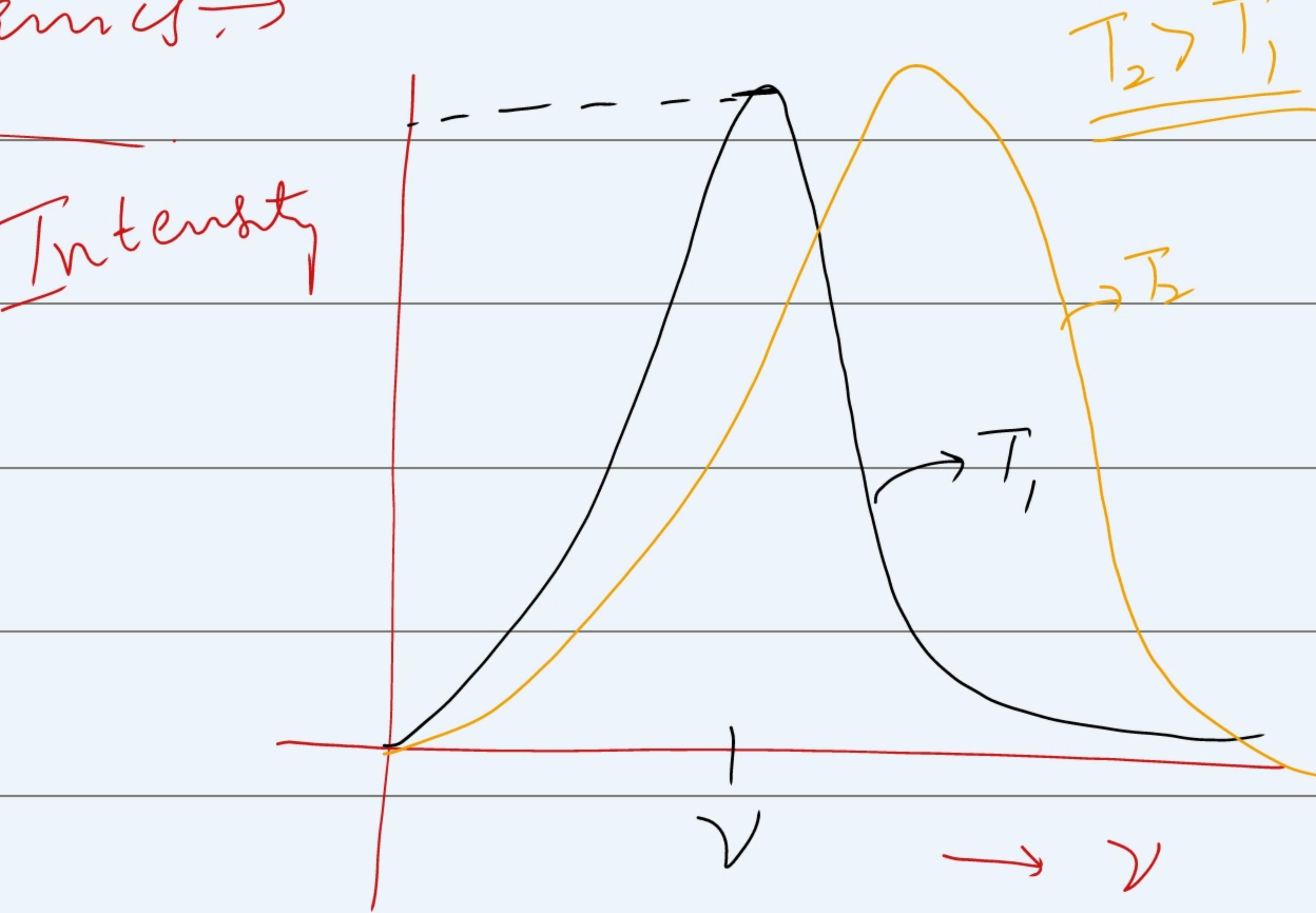
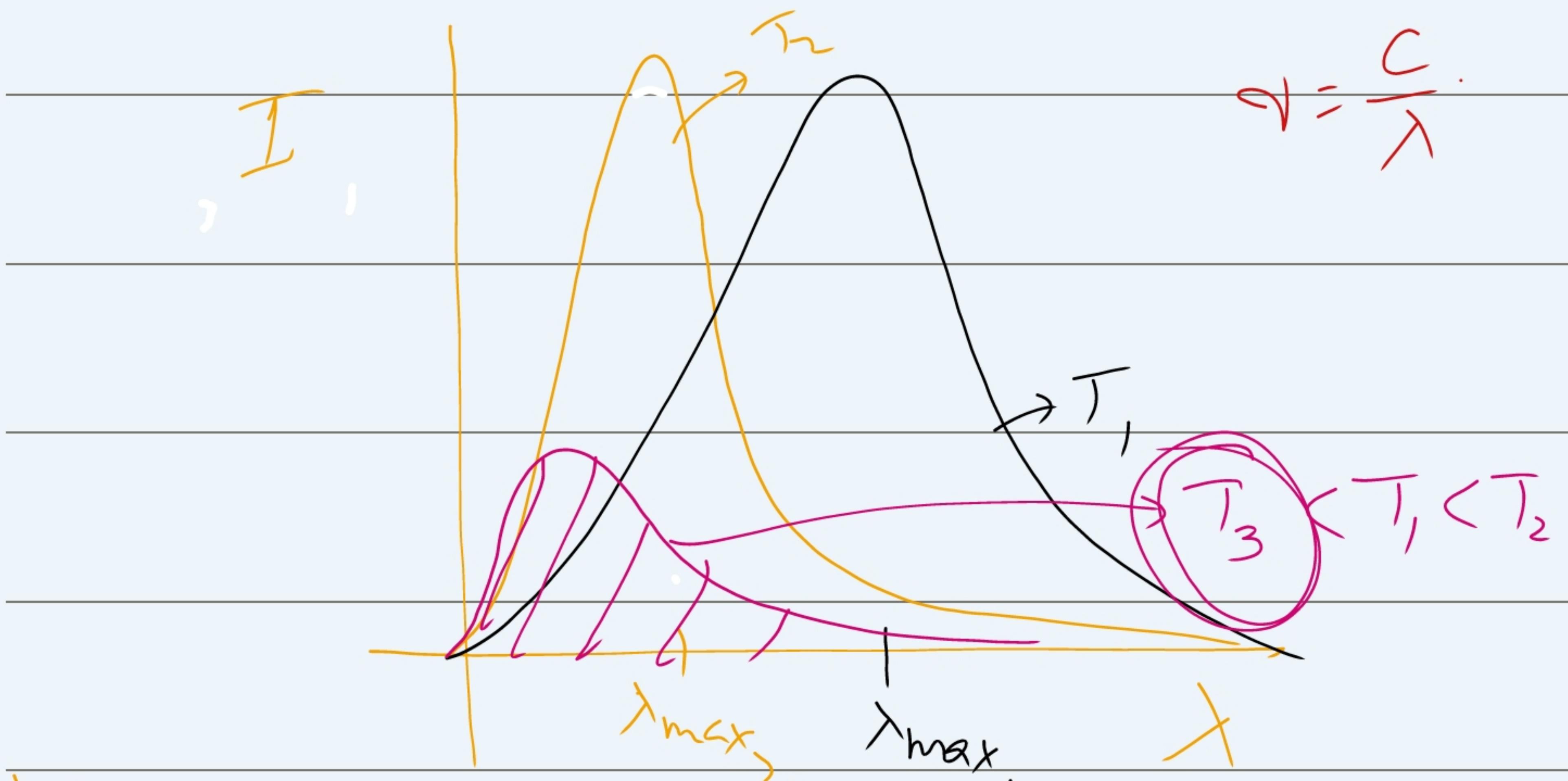
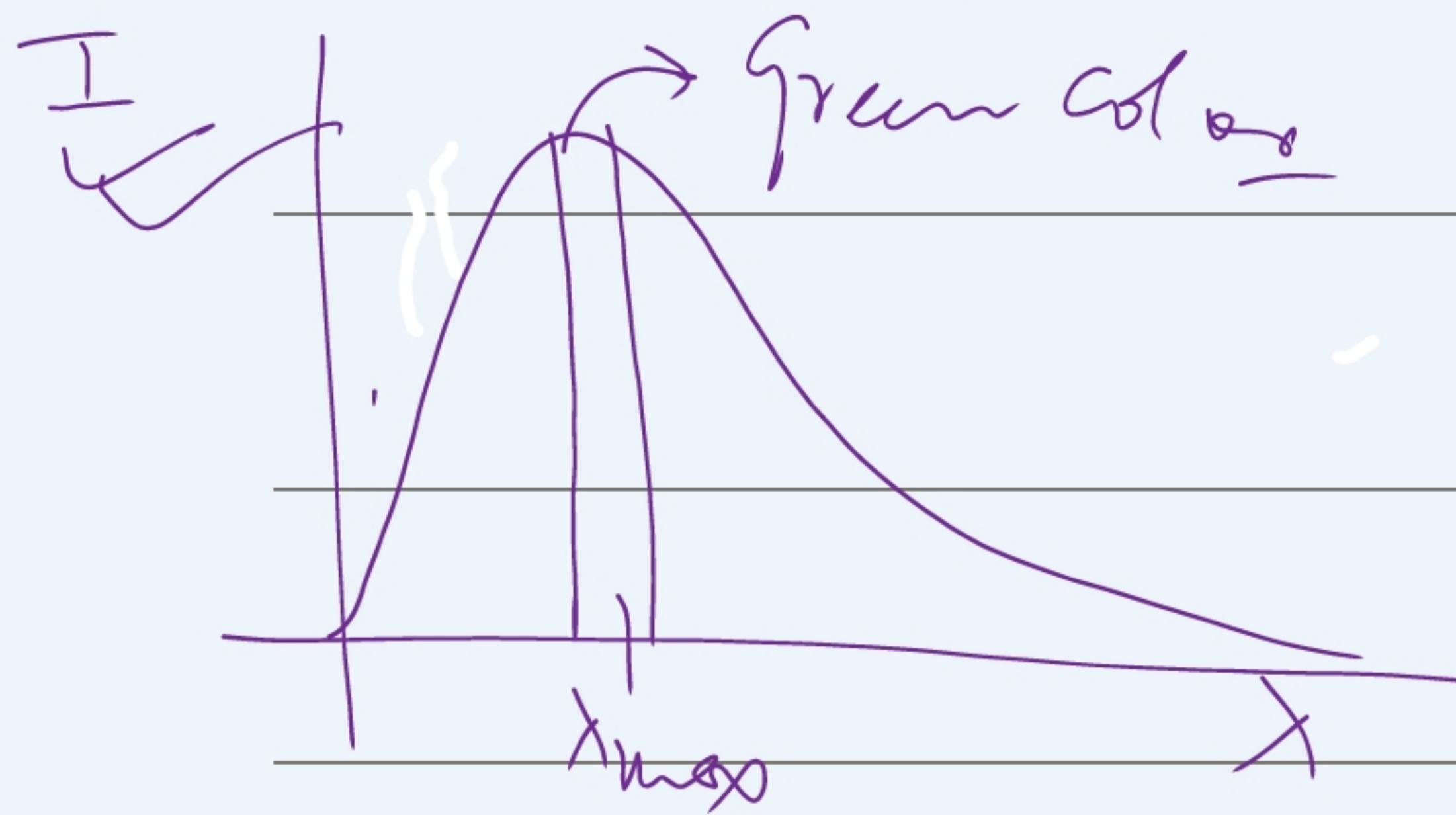


01/11/2023

Failures of Classical Mechanics \rightarrow



Wien's Displacement



$$\lambda_{\text{max}} T = 2.898 \times 10^{-3} \text{ mK}$$

Stefan Boltzmann Law

$$R(T) \propto T^4$$

$$\lambda_{\text{max}} 6000 \text{ K} = 2.898 \times 10^{-3} \text{ m}$$

$$\lambda_{\text{max}} \approx 480 \times 10^{-9} \text{ m}$$

Postulates of Q.M.

1. $\Psi(x, y, z, t)$:

$$= A + iB$$

$|\Psi|^2$ Born Approximation
= P.D $\Psi^* \Psi = A^2 + B^2$

Normalization:



$$\int_a^b \Psi^* \Psi dx = 1 = \int_{-\infty}^{+\infty} \Psi^* \Psi dx$$

Well-behaved:

Single-valued.

Continuous & differentiable

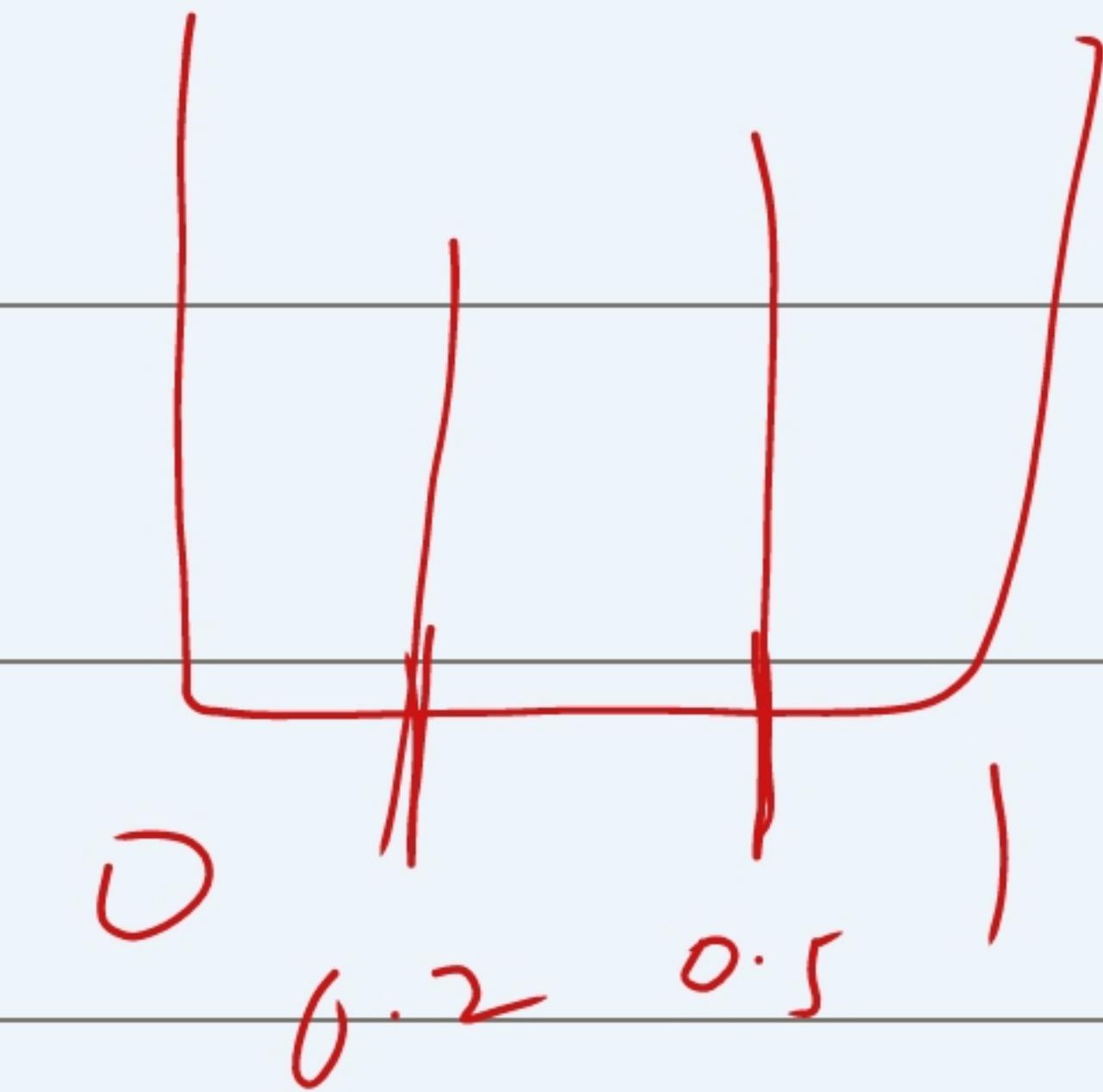
$\frac{\partial \Psi}{\partial x}, \frac{\partial^2 \Psi}{\partial x^2}$ Continuous & differentiation.

