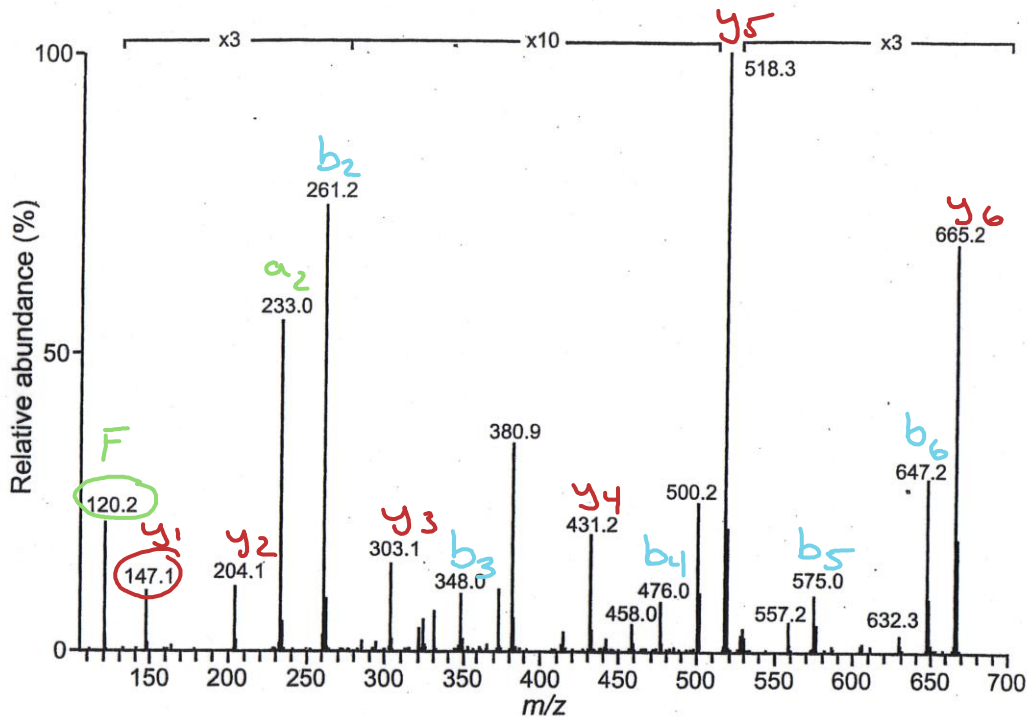


$$[M+H]^+ = 778.6$$



1) Phe immonium ion m/z 120

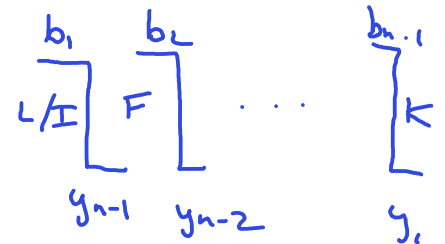
2) b_2 ion 261.2 a_2 ion 233.0 (L/I F)(YP)(EM)

