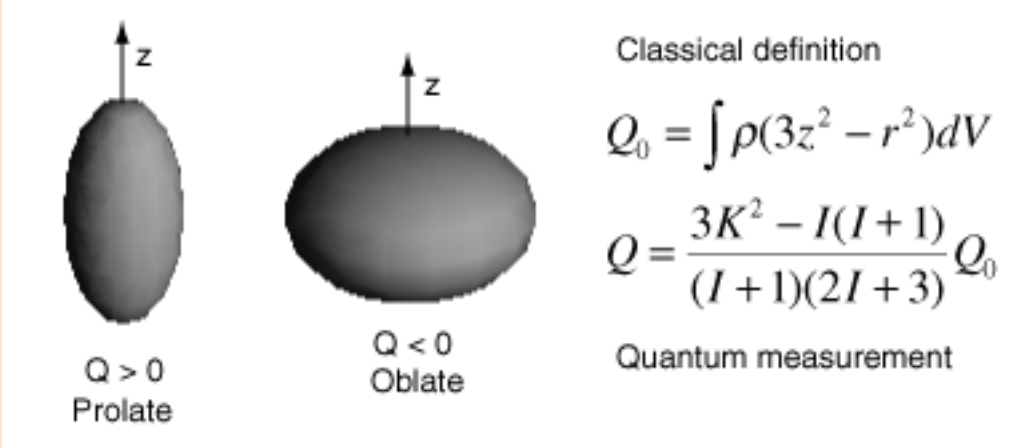


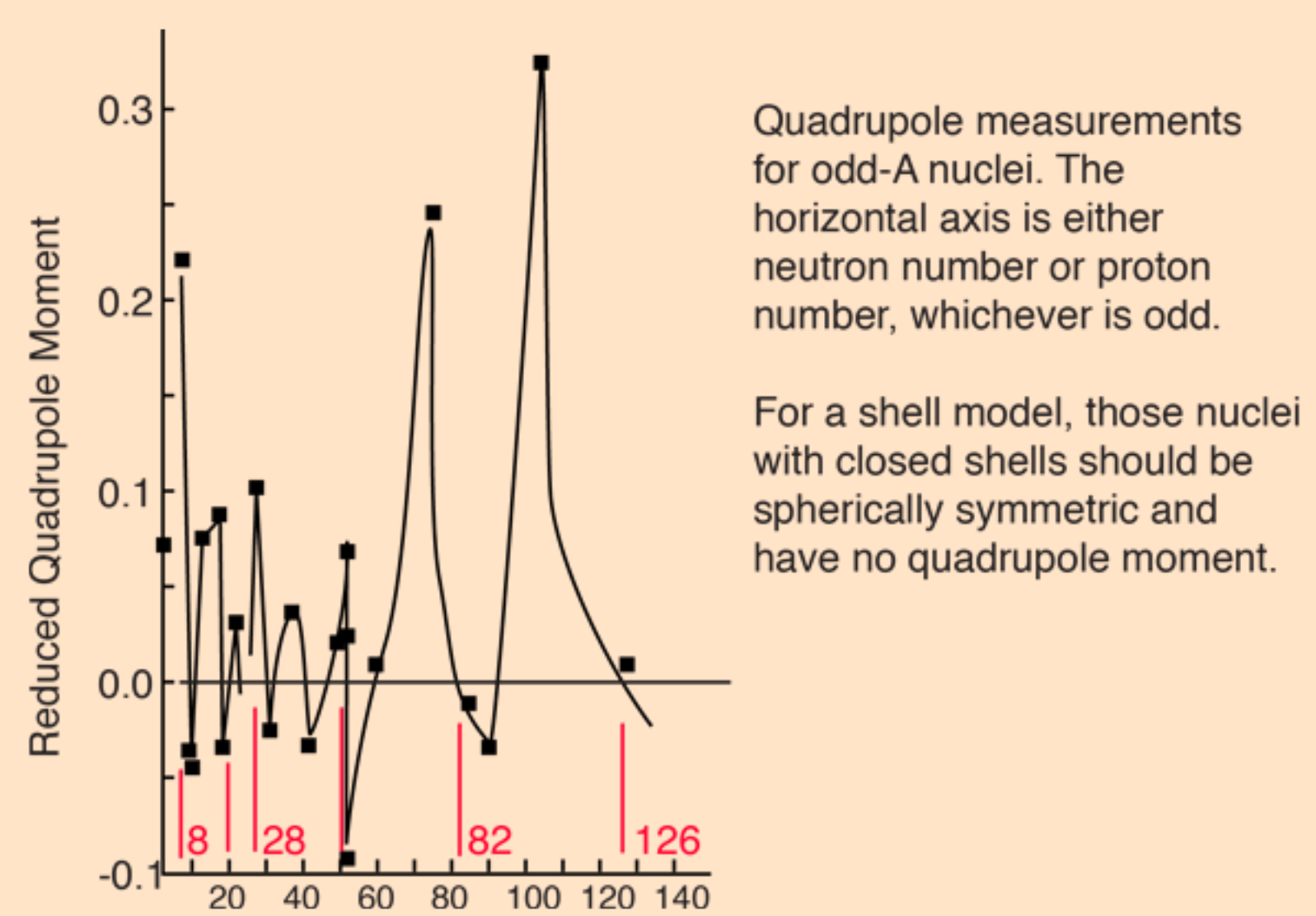
Electric Quadrupole Moments of Nuclei

The nuclear [electric quadrupole moment](#) is a parameter which describes the effective shape of the ellipsoid of nuclear charge distribution. A non-zero quadrupole moment Q indicates that the charge distribution is not spherically symmetric. By convention, the value of Q is taken to be positive if the ellipsoid is prolate and negative if it is oblate.



The quantity Q_0 is the classical form of the calculation represents the departure from spherical symmetry in the rest frame of the nucleus. The expression for Q is the quantum mechanical form which takes takes into account the [nuclear spin](#) I and the projection K in the z-direction.

One of the expectations of the [shell model](#) for the nucleus is that for closed shells the nuclear charge is spherically symmetric. If a nucleus is not spherically symmetric, it will have a non-zero electric quadrupole moment, so the measurement of the quadrupole moment is a test of the shell theory. Since the quadupole moment depends upon the size and charge of the nucleus, a better comparison is obtained by normalizing for those factors, giving what is called a "reduced quadrupole moment". A plot of measured values shows that magic numbers of neutrons or protons correlate with near-zero quadrupole moments.



After Booth & Combley

Electric quadrupole moments of nuclei can be measured brom hyperfine splitting of atomic spectral lines, from quadrupole hyperfine splitting of molecular rotational spectra, and other spectroscopic techniques.

Nuclide	Q (eb)	<p>The electric quadrupole moment has the units of Coulomb x meter². It is sometimes tabulated in units of e x 10⁻²⁴ cm² where e is the electron charge. In nuclear cross-section measurements, the quantity 10⁻²⁴ cm² is called a "barn" and is represented by b. So sometimes the quadrupole unit is written "eb" for electron-barns. Usually the quadrupole moments will be a few tenths in these units until you reach mass numbers around 150. Then quadrupole moments around 2 are common and have been measured as high as 8.</p> <p>Data from V. S. Shirley, Table of Isotopes, Wiley, New York, 1978, Appendix VII.</p>
² H (D)	+0.00288	
¹⁷ O	-0.02578	
⁵⁹ Co	+0.40	
⁶³ Cu	-0.209	
¹³³ Cs	-0.003	
¹⁶¹ Dy	+2.4	
¹⁷⁶ Lu	+8.0	
²⁰⁹ Bi	-0.37	

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