Name: Key

Tip: Go to Desmos.com to plot implicit functions!

1. Find the derivative y' = y'(x) of y with respect to x.

(a) 
$$x^2 + y^2 = 4$$

(b) 
$$x^2 + y^2 = 9$$

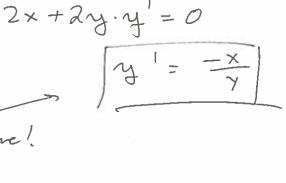
$$2x + 2y \cdot y' = 0$$

$$2y y' = -2x$$

$$y' = -\frac{2}{2}y$$

$$y' = -\frac{2}{2}x$$





(c)  $2\sqrt{x} + 2\sqrt{y} = 1$ 

$$2^{-1/2} \times 2^{-1/2} + 2 \cdot \frac{1}{2} \cdot y^{-1/2} \cdot y^{-1/2} = 0$$

$$x^{-1/2} + y^{-1/2} \cdot y^{-1/2} = 0$$

$$y^{-1/2} \cdot y^{-1/2} = -x$$

$$y^{-1/2} \cdot y^{-1/2}$$

$$(d) 1 + 2x = \sin(y^2)$$

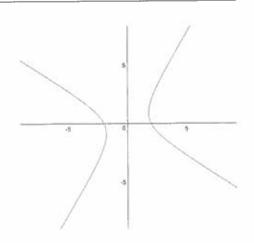
$$(1+2x)' = (sin(y^2))'$$

$$2 = cos(y^2) \cdot 2y \cdot y'$$

$$y' = 2$$

$$cos(y^2) \cdot 2y$$

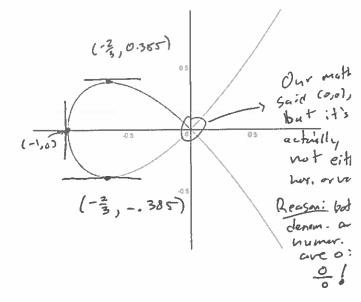
(e) 
$$x^2 + xy - y^2 = 4$$



- line is vertical or horizontal. We will work with the equation  $y^2 = x^3 + x^2$ , plotted below. (You should use this to check your answers!)
  - i. Find y' by implicitly differentiating.

$$2yy' = 3x^2 + 2x$$

$$|y'| = \frac{3x^2 + 2x}{2y}$$



ii. Horiontal Tangents: set the numerator of your fraction equal to 0, and solve.

$$3x^{2}+2x=0$$

$$(3x+2)=0 \Rightarrow \boxed{X=0} \text{ or } \boxed{X=-\frac{2}{3}}$$

iii. Use the numbers you found to get the remaining coordinate.

$$X=0$$
  $Y^2=(0)^3+(0)^3=0$ 

$$\frac{1}{\sqrt{12}} = \frac{1}{\sqrt{12}} =$$

$$(0,0)$$
  
 $(-3/2,0.385)$   
 $(-3/2,-0.385)$ 

iv. Vertical Tangents: set the denominator equal to 0.

v. Use the numbers you found to get the other coordinate.

$$O^{2} = \chi^{2} + \chi^{2}$$

$$O = \chi^{2} (\chi + 1)$$

$$(0,0)$$

$$(-1,0)$$

(g) Apply the same strategy to find the locations (meaning (x, y)) where there are vertical or horizontal tangents for the ellpise:

$$2(x-1)^{2}+2(y-2)^{2}=4.$$

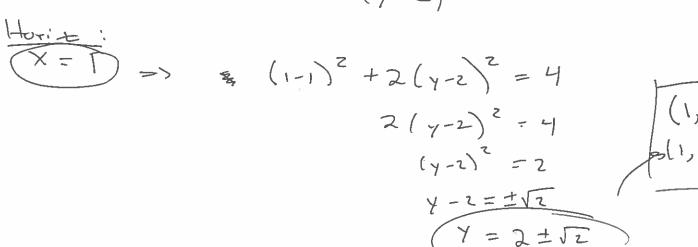
$$(1,2+\sqrt{2})$$

$$4'=-(x-1)$$

$$(3,2-\sqrt{2})$$

$$(4,2-\sqrt{2})$$

$$(4,2-\sqrt{2})$$



Verti 
$$y-2=0 \rightarrow y=2$$
 (3,2), (4,2)  
 $(x-1)^2+2(22)^2=4$   
 $(x-1)^2=4$   
 $x-1=\pm 2$   
 $x=1\pm 2$  3

(h) (Challenge Problem) In advanced chemistry, you may come across the so-called <u>van der Waals</u> equation, which is a generalization of the ideal gas law PV = nRT. The van der Waals equation is the following:

$$\left(P + \frac{n^2 a}{V^2}\right)(V - nb) = nRT$$

where P is the pressure of a gas, T its temperature, and V its volume. R is the universal gas constant, and a and b are constants depending on the chemical. If T remains constant, find  $\frac{dV}{dP}$  using implicit differentiation.

Move to the size 
$$\frac{dV}{dP} \cdot \left[ -\frac{2n^2a}{V^3} \cdot \left( V - nb \right) + \left( P + \frac{n^2a}{V^2} \right) \right] = -\left( V - nb \right)$$

$$\frac{dV}{dP} = -\left[-\frac{2n^{2}a}{V^{3}}(V-nb) + (P+\frac{n^{2}a}{V^{2}})\right]^{-1}(V-nb)$$