

# Software Requirements Specification for Solar Water Heating Systems

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# 1 Reference Material

This section records information for easy reference.

## 1.1 Table of Units

The unit system used throughout is SI (Système International d'Unités). In addition to the basic units, several derived units are also used. For each unit, the table lists the symbol, a description and the SI name.

symbol	unit	SI
m	length	metre
kg	mass	kilogram
s	time	second
°C	temperature	centigrade
J	energy	Joule
W	power	Watt ( $W = J s^{-1}$ )

## 1.2 Table of Symbols

The table that follows summarizes the symbols used in this document along with their units. The choice of symbols was made to be consistent with the heat transfer literature and with existing documentation for solar water heating systems. The symbols are listed in alphabetical order.

symbol	unit	description
$A_C$	$m^2$	heating coil surface area
$A_{in}$	$m^2$	surface area over which heat is transferred in
$A_{out}$	$m^2$	surface area over which heat is transferred out
$C$	$J/(kg\ ^\circ C)$	specific heat capacity
$C^L$	$J/(kg\ ^\circ C)$	specific heat capacity of a liquid
$C_W$	$J/(kg\ ^\circ C)$	specific heat capacity of water
$D$	m	diameter of tank
$E$	J	change in heat energy
$E_W$	J	change in heat energy in the water
$g$	$W/m^2$	volumetric heat generation per unit volume
$h$	$W/(m^2\ ^\circ C)$	convective heat transfer coefficient
$h_C$	$W/(m^2\ ^\circ C)$	convective heat transfer coefficient between coil and water

$L$	m	length of tank
$m$	kg	mass
$m_W$	kg	mass of water
$q$	W/(m <sup>2</sup> °C)	heat flux
$\mathbf{q}$	W/m <sup>2</sup>	thermal flux vector
$q_C$	W/m <sup>2</sup>	heat flux into the water from the coil
$q_{\text{in}}$	W/m <sup>2</sup>	heat flux in
$q_{\text{out}}$	W/m <sup>2</sup>	heat flux out
$t$	s	time
$T$	°C	temperature
$T_C$	°C	temperature of the heating coil
$T_{\text{env}}$	°C	temperature of the environment
$t_{\text{final}}$	s	final time
$T_{\text{init}}$	°C	initial temperature
$T_W$	°C	temperature of the water
$\Delta T$	°C	change in temperature
$V$	m <sup>3</sup>	volume
$V_W$	m <sup>3</sup>	volume of water
$\rho$	kg/m <sup>3</sup>	density
$\rho_W$	kg/m <sup>3</sup>	density of water
$\tau$	s	dummy variable for integration over time

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### 1.3 Abbreviations and Acronyms

symbol	description
A	Assumption
DD	Data Definition
GD	General Definition
GS	Goal Statement
IM	Instance Model
LC	Likely Change
ODE	Ordinary Differential Equation
PS	Physical System Description
R	Requirement
SRS	Software Requirements Specification
SWHS	Solar Water Heating System
T	Theoretical Model
TU	Typical Uncertainty

## 2 Introduction

Due to the increasing cost, diminishing availability, and negative environmental impact of fossil fuels, there is a higher demand for renewable energy sources and energy storage technology. Solar water heating systems provide a novel way of storing energy.

The following section provides an overview of the Software Requirements Specification (SRS) for a solar water heating system. The developed program will be referred to as Solar Water Heating System (SWHS). This section explains the purpose of this document, the scope of the system, the organization of the document and the characteristics of the intended readers.

### 2.1 Purpose of Document

The main purpose of this document is to describe the modelling of solar water heating systems. The goals and theoretical models used in the SWHS code are provided, with an emphasis on explicitly identifying assumptions and unambiguous definitions. This document is intended to be used as a reference to provide ad hoc access to all information necessary to understand and verify the model. The SRS is abstract because the contents say *what* problem is being solved, but do not say *how* to solve it.

This document will be used as a starting point for subsequent development phases, including writing the design specification and the software verification and validation plan. The design document will show how the requirements are to be realized, including decisions on the numerical algorithms and programming environment. The verification and validation plan will show the steps that will be used to increase confidence in the software documentation and the implementation. Although the SRS fits in a series of documents that follow the so-called waterfall model, the actual development process is not constrained in any way. Even when the waterfall model is not followed, as Parnas and Clements [5] point out, the most logical way to present the documentation is still to “fake” a rational design process.

