

# Earth Materials - Problem Set

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Assigned 1/28/16 Due 2/9/16 10pts (out of 100) deducted for each day late

**Please show ALL of your work.** Write clearly and organize your work so that I can easily see what you did. I **strongly encourage** you to use a spreadsheet (like Excel) and print out your organized calculations. You may also email me your homework as a clearly organized Excel file. Consult your book for information about the basic formulas for these minerals. Pages 198-200 are especially helpful for reminding you how to do these calculations.

## 1 1

Compute the structural formula (on the basis of 6 Oxygens) for the ANKERITE compositions shown below.

Oxide	Weight Percent
FeO	12.83
MgO	12.85
CaO	29.23
CO <sub>2</sub>	44.70
total	99.61

## 2 2

Compute the structural formula (on the basis of 12 Oxygens) for the GARNET composition shown below.

Oxide	Weight Percent
SiO <sub>2</sub>	37.08
TiO <sub>2</sub>	0.03
Al <sub>2</sub> O <sub>3</sub>	20.95
Cr <sub>2</sub> O <sub>3</sub>	0.02
FeO	30.21
MnO	3.64
MgO	2.04
CaO	5.55
Na <sub>2</sub> O	0.01
Total	99.51

## 3 3

Garnet may often be described by continuous solid-solution between four main end-members: a) Almandine - Fe<sub>3</sub>Al<sub>2</sub>Si<sub>3</sub>O<sub>12</sub> b) Pyrope - Mg<sub>3</sub>Al<sub>2</sub>Si<sub>3</sub>O<sub>12</sub> c) Spessartine - Mn<sub>3</sub>Al<sub>2</sub>Si<sub>3</sub>O<sub>12</sub> d) Grossular - Ca<sub>3</sub>Al<sub>2</sub>Si<sub>3</sub>O<sub>12</sub>

Compute the percent end-member composition of the garnet in #2 above. Report it as: AlmXX,PyXX,SpXX,GrXX, where "XX" is the percent of that end-member.

## 4 4

Compute the structural formula (on the basis of 11 Oxygens - this counts the two (OH)<sup>1-</sup> groups as one O<sup>2-</sup>) for the BIOTITE composition below.

Oxide	Weight Percent
SiO <sub>2</sub>	35.55
TiO <sub>2</sub>	2.81
Al <sub>2</sub> O <sub>3</sub>	16.71
FeO	21.38
MnO	0.36
MgO	8.24
BaO	0.01
CaO	0.02
Na <sub>2</sub> O	0.05
K <sub>2</sub> O	9.64
F-	0.35
Total	95.15

Hints:

- use ALL Si and some of the Al to completely fill the tetrahedral site (to a total of 4 formula cations)
- remaining Al must go into the octahedral site along with all the Ti, Fe, Mn, Mg.
- All K, Na, Ba, Ca ust go into the interlayer "A" site.
- Flourine substitutes in the (OH) site. It doesn't have any oxygens related to it (it is not an oxide).

## 5 5

Why is the total on the biotite analysis so low?