	d) 1.837 cm
57. When a cylinder is heated,	its length increases	by 2%, the area of its base will increase by:
a) 0.5% b) 1%	c) 2%	d) 4%
58. In a pressure cooker food is	s cooked usually at	a pressure
a) One atmosphere	b) half atmos	
•	d) 50 atmospl	
59. The mean free path of a ga		
a) T ⁴ b) 1/T ⁴	c) 1/T	d) T
60. The efficiency of Carnot en	gine working betwe	een steam point and ice point is:

	•	16.8 %	b) 26.81%	•	d) 46.8%				
61.	. The bending of sound waves through the corners of the obstacle is called a) Refraction b) interference c) beatsd) diffraction								
62	When a tuning fork is vibrated its prongs have a phase difference of:								
02.	a)		b) π/4	c) π	d) 2π				
	a)	U	0) 10/4	C) II	a) zit				
63.	The	e image of a	n object placed a	at the focus of co	onvex mirror is at a distance				
	a)	∞	b) f	c) 2f	d) f/2				
64.	If s	ky is seen fro	om moon's surfa	ace, it will appea	r				
	a) Black b) blue c) whited) red								
65.	65. A double convex air bubble in a glass sheet would behave as a a) Convergent lens b) concave mirror								
		Convergent		ror					
	-	d) plane mirror							
66.	66. When two prisms are combined it gives a) Reflection only								
	•	lateral shift							
		-	and deviation b	otn					
c 7	•	•	or deviation						
67.				used in an electr	on microscope e nature				
		Negative ch	iarge	c) sma					
68	c)	•	ee charges may	•	n. If they are placed				
08.		On a straig		be in equilibriui	ii. Ii tiley are placed				
	-	_	of equilateral tr	riangle					
			orners of a recta						
	-		ners of a cube						
69.	•			The capacitance	e is				
		6400 MF		c) 1 F	d) 711 μF				
70.	0. Two wires of same material have lengths L and 2L and cross reaction areas 4A and A								
	respectively. The ratio of their specific resistances would be								
	a)	1:1	b) 1:8	c) 8:1	d) 1:2				
71.	The	e magnetic f	ield inside a long	g solenoid is					
	•	Zero	b)	c) μ ₀ ni	d) μni				
72. Electrons in the atoms are held due to									
	•	Nuclear for		b) Coulomb for					
	•	Van der Wa		c) Gravitationa	I forces				
/3.			f H atom lies in v	_	Develope				
	-	Lyman		b) Bracket and	Pascnen				
74	,	Balmer	a substance is 1	c) Paschen	y years. The mean life is				
74.	a)		a substance is 1	b) 3200 years	ilean life is				
	•	800 years		d) 2309 years					
75.	-	lberg consta	int is	a, 2505 years					
	•	Same for al							
	,		or different elem	ients					
	c)								
	d)	Is different	for lighter elem	ents but same fo	or heavier elements				
				Chemistry	<i>I</i>				
76	Нα	w many orh	itals are nossible		uantum number 3?				
70.	a)	-	b) 4	c) 6	d) 9				
77.	,		•	present in 18g o	•				
		18	b) 1	c) 1.8	d) 180				
			,		•				

•	3. The half-life of an radioisotope is 30 minutes. After how much time the amount of radioisotope left is 1/4th of the original amount.							
a) 4 minutes	b) 30 minutes							
c) 60 minutes	d) 120 minutes							
79. What is oxidation of chlorine is								
a) +1 b) 0	c) – 1 d) +3							
•	f iron by weight. Molecular weight of hemoglobin is 67200.							
_	wt of Fe = 55.85) ion one molecule of hemoglobin is							
a) 2 b) 4	c) 6 d) 8							
	•							
 The gas law which deals with the effect of pressure in a given volume of gas constant temperature is known as 								
a) Boyle's Law c) Charle's Law								
c) Graham's Law								
•	r elements A, B, C, D are as follow, which will be most							
metallic?	referrence ry by cy b are as renow, which will be most							
a) A=2,8, 4 b) B= 2, 8, 8	c) C= 2, 8, 1 d) D= 2, 8, 7							
	bound the molecular geometry cannot be predicted by the use							
of VSEPR theory	, , , , ,							
•	c) CH ₄ d) NH ₃							
-, -	C + D. If the concentration of the reactants are increase by							
three times, the rate of reaction	•							
a) 3 times b) 6 times	c) 9 times d) 27 times							
85. An aqueous solution of NaCl w	vould be							
a) Acidic								
b) basic								
c) neutral								
d) could be acidic or basic dep	ending upon temperature							
86. The pH of pure water at 50°C v	would be							
a) 7.0	b) more than 7.0							
c) less than 7.0	d) cannot be predicted							
	generate H_2 on treatment with dil. HNO_3 .							
a) Zn b) Mn	c) Cu d) Pb							
88. 1 M H₂SO₄ is equal to								
-,	c) 0.5 N H ₂ SO ₄ d) 4 N H ₂ SO ₄							
	parallel spin of two nuclei is called							
a) Ortho hydrogen	b) para hydrogen							
c) heavy hydrogen	d) nascent hydrogen							
90. Ammonia is manufactured by								
a) Solvoy's process	b) Haber's process							
c) Ostwald's process	d)Down's process							
91. Hematite is an ore of	م/ 7م م/ ۱۱ م							
a) Feb) Cu92. Which of the following is not a	c) Zn d) Hg							
a) BF ₃ b) NH ₃	c) H ₂ O d) OH ⁻							
93. Anti-MarkowniKoff addition of	· -							
a) Propene b) 1-b								
•	glycerol are mixed and heated to about 110°C, the compound							
obtained is	gryceror are mixed and heated to about 110°C, the compound							
a) Chloroform	b) diethylether							
c) Acetone	d) formic acid							
95. Which of the following is a me	•							
a) $-NH_2$ b) $-OH$ c) $-CO$								
-, ₂ 2, 3 c, co	· - · · · · · · · · · · · · · · · · · ·							

96. How many position isomers are possible for dichlorobenzene

a) 2 b) 3 c) 4 d) 6 97. Which one of the following is a condensation polymer a) Polyethene b) Terylene d) Teflon c) PVC 98. Penicillin is an example of a) Antiseptic drug b) antipyretic drug c) Antibiotic drug d) analgesic drug 99. Enzymes are a) Carbohydrates b) proteins d) nucleic acids c) lipids 100. Which base is found only in the nucleotides of RNA? a) Adenine b) guanine b) uracil d) cytosine

Answers

1	L. c	2. a	3. b	4. b	5. c	6. b	7. a	8. b	9. d	10. c	11. d	12. a	13. b
	14. a	15. b	16. a	17. d	18. a	19. c	20. d	21 . a	22. d	23. a	24. b	25. b	26. a
	27. a	28. a	29. a	30. a	31. a	32. a	33. a	34. a	35. a	36. a	37. a	38. a	39. a
	40. a	41. a	42. a	43. a	44. a	45. a	46. a	47. a	48. a	49. a	50. a	51. b	52. b
	53. a	54. a	55. d	56. a	57. c	58. c	59. a	60. c	61. d	62. c	63. d	64. a	65. b
	66. b	67. a	68. b	69. d	70. a	71. d	72. a	73. c	74. d	75. b	76. c	77. b	78. c
	79. a	80. b	81. a	82. c	83. a	84. d	85. c	86. c	87. a	88. b	89. a	90. b	91. a
			92. a	93. a	94. d	95. c	96. b	97. a	98. c	99. b	100. c		