### 2.2 Scope of Requirements

The scope of the requirements includes thermal analysis of a single solar water heating tank. Given the appropriate inputs, SWHS predicts the temperature and thermal energy histories for the water.

### 2.3 Characteristics of Intended Reader

Reviewers of this documentation should have a strong knowledge in heat transfer theory. A third or fourth year Mechanical Engineering course on this topic is recommended. The reviewers should also have an understanding of differential equations, as typically covered in first and second year Calculus courses. The users of Solar Water Heating Systems can have a lower level of expertise, as explained in Section 3.2.

## 2.4 Organization of Document

The organization of this document follows the template for an SRS for scientific computing software proposed by [3] and [6]. The presentation follows the standard pattern of presenting goals, theories, definitions, and assumptions. For readers that would like a more bottom up approach, they can start reading the instance models in Section 4.2.5 and trace back to find any additional information they require. The goal statements are refined to the theoretical models, and the theoretical models to the instance models. The instance model (Section 4.2.5) to be solved is referred to as IM1. The instance model provides the Ordinary Differential Equation (ODE) that model the solar water heating system. SWHS solves this ODE.

## 3 General System Description

This section provides general information about the system including identifying the interfaces between the system and its environment (system context), describing the user characteristics and listing the system constraints.

### 3.1 System Context

Figure 1 shows the system context. A circle represents an external entity outside the software, the user in this case. A rectangle represents the software system itself (SWHS). Arrows are used to show the data flow between the system and its environment.

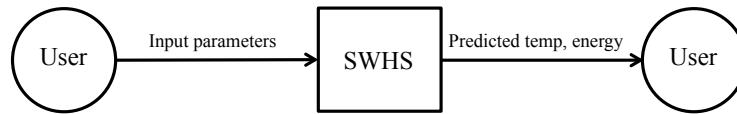


Figure 1: System Context

SWHS is mostly self-contained. The only external interaction is through the user interface. The responsibilities of the user and the system are as follows:

- User Responsibilities:
  - Provide the input data to the system, ensuring no errors in the data entry
  - Take care that consistent units are used for input variables
- SWHS Responsibilities:
  - Detect data type mismatch, such as a string of characters instead of a floating point number

- Determine if the inputs satisfy the required physical and software constraints
- Calculate the required outputs

### **3.2 User Characteristics**

The end user of SWHS should have an understanding of undergraduate Level 1 Calculus and Physics.

### **3.3 System Constraints**

There are no system constraints.

## **4 Specific System Description**

This section first presents the problem description, which gives a high-level view of the problem to be solved. This is followed by the solution characteristics specification, which presents the assumptions, theories, definitions and finally the instance model (ODE) that models the solar water heating tank.

### **4.1 Problem Description**

SWHS is a computer program developed to investigate the heating of water in a solar water heating tank.

#### **4.1.1 Terminology and Definitions**

This subsection provides a list of terms that are used in the subsequent sections and their meaning, with the purpose of reducing ambiguity and making it easier to correctly understand the requirements:

- Heat Flux: the rate of thermal energy transfer through a given surface per unit time.
- Specific heat capacity: the amount of energy required to raise the temperature of the unit mass of a given substance by a given amount.
- Thermal Conduction: the transfer of heat energy through a substance.
- Transient: Changing with time.



### 4.1.2 Physical System Description

The physical system of SWHS, as shown in Figure 2, includes the following elements:

PS1: Tank containing water.

PS2: Heating coil at bottom of tank. ( $q_C$  represents the heat flux into the water from the coil.)

