CLASSROOM CONTACT PROGRAMME

(Academic Session: 2019 - 2020)

LEADER & ACHIEVER COURSE

PHASE: MLI,J,K,M,N,O,R,S,MAZG,H,I,J,K,L,M,T,U,M4AA2A,M2AP1A,M2AP1B **TARGET: PRE-MEDICAL 2020**

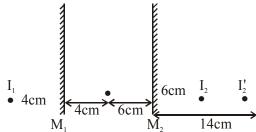
Test Type: MAJOR Test Pattern: NEET(UG)

TEST DATE: 11 - 03 - 2020

TEST SYLLABUS: 06

HINT - SHEET

1.



Distance between $I_1 \& I_2'$ is 28cm

Distance between to minimize chromatic 2.

abberration =
$$\frac{f_1 + f_2}{2} = \frac{8+4}{2} = 6 \text{ cm}$$

3. Intensity becomes 4 times because area is

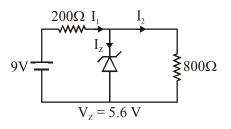
reduced to
$$\left(\frac{1}{4}\right)^{th}$$
.

By using $E = -\frac{13.6}{n^2} \text{eV}$ (for H₂ atom)

$$\Rightarrow 0.544 = -\frac{13.6}{n^2} \Rightarrow n^2 = 25 \Rightarrow n = 5$$

$$\therefore \quad \text{Angular momentum } = n \frac{h}{2\pi} = \frac{5h}{2\pi} \,.$$

5.



In first loop

$$9V = 200 I_1 + 5.6 V$$

$$I_1 = \frac{3.4}{200} = 17 \text{ mA}$$

In second loop

$$I_2 = \frac{5.6}{800} = 7 \text{ mA}$$

$$I_z = I_1 - I_2$$

= (17 - 7) mA = 10 mA

Rotation in Reflected ray

$$10^{\circ} + 40^{\circ} = 30^{\circ}$$

7. $m = m_0 \times m_e$

$$= 25 \times 6$$

$$= 150$$

$$8. d = \frac{1.22\lambda}{2(NA)}$$

$$\Rightarrow$$
 NA = 0.5

The characteristic X-ray depends on the material used.



10. Voltage on R is 3V

$$V = IR \Rightarrow 3 = \frac{100}{1000} \times R \Rightarrow R = 30\Omega$$

11.
$$\sin \theta_C = \frac{\mu_R}{\mu_D} = \frac{v_D}{v_R}$$

$$\frac{6}{10} = \frac{v_D}{3 \times 10^8}$$

$$v_{\rm D} = 1.8 \times 10^8 \text{ m/s}$$

13.
$$\frac{3}{8}I_0 = I = \frac{I_0}{2} \cos^2 \theta$$

14. outside the nucleus, neutron is unsatble.

16.
$$v = \frac{uf}{u - f} = -60 \text{ cm}$$

$$\frac{h_i}{5} = -\frac{v}{u} \Rightarrow h_i = -25cm$$

19.
$$\sigma = [(Zm_p + NM_n) - M] \times 931 \text{ MeV}$$

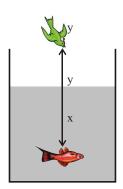
$$\sigma = [26 \times 1.00783 + 30 \times 1.00867) - 55.9349] \times 931$$

$$\sigma = 492 \text{ MeV}$$

20.
$$V_0 = (A_1 \times A_2)V_s$$

= $(10 \times 20) \times 0.01 = 2V$

21. Here
$$x_{bf} = \mu y + x$$



$$\frac{dx_{bf}}{dt} = \mu \frac{dy}{dt} + \frac{dx}{dt} = \frac{4}{3}(-12) + 24$$

$$\Rightarrow$$
 $v_{bf} = 8$ m/s upward

22.
$$\frac{I_1}{I_2} = \frac{1}{4} \implies I_1 = k \text{ and } I_2 = 4k$$

$$\therefore \text{ Fringe visibility V} = \frac{2\sqrt{I_1I_2}}{(I_1 + I_2)} = \frac{2\sqrt{k \times 4k}}{(k + 4k)} = 0.8$$

