## Chem231B: Quiz #4

## March 1, 2020

- 1. What is the spin multiplicity of the ground state of  $H_2$  and of  $H_2^+$ ?  $H_2$  spin multiplicity is 1;  $H_2^+$  is 2.
- 2. Give the electronic Hamiltonian for  $H_2$ .

 $H_{\rm el}=\hat{h}(1)+\hat{h}(2)+V_{ee}$  where  $\hat{h}(i)$  is the one electron Hamiltonian for *i*-th electron and  $V_{ee}$  is the coulombic interaction:

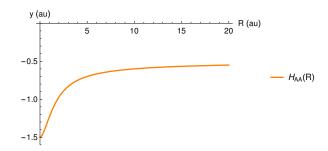
$$\hat{h}(i) = -\frac{\nabla_i^2}{2} - \frac{1}{|r_i - R\hat{z}|} - \frac{1}{|r_i|}$$
 and  $V_{ee} = \frac{1}{|r_1 - r_2|}$ 

r is the position of the electron and R is the position of the nucleus. The  $H_2$  is placed along the z-axis where a proton is at the origin.

- 3. Which one of the following changes significantly when going from H<sub>2</sub> to D<sub>2</sub>:  $R_e$ ,  $D_e$ ,  $\omega$ ?  $\omega$ .
- 4. Within the harmonic approximation, say how your answer to the previous question will change?
- 5. Give an expression for the matrix element  $H_{AA}$  in  $H_2^+$  for 1s orbitals ( $\gamma = 1$ ).

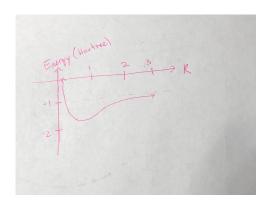
$$h_{AA} = \gamma^2/2 - \gamma f(x) = 1/2 - f(x)$$
, where  $f(x) = 1 - \frac{(1+x)e^{-2x} - 1}{x}$ .

6. Sketch how the matrix element  $H_{AA}$  should depend on R, giving its value as  $R \to \infty$ .



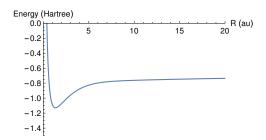
 $H_{AA}$  should go to -0.5 Hartree as  $R \to \infty$ .

7. Sketch a molecular energy curve for  $H_2$  as a function of R, giving its value as  $R \to \infty$ .



## -1 Hartree as $R \to \infty$

8. Sketch the curve within the Hartree-Fock approximation. What qualitative error does it make? Sketch which curve within HF?  $H_2$ ?



## -11/16 or -0.6875 Hartree as $R \to \infty$

9. For a molecule with  $y_e=0$ , deduce a formula for the number of states it will bind in terms of  $D_e,\,\omega,$  and  $x_e.$ 

$$E_{\text{vib}}(\nu) = \nu_e[(\nu + 1/2) - x_e(\nu + 1/2)^2]$$

Dissociation is when  $E_{\rm vib}=D_e$  and hence, solve for  $\nu$  which will yield the max number of bounded states  $\nu_{\rm max} \approx \frac{1}{2x_e} - \frac{1}{2} + \frac{\sqrt{(1-x_e)^2 - 4x_eD_e/\nu_e}}{2x_e}$ 

10. Deduce an expression for  $x_e$  in terms of  $D_e$  and  $\omega$  for the Morse potential, for which  $\epsilon_n = -V_0 \left(1 - \frac{\alpha(n + \frac{1}{2})}{\sqrt{2} \mu V_0}\right)^2$   $x_e = \frac{\omega}{8D_e}$