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Title:	Manganese(II) and zinc(II) complexes of 4-phenyl(2-methoxybenzoyl)-3-thiosemicarbazide: Synthesis, spectral, structural characterization, thermal behavior and DFT study				
Authors:	Singh, Sanjay (/jspui/browse?type=author&value=Singh%2C+Sanjay)				
Keywords:	DFT Thiosemicarbazide Mn(II) and Zn(II) complexes Crystal structure Supramolecular architecture TGA				
Issue Date:	2014				
Publisher:	Elsevier				
Citation:	Polyhedron,73, pp.98-109.				
Abstract:	The ligand 4-phenyl(2-methoxybenzoyl)-3-thiosemicarbazide (Hpmt), forms isostructural [Mn(pmt)2(o-phen)] (1) and [Zn(pmt)2(o-phen)] (2) complexes containing o-phen as coligand which have been characterized by analytical, spectroscopic (IR, UV-Vis, NMR), magnetic susceptibility, TGA and single crystal X-ray data. Both complexes crystallize in monoclinic systems with the space group P2/n. The complexes have distorted octahedral geometry around the metal center. The ligand in the complexes is coordinated through the deprotonated hydrazinic nitrogen and carbonyl oxygen. The hydrazinic nitrogen coordinates with a shorter M-N distance than the o-phen nitrogen and bond lengths in the chelate ring systems are intermediate between single and double bond distances, suggesting considerable delocalization of charge. There is a good agreement between the geometrical parameters obtained by X-ray crystallography to those generated by DFT method. The thermal degradations of complexes 1 and 2 have been investigated by thermogravimetric analyses which indicate that the final residues left are Mn(NCO)2 and Zn(NCSNH) 2. The small HOMO-LUMO energy gap suggests low excitation energy for the complexes.				
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