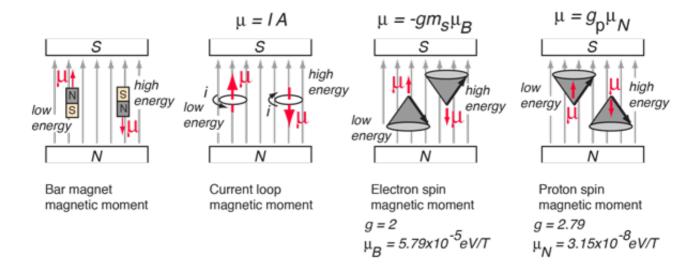
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Nuclear Magnetic Resonance

When the <u>nuclear magnetic moment</u> associated with a <u>nuclear spin</u> is placed in an external magnetic field, the different spin states are given different <u>magnetic potential energies</u>. In the presence of the static magnetic field which produces a small amount of <u>spin polarization</u>, a radio frequency signal of the proper frequency can induce a transition between spin states. This "spin flip" places some of the spins in their higher energy state. If the radio frequency signal is then switched off, the <u>relaxation</u> of the spins back to the lower state produces a measurable amount of RF signal at the resonant frequency associated with the spin flip. This process is called Nuclear Magnetic Resonance (NMR).

A magnetic dipole moment (usually just called "magnetic moment") in a magnetic field will have a potential energy related to its orientation with respect to that field.



Note that the <u>electron spin magnetic moment</u> is opposite to the <u>electron spin</u> while the proton spin magnetic moment is in the direction of the proton spin. The electron spin or proton spin will tend to precess around the magnetic field with a frequency traditionally called the <u>Larmor frequency</u>. For a 1 Tesla magnetic field this Larmor frequency would be

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$$\omega_{electron\,spin} = \frac{2\mu_{e}B}{\hbar} = \frac{2 \cdot 2 \cdot \frac{1}{2} (5.79 \times 10^{-5} eV / T)(1T)}{6.58 \times 10^{-16} eV \cdot s} = 1.7608 \times 10^{11} s^{-1}$$

$$v = \frac{\omega}{2\pi} = 28.025 \, GHz \quad Larmor\, frequency$$

$$\omega_{proton\,spin} = \frac{2\mu_{p}B}{\hbar} = \frac{2(2.79)(3.15 \times 10^{-8} eV / T)(1T)}{6.58 \times 10^{-16} eV \cdot s} = 2.6753 \times 10^{8} s^{-1}$$

$$v = \frac{\omega}{2\pi} = 42.5781 \, MHz \quad Larmor\, frequency$$

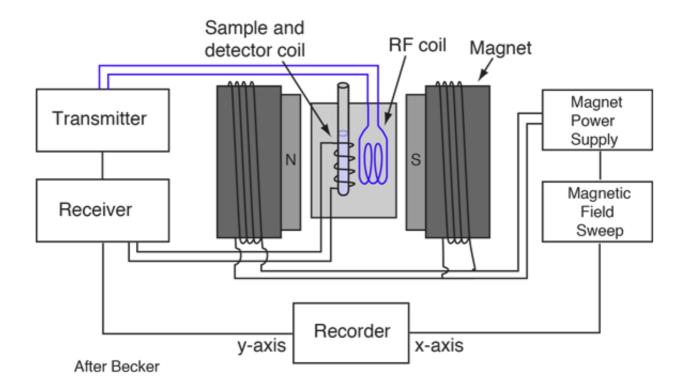
The Larmor frequency can be visualized classically in terms of the precession of the magnetic moment around the magnetic field, analogous to the precession of a spinning top around the gravity field. It can also be visualized quantum mechanically in terms of the quantum energy of transition between the two possible spin states for spin 1/2. This can be expressed as a photon energy according to the Planck relationship. The magnetic potential energy difference is hn = 2mB. The short table of Larmor frequencies below is from Hobbie, Ch 17 and Becker. An extensive list including the magnetic moments and Larmor frequencies of most elements can be found in Appendix A of Becker.

Particle	Spin	w _{Larmor} /B s ⁻¹ T ⁻¹	n/B
Electron	1/2	1.7608 x 10 ¹¹	28.025 GHz/T
Proton	1/2	2.6753 x 10 ⁸	42.5781 MHz/T
Deuteron	1	0.4107 x 10 ⁸	6.5357 MHz/T
Neutron	1/2	1.8326 x 10 ⁸	29.1667 MHz/T
²³ Na	3/2	0.7076 x 10 ⁸	11.2618 MHz/T
³¹ P	1/2	1.0829 x 10 ⁸	17.2349 MHz/T
¹⁴ N	1	0.1935 x 10 ⁸	3.08 MHz/T
¹³ C	1/2	0.6729 x 10 ⁸	10.71 MHz/T
¹⁹ F	1/2	2.518 x 10 ⁸	40.08 MHz/T

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The Larmor frequency of the electron spin is in the <u>microwave region</u> of the electromagnetic spectrum and is used in <u>electron spin resonance</u>.

The precession of the proton spin in the magnetic field is the interaction which is used in proton NMR. As a practical technique, a sample containing protons (hydrogen nuclei) is placed in a strong magnetic field to produce <u>partial polarization</u> of the protons. A strong RF field is also imposed on the sample to excite some of the nuclear spins into their higher energy state. When this strong RF signal is switched off, the spins tend to return to their lower state, producing a small amount of radiation at the Larmor frequency associated with that field. The emission of radiation is associated with the "<u>spin relaxation</u>" of the protons from their excited state. It induces a radio frequency signal in a detector coil which is amplified to display the NMR signal.



Since the Larmor frequency of the detected signal is proportional to the applied magnetic field, changing the magnitude of that field produces a different detected frequency. Placing a magnetic field gradient across a sample allows you to locate the source of the proton NMR signal in the sample. This is used to great advantage in the medical imaging process