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Title:	Engineering a Nanoscale Primary Amide-Functionalized 2D Coordination Polymer as an Efficient and Recyclable Heterogeneous Catalyst for the Knoevenagel Condensation Reaction
Authors:	Markad, D. (/jspui/browse?type=author&value=Markad%2C+D.) Mandal, S.K. (/jspui/browse?type=author&value=Mandal%2C+S.K.)
Keywords:	Primary amide-functionalized coordination polymer 2D Cd(II) coordination polymer Knoevenagel Condensation reaction
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Abstract:	<p>This work reports the design, structural characterization, and catalytic behavior of the first example of primary amide-functionalized coordination polymers (CPs), namely, $\{[\text{Cd}_2(2\text{-bpbg})(\text{fum})_2(\text{H}_2\text{O})_2] \cdot 8.5\text{H}_2\text{O}\}_n$ (1) (where 2-bpbg = N,N'-bis(2-pyridylmethyl)-1,4-diaminobutane-N,N'-diacetamide and fum = fumarate). CP 1 is synthesized from a one-pot self-assembly of starting materials in methanol under ambient conditions in excellent yield and purity, allowing an easy access to multigram quantities of it within few hours. As an example, CP 1 was used as a highly efficient heterogeneous catalyst in the carbon-carbon bond-forming Knoevenagel condensation reaction for the conversion of benzaldehyde to benzylidene malononitrile. CP 1 possesses both Lewis acidic and Brønsted basic character for the presence of unsaturated metal sites and primary amide groups, respectively, making it a highly competent bifunctional catalyst for such type of reactions. Surprisingly, on the one hand, 100% conversion was observed using only 2 mol % catalyst within 1 h at 27 °C in methanol. On the other hand, 2 mol % and 3 mol % catalyst loadings but without a solvent gives 93% and 100% conversions, respectively, in 1 h at 27 °C. CP 1 is far better than those reported in the literature. To prove the uniqueness and efficiency of the primary amide-based ligand, a similar compound with a pyridyl-based ligand was also synthesized, $\{[\text{Cd}_2(\text{tpbn})(\text{fum})_2] \cdot 6\text{H}_2\text{O}\}_n$ (2) (where tpbn = N',N'',N''',N''-tetrakis(pyridin-2-ylmethyl)butane-1,4-diamine). With CP 2 under the same catalyst loading and conditions (2 mol %, 27 °C, 1 h), only 28% conversion was observed. This demonstrates that selective heterogeneous catalytic properties of 1 over 2 are due to the presence of the primary amide moieties and open metal sites. Moreover, CP 1 can easily be separated from the reaction mixture and reused for five consecutive cycles without significant loss of its activity. Both 1 and 2 were fully characterized by elemental analysis, infrared spectroscopy, thermogravimetric analysis, and single-crystal and powder X-ray diffraction. These crystallize in the triclinic $P\bar{1}$ space group, showing their isostructural nature and three-connected, uninodal {63} honeycomb net topology.</p>
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