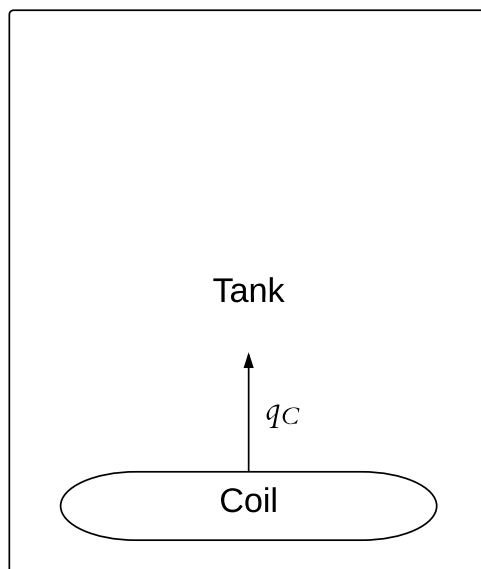


Figure 2: Solar water heating tank, with heat flux from coil of  $q_C$

### 4.1.3 Goal Statements

Given the temperature of the heating coil, initial temperature of the water and, material properties, the goal statements are:

GS1: predict the temperature of the water over time

GS2: predict the change in heat energy in the water over time

## 4.2 Solution Characteristics Specification

The instance model (ODE) that governs SWHS is presented in Subsection 4.2.5. The information to understand the meaning of the instance model and its derivation is also presented, so that the instance model can be verified.

### 4.2.1 Assumptions

This section simplifies the original problem and helps in developing the theoretical model by filling in the missing information for the physical system. The numbers given in the square brackets refer to the theoretical model [Section 4.2.2], general definition [Section 4.2.3], data definition [Section 4.2.4], instance model [Section 4.2.5], or likely change [Section 6], in which the respective assumption is used.

- A1: The only form of energy that is relevant for this problem is thermal energy. All other forms of energy, such as mechanical energy, are assumed to be negligible [T1].
- A2: All heat transfer coefficients are constant over time [GD1].
- A3: The water in the tank is fully mixed, so the temperature is the same throughout the entire tank [GD2].
- A4: The density of the water has no spatial variation; that is, it is constant over the entire volume [GD2, LC1].
- A5: The specific heat capacity of the water has no spatial variation; that is, it is constant over the entire volume [GD2].
- A6: Newton's law of convective cooling applies between the heating coil and the water [DD1].
- A7: The temperature of the heating coil is constant over time [DD1, LC1].
- A8: The temperature of the heating coil does not vary along its length [DD1, LC2].
- A9: The model only accounts for charging of the tank, not discharging. The temperature of the water can only increase, or remain constant; it cannot decrease. This implies that the initial temperature is less than (or equal to) the temperature of the coil [IM1, LC3].
- A10: The operating temperature range of the system is such that the material (water in this case) is always in a liquid state. That is, the temperature will not drop below the melting point of water, or rise above its boiling point temperature [T2, IM1, IM2].
- A11: The tank is perfectly insulated so that there is no heat loss from the tank [IM1, LC4].
- A12: No internal heat is generated by the water; therefore, the volumetric heat generation per unit volume is zero [IM1].
- A13: The pressure in the tank is atmospheric, so the melting and boiling points of water are 0°C and 100°C, respectively [IM1, IM2].

### 4.2.2 Theoretical Models

This section focuses on the general equations and laws that SWHS is based on.

Number	T1
Label	<b>Conservation of thermal energy</b>
Equation	$-\nabla \cdot \mathbf{q} + g = \rho C \frac{\partial T}{\partial t}$
Description	The above equation gives the conservation of energy for time varying heat transfer in a material of specific heat capacity $C$ and density $\rho$ , where $\mathbf{q}$ is the thermal flux vector, $g$ is the volumetric heat generation, $T$ is the temperature, $t$ is time, and $\nabla$ is the degree of steepness of a graph at any point. For this equation to apply, other forms of energy, such as mechanical energy, as assumed to be negligible in the system (A1).
Source	[1]
Ref. By	GD2

Number	T2
Label	<b>Sensible heat energy (No State Change)</b>
Equation	$E = Cm\Delta T$
Description	$E$ is the change in sensible heat energy (J). $C$ is the specific heat capacity ( $\text{J kg}^{-1} \text{ }^{\circ}\text{C}^{-1}$ ). $m$ is the mass (kg). $T$ is the temperature ( $^{\circ}\text{C}$ ), and $\Delta T$ is the change in temperature ( $^{\circ}\text{C}$ ). Sensible heating occurs as long as the material does not reach a temperature where a phase change occurs, as assumed in A10.
Source	<a href="http://en.wikipedia.org/wiki/Sensible_heat">http://en.wikipedia.org/wiki/Sensible_heat</a>
Ref. By	IM2

### 4.2.3 General Definitions

This section collects the laws and equations that will be used in deriving the data definitions, which in turn are used to build the instance models.