Square integrable

$$\underline{\psi}(x) = \underline{a}^x, [0, 1]$$

$$\int_{0.2}^{0.5} a^2 x^2 dx = a^2 \frac{x^3}{3} \Big|_{0.2}^{0.5}$$

$$\int_a^b \psi^* \varphi dx$$



2. How to find wavefunction: Schrödinger Eqn:

$$\frac{\partial^2 \psi}{\partial x^2} = \frac{1}{V^2} \frac{\partial^2 \psi}{\partial t^2}, \quad \frac{A \sin(Kx - \omega t)}{A \cos(Kx - \omega t)}$$

$$E = K.E + P.E$$

$$\psi(x, t) = A e^{i [Kx - \omega t]}$$

$$= \frac{p^2}{2m} + V(x, y, z, t)$$

$$= A e^{-i [\omega t - Kx]}$$

$$= A e^{-i \frac{2\pi}{\lambda} \left[\gamma t - \frac{x}{\lambda} \right]}$$

$$\frac{2\pi}{\lambda} = \frac{2\pi p}{h}$$
$$= \frac{p}{\hbar}$$

$$e^{i\varphi} = \cos \theta + i \sin \theta$$

$$\omega = 2\pi\gamma, K = 2\pi/\lambda$$

$$\boxed{\psi(x, t) = A e^{-i \left[\frac{E}{\hbar} t - \frac{p}{\hbar} x \right]}}$$

$$E = h\nu \Rightarrow \nu = \frac{E}{h}$$

$$2\pi\gamma = \frac{2E\pi}{h} = \frac{E}{t}$$

$$t_h = \hbar/2\pi$$

$$\Psi(x, t) = A e^{-\frac{i}{\hbar} [E t - Px]}$$

$$\frac{\partial \Psi}{\partial x} = \frac{iP}{\hbar} \Psi$$

$$\Rightarrow P\Psi = \frac{\hbar}{i} \frac{\partial \Psi}{\partial x}, \boxed{P = \frac{\hbar}{i} \frac{\partial}{\partial x}}$$

$$\frac{\partial \Psi}{\partial t} = -\frac{iE}{\hbar} \Psi \Rightarrow E\Psi = \frac{\hbar}{i} \frac{\partial \Psi}{\partial t}$$

$$\boxed{E\Psi = i\hbar \frac{\partial \Psi}{\partial t}}$$

$$\boxed{\hat{E} = i\hbar \frac{\partial}{\partial t}}$$

4. Measurement of physical quantity \rightarrow

$$\hat{P} \Psi = a \Psi$$

$$\hat{E} \Psi = b \Psi$$

$$\hat{P}^2 = \left(\frac{\hbar}{i}\right)^2 \frac{\partial}{\partial x} \frac{\partial}{\partial x}$$

Free particle: $E = \frac{p^2}{2m} + V(x, y, z, t)$

For any quantum system: $\hat{E}\psi = \frac{\vec{p}^2}{2m}\psi + V\psi$

$$i\hbar \frac{\partial \psi_{(n,t)}}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} \psi_{(n,t)} + V\psi_{(n,t)}$$

Time-dependent
Schrodinger
Eqn.

$$\Psi(x,t) = A e^{-\frac{i}{\hbar} [E t - p_n]} \\ = \underbrace{A e^{-\frac{i}{\hbar} E t}}_{\text{---}} e^{+\frac{i}{\hbar} p_n}$$

~~$$\Psi(x,t) = \underbrace{\Psi(x)}_{\text{---}} e^{-\frac{i}{\hbar} E t}$$~~

$$\frac{\partial \Psi}{\partial t} = -\frac{i}{\hbar} E \Psi(x) e^{-\frac{i}{\hbar} E t} \\ -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} \approx -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} e^{-\frac{i}{\hbar} E t}$$

TDSE

$$i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + U \Psi(x,t)$$

$$\Rightarrow \left(i\hbar \right) \left(-\frac{i}{\hbar} E \right) \Psi(x,t) = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x,t)}{\partial x^2} + U \Psi(x,t)$$

TDSE

$$E \Psi(x,t) = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x,t)}{\partial x^2} + U \Psi(x,t)$$

TDSE

$$\Psi''(x) + \frac{2m}{\hbar^2} \left[E - U \right] \Psi(x) = 0$$

3. Linear & Superposition (morph)

Intensity

$$e^- \times \psi_1 \quad P_1 = \psi_1^* \psi_1$$

$$e^- \times \psi_2 \quad P_2 = \psi_2^* \psi_2$$

$$\frac{\partial^2 y}{\partial x^2} = \frac{1}{b^2} \frac{\partial^2 y}{\partial t^2}$$

① $A \sin(kx - \omega t)$

② $A \cos(kx - \omega t)$

③ ~~$A \sin(kx - \omega t) + B \cos(kx - \omega t)$~~

ψ_1

$$(\psi_1 + \psi_2)^* (\psi_1 + \psi_2)$$

 ψ_2

$$= \underbrace{\psi_1^* \psi_1}_{\text{}} + \underbrace{\psi_1^* \psi_2}_{\text{}} + \underbrace{\psi_2^* \psi_1}_{\text{}} + \underbrace{\psi_2^* \psi_2}_{\text{}}$$

$$= |\psi_1|^2 + |\psi_2|^2 + \underline{\text{Interference term}}$$

5. Expectation Values \rightarrow

$$q_f = \int_{-\infty}^{+\infty} \psi^* \psi \left(\frac{\hbar}{i} \frac{\partial}{\partial x} \right) dx$$

$$= \frac{\hbar}{i} \int_{-\infty}^{+\infty} |\psi|^2 dx$$

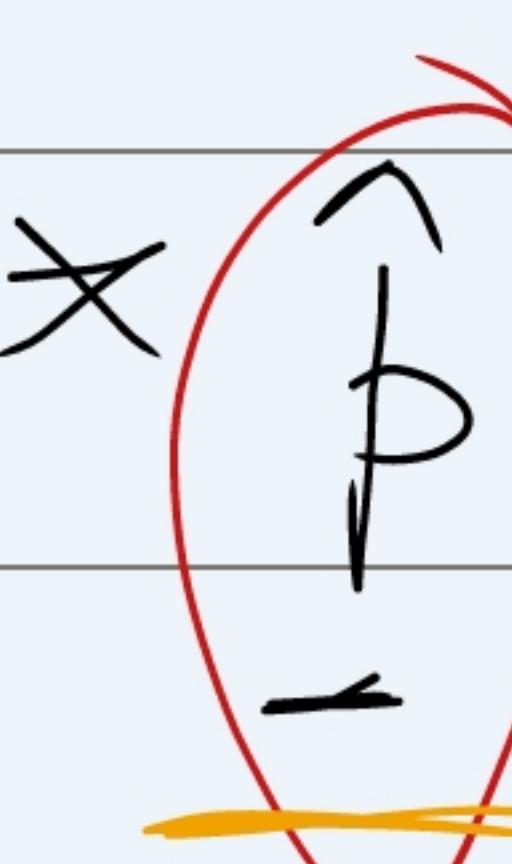
$$\langle x \rangle =$$

$$= \int_{-\infty}^{+\infty} \psi^* x \psi dx$$

$$= \int_{-\infty}^{+\infty} \psi^* \psi dx$$

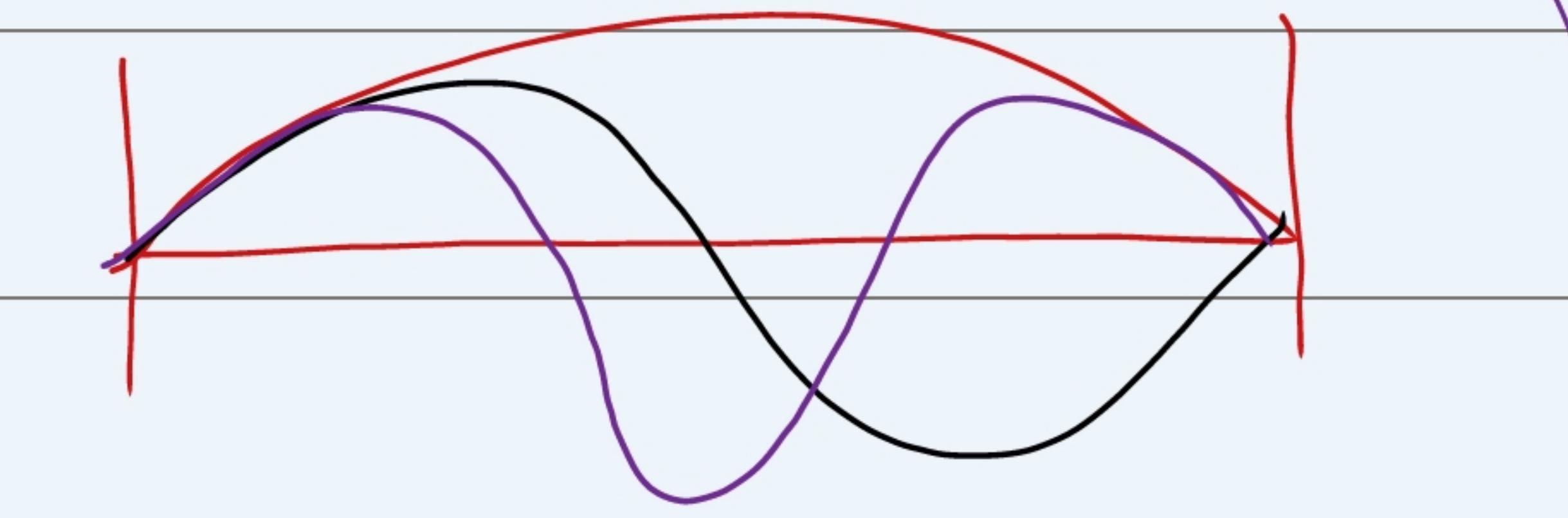
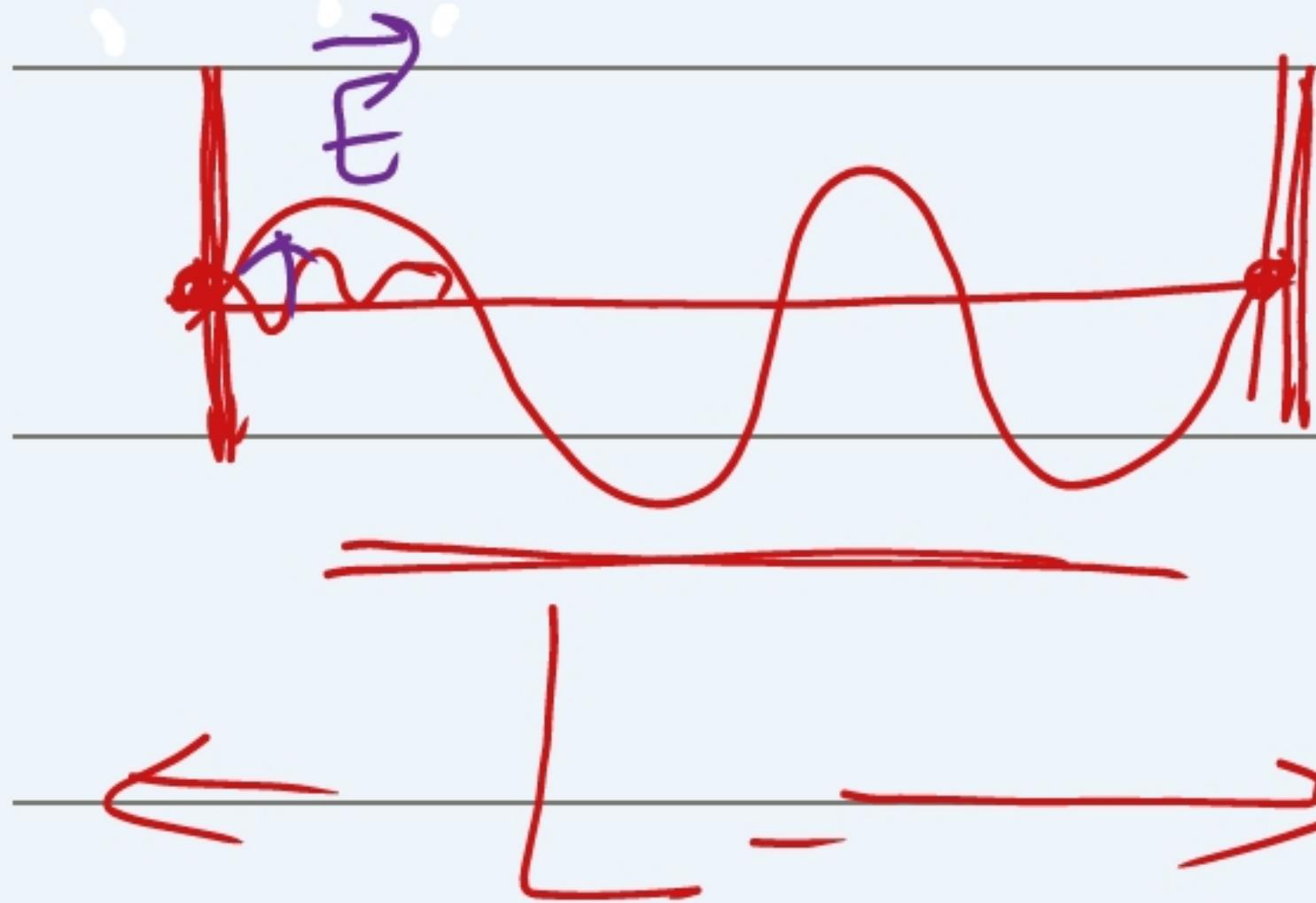
(1)

$$\langle p \rangle = \int_{-\infty}^{+\infty} \psi^* \hat{p} \psi dx$$



$$\begin{aligned} & \text{gf} \\ & \int_{-\infty}^{+\infty} \hat{p} \psi^* \psi dx \\ &= \int_{-\infty}^{+\infty} \left(\frac{\hbar}{i} \frac{\partial}{\partial x} \right) |\psi|^2 dx \\ &= \frac{\hbar}{i} \cdot |\psi|^2 \Big|_{-\infty}^{+\infty} \\ &= 0 \end{aligned}$$

