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Title:	Coordination driven or/and H-bonded Cu(II)-N,N-dialkylisonicotinamide frameworks
Authors:	Kapoor, Ramesh (/jspui/browse?type=author&value=Kapoor%2C+Ramesh)
Keywords:	Coordination polymers
	Hydrogen bonding
	N,N-dialkylisonicotinamides
Issue Date:	2011
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Citation:	Polyhedron, 30 (10), pp. 1691-1702
Abstract:	Reactions of N,N-diisopropylisonicotinamide (L) with anhydrous CuCl2, CuBr2 and Cu(ClO4)2-6H2O yielded: (a) an ionic product with the molecular composition of [CuL2(H2O)4]·2[CuLCl3] 1; (b) a coordination polymer [CuL2Br2]n2 and (c) a mononuclear complex [CuL4(C3H8O)2]·(ClO4)2]3, respectively. Similarly, the reaction of N,N-diisobutylisonicotinamide (L') with Cu(NO3)2·3H2O yielded a coordination polymer [{CuL'2(H2O)2}(NO3) 2] n 4. The ligand L and L' coordinate in a monodentate fashion through the pyridine nitrogen atom to the metal centers in complexes 1 and 3. However, complexes 2 and 4 are coordination polymers in which the corresponding ligands (L and L') act as bridging bidentate between metal centers to form 1D double chains. All the four complexes form networks through coordination polymerization and/or hydrogen bonding.
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