Number	GD1
Label	<b>Newton's law of cooling</b>
SI Units	$\text{W m}^{-2}$
Equation	$q(t) = h\Delta T(t)$
Description	<p>Newton's law of cooling describes convective cooling from a surface. The law is stated as: the rate of heat loss from a body is proportional to the difference in temperatures between the body and its surroundings.</p> <p><math>q(t)</math> is the thermal flux (<math>\text{W m}^{-2}</math>).</p> <p><math>h</math> is the heat transfer coefficient, assumed independent of <math>T</math> (A2) (<math>\text{W m}^{-2} \text{ }^{\circ}\text{C}^{-1}</math>).</p> <p><math>\Delta T(t) = T(t) - T_{\text{env}}(t)</math> is the time-dependent thermal gradient between the environment and the object (<math>^{\circ}\text{C}</math>).</p>
Source	[2, p. 8]
Ref. By	DD1

Number	GD2
Label	<b>Simplified rate of change of temperature</b>
Equation	$mC\frac{dT}{dt} = q_{\text{in}}A_{\text{in}} - q_{\text{out}}A_{\text{out}} + gV$
Description	<p>The basic equation governing the rate of change of temperature, for a given volume <math>V</math>, with time.</p> <p><math>m</math> is the mass (kg).</p> <p><math>C</math> is the specific heat capacity (<math>\text{J kg}^{-1} \text{ }^\circ\text{C}^{-1}</math>).</p> <p><math>T</math> is the temperature (<math>^\circ\text{C}</math>) and <math>t</math> is the time (s).</p> <p><math>q_{\text{in}}</math> and <math>q_{\text{out}}</math> are the in and out heat transfer rates, respectively (<math>\text{W m}^{-2}</math>).</p> <p><math>A_{\text{in}}</math> and <math>A_{\text{out}}</math> are the surface areas over which the heat is being transferred in and out, respectively (<math>\text{m}^2</math>).</p> <p><math>g</math> is the volumetric heat generated (<math>\text{W m}^{-3}</math>).</p> <p><math>V</math> is the volume (<math>\text{m}^3</math>).</p>
Ref. By	IM1

### Detailed derivation of simplified rate of change of temperature

Integrating (T1) over a volume ( $V$ ), we have

$$-\int_V \nabla \mathbf{q} dV + \int_V g dV = \int_V \rho C \frac{\partial T}{\partial t} dV.$$

Applying Gauss's Divergence theorem to the first term over the surface  $S$  of the volume, with  $\mathbf{q}$  as the thermal flux vector for the surface, and  $\hat{\mathbf{n}}$  is a unit outward normal for the surface,

$$-\int_S \mathbf{q} \cdot \hat{\mathbf{n}} dS + \int_V g dV = \int_V \rho C \frac{\partial T}{\partial t} dV. \quad (1)$$

We consider an arbitrary volume. The volumetric heat generation per unit volume is assumed constant. Then (1) can be written as

$$q_{\text{in}}A_{\text{in}} - q_{\text{out}}A_{\text{out}} + gV = \int_V \rho C \frac{\partial T}{\partial t} dV,$$

where  $q_{\text{in}}$ ,  $q_{\text{out}}$ ,  $A_{\text{in}}$ , and  $A_{\text{out}}$  are explained in GD2. Assuming  $\rho$ ,  $C$  and  $T$  are constant over the volume, which is true in our case by assumption (A3), (A4), and (A5), we have

$$\rho CV \frac{dT}{dt} = q_{\text{in}}A_{\text{in}} - q_{\text{out}}A_{\text{out}} + gV. \quad (2)$$

Using the fact that  $\rho = m/V$ , (2) can be written as

$$mC \frac{dT}{dt} = q_{\text{in}} A_{\text{in}} - q_{\text{out}} A_{\text{out}} + gV.$$

#### 4.2.4 Data Definitions

This section collects and defines all the data needed to build the instance models. The dimension of each quantity is also given.