- 23. For emission of electrons incident energy of each photon must be greater than work function (threshold energy).
- **26.** A real image \Rightarrow v = R/2

$$U = +\frac{R}{3}$$

from
$$\frac{n_2}{v} - \frac{n_1}{u} \quad \frac{n_2 - n_1}{R}$$

$$\Rightarrow \frac{2n_2}{R} - \frac{3n_1}{+R} = \frac{n_2 - n_1}{R} \Rightarrow 2n_2 - 3n_1 = n_2 - n_1$$
$$\Rightarrow n_2 = 2n_1$$

$$\therefore \frac{n_1}{n_2} = \frac{1}{2}$$

27.
$$I_{\text{max}} = \left(\sqrt{I_1} + \sqrt{I_2}\right)^2 = \left(\sqrt{9} + \sqrt{4}\right)^2 = 25$$

$$I_{min} = (\sqrt{I_1} - \sqrt{I_2})^2 = (\sqrt{9} - \sqrt{4})^2 = 1$$

28. Energy of incident light $E = \frac{12375}{2000} = 6.18 \text{ eV}$

According to relation $E = W_0 + eV_0$

$$\Rightarrow V_0 = \frac{(E - W_0)}{e} = \frac{(6.18 \text{ eV} - 5.01 \text{ eV})}{e} = 1.17 \text{ V} \approx 1.2 \text{ V}$$

29.
$$N = N_0 \left(\frac{1}{2}\right)^{\frac{15}{5}} = \frac{N_0}{8}$$

31. Focal length of plano convex lens

$$f = \frac{R}{\mu - 1} = \frac{R}{(1.5 - 1)} = 2R$$



35. From input signals, we have,

A	В	Output NAND gate
0	0	1
1	0	1
0	0	1
1	1	0
0	0	1

36. $m = \frac{f - v}{f} = \frac{+10 - (-25)}{+10} = \frac{35}{10}$

$$m = \frac{7}{2}$$

37. $\lambda_1 = 5893 \text{ Å}, n_1 = 62, \lambda_2 = 4358 \text{ Å}$

As field of view in case of both the wavelengths is same, hence

$$n_1\beta_1 = n_2\beta_2$$

or
$$n_1 \left(\frac{D\lambda_1}{d} \right) = n_2 \left(\frac{D\lambda_2}{d} \right)$$

or
$$n_2 = n_1 \left(\frac{\lambda_1}{\lambda_2} \right) = 62 \times \frac{5893}{4358} \cong 84$$

38. For an electron,

de Broglic wavelength, $\lambda = \frac{h}{\sqrt{2 \text{ mK}}}$

where h = Planck constant, m = mass of an electron, K = kinetic energy of an electron Since, h, m remains the same,

$$\therefore \qquad \qquad \lambda \propto \frac{1}{\sqrt{K}}$$

or
$$\frac{\lambda}{\lambda'} = \sqrt{\frac{K'}{K}} = \sqrt{\frac{3K}{K}}$$

or
$$\frac{\lambda'}{\lambda} = \frac{1}{\sqrt{3}}$$
.

39. In positive half cycle one diode is in forward biasing and other is in reverse biasing while in negative half cycle their polarity reverses, and direction of current is opposite through R for positive and negative half cycles so output is not rectified.

Since R_1 and R_2 are different hence the peaks during positive half and negative half of the input signal will be different.

- **40.** For NAND gate, $\overline{1.1} = \overline{1} = 0$
- 42. $2\theta = \frac{2\lambda}{d}$ (where d = slit width)

As d decreases, θ increases.

43. According to Einstein's photoelectric equation

$$V_0 = \frac{hc}{e} \left[\frac{1}{\lambda} - \frac{1}{\lambda_0} \right]$$

Hence if λ decreases V_0 increases

- **44.** $E = \frac{V}{d} = \frac{0.5}{5 \times 10^{-7}} = 10^6 \text{ V/m}$
- **45.** This is a truth table for NAND gate because $X = \overline{A \cdot B}$.

46.
$$CH_2-CH_2 \xrightarrow{Alc. KOH} CH=CH_2 \xrightarrow{NaNH_2} H-C=C-H$$
Br Br

- **47.** Rate of $SN_1 \propto Stability of C^{\oplus}$
- 48. Maltose = α -D-glucose + α -D-glucose
- 49.