Kayleigh-Jeans Law $\Rightarrow \frac{c}{\lambda_m} = \lambda_m = \frac{2L}{j}$



$$1 \text{ mode } \frac{\lambda}{2} = L \Rightarrow j_n \lambda = 2L$$

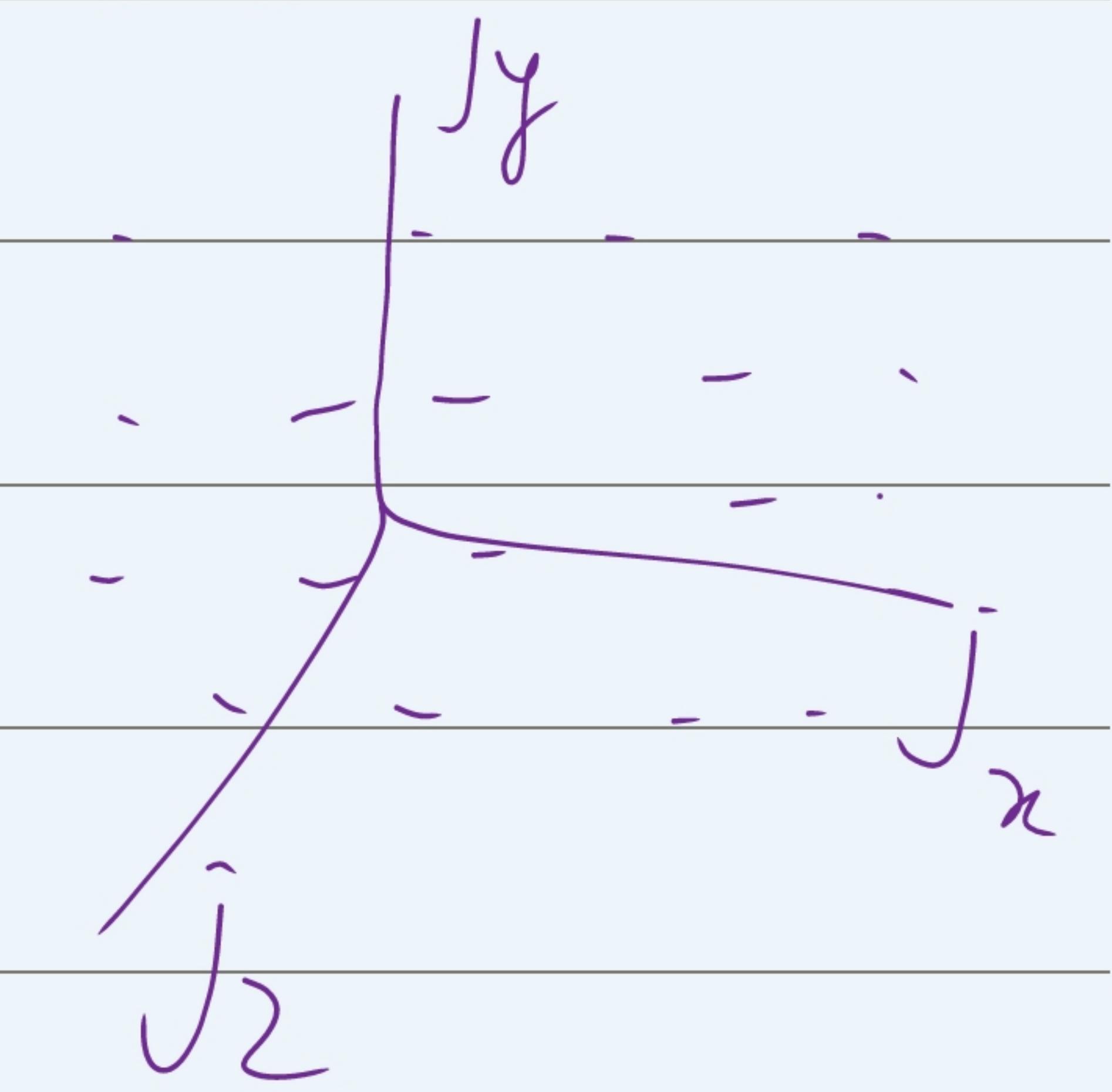
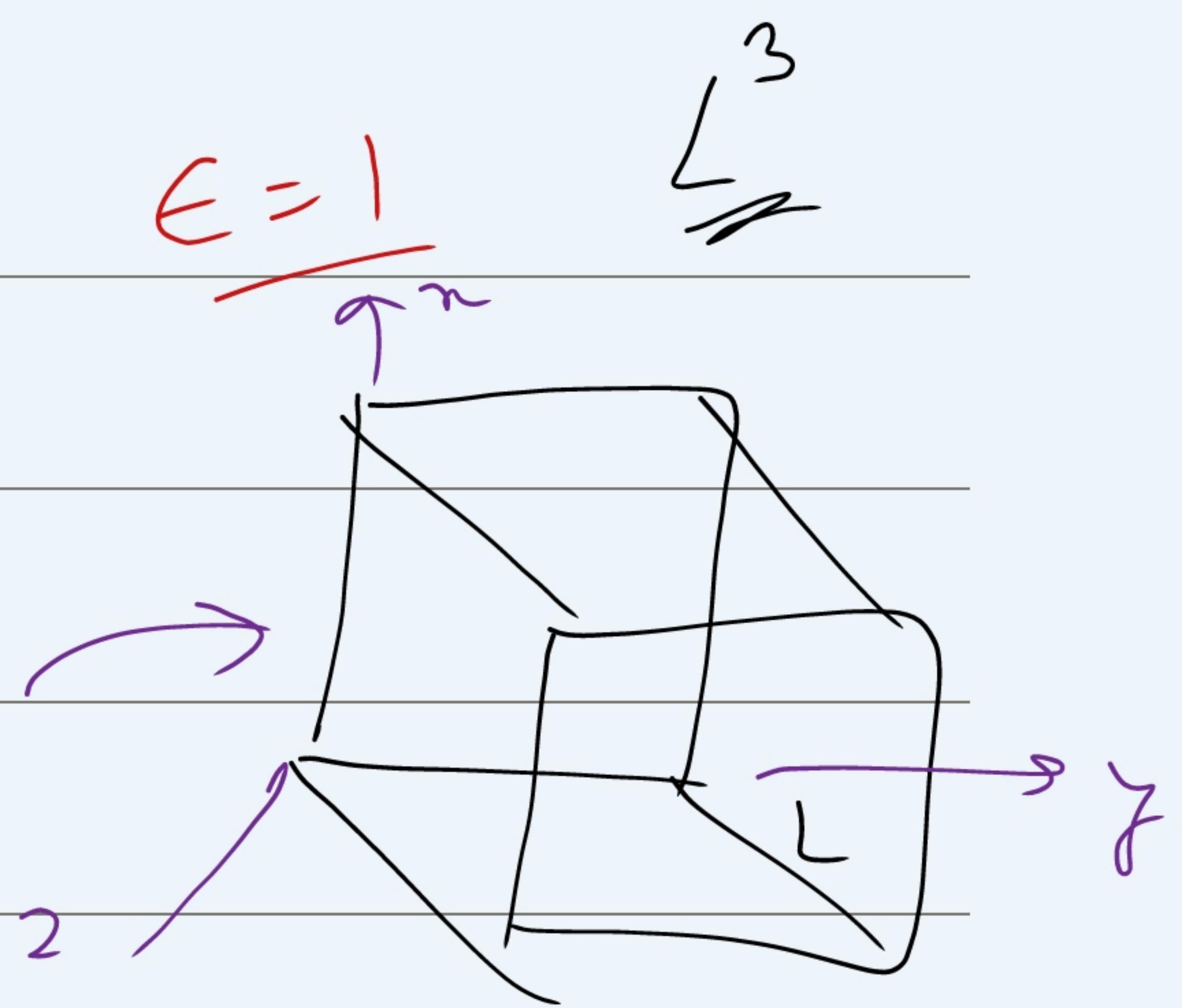
$$\lambda = L$$

$$\frac{3\lambda}{2} = L$$

$$j_y \lambda = 2L$$

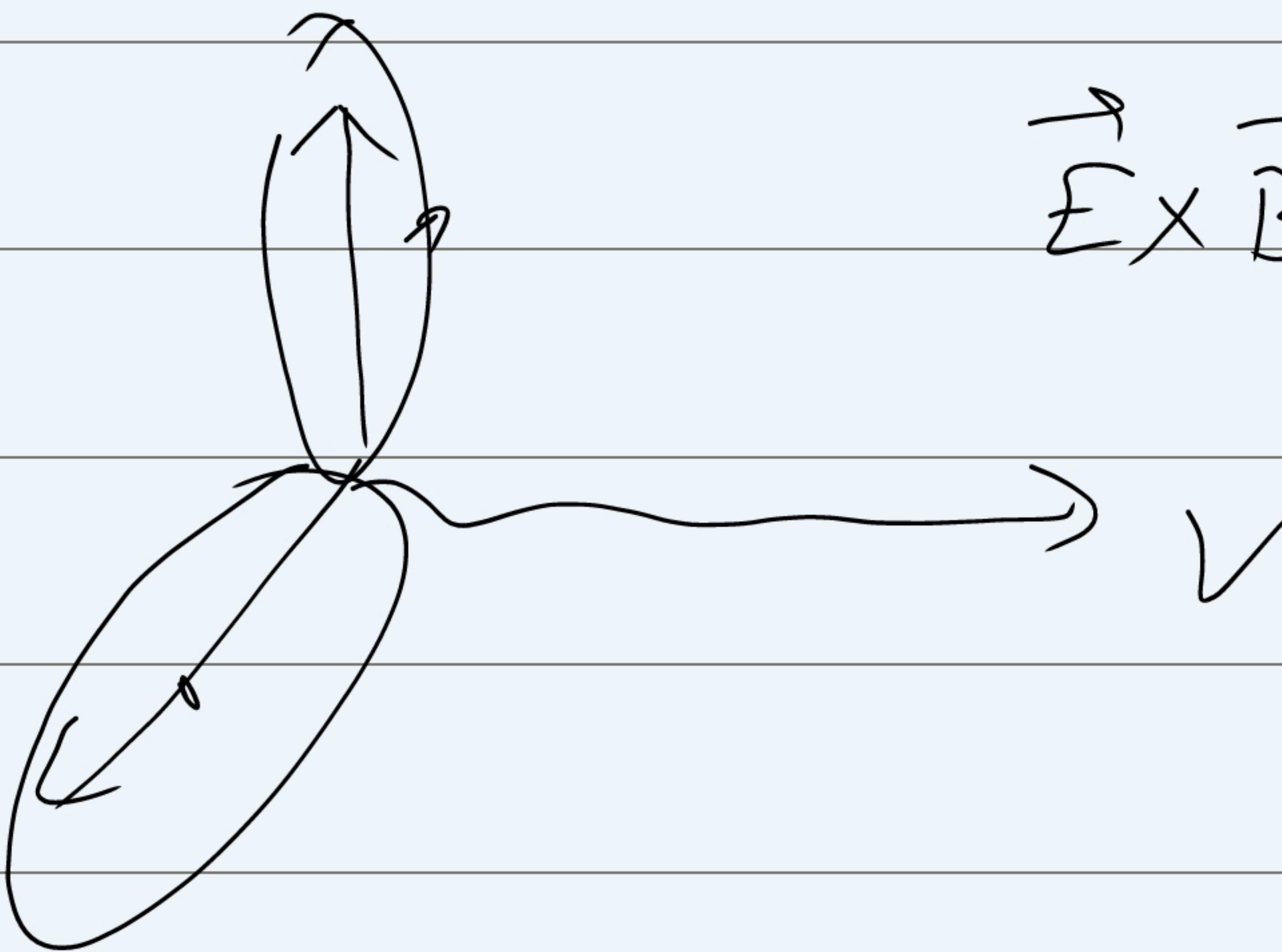
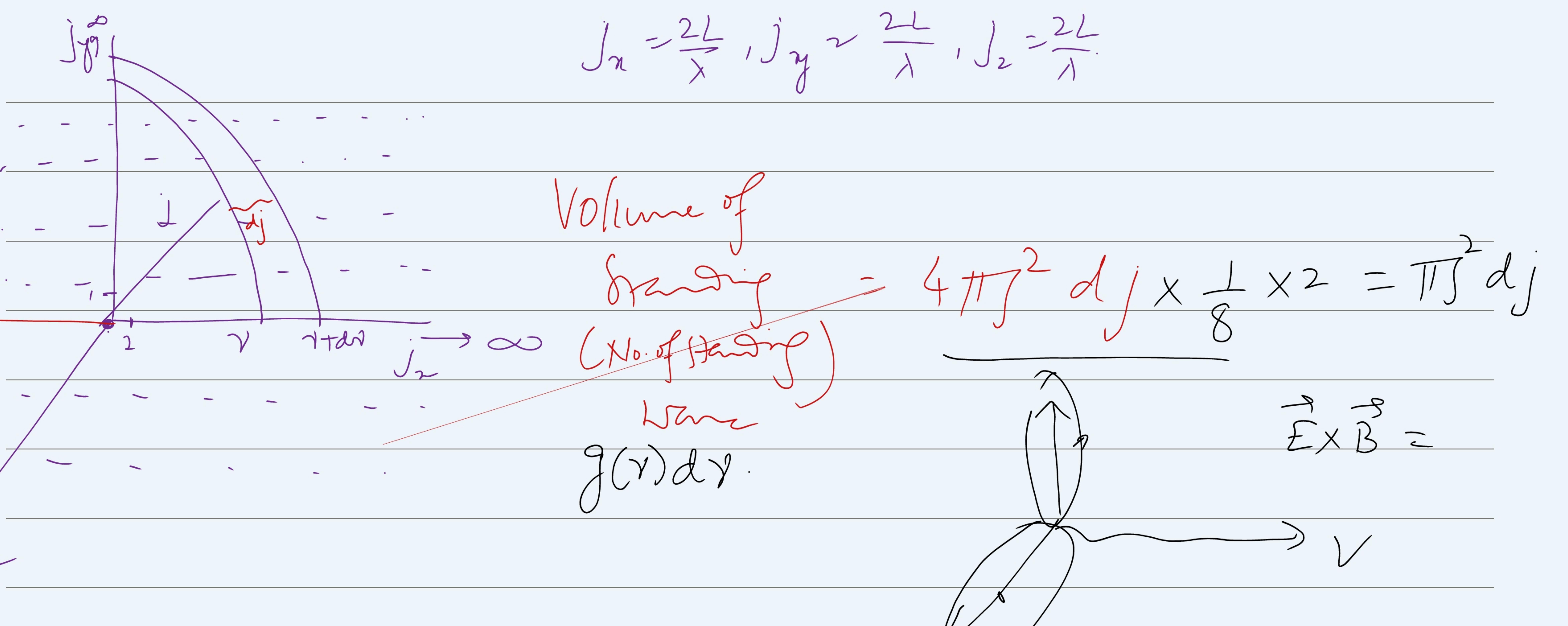
$$j_z \lambda = 2L$$

$$j_x^2 + j_y^2 + j_z^2$$



For any arbitrary direction, Standing Wave

$$= \sqrt{j_x^2 + j_y^2 + j_z^2}$$



$$g(\gamma) d\gamma = \pi j^2 dj$$

$$= \pi \left(\frac{2L\gamma}{c} \right)^2 \cdot \left(\frac{2L}{c} \right) d\gamma$$

$$j = \frac{2L}{\pi} = \frac{2L\gamma}{c}$$

$$dj = \frac{2L}{c} d\gamma$$

$$g(\gamma) d\gamma = \frac{8\pi L^3}{c^3 k} \gamma^2 d\gamma$$

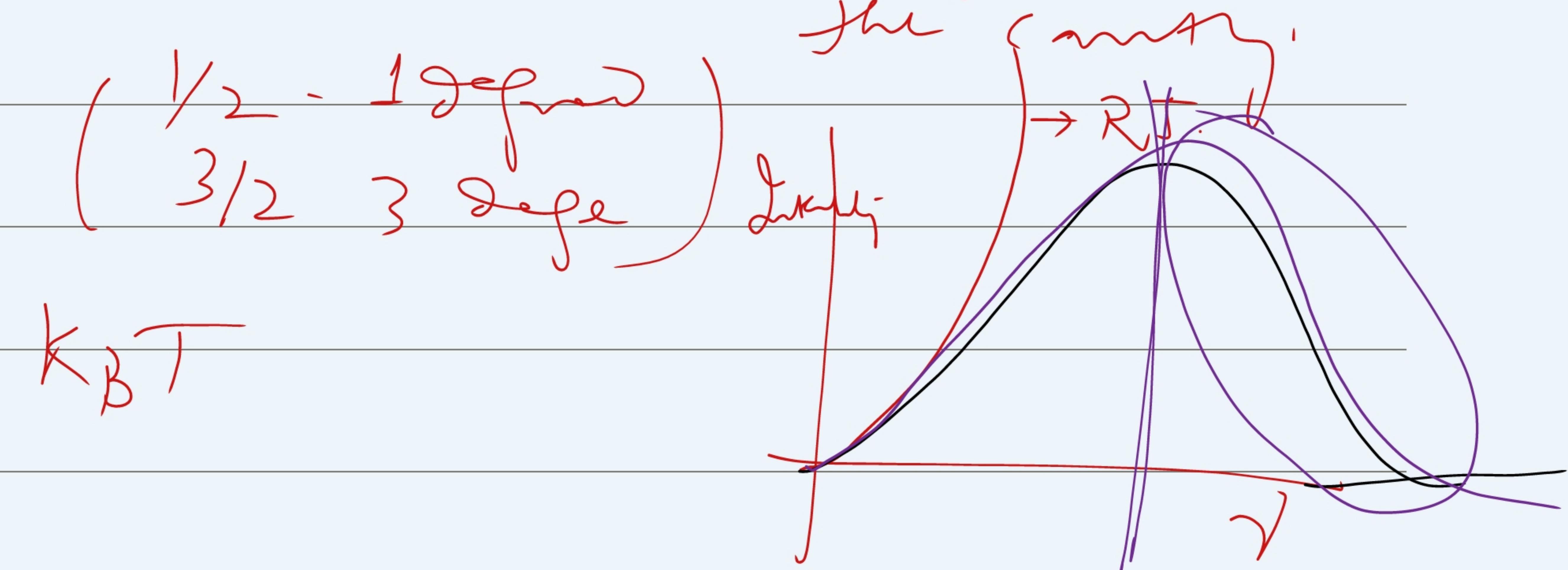
No. of Standing waves.