Number	DD1
Label	<b>Heat flux into the water from the coil</b>
Symbol	$q_C$
SI Units	$\text{W m}^{-2}$
Equation	$q_C = h_C(T_C - T_W(t))$ , over area $A_C$
Description	$q_C$ is the heat flux into the water from the coil ( $\frac{\text{W}}{\text{m}^2}$ ) $h_C$ is the convective heat transfer coefficient between coil and water ( $\frac{\text{W}}{(\text{m}^2 \text{ } ^\circ\text{C})}$ ) $T_C$ is the temperature of the heating coil ( $^\circ\text{C}$ ) $T_W$ is the temperature of the water ( $^\circ\text{C}$ ) $t$ is the time (s)
Sources	[4]
Ref. By	IM1

#### 4.2.5 Instance Models

This section transforms the problem defined in the Section 4.1 into one which is expressed in mathematical terms. It uses concrete symbols defined in Section 4.2.4 to replace the abstract symbols in the models identified in the Sections 4.2.2 and 4.2.3.

Number	IM1
Label	<b>Energy balance on water to find <math>T_W</math></b>
Input	$m_W, C_W, h_C, A_C, t_{\text{final}}, T_C, T_{\text{init}}$ The input is constrained so that $T_{\text{init}} \leq T_C$ (A9)
Output	$T_W(t), 0 \leq t \leq t_{\text{final}}$ , such that $\frac{dT_W}{dt} = \frac{1}{\tau_W}(T_C - T_W(t))$ , $T_W(0) = T_{\text{init}}$
Description	$T_W$ is the water temperature ( $^{\circ}\text{C}$ ). $T_C$ is the coil temperature ( $^{\circ}\text{C}$ ). $\tau_W = \frac{m_W C_W}{h_C A_C}$ is a constant (s). The above equation applies as long as the water is in liquid form, $0 < T_W < 100^{\circ}\text{C}$ , where $0^{\circ}\text{C}$ and $100^{\circ}\text{C}$ are the melting and boiling points of water, respectively (A10).
Sources	Original SRS with PCM removed
Ref. By	

### Derivation of the energy balance on water

To find the rate of change of  $T_W$ , we look at the energy balance on water. The volume being considered is the volume of water in the tank  $V_W$ , which has mass  $m_W$  and specific heat capacity,  $C_W$ . Heat transfer occurs in the water from the coil as  $q_C$ , over area  $A_C$ . No heat transfer occurs to the outside of the tank, since it has been assumed to be perfectly insulated (A11). Assuming no volumetric heat generation per unit volume (A12),  $g = 0$ . Therefore, the equation for GD2 can be written as:

$$m_W C_W \frac{dT_W}{dt} = q_C A_C.$$

Using DD1 for  $q_C$ , this can be written as

$$m_W C_W \frac{dT_W}{dt} = h_C A_C (T_C - T_W). \quad (3)$$

Dividing (3) by  $m_W C_W$ , we obtain

$$\frac{dT_W}{dt} = \frac{h_C A_C}{m_W C_W} (T_C - T_W). \quad (4)$$

Setting  $\tau_W = \frac{m_W C_W}{h_C A_C}$ , Equation (4) can be written in its final form as:

$$\frac{dT_W}{dt} = \frac{1}{\tau_W}(T_C - T_W).$$

Number	IM2
Label	<b>Heat energy in the water</b>
Input	$C_W, m_W, T_{\text{init}}, T_W(t)$
Output	$E_W(t), 0 \leq t \leq t_{\text{final}}, \text{ such that}$ $E_W(t) = C_W m_W (T_W(t) - T_{\text{init}})$
Description	The above equation is derived using T2. $E_W$ is the change in thermal energy of the liquid water relative to the energy at the initial temperature ( $T_{\text{init}}$ ). $C_W$ is the specific heat capacity of liquid water and $m_W$ is the mass of the water. The change in temperature is the difference between the temperature at time t, $T_W$ , and the initial temperature, $T_{\text{init}}$ , this equation applies as long as $0 < T_W < 100^\circ\text{C}$ (A10).
Sources	[4]
Ref. By	–

#### 4.2.6 Data Constraints

Table 1 and 2 show the data constraints on the input and output variables, respectively. The column for physical constraints gives the physical limitations on the range of values that can be taken by the variable. The uncertainty column provides an estimate of the confidence with which the physical quantities can be measured. This information would be part of the input if one were performing an uncertainty quantification exercise. The constraints are conservative, to give the user of the model the flexibility to experiment with unusual situations. The column of typical values is intended to provide a feel for a common scenario. The column for software constraints restricts the range of inputs to reasonable values.

