Tribhuvan University

Institute of Science and Technology

4 Years Bachelor in Computer Science and Information Technology

Entrance Examination

2071

Mathematics

d) 1

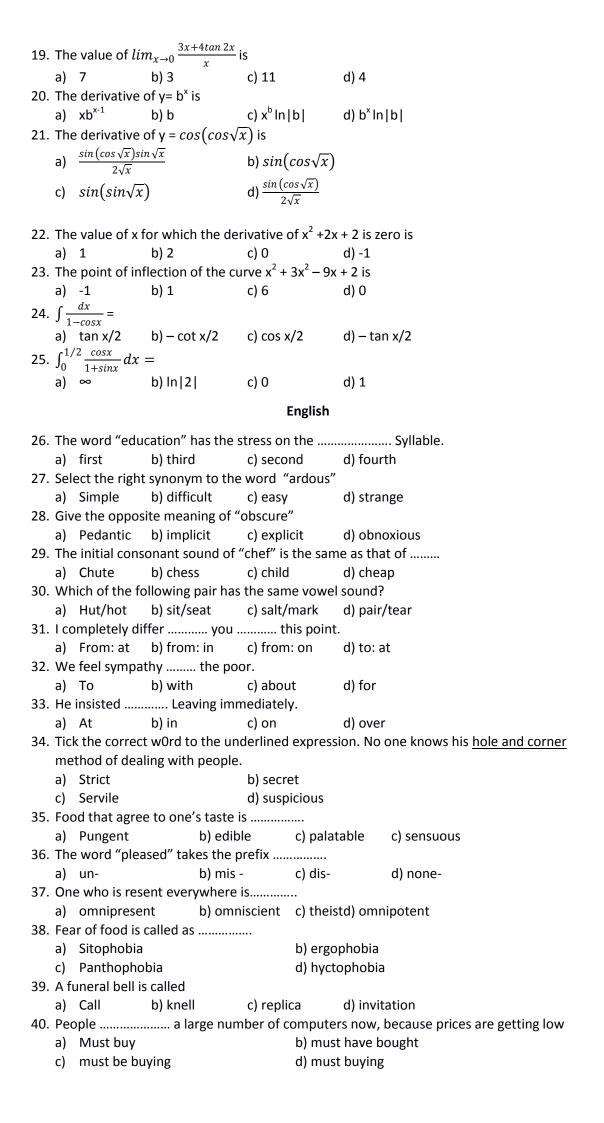
c) 7

1. If (X-3) = 4 then X is a) 7 and -1 b) 7

2. If $\{X-X^2=9 \text{ and } 2X=6\}$

b) 7 and 1

	(
	a) Null set b) {3}		d) {4,3}			
3.	3. If: $R \rightarrow R$ defined by $f(X)=X^2+3$ then $f^{-1}(4)$ is					
	a) Null set b) {-1,1}	c) {1,0}	d) {4,3}			
4.	The value of $\log_2 \sqrt{2} + \log_3 \sqrt{3}$	$\overline{3} + log_{\Lambda}\sqrt{4}$ is				
			d) $\pi/6$			
5.	If $\cos^2 \theta = 1/2$, then θ is	-,, -	2,, 2			
٥.	a) $n\pi \pm \pi/4$ b) $n\pi + \pi/4$	c) nπ + π/3	d) 2nπ			
6	Cos ($\sin^{-1}3/5$) =	c) 11/1 ± 11/3	a) zinc			
0.		c) 0/2E	d) 4/2E			
7	a) $3/5$ b) $4/5$	C) 9/25	u) 4/23			
/.	If a^2 , b^2 , c^2 are in H.P. then a) $2 a^2 c^2 = b^2 (a^2 + c^2)$	L \ -2-2 L-2 /-	22\			
	a) $2ac = b(a + c)$	b) a c = b (a	+ C)			
	c) $2b^2c^2 = a^2(b^2 + c^2)$	d) c a = 2 b	(a + c)			
8.	If $A = \begin{bmatrix} 1 & 2 \\ 2 & 3 \end{bmatrix}$ and $A^2 - nA - I =$					
	a) 0 b) 4	c) 1	d) -4			
	If $A = \begin{bmatrix} 1 & 0 & 1 \\ 0 & 1 & 1 \\ 1 & 0 & 0 \end{bmatrix}$, then A is					
9.	If $A = \begin{bmatrix} 0 & 1 & 1 \end{bmatrix}$, then A is					
		1.3.1				
	a) Symmetric	b) skew-symr				
	c) Singular	d) Non singul				
10	The system of equations 2x +		3 is			
	a) Consistent and dependen					
	b) Inconsistent and independent					
	c) Inconsistent and depende					
	d) Consistent and independe	ent				
11	. The value of i ⁷ + 1/i ⁴ is					
	a) 1 b)-1	c) —i	d) –i +1			
12	. The value of $\frac{x-wy+w^2z}{y+wz+w^2x}$, w is t	he cube root of	unity, is			
	a) w b) $x + y + z$		d) 1			
13	If the roots of $x^2 + 4x + k = 0$, a		is			
		c) -4				
14	The two lines $a_1x + b_1y + c_1 = 0$	•	•			
	The two lines $2x^2 - 2xy + ky^2 =$					
13	a) 2 b) -2	c) 0	d) 1			
16	,	•	•			
10	16. The equation of the circle touching the positive x- axis and y- axis with radius 2 units is a) $x^2 + y^2 - 4x - 4y + 4 = 0$					
	b) $x^2 + y^2 + 4x + 4y + 4 = 0$					
	c) $x^2 + y^2 - 4x + 4y + 4 = 0$					
	d) $x^2 + y^2 - 4x - 4y - 4 = 0$		2 2 2 2 3 4 2 3 4 2 4 3			
17	If the line $4x + 3y = k$ is tanger					
	a) 25 b) -25	c) ±25	d) ±11			
18. The value of $\lim_{x\to 2} \frac{x^5-2^5}{x-2}$ is						
	a) 80 b) 40 ²	c) 10	d) 0			
	-					



