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Title:	Stabilization of the C20 cage by encapsulation of H+ and He2+ ions					
Authors:	Abhijith Kumar, R.P.S. (/jspui/browse?type=author&value=Abhijith+Kumar%2C+R.P.S.)					
	Dev, Sagarika (/jspui/browse?type=author&value=Dev%2C+Sagarika)					
	Sathyamurthy, N. (/jspui/browse?type=author&value=Sathyamurthy%2C+N.)					
Keywords:	Atoms-in-molecules					
	Encapsulation of protons					
	Fullerene					
	Natural bond orbital					
	Perturbation theory					
Issue Date:	2014					
Publisher:	Indian Academy of Sciences					
Citation:	Current Science, 106(9), pp.1255-1259.					
Abstract:	It is shown by ab initio calculations using secondorder Møller-Plesset perturbation theory and 6 31g*, aug-cc-pVDZ and aug-cc-pVTZ basis sets and coupled cluster singles, doubles and perturbative triples method with the 6-31g* basis set, that it is possible to stabilize the cage					
	structure of the smallest fullerene C20 by encapsulating small cations like H+ and He2+. While					
	the latter stabilizes the cage structure by charge transfer, the former prefers to form a covalent					
	linkage with the carbon atoms.					

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