$\frac{8\pi}{c^3} \gamma^2 d\gamma$ = Density of Standing wave formed in the cavity.

Volume $h\nu$

$RJ. \frac{e^{h\nu/kT}}{C^3} = \bar{\epsilon} = K_B T$

Radiant Intensity = $\frac{8\pi \gamma^2 d\gamma \cdot K_B T}{C^3}$



Planck's Radiation Law:-

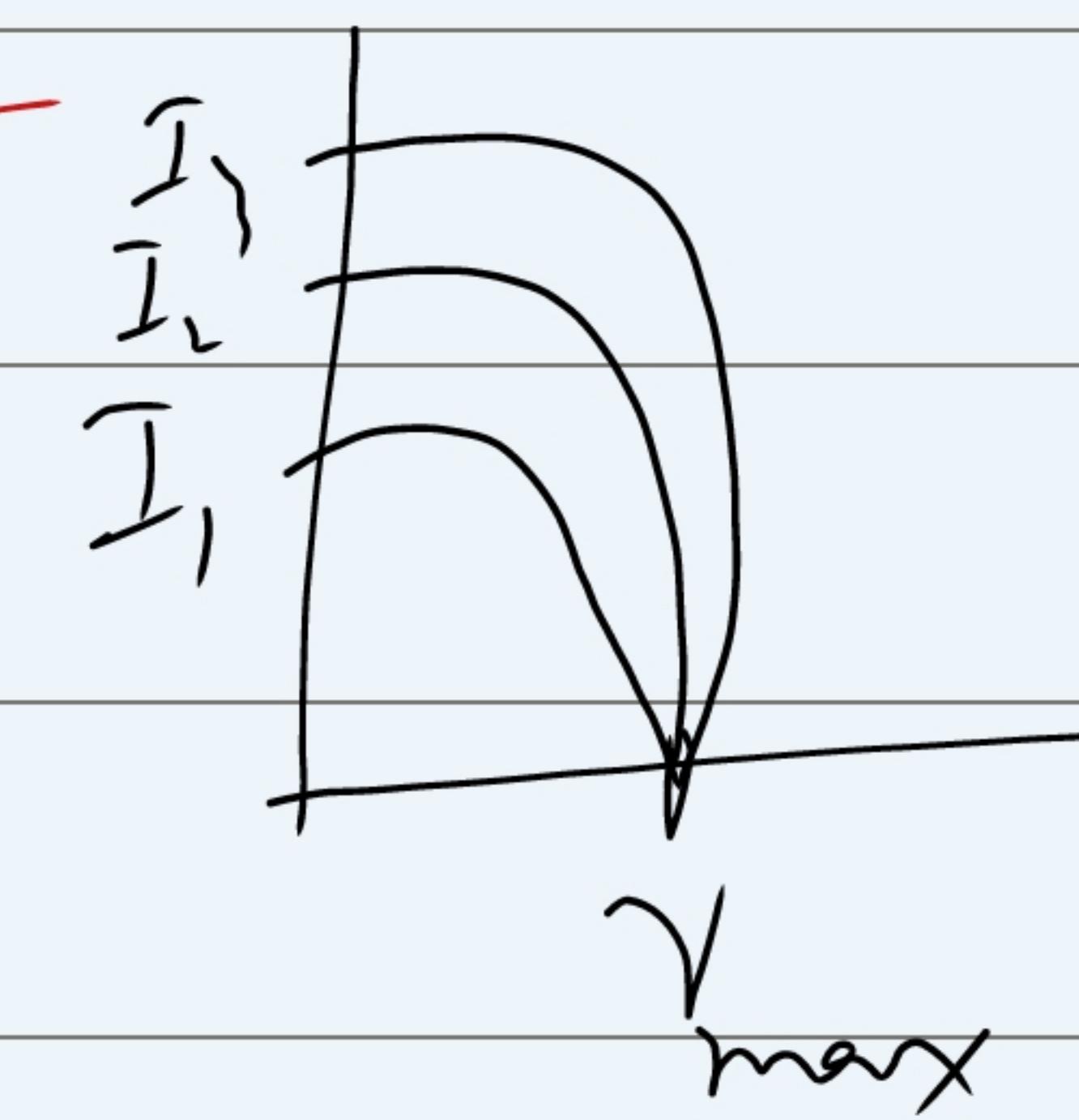
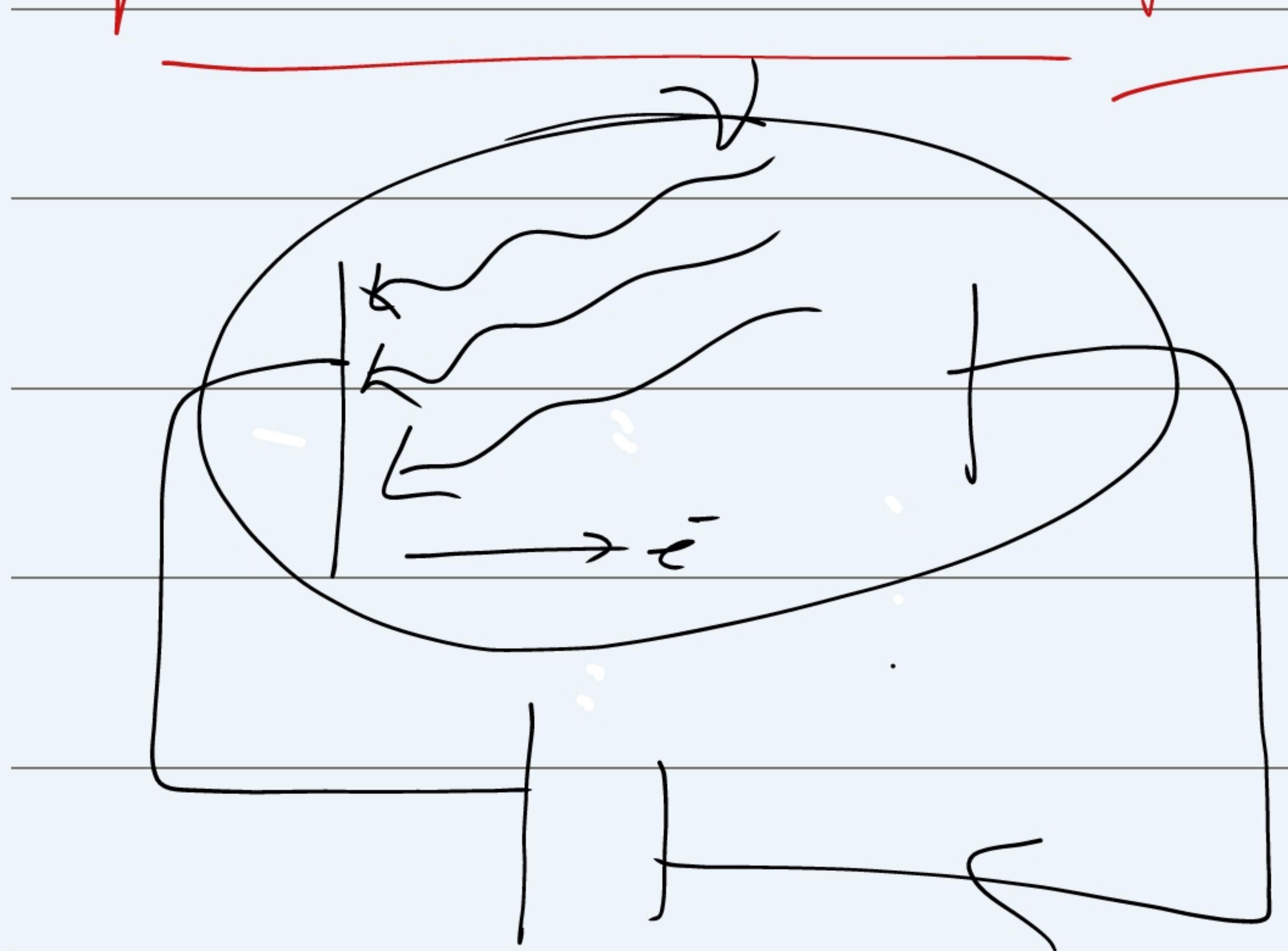
$$I = \left(\frac{8\pi\nu^2 d\nu}{c^3} \right) \cdot e^{-\frac{h\nu}{kT}}$$

① For large ν , $\frac{h\nu}{kT} \gg 1$, $h\nu \gg kT$, $e^{-\frac{h\nu}{kT}} \rightarrow 0$

② For small ν , $\frac{h\nu}{kT} \ll 1$, $h\nu \ll kT$, $e^{-\frac{h\nu}{kT}} \approx 1 + \frac{h\nu}{kT}$

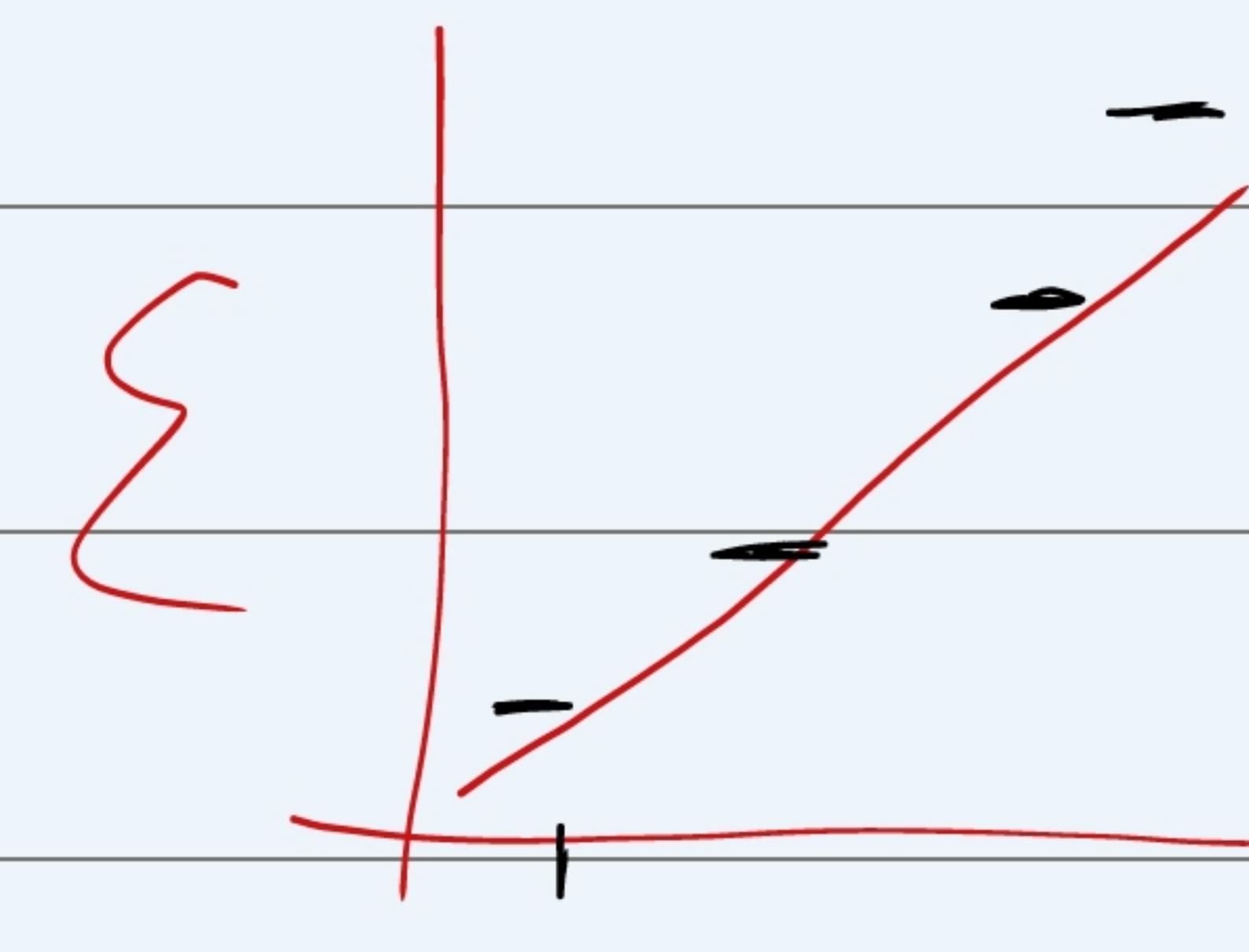
$$e^{-\frac{h\nu}{kT}} = 1 + \alpha + \frac{\alpha^2}{2!} + \dots \approx 1 + \frac{h\nu}{kT}, I =$$