51. Chloro benzene have partial double bond character between C---C1, So it does not show SN² reaction easily.

52.
$$\begin{array}{c}
CH_{3} \\
CH_{3}-CH \\
O_{3} \\
\oplus \\
H/H_{2}O
\end{array}$$

$$\begin{array}{c}
OH \\
O\\
CH_{3}-C-CH_{3}
\end{array}$$

$$\begin{array}{c}
(A)
\end{array}$$

53.

55. Hinsberg test to distinguish 1°, 2°, 3° amine.

57.
$$\bigcap_{\substack{\Theta \\ CH_3}}^{CH_3} \bigcap_{\alpha}^{\beta} \bigcap_{\Delta}^{\Theta_{OH}} \bigcap_{\Delta}^{CH_3} \bigcap_{CH_3}^{CH_3} \bigcap_{CH_3}^{$$

58. CH₃CHO gives Iodoform test

59.
$$\bigcirc$$
 CH₂—O \bigcirc + HI \rightarrow

$$\bigcirc$$
 CH₂I+ \bigcirc OH

60.
$$COOH$$

CH₃COCI

CH₃COCI

Asperin

Antipyretic and Analgesic

61. Nylon-6,6 is a poly amide

62. Fact

63.
$$COCl \xrightarrow{H_2} CHO$$

65.
$$H \xrightarrow{CH_3} OH \xrightarrow{HIO_4} 2CH_3-C-H$$
 CH_3

66. Fact

67. CH_3 -C-O-H does not give silver mirror test

$$CH_3-C-CH_2-C-O-H+CH_3-OH$$

$$O \qquad O$$

$$\downarrow \Delta$$

$$CH_3-C-CH_3+CO_2$$

69.
$$\begin{array}{c} CH_3 \\ -C-NC \\ CH_3 \\ -C-NC \\ CH_3 \end{array} \xrightarrow{\text{Reduction}} \begin{array}{c} CH_3 \\ -C-NH-CH_3 \\ CH_3 \end{array}$$

t-butyl methyl amine

70.
$$CH_2 \xrightarrow{\Delta} CH_3COOH + CO_2(A)$$



71. Fact

72.
$$\begin{array}{c}
O \\
NaBD_4 \\
H^{\oplus}
\end{array}$$

73.
$$\begin{array}{c}
\text{Br} \\
\text{CH}_2-\text{CONH}_2 \\
\text{COOCH}_3
\end{array}$$

74. Hydrolysis of ester $\alpha \frac{-M/-H/-I}{+M/+H/+I}$

75.
$$HC \equiv CH \xrightarrow{\text{Red hot Fe}} (A) \xrightarrow{\text{CH}_3\text{COCl}} (A)$$

76. CH₃CHO+CH₃CH₂CHO dilute NaOH CH₂-CH=CH-CHO CH₃-CH=C-CHO CH₃-CH=C-CHO CH₃

$$\begin{cases} \begin{cases} \begin{cases}$$

77.
$$O \xrightarrow{NH_2} \xrightarrow{NaNO_2 + HCl} O \xrightarrow{N_2Cl} OH$$

$$pH = 8 - 10$$

$$\bigcirc$$
N=N \bigcirc OH

78.
$$\underbrace{NH_2}_{NaNO_2/HCl} \underbrace{N_2Cl}_{pH = 8-10}$$

ethyl amine

an Azo dye

79.
$$CH_3-C-O-C-CH_3 \xrightarrow{H_2O} 2CH_3-C-OH$$

$$C_2H_5-OH \downarrow H / \Delta$$

$$CH_3-C-O-C_2H_5$$

80. Cetyltrimethyl ammonium bromide is cationic detergent

ester

81. $R-NH_2 + CHCl_3 + KOH \rightarrow R-NC$

82.

$$\begin{array}{c} O \\ \parallel \\ R-C-NH-H \\ \hline OH \\ \hline -H_2O \\ \hline OH \\ \hline -H_2O \\ \hline OH \\ \hline OH \\ \hline OH \\ \hline R-C-NH-B1 \\ \hline -H_2O \\ \hline OH \\ \hline$$

83. Fact

84.
$$R-C=N \xrightarrow{(i) DIBAL-H} R-CHO + NH_3$$

85. Fact

86.
$$CH_3-C-Cl$$
 $CH_3-NH_2 \longrightarrow CH_3-C-NH-CH_3$
Acetylchloride Methyl amine

87. Neoprene is

n
$$CH_2=C-CH=CH_2 \longrightarrow (CH_2-C=CH-CH_2)$$
Cl
Cl
Cl
neoprene



89.
$$CH_3$$
 CH_3 CH_3 CH_3 CH_3 CH_3 CH_3

- **90.** Fact
- **91.** NCERT Pg.# 169
- **92.** Module-3 Pg.#2

NCERT XII, Pg.# 233 (E), 254 (H)

The term 'Keystone Predator' was coined by paine during his studies on Pisaster.

