

Worksheet 6

Math 251, Summer 2017

Name: _____

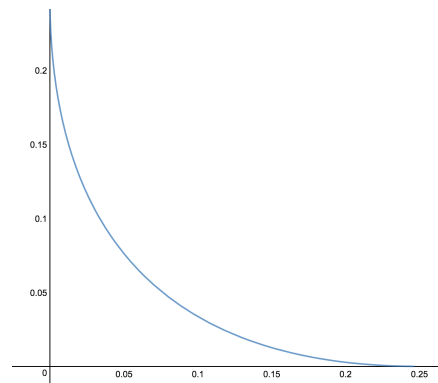
Tip: Go to [Desmos.com](https://www.desmos.com) to plot implicit functions!

1. Find the derivative $y' = y'(x)$ of y with respect to x .

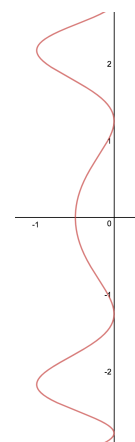
(a) $x^2 + y^2 = 4$

(b) $x^2 + y^2 = 9$

(c) $2\sqrt{x} + 2\sqrt{y} = 1$



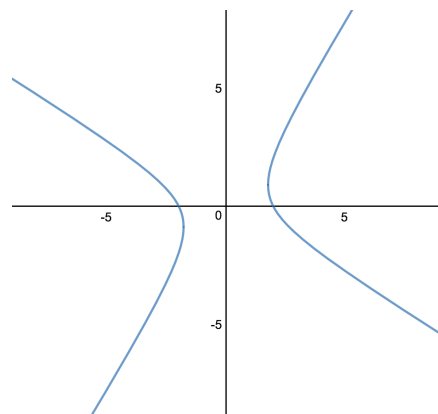
(d) $1 + 2x = \sin(y^2)$



Worksheet 6

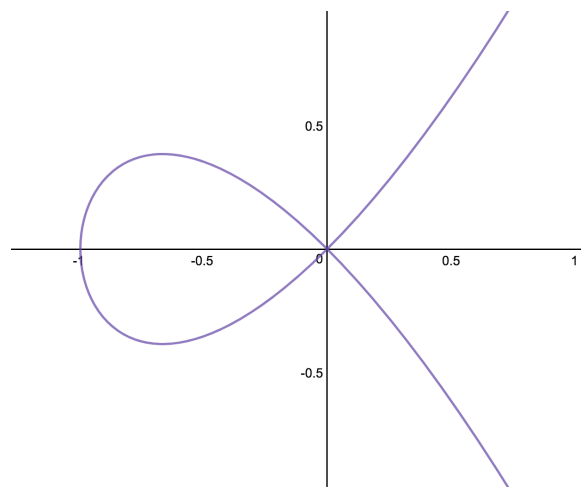
Math 251, Summer 2017

(e) $x^2 + xy - y^2 = 4$



- (f) For the following problem, you will be guided through the process of finding all points (x, y) where the tangent line is vertical or horizontal. We will work with the equation $y^2 = x^3 + x^2$, plotted below. (You should use this to check your answers!)

i. Find y' by implicitly differentiating.



ii. Horizontal Tangents: set the numerator of your fraction equal to 0, and solve.

iii. Use the numbers you found to get the remaining coordinate.

Worksheet 6

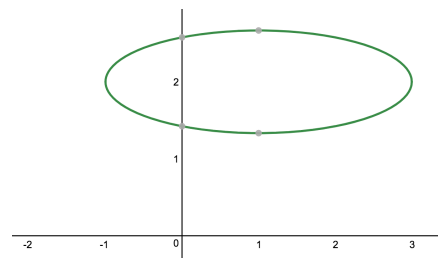
Math 251, Summer 2017

iv. Vertical Tangents: set the denominator equal to 0.

v. Use the numbers you found to get the other coordinate.

(g) Apply the same strategy to find the locations (meaning (x, y)) where there are vertical or horizontal tangents for the ellipse:

$$(x - 1)^2 + 2(y - 2)^2 = 4.$$



Worksheet 6

Math 251, Summer 2017

-
- (h) (Challenge Problem) In advanced chemistry, you may come across the so-called *van der Waals equation*, which is a generalization of the ideal gas law $PV = nRT$. The van der Waals equation is the following:

$$\left(P + \frac{n^2a}{V^2}\right)(V - nb) = nRT$$

where P is the pressure of a gas, T its temperature, and V its volume. R is the universal gas constant, and a and b are constants depending on the chemical. If T remains constant, find $\frac{dV}{dP}$ using implicit differentiation.