3) C-terminal lysine (147) $y_1 \rightarrow b_{n-1} = 632.3$

$$y_{n-2} = 778.6 - 261.2 + H = 518.4$$

$$y_{n-1} = 778.6 - \text{Leu/Ile} = 665.6$$

$$y_{n-1} - y_{n-2} = 147 = \text{Phe}$$



$$5) 518.3 - 431.2 = 87.3 = \text{Ser} = y_4$$

$$778.6 - 431.2 + H = 348.4 = b_3$$



$$431.2 - 303.1 = 128.1 = Q/K = y_3$$

$$778.6 - 303.1 + H = 476.5 = b_4$$



$$303.1 - 204.1 = 99 = V$$

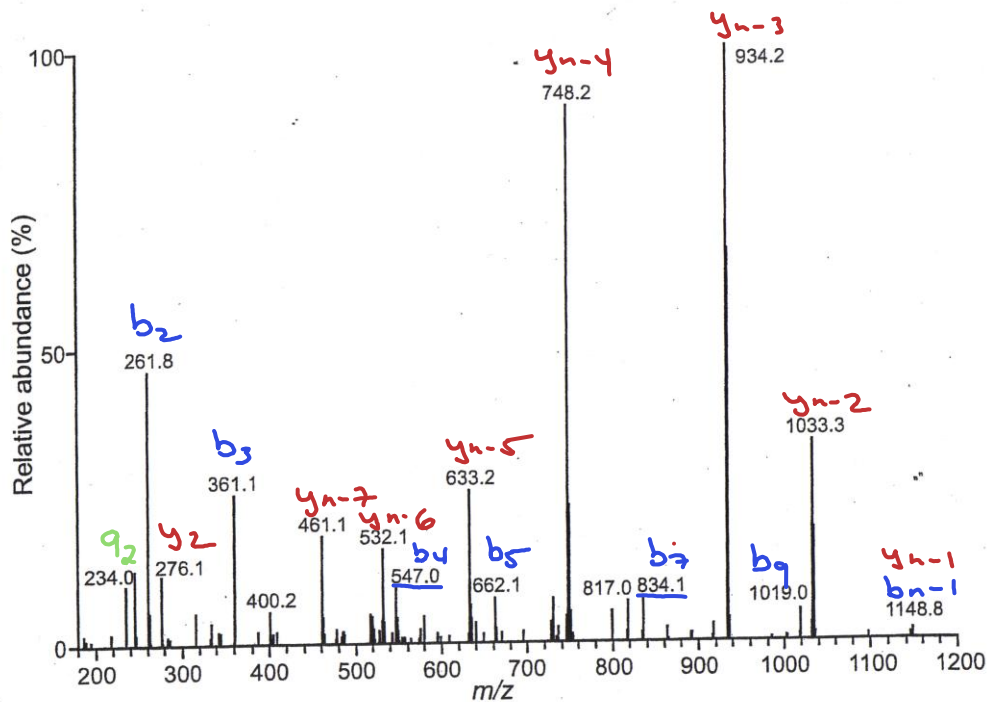
$$778.6 - 204.1 + H = 575.5 = b_5$$



$$6) 632.6 - 575.0 = 57.6 = G$$



$[M+H]^+ = 1295.0$



- 1) No ammonium ions b/c of low mass cutoff
- 2) b_2 ion 262 a_2 ion 234 $(C^+S)(C^+T)(FN)$
- $y_{n-2} = 1295 - 261.8 + H = 1034.2$

3) not there although $1295 - 147 = 1148 = 4 \times 287$

c) $y_{n-1} = 1295 - \text{Phe} = 1148$

$$5) 1033.3 - 934.2 = 99.1 = V$$
$$1295 - 934.2 + 4 = 361.8 = b_3$$
$$934.2 - 748.2 = 186 = W$$
$$1295 - 748.2 + 11 = 547.2 = b_4$$
$$748.2 - 633.2 = 115 = D$$
$$1295 - 633.2 + 4 = 662.8 = b_5$$
$$633.2 - 532.1 = 101.1 = T$$
$$1295 - 532.1 = 763.9 \text{ (Not there)}$$
$$\begin{aligned} \gamma &= L \gamma_s \\ & \left[\begin{array}{c} F_N \\ \vdots \\ F_N \end{array} \right] \left[\begin{array}{c} V \\ \vdots \\ V \end{array} \right] \left[\begin{array}{c} W \\ \vdots \\ W \end{array} \right] \left[\begin{array}{c} D \\ \vdots \\ D \end{array} \right] \left[\begin{array}{c} T \\ \vdots \\ T \end{array} \right] \end{aligned}$$

$$532.1 - 461.1 = 71 = A|a$$

F[N][V][W][D][T][A]]K

$$1295 - 461.1 + 4 = 834.9 = b_7$$

$$y_1 = 147$$

$$y_2 = 147 + 129.1 = 276.1$$

Thus $y_2 = E$

$$1295 - 276.1 + 4 = 1019.9 = b_8$$

$$\begin{array}{ccccccc}
 & & & & 834 & & 1619 \\
 F & [& N &] & V & [& W &] & D & [& T &] & A &] & 185 & [& E &] & K \\
 & & & & & & & & & & & & 461 & & & & 276
 \end{array}$$

$$\hookrightarrow 185 = -G \ Q - = 57 + 128$$