



# Library Indian Institute of Science Education and Research Mohali



**DSpace@IISERMohali (/jspui/)**  
**/ Publications of IISER Mohali (/jspui/handle/123456789/4)**  
**/ Research Articles (/jspui/handle/123456789/9)**

Please use this identifier to cite or link to this item: <http://hdl.handle.net/123456789/3173>


Title:	Two-Coordinate Cu(I) and Au(I) Complexes Supported by BICAAC and CAAC Ligands
Authors:	Manar, K.K. (/jspui/browse?type=author&value=Manar%2C+K.K.) Chakraborty, Soumyadeep (/jspui/browse?type=author&value=Chakraborty%2C+Soumyadeep) Porwal, Vishal Kumar (/jspui/browse?type=author&value=Porwal%2C+Vishal+Kumar) Prakash, D. (/jspui/browse?type=author&value=Prakash%2C+D.) Thakur, Sandeep Kumar (/jspui/browse?type=author&value=Thakur%2C+Sandeep+Kumar) Choudhury, A.R. (/jspui/browse?type=author&value=Choudhury%2C+A.R.) Singh, Sanjay (/jspui/browse?type=author&value=Singh%2C+Sanjay)
Keywords:	BICAAC Carbenes Copper Gold Complexes
Issue Date:	2020
Publisher:	Wiley-Blackwell
Citation:	ChemistrySelect, 5(32), pp.9900-9907.
Abstract:	Two-coordinate Cu(I) and Au(I) complexes supported by bicyclic (alkyl)(amino)carbene, [BICAAC-CuCl] (1), [BICAAC-CuI] (2), [(BICAAC)2Cu]+[PF6]- (3) and [(BICAAC)2Au]+[AuCl2]- (6) have been synthesized. The reaction of cyclic (alkyl)(amino)carbene, CAACcy with CuCl afforded [CAACcy-CuCl] (4) and its further reaction with KPF6 gave [(CAACcy)2Cu]+[PF6]- (5). Complexes 1–6 have been characterized by multinuclear NMR, IR and UV-Vis., spectroscopic method and high-resolution mass spectrometry (HRMS). Single crystal X-ray structure of heteroleptic complexes 1 and 4 and homoleptic complexes [(BICAAC)2Cu]+[CuI2]- (2'), 3 and 6 have also been determined. The crystal structure of these complexes confirmed linear two-coordinate geometry around the metal centers. In the solid- state, complexes 1, 2', 4 and 6 displayed C–H...M (M=Cu, Au) and weak non-covalent C–H...X (X=Cl, I) and C–H...H–C interactions. Computational calculations correlate well to the experimentally observed geometry and help elucidate the absorption characteristics type of transitions and the frontier orbitals involved in them.
URI:	<a href="https://chemistry-europe.onlinelibrary.wiley.com/doi/10.1002/slct.202002295">https://chemistry-europe.onlinelibrary.wiley.com/doi/10.1002/slct.202002295</a> ( <a href="https://chemistry-europe.onlinelibrary.wiley.com/doi/10.1002/slct.202002295">https://chemistry-europe.onlinelibrary.wiley.com/doi/10.1002/slct.202002295</a> ) <a href="http://hdl.handle.net/123456789/3173">http://hdl.handle.net/123456789/3173</a> ( <a href="http://hdl.handle.net/123456789/3173">http://hdl.handle.net/123456789/3173</a> )
Appears in	Research Articles (/jspui/handle/123456789/9)
Collections:	

## Files in This Item:

File	Description	Size	Format
Need to add pdf.odt (/jspui/bitstream/123456789/3173/1/Need%20to%20add%20pdf.odt)		8.63 kB	OpenDocument Text

[View/Open \(/jspui/bitstream/123456789/3173/1/Need%20to%20add%20pdf.odt\)](#)

Show full item record (</jspui/handle/123456789/3173?mode=full>)

 (</jspui/handle/123456789/3173/statistics>)

Items in DSpace are protected by copyright, with all rights reserved, unless otherwise indicated.