- **93.** NCERT XII, Pg.# 245 (E), 267 (H)
- **94.** NCERT XII, Pg.# 266,267 (E), 291 (H)
- **95.** NCERT XII, Pg.# 229 (E), 249 (H)
- **96.** Anaerobic environment leads to slow decomposition
- **98.** DDT is a non biodegradable pollutant. So does not affects BOD and DO even do not cause eutrophication as DDT is not a nutrient for plants in water body.
- **99.** NCERT XII, Pg# 63
- **101.** NCERT Pg.# 168
- **102.** Module-3 Pg.# 4
- **103.** NCERT XII, Pg.# 224 (E), 243 (H)
- **104.** NCERT XII, Pg.# 272 (E), 297 (H)
- **105.** Number of reproductive individuals (adults) are higher than prereproductive individuals (Juvaniles) means population is decending.
- **106.** Chemosynthetic bacteria (Chemoautotrophs) get energy from oxidation reduction of inorganic compounds.
- **107.** 'Z'-represents slope of line. If value of 'Z' is high per unit area species diversity is high.
- 109. NCERT XII, Pg# 158
- **111.** NCERT Pg.# 168

- 112. NCERT XII, Pg.# 252 (E), 275 (H)
- 113. NCERT XII, Pg.# 223 (E), 243 (H)
- **114.** NCERT XII, Pg.# 280 (E), 305 (H)
- 119. NCERT XII, Pg# 158, 159
- **121.** NCERT Pg.# 168
- 122. NCERT XII, Pg.# 238 (E), 259 (H)
- **123.** NCERT XII, Pg.# 226 (E), 245 (H)
- 124. NCERT XII, Pg.# 280 (E), 305 (H)
- 125. 'r'- stratagist population like oyster, pelagic fishes does not produce few number of large sized offsprings.
- **127.** David Tillman experimentally proved that more the diversity, more the productivity, more the stability of ecosystem.
- 128. NCERT XII, Pg# 146
- 131. NCERT Pg.# 169
- **132.** Module-3 Pg.# 13
- 133. NCERT XII, Pg.# 254 (E), 277 (H)
- 134. NCERT XII, Pg.# 283 (E), 309 (H)
- **135.** Ecological niche of a species is functional role of species in community which is created by resources utilized and conditions tolerated by a species in a community.
- 136. A, B, C and E are correct
- 138. NCERT XII, Pg# 151
- **141.** NCERT Pg.# 182
- **142.** Except *Agaricus*, all others are photosynthetic organisms
- 143. Module-3, Pg.# 33
- **144.** NCERT XII, Pg.# 279 (E), 304 (H)
- **145.** Predator exploits their prey species but never over exploits them as extinction of their prey species may leads to extinction of predator species.

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- 148. NCERT XII, Pg# 153
- **151.** NCERT Pg.# 167
- **152.** NCERT XII, Pg.# 242 (E), 264 (H)
- 153. NCERT XII, Pg.# 260 (E), 284 (H) Fig. 15.1
- **154.** Module-3 Pg.# 49
- 158. NCERT XII, Pg# 152
- **161.** NCERT Pg.# 167
- **162.** NCERT XII, Pg.# 242 (E), 268 (H)
- 163. NCERT XII, Pg.# 262 (E), 286 (H) Fig. 15.2
- **164.** Module-3 Pg.# 36
- **165.** Antibiosis is a kind of amensalism (0/–)

- **166.** Age pyramid is characterstic of population.
- 167. In troposphere bad O_3 is created due to secondary pollution photochemical smog which harms organisms.
- 168. NCERT XII, Pg# 153
- **171.** Module-3 Pg.# 1
- **172.** NCERT XII, Pg.# 243 (E), 265 (H)
- 173. NCERT XII, Pg.# 264 (E), 288 (H)
- 174. NCERT XII, Pg.# 228 (E), 249 (H)
- 175. Lack of light in deep aquetic ecosystems
- 178. NCERT XII, Pg# 158

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