## 5 Requirements

This section provides the functional requirements, the business tasks that the software is expected to complete, and the nonfunctional requirements, the qualities that the software is expected to exhibit.

### 5.1 Functional Requirements



Table 1: Input Variables Constraints

Var	Physical Constraints	Software Constraints	Typical Value	Uncertainty
$L$	$L > 0$	$L_{\min} \leq L \leq L_{\max}$	1.5 m	10%
$D$	$D > 0$	$\frac{D}{L}_{\min} \leq \frac{D}{L} \leq \frac{D}{L}_{\max}$	0.412 m	10%
$A_C$	$A_C > 0$ (*)	$A_C \leq A_C^{\max}$	0.12 m <sup>2</sup>	10%
$T_C$	$0 < T_C < 100$ (+)		50.0 °C	10%
$\rho_W$	$\rho_W > 0$	$\rho_W^{\min} < \rho_W \leq \rho_W^{\max}$	1000 kg/m <sup>3</sup>	10%
$C_W$	$C_W > 0$	$C_W^{\min} < C_W < C_W^{\max}$	4186 J/(kg °C)	10%
$h_C$	$h_C > 0$	$h_C^{\min} \leq h_C \leq h_C^{\max}$	1000.0 W/(m <sup>2</sup> °C)	10%
$T_{\text{init}}$	$0 < T_{\text{init}} < 100$ (+)		40.0 °C	10%
$t_{\text{final}}$	$t_{\text{final}} > 0$	$t_{\text{final}} < t_{\text{final}}^{\max}$ (**)	50000.0 s	10%

Table 2: Output Variables Constraints

Var	Physical Constraints
$T_W$	$T_{\text{init}} \leq T_W \leq T_C$ (by A9)
$E_W$	$E_W \geq 0$

R1: Input the following quantities, which define the tank parameters, material properties and initial conditions:

symbol	unit	description
$L$	m	length of tank
$D$	m	diameter of tank
$A_C$	m <sup>2</sup>	heating coil surface area
$T_C$	°C	temperature of the heating coil
$\rho_W$	kg/m <sup>3</sup>	density of water
$C_W$	J/(kg °C)	specific heat capacity of water
$h_C$	W/(m <sup>2</sup> °C)	convective heat transfer coefficient between coil and water
$T_{\text{init}}$	°C	initial temperature of water
$t_{\text{final}}$	s	final time

R2: Use the inputs in R1 to find the mass needed for IM1, as follows:

$$m_W = V_W \rho_W = \pi(D/2)^2 L \rho_W$$

where  $V_W$  is the volume of water in the tank.

- R3: Verify that the inputs satisfy the required physical constraints shown in Table 1.
- R4: Output the input quantities and derived quantities in the following list: the quantities from R1, the mass from R2 and  $\tau_W$  (from IM1).
- R5: Calculate and output the temperature of the water ( $T_W(t)$ ) over the simulation time (from IM1).
- R6: Calculate and output the energy in the water ( $E_W(t)$ ) over the simulation time (from IM2).

## 5.2 Nonfunctional Requirements

This problem is small in size and relatively simple, so performance is not a priority. Any reasonable implementation will be very quick and use minimal storage. Rather than performance, the non-functional requirement priorities are correctness, verifiability, understandability, reusability, and maintainability.

## 6 Likely Changes

- LC1: A7 - The temperature of the heating coil will change over the course of the day, depending on the energy received from the sun.
- LC2: A8 - The temperature of the water in the coil will actually change along its length as the water cools.
- LC3: A9 - The model currently only accounts for heating of the tank. That is, increasing the temperature of the water to match the temperature of the coil. A more complete model would also account for cooling of the tank.
- LC4: A11 - Any real tank cannot be perfectly insulated and will lose heat.

## 7 Unlikely Changes

- UC1: It is unlikely for the state change of water from liquid to a solid to be considered.
- UC2: For the derivation of IM1, A12 must be held as true.

## References

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