41. Although they are similar, o	hipmunks are	most other ground squirrels.
a) Alike b) like as	c) like to	d) like
42. She the whole day	long.	
a) Has been working	b) wo	rks
c) had worked	d) wo	
43. Now a days Shyam is	•	
a) Very b) to	c) so	d) extremely
44. She demandedhe		N
- · · · · · · · · · · · · · · · · · · ·	c) to know	,
45. The mail comes at ten O'clo		
	and c) also	o d) but
46. In my office, salary is paid be a) The b) a	-	d) some
a) The	a c) an	d) some
47. Every boy and every girl	very respons	ihle
a) Are b) i		
48. Which of the following is ac	•	a, nave
a) Some of rice is still left		
b) Time and tide wait for i	none	
c) She absents herself fro	m the college today	/
d) None of the squares wa	as properly drawn	
49. Which of the following is a	complex sentence?	
a) However, hard she wor	ked, she failed in t	he exams.
b) It rained heavily, howe	ver he came to sch	ool
c) In spite of heavy rain, h		
d) Neither she nor he will		
50. Which of the following is an	•	
a) Sympathize b)	lighten c) per	ception d) moral
	Dh	
51. The dimensions of (h/e) are		ysics
a) Magnetic field	b) electric fiel	d
c) magnetic flux		
52. The number of significant f	<u>-</u>	
a) 5 b) 4	c) 11	d) 2
•	•	ictions between two bodies. Then
a) $\mu_s \geq \mu_k$	b) $\mu_s < \mu_k$	
b) $\mu_s = 0$ and $\mu_k > 0$		ı _k < 0
	body, the change in	n angular momentum () of the body is
a) Perpendicular to $ec{L}$	b) parallel to	$ec{L}$
c) always zero	d) $180^{\circ}\vec{L}$	
55. The Poisson's ratio of mate	•	
a) 0.10 b) 0.20	c) 0.40	d) 0.65
56. The depth of water at whic	h air-bubble of rad	ius 0.4 mm may remain in equilibrium (surface
tension of water = 72×10^{-5}	3 N/m, g= 9.8 m/s 2	is given by
a) 7.348 cm b) 0.918 cm	n c) 3.674 cm	d) 1.837 cm
57. When a cylinder is heated,	its length increases	by 2%, the area of its base will increase by:
a) 0.5% b) 1%	c) 2%	d) 4%
58. In a pressure cooker food is	s cooked usually at	a pressure
a) One atmosphere	b) half atmos	
•	d) 50 atmospl	
59. The mean free path of a ga		
a) T ⁴ b) 1/T ⁴	c) 1/T	d) T
60. The efficiency of Carnot en	gine working betwe	een steam point and ice point is:

	•	16.8 %	b) 26.81%	•	d) 46.8%
61.		_		_	ers of the obstacle is called
62	a) Refraction b) interference c) beatsd) diffraction. When a tuning fork is vibrated its prongs have a phase difference of:				
02.	a)		b) π/4	c) π	d) 2π
	a)	U	0) 10/4	C) II	a) zit
63.	The	e image of a	n object placed a	at the focus of co	onvex mirror is at a distance
	a)	∞	b) f	c) 2f	d) f/2
64.	If s	ky is seen fro	om moon's surfa	ace, it will appea	r
	a)	Black	b) blue	c) whited) red	
65.	A d	ouble conve	ex air bubble in a	glass sheet wou	uld behave as a
		Convergent		b) concave mir	ror
	-	Divergent l		d) plane mirror	•
66.		•	ms are combine	d it gives	
	-	Reflection	•		
	•	lateral shift			
		-	and deviation b	otn	
c 7	•	•	or deviation		
67.				used in an electr	on microscope e nature
		Negative ch	iarge	c) sma	
68	c)	•	ee charges may	•	n. If they are placed
08.		On a straig		be in equilibriui	ii. Ii tiley are placed
	-	_	of equilateral tr	riangle	
			orners of a recta		
	-		ners of a cube		
69.	•			The capacitance	e is
		6400 MF		c) 1 F	d) 711 μF
70.	Tw	o wires of sa	me material ha	ve lengths L and	2L and cross reaction areas 4A and A
	res	pectively. Th	ne ratio of their	specific resistand	ces would be
	a)	1:1	b) 1:8	c) 8:1	d) 1:2
71.	The	e magnetic f	ield inside a long	g solenoid is	
	•	Zero	b)	c) μ ₀ ni	d) μni
72.			e atoms are held		
	•	Nuclear for		b) Coulomb for	
	•	Van der Wa		c) Gravitationa	I forces
/3.			f H atom lies in v	_	Develope
	-	Lyman		b) Bracket and	Pascnen
74	,	Balmer	a substance is 1	c) Paschen 600 years. The n	agan life is
74.	a)		a substance is 1	b) 3200 years	ilean life is
	•	800 years		d) 2309 years	
75.	-	lberg consta	int is	a, 2505 years	
	•	Same for al			
	,		or different elem	ients	
	c)				
	d) Is different for lighter elements but same for heavier elements				
Chemistry					
76How many orbitals are possible for principle quantum number 3?					
70.	a)	-	b) 4	c) 6	d) 9
77.	,		•	present in 18g o	•
		18	b) 1	c) 1.8	d) 180
			,		•

3. The half-life of an radioisotope is 30 minutes. After how much time the amount of radioisotope left is 1/4th of the original amount.				
a) 4 minutes	b) 30 minutes			
c) 60 minutes	d) 120 minutes			
79. What is oxidation of chlorine is				
a) +1 b) 0	c) – 1 d) +3			
•	f iron by weight. Molecular weight of hemoglobin is 67200.			
_	wt of Fe = 55.85) ion one molecule of hemoglobin is			
a) 2 b) 4	c) 6 d) 8			
	the effect of pressure in a given volume of gas constant			
temperature is known as	the effect of pressure in a given volume of gas constant			
a) Boyle's Law	c) Charle's Law			
c) Graham's Law				
•	r elements A, B, C, D are as follow, which will be most			
metallic?	referrence ry by cy b are as renow, which will be most			
a) A=2,8, 4 b) B= 2, 8, 8	c) C= 2, 8, 1 d) D= 2, 8, 7			
	bound the molecular geometry cannot be predicted by the use			
of VSEPR theory	, , , , ,			
•	c) CH ₄ d) NH ₃			
-, -	C + D. If the concentration of the reactants are increase by			
three times, the rate of reaction	•			
a) 3 times b) 6 times	c) 9 times d) 27 times			
85. An aqueous solution of NaCl w	vould be			
a) Acidic				
b) basic				
c) neutral				
d) could be acidic or basic dep	ending upon temperature			
86. The pH of pure water at 50°C v	would be			
a) 7.0	b) more than 7.0			
c) less than 7.0	d) cannot be predicted			
	generate H_2 on treatment with dil. HNO_3 .			
a) Zn b) Mn	c) Cu d) Pb			
88. 1 M H₂SO₄ is equal to				
	c) 0.5 N H ₂ SO ₄ d) 4 N H ₂ SO ₄			
	parallel spin of two nuclei is called			
a) Ortho hydrogen	b) para hydrogen			
c) heavy hydrogen	d) nascent hydrogen			
90. Ammonia is manufactured by				
a) Solvoy's process	b) Haber's process			
c) Ostwald's process	d)Down's process			
91. Hematite is an ore of	م/ 7م م/ ۱۱ م			
a) Feb) Cu92. Which of the following is not a	c) Zn d) Hg			
a) BF ₃ b) NH ₃	c) H ₂ O d) OH ⁻			
93. Anti-MarkowniKoff addition of	· -			
a) Propene b) 1-b				
•	glycerol are mixed and heated to about 110°C, the compound			
obtained is	gryceror are mixed and heated to about 110°C, the compound			
a) Chloroform	b) diethylether			
c) Acetone	d) formic acid			
95. Which of the following is a meta directing group				
a) $-NH_2$ b) $-OH$ c) $-CO$				
-, ₂ 2, 3 c, co	· - · · · · · · · · · · · · · · · · · ·			

96. How many position isomers are possible for dichlorobenzene

a) 2	b) 3	c) 4	d) 6		
97. Which one of the following is a condensation polymer					
a)	Polyethene	9	b) Terylene		
c)	PVC		d) Teflon		
98. Penicillin is an example of					
a) Antiseptic drug			b) antipyretic drug		
c)	Antibiotic (drug	d) analgesic drug		
99. Enzymes are					
a)	a) Carbohydrates		b) proteins		
c)	lipids		d) nı	ucleic acids	
100. Which base is found only in the nucleotides of RNA?					
a)	Adenine	b) guanine	b) uracil	d) cytosine	