Photoelectric Effect \rightarrow



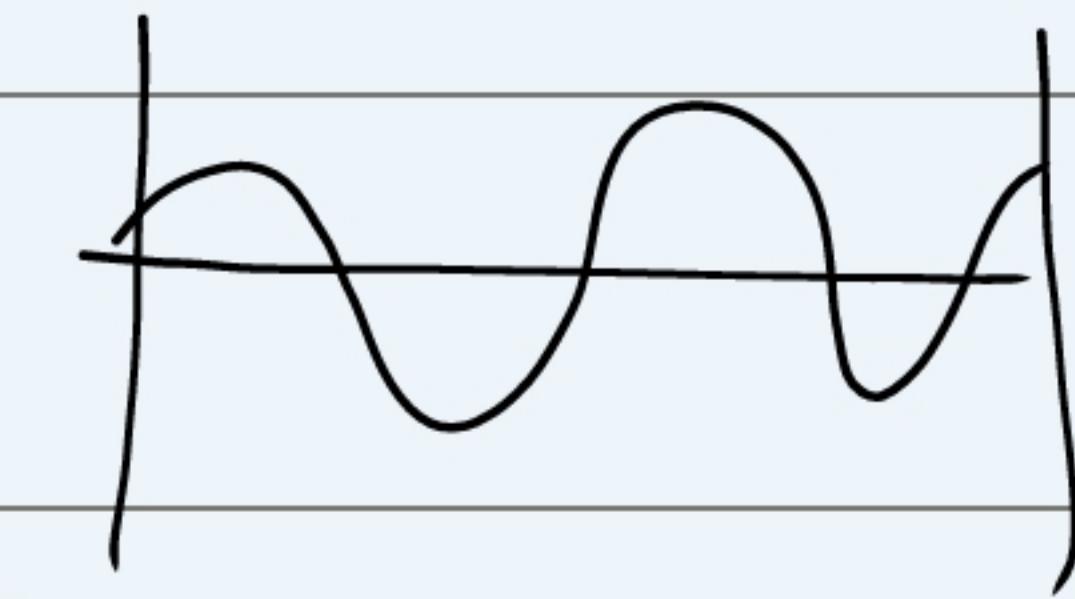
$$\Sigma = k_B T$$

$$\Sigma = nh\nu$$

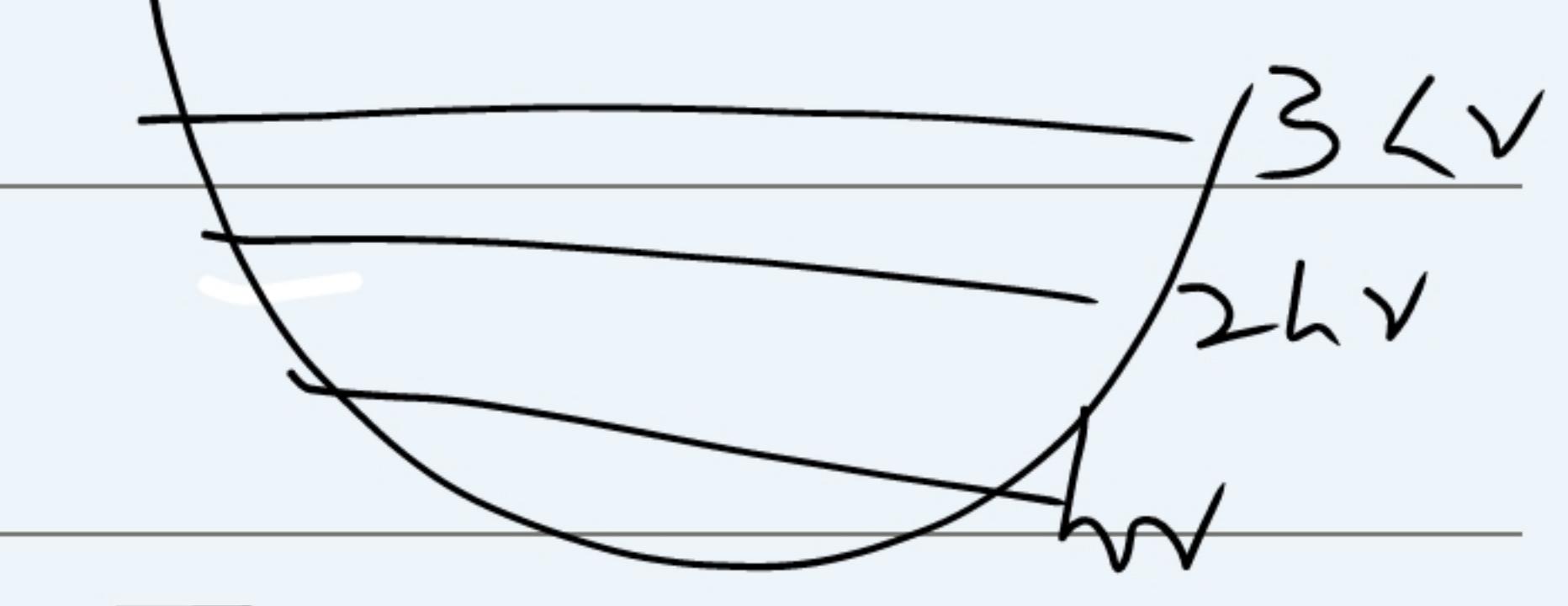
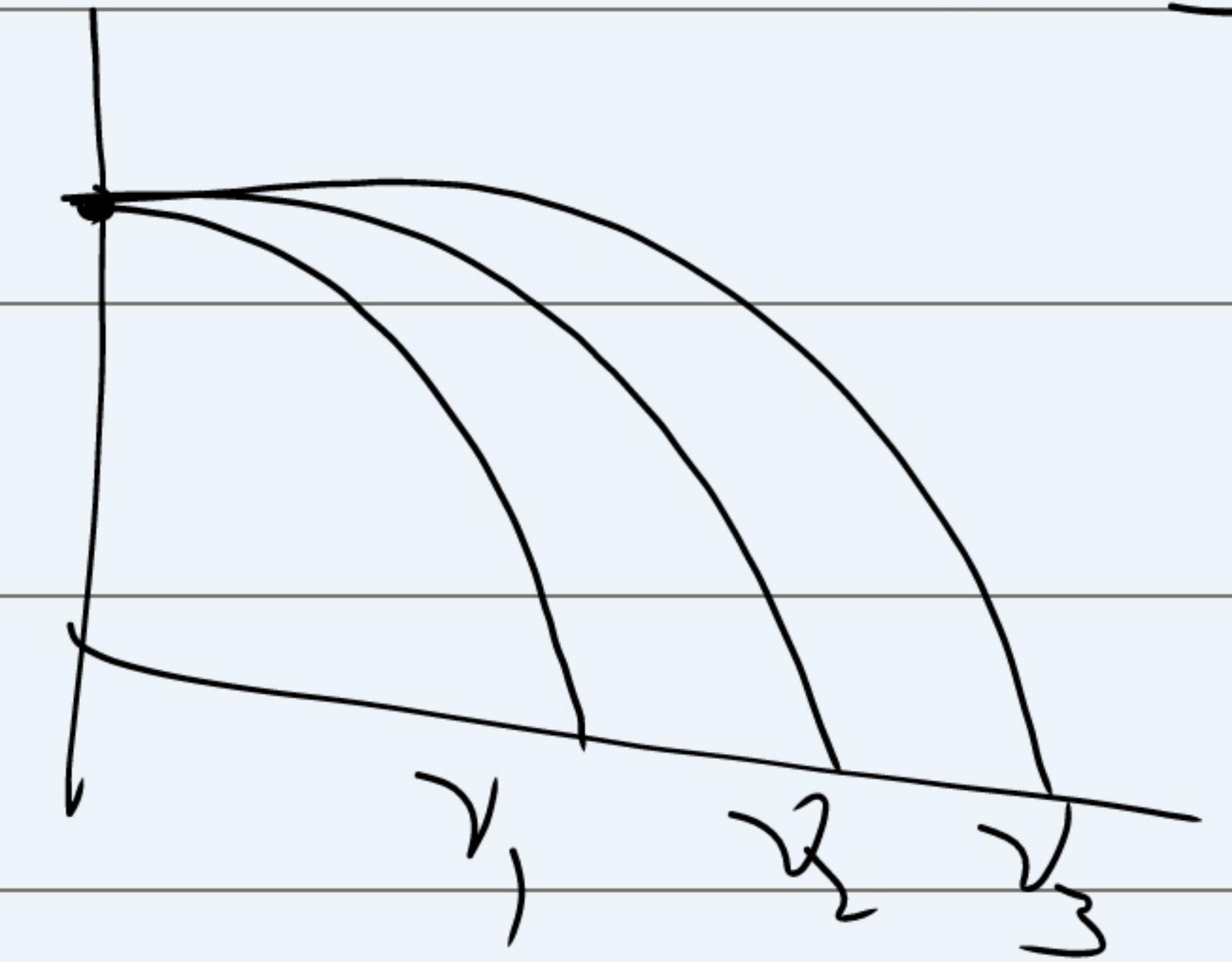


