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Title:	Borrowing Hydrogen-Mediated N-Alkylation Reactions by a Well-Defined Homogeneous Nickel Catalyst
Authors:	Bains, A.K. (/jspui/browse?type=author&value=Bains%2C+A.K.) Kundu, A. (/jspui/browse?type=author&value=Kundu%2C+A.) Yadav, Sudha (/jspui/browse?type=author&value=Yadav%2C+Sudha) Adhikari, D. (/jspui/browse?type=author&value=Adhikari%2C+D.)
Keywords:	Hydrogen-borrowing catalysis Ni(azo-phenolate) ₂ complex N-alkylation
Issue Date:	2019
Publisher:	American Chemical Society
Citation:	ACS Catalysis, 9(10), pp.9051-9059.
Abstract:	We report herein a well-defined and bench-stable azo-phenolate ligand-coordinated nickel catalyst which can efficiently execute N-alkylation of a variety of anilines by alcohol. We demonstrate that the redox-active azo ligand can store hydrogen generated during alcohol oxidation and redelivers the same to an in-situ-generated imine bond to result in N-alkylation of amines. The reaction has wide scope, and a large array of alcohols can directly couple to a variety of anilines. Mechanistic studies including deuterium labeling to the substrate establishes the borrowing hydrogen method from alcohols and pinpoints the crucial role of the redox-active azo moiety present on the ligand backbone. Isolation of the ketyl intermediate in its trapped form with a radical quencher and higher k_H/k_D for the alcohol oxidation step suggest altogether a hydrogen-atom transfer (HAT) to the reduced azo backbone to pave alcohol oxidation as opposed to the conventional metal-ligand bifunctional mechanism. This example clearly demonstrates that an inexpensive base metal catalyst can accomplish an important coupling reaction with the help of a redox-active ligand backbone.
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