$$\Sigma_1 = h\nu$$

$$\Sigma_2 = 2h\nu$$



$$eV_{\text{max}} = h\nu_0 + h\nu$$



$$E = \frac{1}{2}KA^2$$

③ Compton Effect \rightarrow

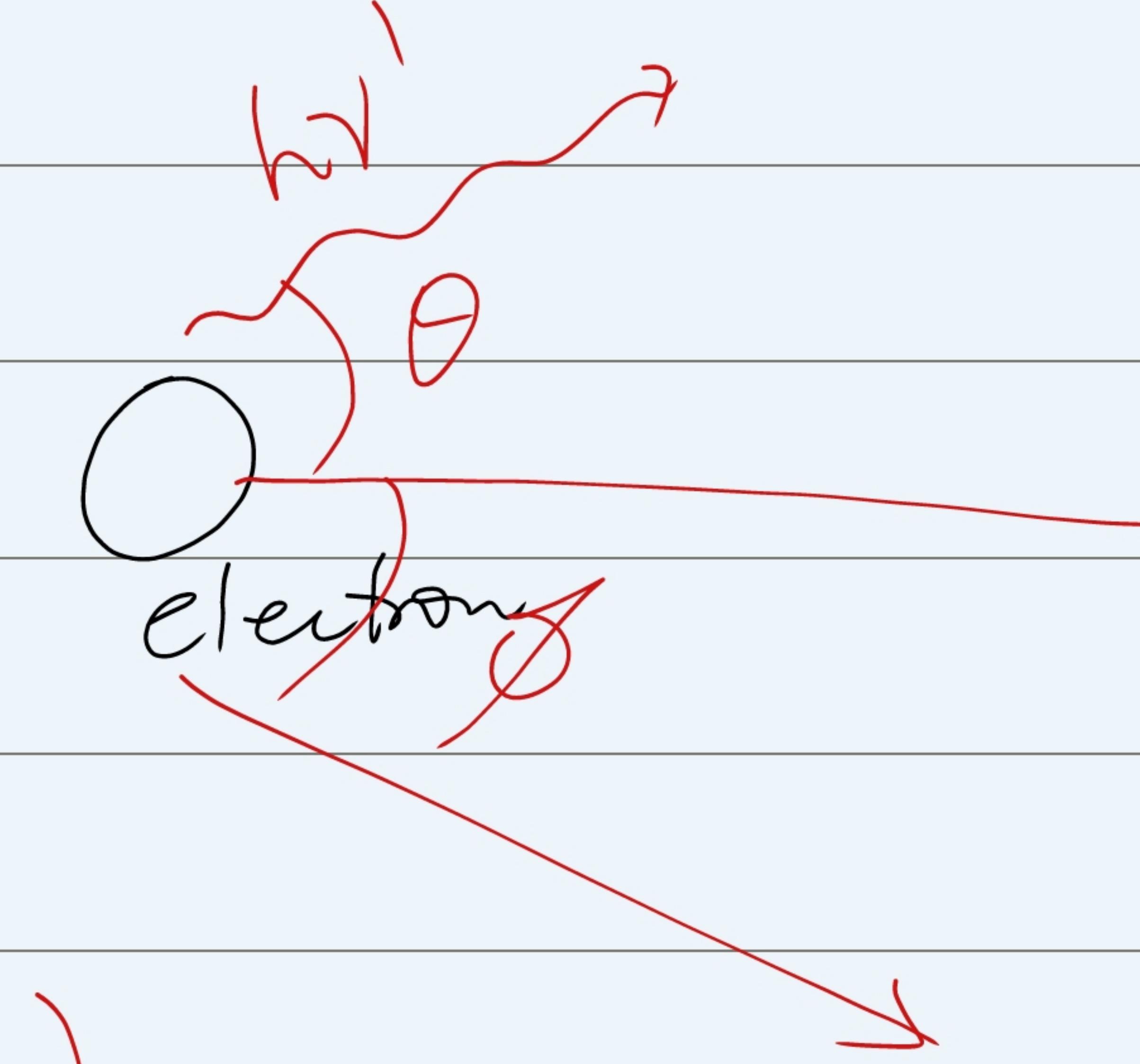
Planck's

$$E = nh\nu$$

$$p = \frac{E}{c}$$

$$= \frac{h\nu}{c}$$

$$h\nu$$



$$h(\nu - \nu') \rightarrow \text{Quantized}$$

$$\frac{h}{mc} [1 - \cos\theta]$$

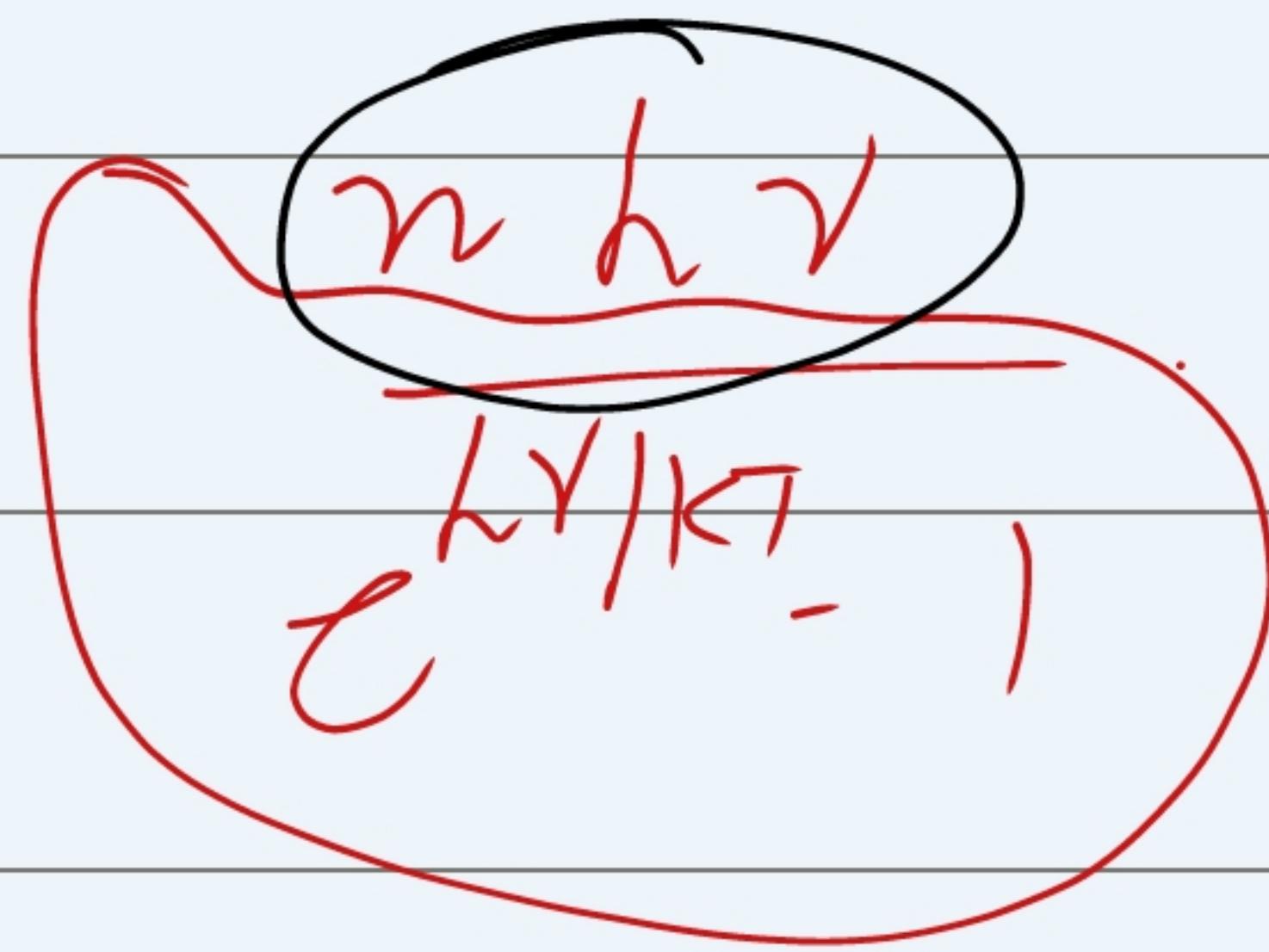
$$\frac{8\pi\nu^2}{c^3}$$

$$\text{Av. Eng} = k_B T \quad \leftarrow R T : I \propto \nu^2 \quad I$$

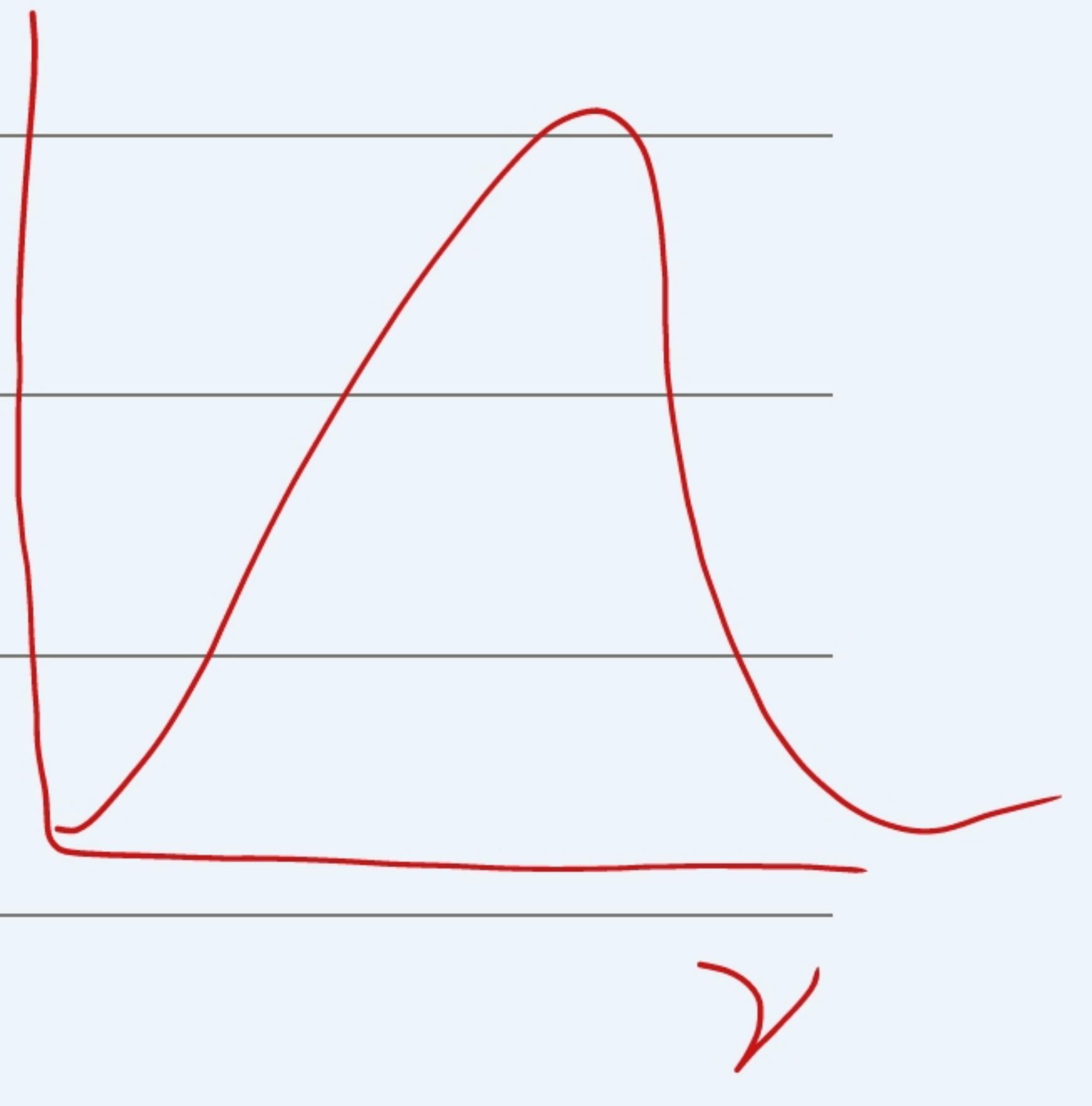
Classical System:

H0:

$$E = \frac{1}{4} K A^2$$



Planck:



$$E_n = n h \nu$$

Compton Effect \Rightarrow

$$E = h\nu$$

A. Beiser

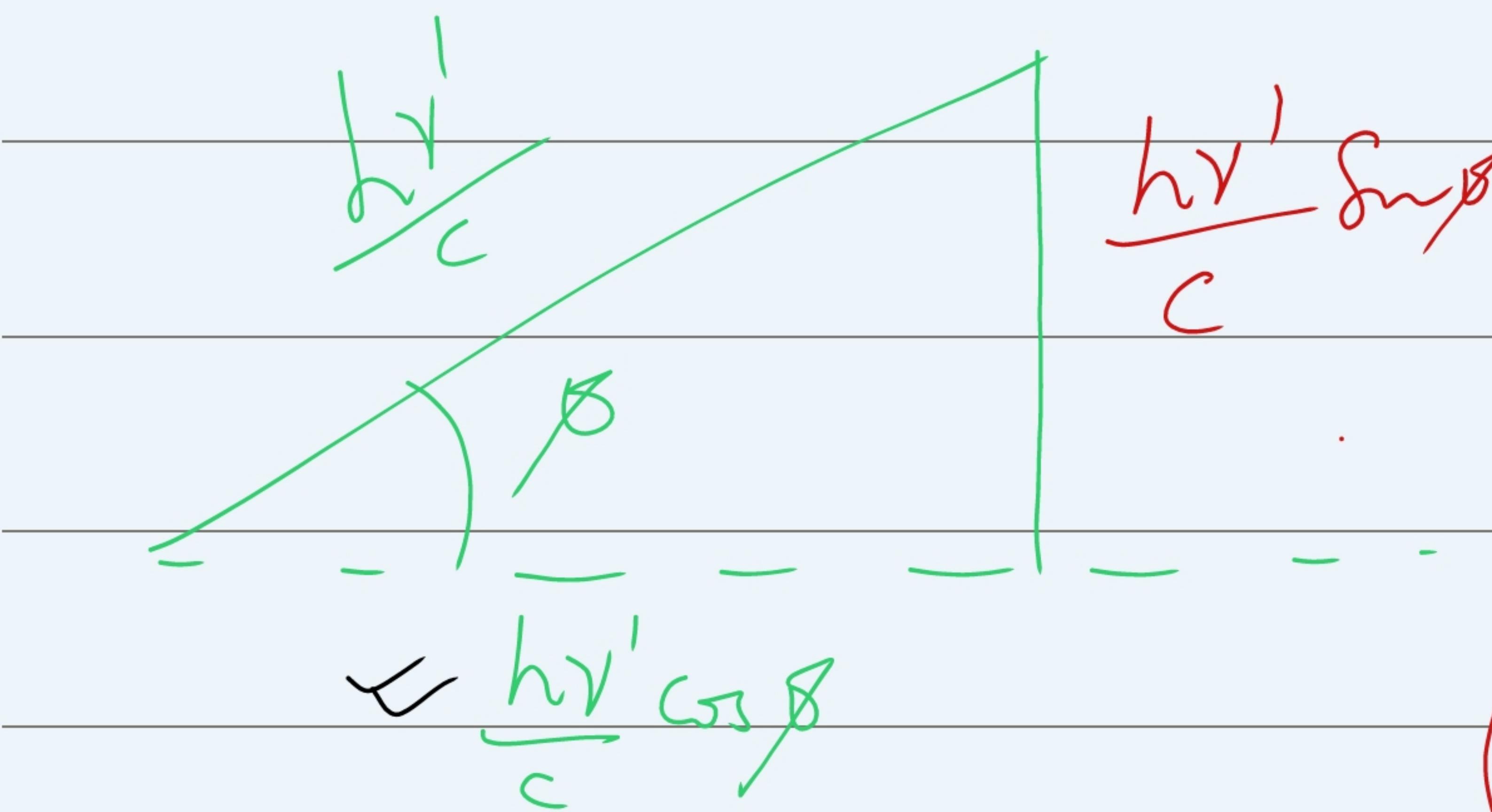
$$p = \frac{E}{c} = \frac{h\nu}{c}$$

① $h\nu'$, $\frac{h\nu}{c}$

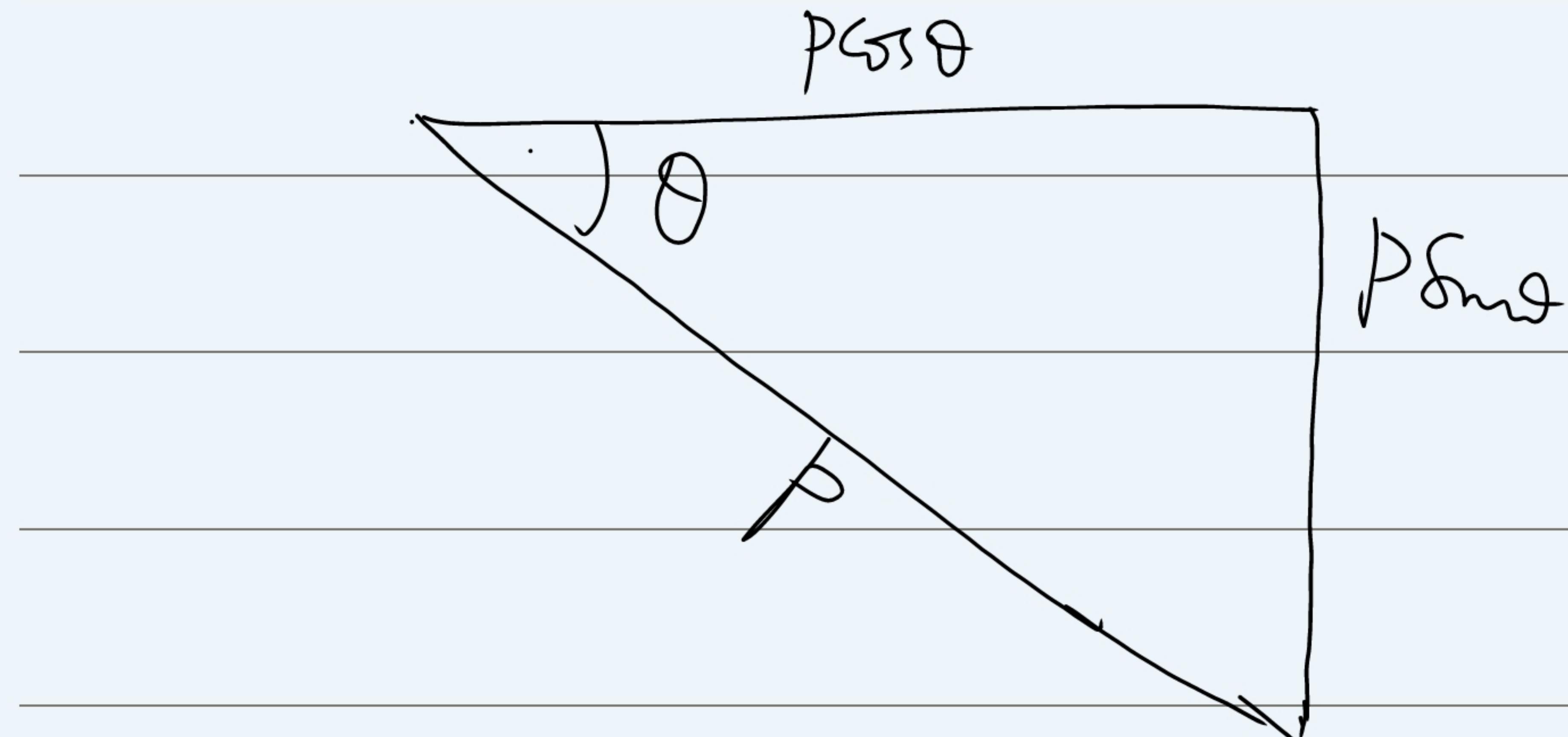
$h\nu$

② θ
Electron

(Particulate nature of wave):



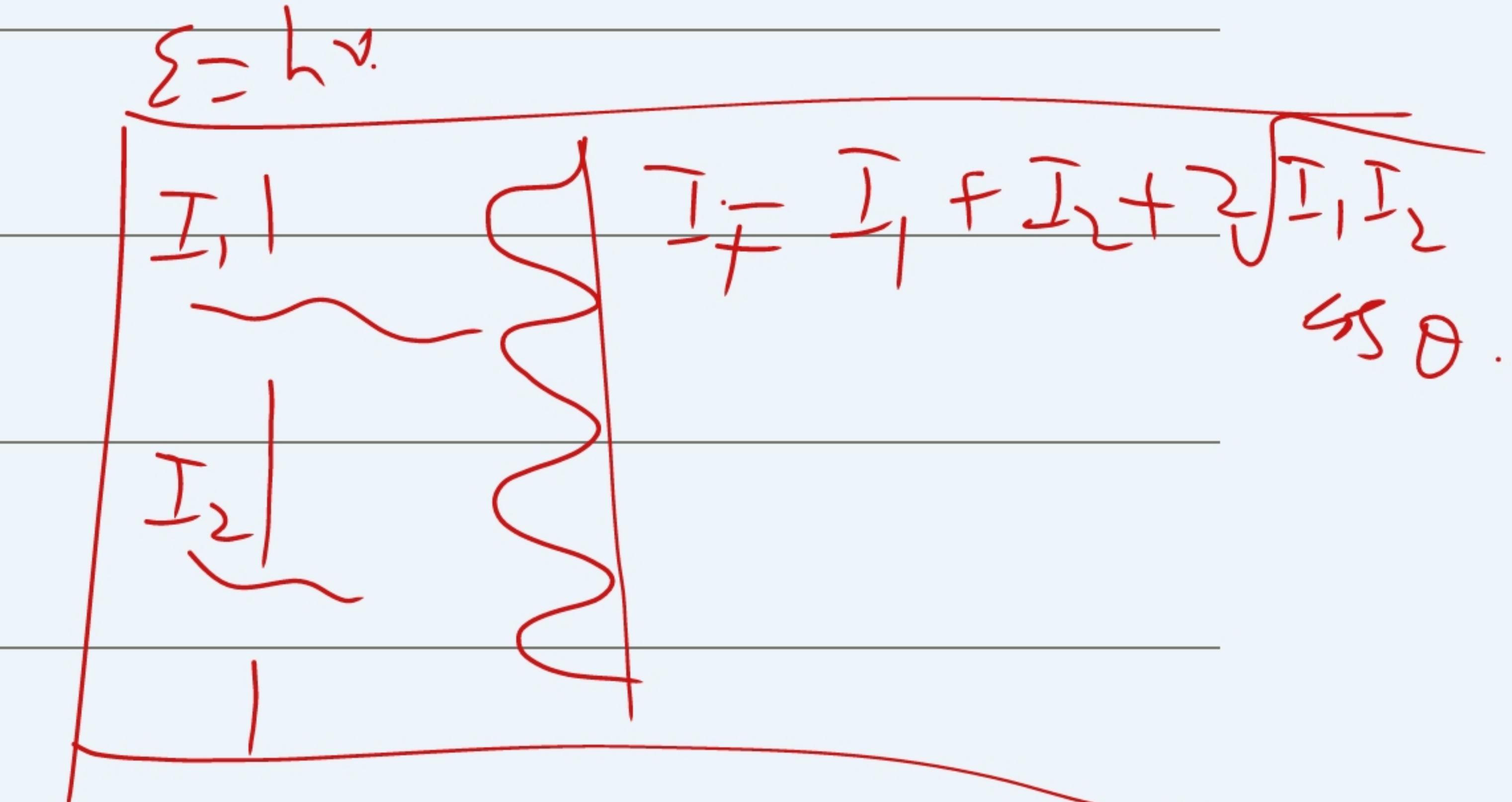
$$(\lambda' - \lambda) = \text{constant} [1 - \cos \theta]$$



STR

$$E = \sqrt{p_c^2 + m_c^2 c^4}$$

Scattered Electron



Horizontal:

$$\frac{hr}{c} + O = \frac{hr' \cos\phi + pc \cos\theta}{c} - \textcircled{1}$$

$$\hookrightarrow hr = hr' \cos\phi + pc \cos\theta \Rightarrow pc \cos\theta = hr - hr' \cos\phi$$

Vertical:

$$O = \frac{hr' \sin\phi}{c} - pc \sin\theta - \textcircled{2}$$

$$C \times \textcircled{1} \& \textcircled{2} \quad O = hr' \sin\phi - pc \sin\theta \Rightarrow pc \sin\theta = hr' \sin\phi.$$

$$\text{L.H.S} \Rightarrow \boxed{pc^2 - (hr)^2 = (hr')^2 + (hr')^2 - 2(hr)(hr') \cos\phi}$$

$$E = KE + mc^2 = \sqrt{m^2 c^4 + p^2 c^2}$$

$$KE = h(\gamma - \gamma')$$

$$\Rightarrow (KE + mc^2)^2 = m^2 c^4 + p^2 c^2$$

$$\Rightarrow KE^2 + \cancel{m^2 c^4} + 2(KE)(mc^2) = \cancel{m^2 c^4} + p^2 c^2$$

$$\Rightarrow p^2 c^2 = KE^2 + 2(KE)(mc^2)$$

$$(h\nu)^2 + (h\nu')^2 - 2(h\nu)(h\nu')\cos\phi = (\cancel{(h\nu)^2} + \cancel{(h\nu')^2}) - \underbrace{2(h\nu)(h\nu')} + \sqrt{2mc^2} \cdot (h\nu - h\nu')$$

$$\gamma = \frac{c}{\lambda}$$

$$(\lambda' - \lambda) = \left(\frac{h}{mc} \right) [1 - \cos\phi]$$

Divide by $2h^2 c^2$ $\Rightarrow \frac{1}{2} \frac{mc^2}{h^2 c^2} (h\nu - h\nu') = \frac{1}{2} \frac{(h\nu)(h\nu')}{h^2 c^2} [1 - \cos\phi]$

$$\Rightarrow \frac{mc}{h^2} \left[\frac{c}{\lambda} - \frac{c}{\lambda'} \right] = \frac{1}{2} \cdot \frac{c}{\lambda} \cdot \frac{c}{\lambda'} [1 - \cos\phi] \Rightarrow \frac{mc}{h} [\lambda' - \lambda] = (1 - \cos\phi)$$

Wave properties of particles \rightarrow

$$\beta = \frac{h\gamma}{c} = \frac{h}{\lambda} \Rightarrow \lambda = \frac{h}{\beta}$$

De Broglie: material: $P = \gamma m \vec{v}$, $\gamma = \frac{1}{\sqrt{1 - v^2/c^2}}$, $v \equiv \frac{\Delta \omega}{\Delta k} = \frac{d\omega}{dk}$.

$$\lambda_{dB} = \frac{h}{\gamma m v}$$

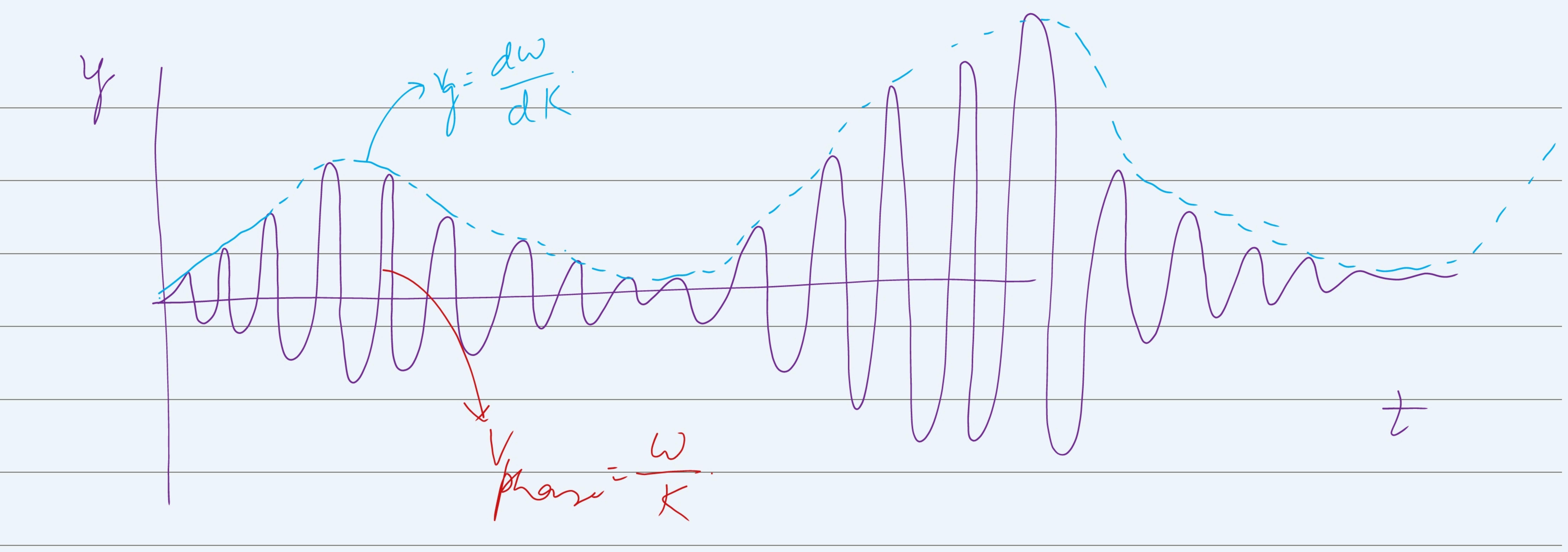
$$y = A \cos(\omega t - kx)$$

+ B

$$y = y_1 + y_2 = A \cos[(\omega + \delta\omega)t - (k + \delta k)x]$$

$$v \approx \frac{2\omega}{2k} = \frac{\omega}{k}$$

$$= \cancel{2A} \cos \frac{1}{2} \left[\left(\frac{2\omega + \delta\omega}{2} \right) t - \left(\frac{2k + \delta k}{2} \right) x \right] \cos \frac{1}{2} \left[\frac{\delta\omega}{2} t - \frac{\delta k}{2} x \right]$$



Davison & Germer / Partikel Diffraction

electron
↓

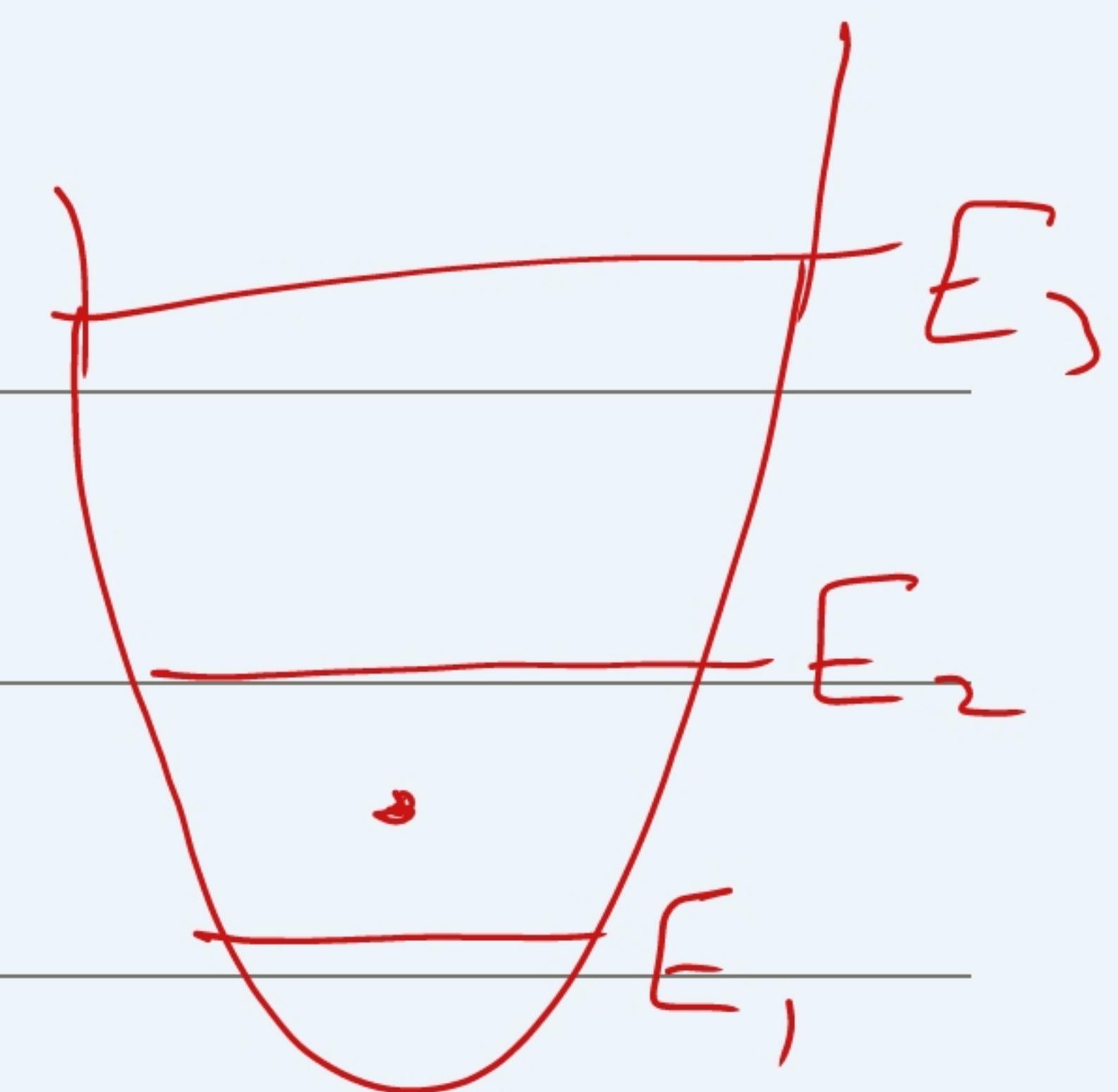
$$\frac{K.E = p^2}{2m} = \frac{h^2}{2m\lambda^2}$$

Nickel
Copper

$$K.E = \frac{1}{2}mv^2 = \frac{p^2}{2m} = \frac{\hbar^2}{2m\lambda^2},$$

$$\frac{\lambda}{2} = L$$

$$\Rightarrow \lambda = \underline{\frac{2L}{n}}, n=1,2,\dots$$



$$\geq \frac{n^2 \hbar^2}{8mL^2}$$

$$+ y_1 = \underline{A} \cos(\omega t - k_x)$$

$$n = 1, 2, 3, \dots$$

$$y_2 = \underline{A} \cos[(\omega + \delta\omega)t - (k + \delta k)x]$$

$$\underline{y} \approx 2A \cos[2\omega t - 2kt] \cos[\delta\omega t - \delta kx]$$

$$V_{\text{phase}} = \frac{\omega}{K} = \frac{2\pi\nu}{(2\pi)} = \underline{\underline{\lambda\nu}} = \frac{h}{\gamma m\nu} \cdot \frac{\gamma m c^2}{h} = \underline{\underline{c^2}}$$

$$E = h\nu = \underline{\underline{\gamma m c^2}}$$

$$\gamma = \underline{\underline{\frac{\gamma m c^2}{h}}}$$

$$V_{\text{group}} = \frac{d\omega(\nu)}{dK}$$

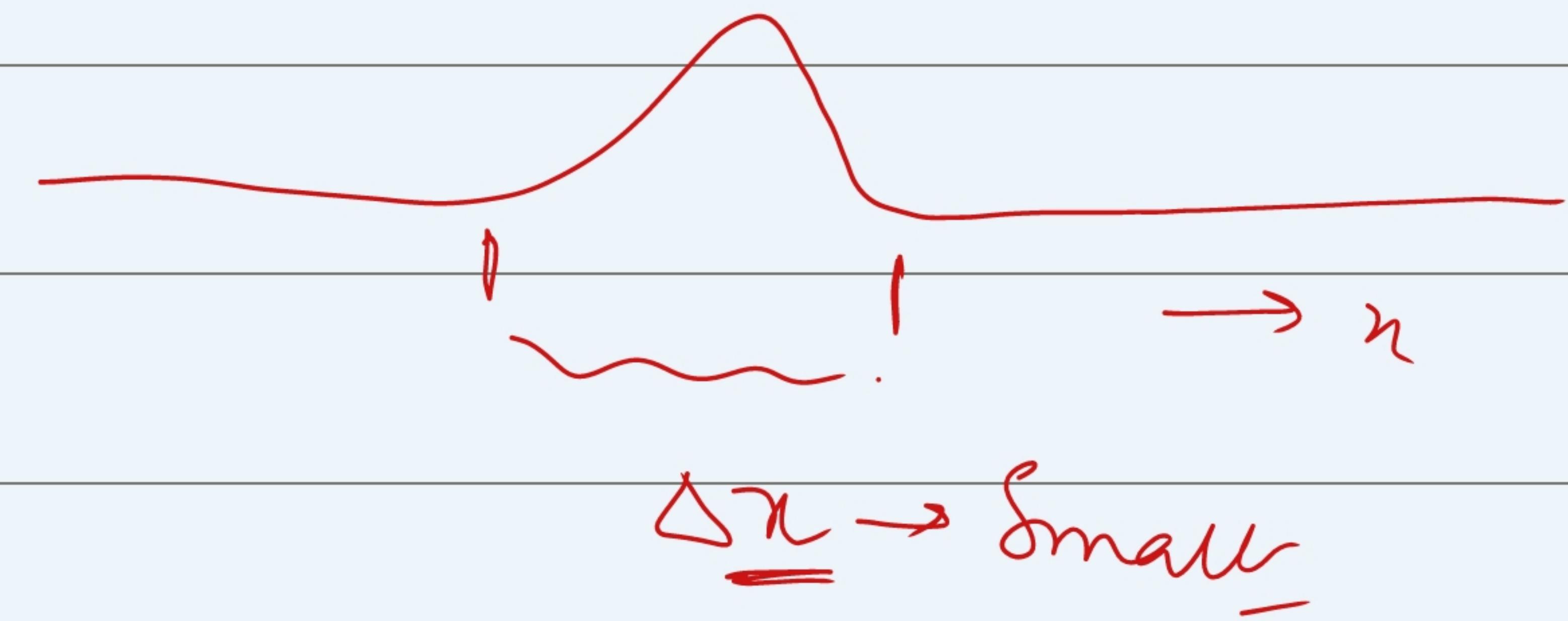
$$\underline{\underline{\omega}} = 2\pi\nu = \frac{2\pi\nu}{\lambda}$$

$$\boxed{\frac{d\omega}{dK} = \nu} \quad d\omega = -\frac{2\pi c}{\lambda^2} \cdot d\lambda$$

$$dK = \frac{2\pi}{\lambda}, \quad dK = -\frac{2\pi}{\lambda^2} d\lambda$$

Heisenberg Uncertainty Principle

$$\hbar = \frac{h}{\lambda}$$



$$\Delta p \approx \frac{\hbar}{\lambda}$$



$$\Delta x \Delta p \approx \cancel{\Delta} \lambda$$

$\Delta p \rightarrow \text{large}$:



Postulates of Quantum Mechanics →

1. All physical systems have a wavefunction

$$e^{i\theta} = \cos \theta + i \sin \theta$$

$$= \cos \underline{\theta}.$$

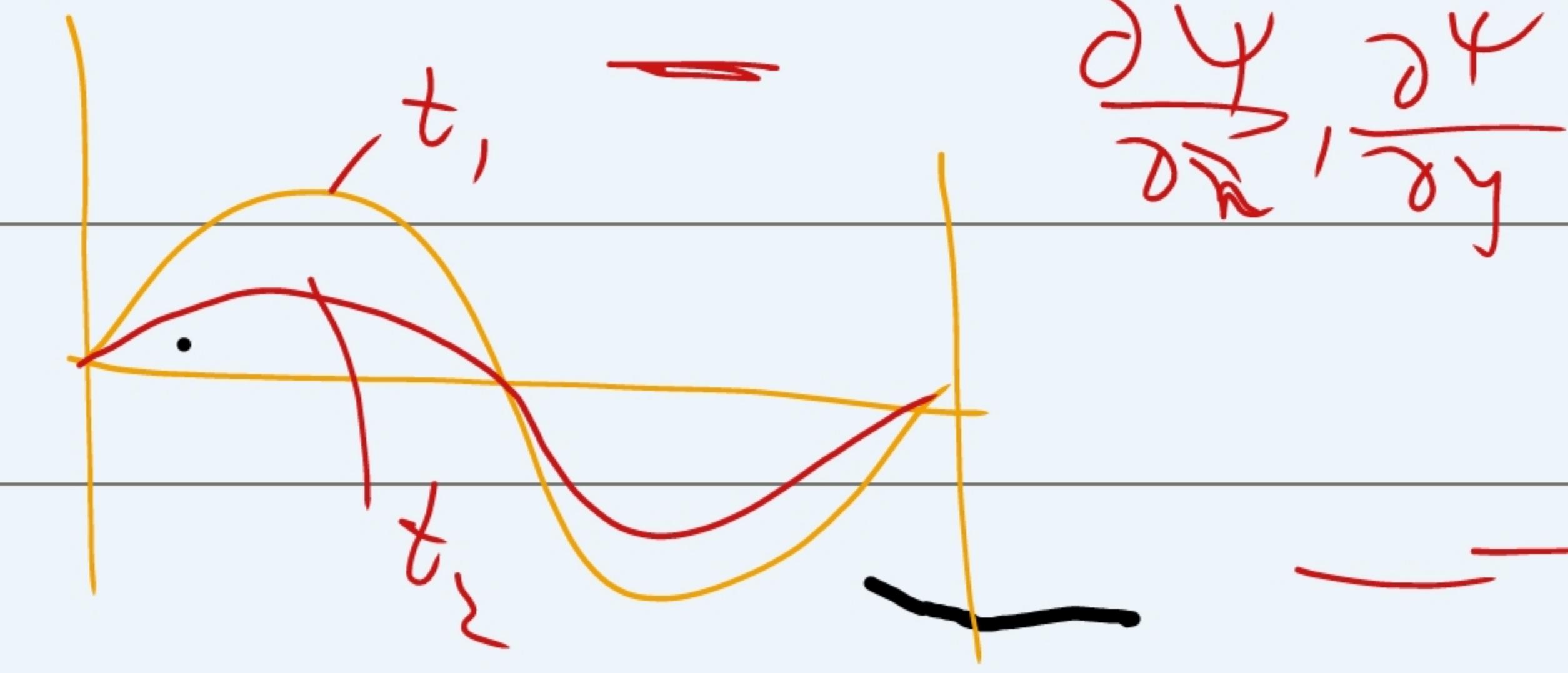
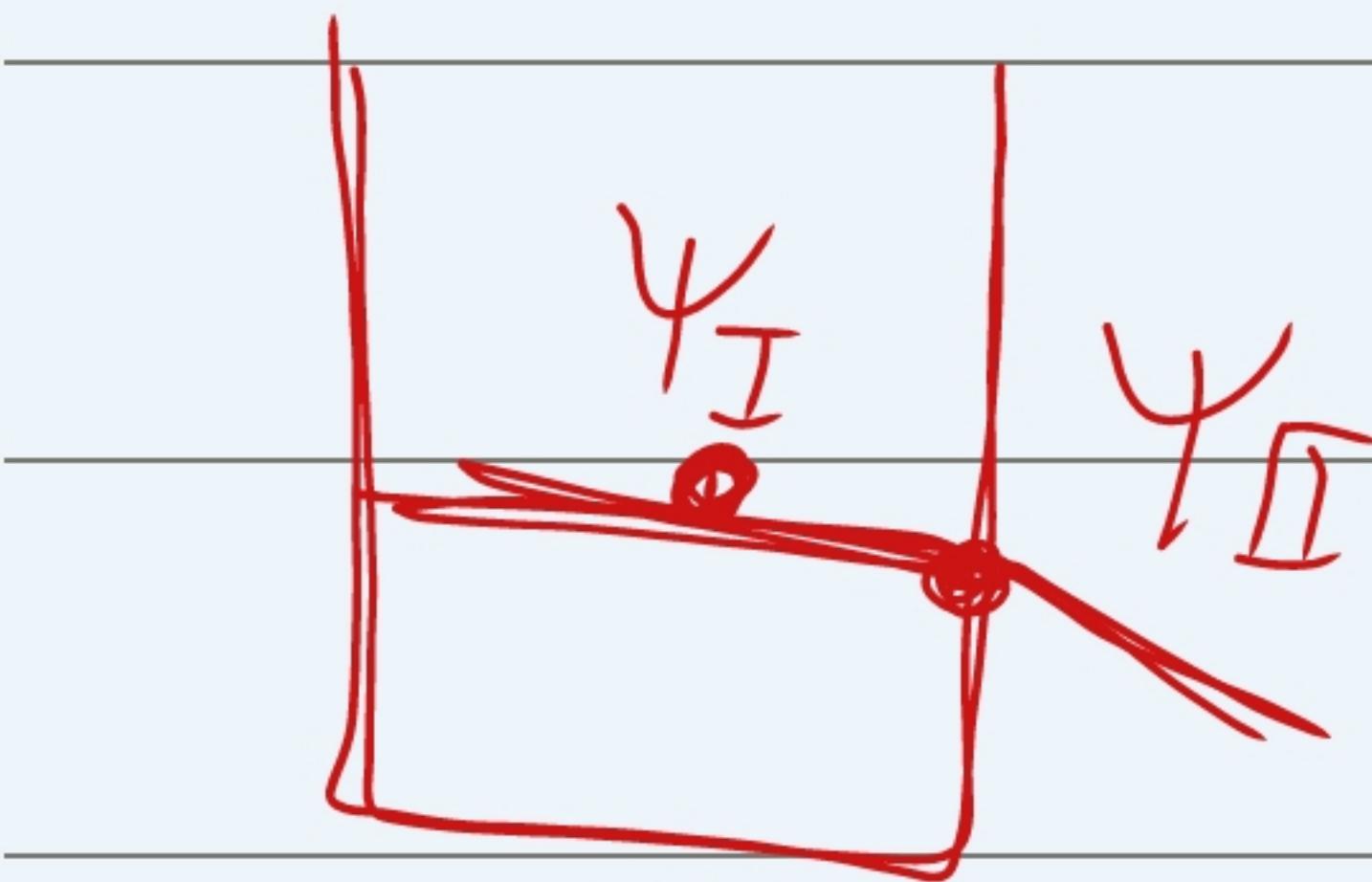
+ libra

Well-behaved :-

$$Y = A + i B$$

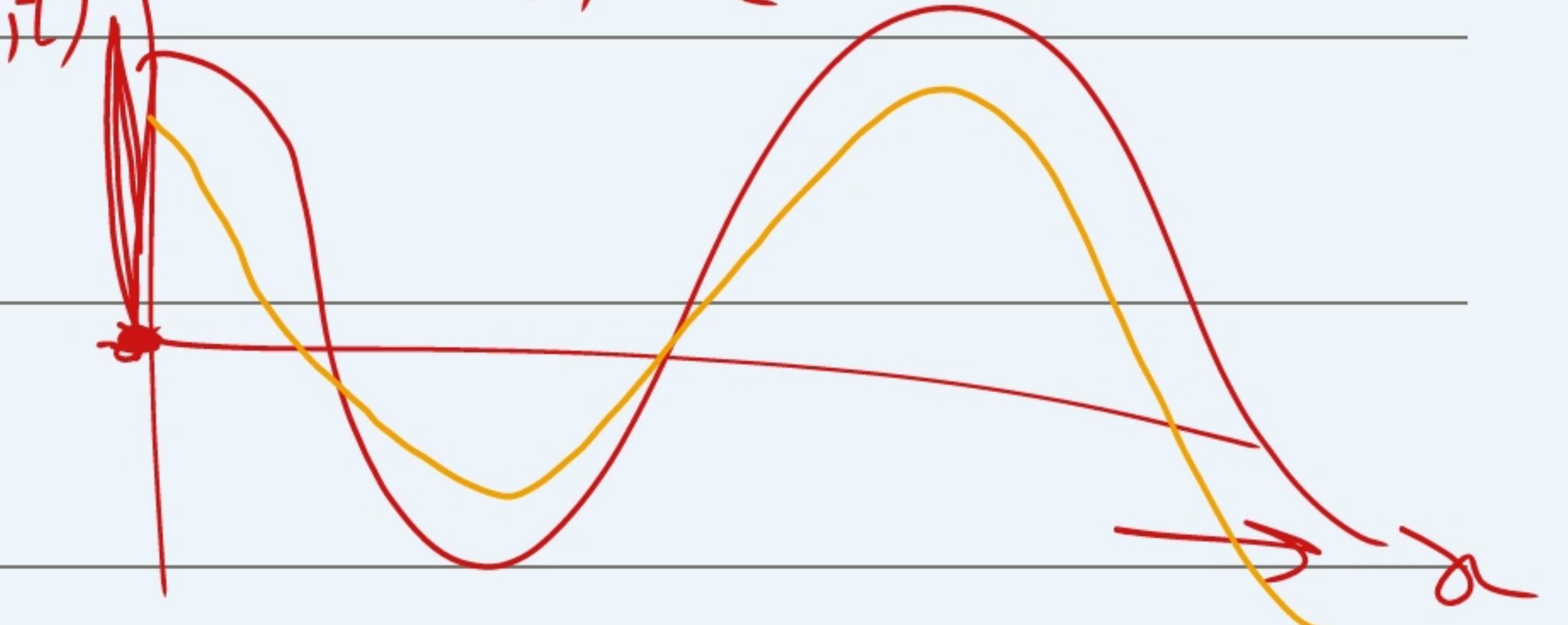
I Single valued (a, t)

Continuous and differentiable

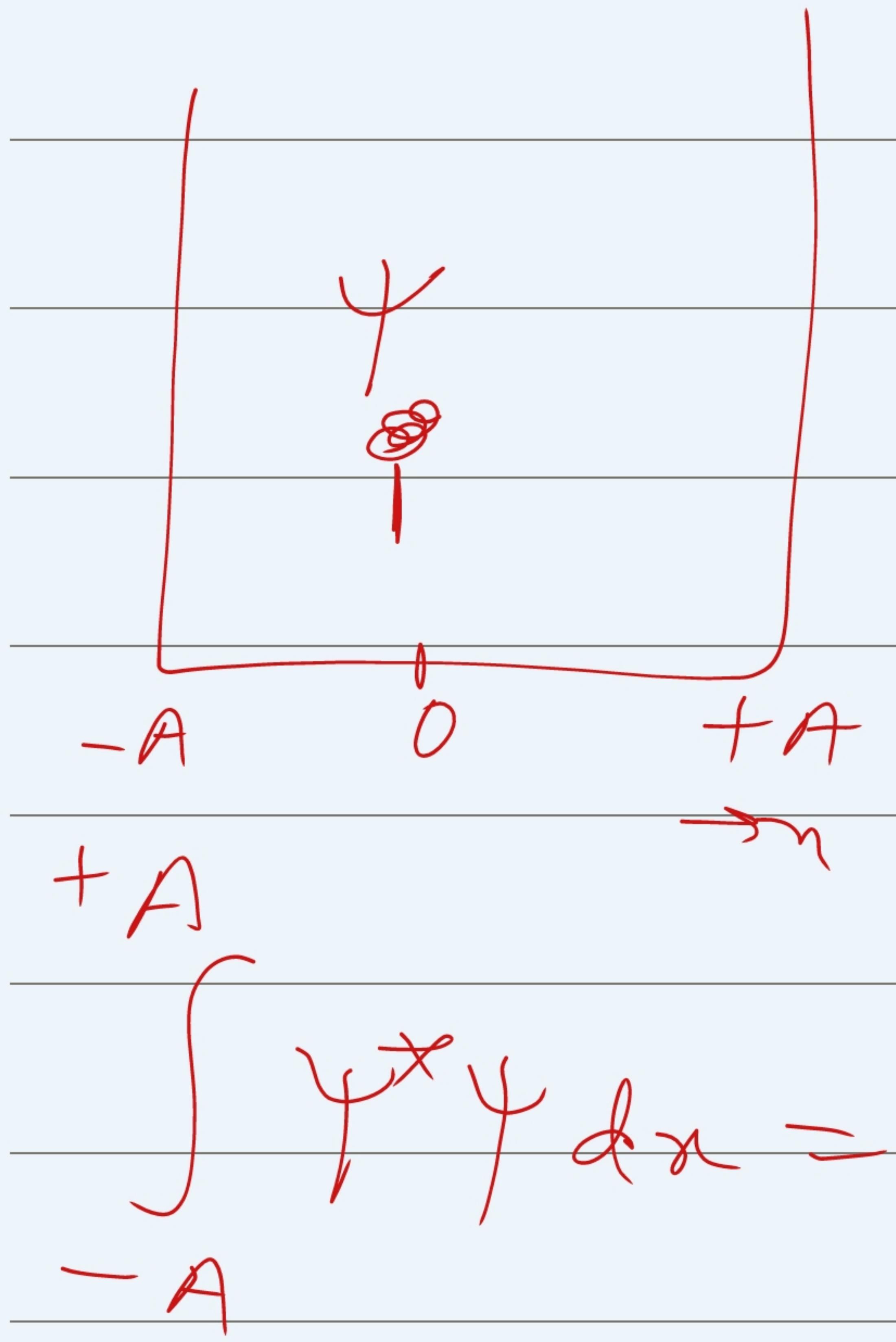


$$\frac{\partial Y}{\partial x}, \frac{\partial Y}{\partial y}, \frac{\partial Y}{\partial z}, \frac{\partial Y}{\partial t}$$

$$y(x,t) \quad t = t_1, t_2$$



~~→~~ Square Integration



$$\psi = A + iB$$

$$\psi^* \psi = A^2 + B^2$$

$$|\psi|^2 = A^2 + B^2$$

$$\int_{-A}^{+A} \psi^* \psi dx = 1$$